DEVELOPMENT OF TAILORABLE ELECTRICALLY CONDUCTIVE THERMAL CONTROL MATERIAL SYSTEMS

1N-27-CR 081473

FINAL REPORT

IITRI Project No. C06804-97 Contract No. NAS8-40580

Submitted to

National Aeronautics and Space Administration Space Environments and Effects Program Office George C. Marshall Space Flight Center Alabama 35812 Attention: Mr. Jason Vaughn/Ms. B. Kilpatric

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FOREWORD

This Final Report No. IITRI-C06804-97 under Contract No. NAS8-40580, entitled "Development of Tailorable Electrically Conductive Thermal Control Material Systems", summarizes work initiated in April 1995 and concluded in April 1997. The bulk of the work was performed by August 1996; subsequent work consisted of laboratory studies for formulating candidate thermal control material system (TCMS) to tailor target resistivities using concepts suggested in IIT Research Institute (IITRI) Proposal No. IITRI-94-578AC and space environment simulations studies. These studies have resulted in screening and ranking of developmental electrically conductive thermal control material systems (TCMS) concepts for future scaleup and validation and use on space hardware/space structures.

Personnel at IITRI who contributed to this program include: Messrs. T. J. Curtis, W. Sabato, A. Ray, and D. Jones. Mr. G. Sresty acted as Manager of this program. The screening space simulation studies were carried out at the Aerospace Corporation by Dr. M. Meshishnek and at Goddard Space Flight Center (GSFC) by Ms. Wanda Peters and Dr. L. Kauder. Dr. L. Kauder also performed Electro Static Discharge (ESD) testing on candidate materials. Their valuable support, technical input, advice and guidance has helped this program considerably. The continuous encouragement and interest shown in this work by Mr. Pat Carlin (Air Force Materials Laboratory) and Mr. Cliff Cerbus (UDRI) was also valuable. Their efforts in testing a few concepts to typical GEO conditions through space simulation studies at SPECTRE are acknowledged. Lastly, the patient guidance and advice of Mr. Jason Vaughn, Contracting Officer's Representative, are gratefully acknowledged.

The report has been reviewed and found to be in compliance with contract requirements and our internal quality system.

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EXECUTIVE SUMMARY

The optical characteristics of surfaces on spacecraft are fundamental parameters in controlling its temperature. Passive thermal control coatings with designed solar absorptance and infrared emittance properties have been developed and have been in use for some time. In this total space environment, the coating must be stable and maintain its desired optical properties as well as mechanical properties for the course of the mission lifetime. The mission lifetimes are increasing and in our quest to save weight, newer substrates are being integrated which limit electrical grounding schemes. All of this has added to already existing concerns about spacecraft charging and related spacecraft failures or operational failures. The concern is even greater for thermal control surfaces that are very large. One way of alleviating such concerns is to design new thermal control material systems (TCMS) that can help to mitigate charging via providing charge leakage paths.

The objective of this program was to develop two types of passive electrically conductive TCMS. The first was a highly absorbing/emitting black surface and the second was a low (α_s/ϵ_N) type white surface. The surface resistance goals for the black absorber was 10^4 to 10^9 Ω/\Box , and for the white surfaces it was 10^6 to 10^{10} Ω/\Box . Several material system concepts were suggested and evaluated for space environment stability and electrical performance characterization.

Our efforts in designing and evaluating these material systems have resulted in several developments. New concepts, pigments and binders have been developed to provide new engineering quality TCMS. Some of these have already found application on space hardware, some are waiting to be recognized by thermal designers, and some require further detailed studies to become state-of-the-are for future space hardware and space structures.

Our studies on baseline state-of-the-art materials and conductive concepts have resulted in several important of findings that are of interest to all thermal designers and systems integrators. Some of the important conclusions are as follows:

• All conductive TCMS designs proposed for high absorptivity applications met the program goals of space environment stability and acceptable ESD behavior.

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- Studies of the nonconducting state-of-the-art reformulated TCMS showed that:
 - (1) S13GP/LO-1 and S13GP/LO-41 show excellent stability to a simulated space environment, approaching that of Z-93P.
 - (2) Reformulated YB-71P has essentially the same stability as YB-71.
 - (3) Surface resistance of exposed YB-71 and YB-71P are 5.90E+07 to 8.0E+08 Ω/\Box .
 - (4) Zn₂TiO₄ (ZOT) based TCMS showed a reversal in degradation with UV intensity. This is an interesting finding and needs further conformation study because it may have consequences in material selection for high electron flux orbits.
- Studies of the electrically conductive TCMS concepts showed that:
 - (1) The electrically conductive TCMS concepts, Z-93CXY, Z-93SCXY and Z-9SCLMXY, demonstrated the required stability and acceptable ESD performance.
 - (2) ZOT/Phos Sol and ZOT/DHS-2 met program goals for stability in the space environment along with required electrical properties. Similar success was also noted for Eu₂O₃-based TCMS. The conductive ZOT-based formulations also showed reversal in degradation with UV intensity as was observed for the nonconducting state-of-the-art systems (Item 4 under last bullet). This is interesting and can have consequences in conductive material selection for high electron flux orbits.
 - (3) Two material technologies, doped silicate binder and doped silicate binder based encapsulation, are now available to design the required electrical resistances in the passive TCMS.
 - (4) DS13N/LO-XY, flexible conductive TCMS showed acceptable space environment stability, but showed limited success in ESD behavior. Future development steps have been identified as a result of this program.
 - (5) Promising electrically conductive passive TCMS have been identified for future investment in scaleup and validation as a result of these efforts. They are listed in the tables provided on the next three pages for ready reference along with their space environment effects' performance. Their ESD performance was considered to be acceptable for use on space hardware.

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| Туріса | l Performance of | f Promising Con | ductive Abso | rber Optical Co | atings | |
|----------------------------|-----------------------------|------------------------------------|---------------------------|----------------------------------|--------------------------------|---------|
| Material | Pre- exposure α_S | Post- exposure $\alpha_{\rm S}$ | $\Delta lpha_{	extsf{S}}$ | Surface Resistivity Ω/□ | Charged to (V) ^a | Remarks |
| Inorganic-Conductive | | | | | | |
| MH55-ICP | 0.940 | 0.938 | -0.001 | 10 ⁸ -10 ⁹ | <60 | b |
| MH55-ICP | 0.942 | 0.941 | -0.001 | 10 ⁸ -10 ⁹ | <60 | ь |
| MH55-IC | 0.928 | 0.938 | 0.010 | 108-109 | <60 | С |
| DBG-IP | 0.974 | 0.975 | 0.001 | 10 ⁸ -10 ⁹ | < 80 | b |
| DBG-IP | 0.973 | 0.974 | 0.001 | 10 ⁸ -10 ⁹ | < 80 | ь |
| DBG-IP | 0.957 | 0.951 | -0.006 | 10 ⁸ -10 ⁹ | <80 | С |
| DBG-IP | 0.970 | 0.978 | 0.008 | 10 ⁸ -10 ⁹ | <80 | d |
| DBG-DHS | 0.958 | 0.957 | -0.001 | 10 ⁷ -10 ⁹ | < 50 | ь |
| DBG-DHS | 0.956 | 0.956 | 0 | 10 ⁷ -10 ⁹ | < 50 | b |
| Organic-Flexible-Conductiv | ve | | | | | |
| D21S/LO | 0.980 | 0.980 | 0 | 1010-1011 | N/A | ь |
| D21SC/LO | 0.956 | 0.957 | 0.001 | 104-109 | N/A | b |
| D21SC/LO | 0.955 | 0.956 | 0.001 | 10 ⁴ -10 ⁹ | N/A | b |
| MH21SC/LO | 0.963 | 0.964 | 0.001 | 10 ⁵ -10 ⁹ | < 200 | b |
| MH21SC/LO | 0.965 | 0.965 | 0 | 10 ⁵ -10 ⁹ | < 200 | b |
| MH41SCB/LO | 0.951 | 0.950 | -0.001 | 108-109 | <75 | b |
| D36SCB/LO | 0.962 | 0.967 | 0.005 | < 10 ³ | < 70 | С |
| D36SCBLO | 0.966 | 0.956 | -0.010 | < 10 ³ | <60 | d |
| MH41SCB/LO | 0.939 | 0.947 | 0.008 | 10 ⁸ -10 ⁹ | <75 | С |
| MH21SC/LO | 0.951 | 0.955 | 0.004 | 10 ⁵ -10 ⁹ | <200 | c _ |

Remarks:

- a = Charged to <50 volts at room temperature; indicated voltages are maximum observed at cryogenic temperatures.
- b = Tests performed at Aerospace Corporation
- c = Tests performed at GSFC
- d = Tests performed at AMCL/SCEPTRE

All tests at GSFC involved exposure to 1000 ESH of UV and Vacuum.

All tests carried out at Aerospace Corporation involved exposure to 2000 ESH of UV + 8.26 x 10¹³ é/cm² [100 KeV] + 5.18 x 10¹³ é/cm² [35 KeV] (typical DMSP orbit) and vacuum.

All tests at AFML/SCEPTRE involved exposure at 2.0 Suns of EUV and 6.78 x 10^{15} e'/cm² (1 KeV) + 3.39 x 10^{15} e'/cm² (10 KeV).

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| Тур | ical Performance | of Promising Cond | uctive Whit | te Thermal Control | Coatings | |
|---------------|---|---|----------------------|--|------------------------------|--------------|
| TCMS | Pre-exposure α _s | Post-exposure α_{s} | $\Delta lpha_{ m S}$ | Surface Resistivity Ω/□ | Charged to V ^a | Remarks |
| | sh Calcination of brid, silicate binder | | etain contro | lled Zn interstitials, a | and stabilization | n with doped |
| ZnO(FC)/DHS-2 | 0.139 | 0.141 | 0.002 | 10 ⁸ -10 ⁹ | <60 | b |
| ZnO(FC)/DHS-2 | 0.146 | 0.146 | 0.000 | 10 ⁸ -10 ⁹ | <60 | b |
| ZnO(FC)/DHS-2 | 0.168 | 0.168 | 0.010 | 10 ⁸ -10 ⁹ | <60 | c* |
| ZnO(FC)/DHS-2 | 0.143 | 0.225 | 0.082 | 10 ⁸ -10 ⁹ | <60 | d¹ |
| | | ion of Zn interstitial orid silicate binders. | | encapsulation and inc | corporation in | hybrid |
| S13GP/DHS-1 | 0.147 | 0.153 | 0.006 | 10 ⁹ -10 ¹⁰ | < 200 | b |
| S13GP/DHS-1 | 0.142 | 0.148 | 0.006 | 10 ⁹ -10 ¹⁰ | <200 | b |
| S13GP/SS-55 | 0.125 | 0.140 | 0.015 | 10 ⁹ -10 ¹⁰ | N/A | b |
| S13GP/SS-55 | 0.127 | 0.137 | 0.010 | 10°-10¹0 | N/A | b |
| S13GP/SS-55 | 0.140 | 0.177 | 0.037 | 10 ⁹ -10 ¹⁰ | N/A | c** |
| S13GP/SS-55 | 0.151 | 0.206 | 0.055 | 10 ⁹ -10 ¹⁰ | N/A | c** |
| S13N/DHS-2 | 0.13 | 0.196 | 0.066 | 10 ⁹ -10 ¹⁰ | < 200 | d¹ |
| | | | | nicroencapsulation us doped hybrid silicate | | rid silicate |
| DS13N/SS-55 | 0.136 | 0.173 | 0.037 | 10 ⁸ -10 ⁹ | N/A | ь |
| DS13N/SS-55 | 0.137 | 0.153 | 0.016 | 10 ⁸ -10 ⁹ | N/A | b |
| DS13N/DHS-2 | 0.124 | 0.161 | 0.037 | 10 ⁸ -10 ⁹ | <200 | ь |
| DS13N/DHS-2 | 0.122 | 0.128 | 0.006 | 10 ⁸ -10 ⁹ | < 200 | ь |
| DS13N/DHS-2 | 0.148 | 0.154 | 0.006 | 10 ⁸ -10 ⁹ | < 200 | c* |
| DS13N/DHS-2 | 0.12 | 0.24 | 0.120 | 10 ⁸ -10 ⁹ | <200 | d² |
| DS13N/SS-55 | 0.16 | 0.27 | 0.11 | ~ 10 ⁸ -10 ⁹ | N/A | d³ |

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| Ту | pical Performance of | F Promising Conductive | White Therr | nal Coatings (Co | ontinued) | |
|---------------------------------------|-----------------------------|--------------------------|----------------------|-----------------------------------|------------------------------|---------|
| TCMS | Pre-exposure α _s | Post-exposure α_S | $\Delta lpha_{ m S}$ | Surface Resistivity Ω/□ | Charged to V ^a | Remarks |
| Concept: | Flexible Conductive | White TCC with DS13 | N in stripped | silicone | | |
| DS13N/LO-41 | 0.161 | 0.278 | 0.117 | 10°-10¹0 | < 800 | b |
| DS13N/LO-41 | 0.163 | 0.239 | 0.076 | 10 ⁹ -10 ¹⁰ | N/A | b |
| DS13N/LO-41 | 0.210 | 0.430 | 0.22 | 10 ⁹ -10 ¹⁰ | N/A | c* |
| DS13N/LO-51 | 0.193 | 0.423 | 0.230 | 109-1010 | N/A | d³ |
| Conductive Cond | epts with Other Can | didate Pigments | | | | |
| ZOT/Phos. Sol. | 0.131 | 0.149 | 0.018 | 10 ⁸ -10 ⁹ | <200 | b |
| ZOT/Phos. Sol. | 0.128 | 0.151 | 0.023 | 10 ⁸ -10 ⁹ | < 200 | b |
| ZOT/SS-55 | 0.117 | 0.156 | 0.039 | 10 ⁹ -10 ¹⁰ | N/A | b |
| ZOT/SS-55 | 0.112 | 0.156 | 0.044 | 10 ⁹ -10 ¹⁰ | N/A | b |
| ZOT/SS-55 | 0.129 | 0.180 | 0.051 | 109-10 ¹⁰ | N/A | b |
| Eu ₂ O ₃ /SS-55 | 0.087 | 0.138 | 0.051 | 10 ⁹ -10 ¹⁰ | N/A | b |
| Eu ₂ O ₃ /SS-55 | 0.100 | 0.126 | 0.028 | 109-10 ¹⁰ | N/A | b |
| Eu ₂ O ₃ /DHS-2 | 0.093 | 0.158 | 0.065 | 108-109 | <200 | b |
| Eu ₂ O ₃ /DHS-2 | 0.076 | 0.108 | 0.032 | 10 ⁸ -10 ⁹ | <200 | b |
| Eu ₂ O ₃ /SS-55 | 0.096 | 0.129 | 0.033 | 10°-10¹0 | N/A | c* |
| Eu ₂ O ₃ /SS-55 | 0.076 | 0.161 | 0.085 | 109-10 ¹⁰ | N/A | d² |

Remarks:

- a = Charged to <50 volts at room temperatures; indicated voltages are maximum observed at cryogenic temperatures.
- b = Tests performed at Aerospace Corporation
- c = Tests performed at GSFC
- d = Tests performed at AMCL/SCEPTRE
- *GSFC test involved exposure to 1000 ESH of UV-Vacuum.
- **This GSFC test involves exposure to 4300 ESH of UV-Vacuum.

All tests carried out at Aerospace Corp. involved exposure to 2000 ESH of UV + 8.26 x 10^{13} e'/cm² (100 KeV) + 5.18 x 10^{13} e'/cm² (35 KeV) (typical DMSP orbit) and vacuum.

- (1) Test at AFML/SCEPTRE involved exposure at 1 Sun of EUV-Vacuum (for Z-93CXY and Z-93SCXY) + 6.78 x 10¹⁵ e'/cm² (1 KeV) + 3.39 x 10¹⁵ e'/cm² (10 KeV).
- (2) Test at AFML/SCEPTRE involved exposure at 3.3 to 3.6 Suns of EUV-Vacuum (for DS13N/DHS-2 and $Eu_2O_3*/SS-55) + 6.78 \times 10^{15} \text{ e}'/\text{cm}^2$ (1 KeV) + 3.39 x 10^{15} e'/cm² (10 KeV).
- The in situ test at AFML/SCEPTRE involved exposure at 1.6 to 2.1 Suns of EUV-Vacuum + 6.78 x 10^{15} e'/cm² (1 KeV) + 3.39 x 10^{15} e'/cm² (10 KeV).

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LIST OF ACRONYMS

AFML Air Force Materials Laboratory

AO Atomic oxygen

ASTM American Society for Testing of Materials

BCC Body centered cubic

BG Black Glass
BOL beginning of life

BRDF Bi-directional reflectance distribution function CVCM Collectable Volatile Condensable Material

DoD Department of Defense
ESD Electrical static discharge
ESH Equivalent sun hours
GEO Geosynchronous Orbit
GPS Global positioning satellite
GSFC Goddard Space Flight Center

HTMIAC High Temperature Materials Information Analysis Center

IR&D Internal Research and Development

IITRI IIT Research Institute

JPL Jet Propulsion Laboratory

LEO Low Earth Orbit

NASA National Aeronautics and Space Administration

PBR Pigment-to-binder ratio

SBIR Small Business Independent Research

SEE Space Environmental Effects
TCMS Thermal control material system

TML Total mass loss UV Ultraviolet

UV/Vis Ultraviolet and Visible

Vac Vacuum

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1. INTRODUCTION

The temperature control of satellites and spacecraft is one of the most fundamental engineering problems confronting spacecraft designers and materials engineers. The ultimate objective of the thermal design is to ensure that the spacecraft operates within a prescribed temperature range defined by the temperature limitations of the vehicle's material and components.

One of the primary and successfully used methods to achieve the required thermal control has been the use of coatings with high reflectance or low solar absorptance (α_s) that are resistant to degradation in vacuum-ultraviolet and particulate radiation environment. Several National Aeronautics and Space Administration (NASA) and Department of Defense (DoD) cooperative efforts^{1,2,3} have reviewed the state-of-the-art carefully and pointed out deficiencies in the data available for existing materials as well as new material needs. This report summarizes successful efforts over the past two and a half years carried out to meet the need of electrically conductive passive thermal control material systems (TCMS).

The ultimate goal was to develop engineering quality tailorable electrically conductive, TCMS for coating space hardware and space structures. This was achieved via performing the following two tasks:

Task A. This task was to develop electrically conductive optically absorbing passive TCMS. The required properties goals were as follows:

- $\alpha_{\rm s} \geq 0.90$
- $\epsilon_{\rm N} \geq 0.90$
- Adhesion, thermal shock satisfactory for long term use, ASTM-D-3359A (4A or better)
- Space environment stable satisfactory in 1000 ESH UV-vacuum test
- Surface resistivity $\approx 10^6$ to $10^9 \Omega/\Box$

Task B. This task was to develop white, electrically conductive, passive (low α_s/ϵ_N) TCMS. The required properties goals were as follows:

- α_s Achieve minimum (≤ 0.18 , comparable to the state-of-the-art Z-93P)
- $\epsilon_{\rm N} \geq 0.88$
- Adhesion, thermal shock satisfactory for long term use, ASTM-D-3359A (4A or better)

- Space environment stable satisfactory in 1000 ESH UV-vacuum test
- Surface resistivity $\approx 10^6$ to $10^{10} \Omega/\Box$

These two tasks have benefited enormously through Internal Research and Development (IR&D) efforts at IITRI and partially funded program for the GPS-Block IIR hardware. In the next section, we summarize past efforts that guided us for selection of product concepts for material design, followed by a results section that first summarizes physical, optical and electrical properties data generated for each concept, followed by results from space environment simulation studies carried out at Aerospace Corporation [Dr. M. Meshishnek], Goddard Space Flight Center (GSFC) [Dr. L. Kauder/Ms. W. Peters], and Air Force Materials Laboratory [Messrs. P. Carlin/C. Cerbus]. The data generated in ESD testing carried out at GSFC [Dr. L. Kauder] for the selected concepts are also presented here.

The studies conducted on the selected concepts and new pigments have provided new conductive engineering TCMS. Some of them have already found applications on space hardware, some are waiting to be noticed by thermal designers, and some require further detailed studies to become the state-of-the-art for future space structures and hardware. A qualitative discussion is also presented at the end of this report on integration of electrically conductive thermal control coatings on space structures and space hardware.

2. TECHNICAL BACKGROUND

Several NASA and Department of Defense cooperative efforts have reviewed the state-of-the-art carefully and pointed out deficiencies in data available for existing spacecraft materials as well as the need for development of new coating materials. (See NASA-CP-3035)⁽¹⁾ A particular need identified by the Space Environments and Effects Project (SEE) was for electrically conductive coating which could be tailored to different property requirements to meet different mission needs. The following summary covers past as well as IR&D efforts at IITRI which have resulted in performing this current program.

2.1 PAST RESEARCH AND DEVELOPMENT ACTIVITIES

2.1.1 Electrically Conductive Absorbing TCMS. The need for electrically conductive absorbing TCMS is not new. The very early efforts were more devoted to absorbing TCMS mainly because of ease and availability of conductive phases. The modification of space environment stable nonconducting cosmic black based D111 silicate binder based TCMS to MH11Z was carried out for the Jet Propulsion Laboratory (JPL) in the early '80s. This experience was achieved to add needed conductive phase (graphite) to the material system to tailor volumetric resistivity of $\leq 1 \times 10^6 \ \Omega \cdot \text{cm}$. As expected, there were penalties to the solar absorptance value due to usage of conductive phases. JPL's in-house work concentrated on iron titanate as a pigment and was more resistive than the MH11Z formulation.

In the mid '80s under an Air Force funded program, Black Glass (BG: a carbon doped silica) was developed as atomic oxygen (AO) resistant pigment for baffle applications. These efforts demonstrated that the BG is not only AO resistant, but provides an opportunity to tailor very flat bi-directional reflectance distribution function (BRDF) response that is needed for baffle applications. Based on these efforts, two nonconducting formulations, MH21-I and MH21S/LO, were developed and are in use for baffle applications as well as absorbing TCMS with $\alpha_s = 0.97$ to 0.98 and $\epsilon_N = 0.90 \pm 0.03$. MH21-I uses an inorganic silicate binder whereas MH21S/LO uses stripped polydimethyl siloxane polymer, meeting necessary outgassing requirements of Total Mass Loss (TML) < 1.0% and Collectable Volatile Condensable Material (CVCM) < 0.1%. The typical values are TML = 0.15%, CVCM = 0.03. These developments

provided opportunities to tailor electrical conductivities using MH21-I and MH21S/LO as base nonconducting phases. Again, graphite, milled graphite fibers, SiC whiskers and B_4C , were among the candidates for the conductive phases. The experience gained in the early '80s was applied to develop formulations MH55IC and MH21-IC as electrically conductive inorganic TCMS, and MH21SC/LO as an electrically conducting flexible thermal control material system. The typical properties of these TCMS are given in Table 2-1. Our recent communications with NASA-Lewis Research Center's Dr. Purvis⁽²⁾ provided the most general criteria: Coating system with differential charging up to 10 volts (conservatively) and leakages current < 10 nanoamperes (1 nA preferable). This translates into the resistivity requirement being on the order of $10^9 \Omega/\Box$.

Based on the defined goals, Internal Research and Development (IR&D) work was first undertaken to tailor the electrical resistivity of black TCMS. The scientific basis of this work is best described in a review article by McLachlan et al. 17, who has provided guidance in designing the composite material systems. This work at IITRI used AO resistant BG as a pigment for the required stable optical properties, stripped polydimethyl siloxane (CVCM <0.1% and Total Mass Loss [TML] <1.0%) as a binder, and graphite fiber, graphite, and semiconductor grade boron carbide as conductive phases. Experiments were carried out to build a database to define threshold conductive phase weight fraction requirements and to select optimized processing requirements for each coating concept with desired optimum properties. This work was extended further in a program to develop conductive thermal control coating for GPS Block IIR hardware. 18 The coating systems that have been developed and have shown great potential are: MH55IC, MH21SC/LO, and MH41SCB/LO. The details of the typical properties of these coatings are listed in reference (3). Since each system of these material systems have been known to be space environment stable, the resultant conductive tailored formulations can be considered space environment stable. All of the above materials have tailorable electrical resistivity $\approx 10^6$ to $10^9 \,\Omega/\Box$, $\alpha_s \geq 0.90$, $\epsilon_T \geq 0.90 \pm 0.03$, and meet adhesion, thermal shock and thermal cycling requirements.

2.1.2 Electrically Conductive Low (α_s/ϵ_N) White TCMS. Our quest for electrically conductive TCMS is not new. Efforts in this area at IITRI date back to the early '70s. Initial efforts were to characterize the electrical resistivity of current state-of-the-art thermal control

| | Table 2-1. Sur | Summary of Black Thermal Control/Optical Coatings at AMCL/IITRI | Control/Optical Coa | tings at AMCL/IITRI | |
|-----------|-------------------------------|---|---------------------|---------------------|---|
| Coating | Pigment | Binder | αs | ίτ | Comments |
| D-111P | Carbon black WB-225 | Kasil 2130 | ≃0.95 | 0.90±0.02 | Stable, LDEF data. Not recommended for AO resistance. |
| MH11-ZP | WB-225 graphite | Kasil 2130 | 0.89 to 0.91 | 0.90 ±0.02 | Electrically conductive < 106 fl.cm stable. (JPL) |
| MH21-IP | Black glass | Kasil 2130 | 0.95 to 0.02 | 0.90±0.02 | AO-resistant; stable. |
| MH55-ICP | Black glass | Kasil 2130 | 0.90 to 0.93 | 0.90±0.05 | Electrically conductive 106-10° Ω/□ |
| MH21S/LO | Black glass | MHS/LO stripped polydimethyl siloxane | 0.95 to 0.98 | 0.90±0.02 | Stable, NASA-TM- 100768, recommended for AO resistance. |
| MH21SC/LO | Black glass graphite fiber | MHS/LO stripped polydimethyl siloxane | 0.90 to 0.93 | 0.90±0.02 | Electrically conductive 10⁴.10° Ω/□ |

materials such as Z-93 (a calcined ZnO pigmented potassium silicate), S13G/LO-1 (reactively encapsulated ZnO in stripped polydimethyl siloxane) and YB-71 (flash calcined zinc orthotitanate [ZOT] pigmented potassium silicate). Due to the use of YB-71 in Geosynchronous Orbit (GEO), much attention has been devoted to the non-contact measurements of charge developed on such thermal control surfaces. Such studies were carried out at Aerojet⁽⁴⁾, JPL⁽⁵⁾, and similar measurements were carried out for Z-93 at the NASA-Lewis Research Center, first in 1979⁽⁶⁾ and more recently⁽⁷⁾ in comparing Z-93 and Z-93P electrical resistivity values. In these studies, the electrical resistivity of silicate coatings was reported, along with contributions from moisture due to their porous nature. It was noted that complete removal of moisture requires about two weeks in vacuum of 10-7 to 10-8 Torr to achieve artifact-free resistance measurements. (8) In R&D efforts at IITRI, we have developed sample preparation and drying methods to measure surface and volumetric resistivities using a HP-4329 and HP-4339 resistivity meter. (9) The measurement procedure provided resistivity values that agreed well with measurements conducted with exhaustive efforts to remove moisture via exposure to vacuum. (8) Accordingly, the surface resistivity values for current state-of-the-art materials Z-93 and YB-71 are on the order of $10^{13} \Omega/\Box$ and for S13G/LO-1 is on the order of 10^{13} to $10^{14} \Omega/\Box$.

The earliest program to develop electrically conductive thermal control material was funded through the U.S. Air Force Materials Laboratory (AFML) and was concluded in 1976 at IITRI. These early efforts used sodium and lithium silicate as binders and also evaluated their Ultraviolet/Vacuum (UV/VAC) stability for 1,000 ESH. Efforts were also made to evaluate antimony-doped SnO₂ pigmented potassium silicate coatings. The results of these efforts are summarized in Table 2-2, and UV/VAC stability results are given in Figures 2-1 to 2-3.

There were no additional systematically funded efforts for the development of conductive thermal control coatings until the very recent efforts at JPL for the Cassini Project; at AZ-Technology for Tethered Satellite; and at IITRI for GPS Block IIR (during 1993, 1994). The interplanetary mission of the Cassini Project required the material system to meet the criteria of $\rho_v \propto t < 2 \times 10^{10} \ \Omega \text{-cm}^2$ ($\rho_v = \text{volume resistivity in } \Omega \text{-cm}$, t = thickness in cm). The coating also must meet non-contact charging criteria of 10 volts, when samples were irradiated with 10 KeV electron at a flux of 0.5 nA/cm². Solar absorptance (α_s) needed to be kept as low as

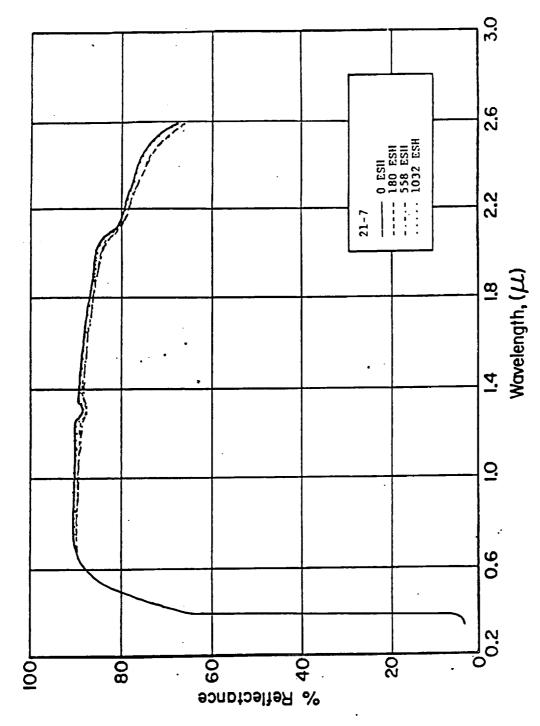


Figure 2-1. Reflectance of ZnO/Lithium Silicate.

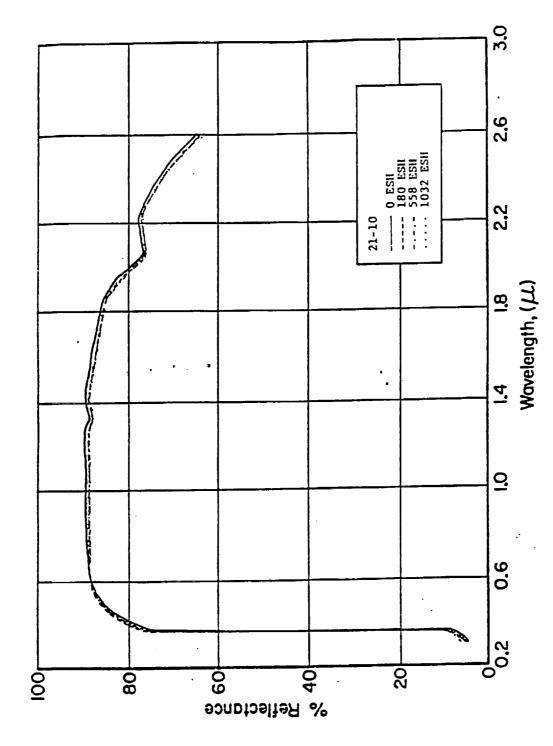
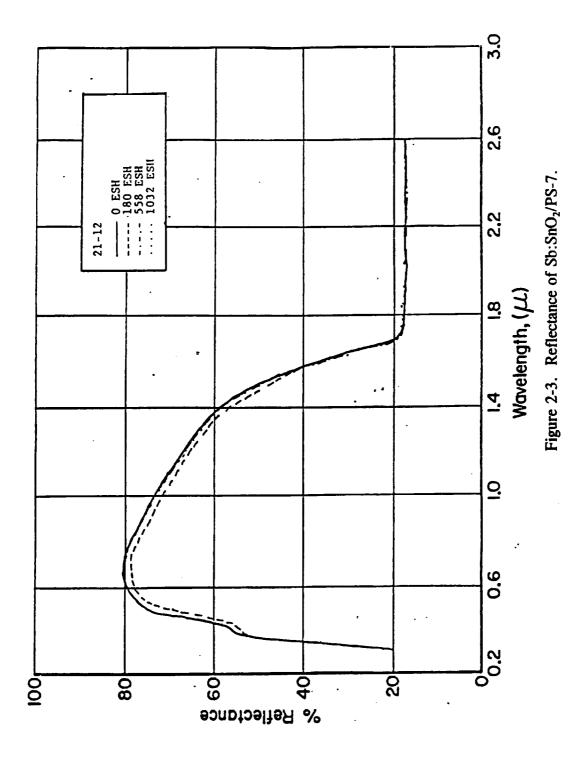


Figure 2-2. Reflectance of ZnO (Calcined)/Lithium Silicate.



| Table 2-2. AFML-TR-76 | -232/IITRI Work (1976) |
|--------------------------------------|------------------------|
| Inorganic Binder | Surface Resistance, Ω |
| ZnO/N ₂ SiO ₃ | 8.0 x 10 ⁸ |
| ZnO/Li ₂ SiO ₃ | 1.5 x 10 ⁹ |
| ZOT/K_2SiO_3 | 2.5 x 10° |
| ZOT/Na_2SiO_3 | 1.7 x 10° |
| ZOT/Li_2SiO_3 | 3.8 x 10° |
| SnO ₂ :SB/PS-7 (High PBR) | 2.1 x 10 ⁷ |
| SnO ₂ :SB/PS-7 (Low PBR) | 1.2 x 10 ⁸ |

Antimony doped SnO₂ pigment was supplied by AFML.

possible with thermal emittance $\epsilon_T \simeq 0.90 \pm 0.02$. In the recently concluded program, material systems considered were Al doped ZnO pigment from Union Miniere, Brussels, Belgium, and NS43G (ZnO with 1% Al₂O₃) yellow pigment from Space Craft Coatings, Inc. of Gambrills, Maryland. (ZnO with 1% Al₂O₃) yellow pigment from Space Craft Coatings, Inc. of Gambrills, Maryland. The French product PCB-Z flexible white conductive thermal control coating was also included in this study. None of these material systems met the required goals. Based on these findings, it was recommended by JPL that some of the requirements be relaxed for the conductive thermal control coatings. The work continued with charging criteria relaxed to 20 volts and α_s (Beginning of Life [BOL]) = 0.22. The resultant coating is now based on a 50:50 mix of NS43G and doped ZnO pigment, the details of which can be found in JPL Spec BS516165A, Rev. A.¹³

The conductive coating used on the Tethered Satellite was developed based on indium stannate as the pigment, and has an α_s (BOL) ≈ 0.28 (<0.30). The electrical conductivity of such a system is considered tailorable and will depend on the pigment-to-binder ratio. The Phase I development is proceeding at AZ Technology under an SBIR Phase I contract. The coating system can provide a viable conductive coating, if a compromise on α_s (BOL) can be made, and if (the change in α_s $\Delta\alpha_s$) due to the space environment effects is shown to be acceptable.

Other efforts made at JPL during Cassini project studies⁽¹⁴⁾ involved controlled doping of ZnO with various dopants and treatments. Among these candidates, the Indium-doped ZnO showed a greater potential in charging tests, but α_s (BOL) was near a relatively high 0.29. The electrical resistivity of such silicate binder material systems was on the order of 10^6 to 10^7 Ω/\Box .^(11,14) Efforts in this direction as well as with different doping approaches were suspended mainly due to high α_s (BOL) and program limitations.⁽¹⁴⁾

Recent efforts at IITRI date back to 1989. (15) We first proposed the concepts of stable, reliable electrical properties development through careful combination of conductive and insulative phases to build a reliable TCMS. Research and development to build reliable microstructurally engineered resistors (during the early '80s) was the basis of this work. The percolation theories, requirements of percolation threshold, morphologies and distribution of conductive phases and their effects on resultant resistor properties and reliability, were the heart of these efforts. The electrical resistivity goals for coatings have not received complete attention. In practice, they are mission dependent, and material system design lacked needed directions from the mission office. The designer often attempted to work with available state-of-the-art materials rather than engaging in the concurrent engineering exercise of developing much needed new materials. We, at IITRI, have more recently received needs related input for various situations. As iterated in Section 2.1.1, the recent communications with NASA-Lewis Research Center's Dr. Purvis⁽¹⁶⁾ provided the most general criteria: Coating system with differential charging up to 10 volts (conservatively) and leakages current < 10 nanoamperes (1 nA preferable). This translates into the resistivity requirement as on the order of 10° Ω/□.

Based on the defined goals, an IITRI IR&D project was first undertaken to tailor the electrical resistivity of black TCMS. The scientific basis of this work is best described in a review article by McLachlan et al.⁽¹⁷⁾, who has provided guidance in designing the composite material systems. The IR&D for white thermal control coating proved that doping of the existing stable pigments like S13GP and ZOT (zinc ortho titanate) is a must.

Before going into a summary of the IR&D efforts and studies conducted during a recent program for GPS hardware needs for white (low α_s/ϵ_T) thermal control coatings, we need to discuss the reliability and coating needs as applied to conduction mechanisms. The charging of spacecraft surfaces as modelled by the NASCAP computational code, need input on material

properties/parameters. There are 19 input variables. One needs to know the following input parameters that are related to the material properties:

- 1. Relative dielectric constant
- 2. Bulk conductivity (volumetric conductivity/resistivity)
- 3. Atomic number
- 4. Maximum secondary yield due to electron impact
- 5. Primary electron energy that produces maximum secondary yield
- 6. Range of electrons and energy dependence
- 7. $R = P_7 E^{P8} + P_9 E^{P10}$ related parameters
- 8. Secondary yield due to 1 KeV protons
- 9. Incident proton energy that products maximum secondary yield
- 10. Photo electron yield for normal incident sun light
- 11. Surface resistivity (surface conductivity)
- 12. Maximum (absolute) potential attainable before a discharge can occur
- 13. Radiation-induced conductivity (K)
- 14. Radiation-induced conductivity power (A)
- 15. Material density

Other parameters deal with thickness of the coating and physical shape and boundary conditions. These input parameters also define some of the characterization needs for the candidate coating material system.

To the best of our knowledge, no systematic effort has been made to measure critical dielectric properties. The charging computational code (NASCAP) needs photo emission and secondary emission properties of a variety of materials. The related databases on materials and coatings have not been systematically assembled. The sensitivity of the NASCAP code output is also very crucial for designing tailorable conductive TCMS. Such sensitivity analyses for chosen orbits have not been performed or published. According to our communications with Dr. Purvis⁽¹⁹⁾, some attempts were made, but results were not published for GEO conditions. The NASCAP charging calculations cannot rank the importance of individual mechanisms. It is apparent that the most important properties will be volumetric and surface resistivity of the coating. The next will be secondary emission yield. Since the thermal control materials are essentially dielectric in nature, they will exhibit leakage current. Since our goal is to limit leakage current (<10 nanoamperes) under differential charging in orbit, they will be maximum in the portions of the orbit where flux currents are maximum (anti-solar side of the orbit). We know these currents are high, and therefore, for materials with high resistivity, one needs to have additional means to manipulate leakage currents. The only available process then, to

control leakage current is <u>secondary emission</u>. The coating material system needs to have the availability of conductive leaky percolation paths and also should have appropriate secondary emission properties to limit the leakage current. Again, for such a dielectric, the maximum differential potential will be induced, when mean <u>energy</u> encountered is at a maximum. This will be found on the solar side of the orbit, i.e., along with photo emission.

Garrett⁽²⁰⁾ has discussed the use of emissions (photo electron, secondary electron emission, back scattered electrons) as current mechanisms for charging control. Such emissions do not reduce charge deposition in dielectrics. They can, depending on the grounding scheme, lead to multipacting and preferential deposition of contaminants. Another effect is pinhole formulation and focussing of electrons to cause punctures in coating systems. Due to these concerns the secondary emissions to manipulate leakage current should be used prudently.

In view of the qualitative inputs from spacecraft charging studies summarized here, a survey was carried out to choose possible candidates. The input provided by the High Temperature Materials Information Analysis Center (HTMIAC) at Purdue University⁽²¹⁾ is provided here in tabular form, along with comments on potential pigment materials (Tables 2-3 and 2-4). The current state-of-the-art pigments and coatings, that are proven stable in space environment were also tested for volumetric resistivity (ρ_v). The product ρ_v xt was calculated as an indicator of possible surface resistivity values. These values are also listed in Table 2-5. These data, when considered with prior work in Europe^(26,27,28) to develop stable electrically conductive coatings, points us toward a more successful selection of concepts. The following discussion from our report for the GPS hardware⁽¹⁸⁾ is provided here to illustrate the care taken in selection of concepts during the recent work that was initiated for GPS hardware.

From the available data in Tables 2-3, 2-4 and 2-5, pigments which can exhibit conductive behavior are ZnO, SnO₂, InSnO_x, Zn₂SnO₄ and Zn₂TiO₄. Of these, ZnO, SnO₂, Zn₂SnO₄, and Zn₂TiO₄ have been studied as potential pigments for TCMS by Zerlaut et al. (22) and Harada (22,30) during initial development of Z-93 and YB-71 thermal control coatings, the current state-of-the-art materials. Coatings incorporating the pigments SnO₂ or Zn₂SnO₄ were not as stable as Z-93 and YB-71. (22,29) It is also noted that the systems based on the pigments SnO₂, Zn₂SnO₄ or InSnO_x, are conductive due to an oxygen deficient defect structure, and the

| | Table 2-3. | Table 2-3. Literature Review Summary ⁽²¹⁾ | mmary ⁽²¹⁾ | |
|----------------------------------|-----------------|--|--|--------------------|
| Material | Density, gms/cc | Volumetric Resistivity (f)·cm) | Relative Dielectric Const. £, at 1 MHz | Comments |
| SnO ₂ | 6.993 | 1.0E+00 | 9.0 | * |
| SiO ₂ | 2.800 | 1.0E+13 | | |
| ZnAl ₂ O ₄ | 4.608 | 1.9E+03 | | |
| TiO_2 | 4.245 | 2.13E+09 | 78.90 | |
| ZnWO4 | 7.872 | | 16.10 | |
| Zn ₂ SiO ₄ | 4.251 | 3.6E+07 | 7.70 | |
| OuZ | 5.676 | 1 to 10 | 12.00 | T. K. Gupta (1990) |
| ZnO-doped (AI) | | 0.1 to 1 | | |
| ZnO (gb) | | 1.0E+12 | | |

*Becomes conductive with defect structures.

| | Table 2-4. Other Potential Pigments |
|-------------------|---|
| Material | Comments |
| SnO ₂ | -vc TCR, conductivity-defect structure dependent, e.g., vac. (10 ⁻⁷ TORR) baking at 100°C. Needs sealing. Gives $\rho_v = 50~\Omega$ ·cm. Doped SnO ₂ = 50 K 50 Ω ·cm (P.Q. Corp. developmental sol) no sealing needed. |
| InSO _x | Conductivity-defect structure dependent. Thick paste - developmental thick paste. (ESL) $\rho_{\rm v}=10^3$ - $10^7~\Omega$ ·cm as polycrystal. |
| InSnO, | -vc TCR, $\rho_v = 10^3 \cdot 10^9 \Omega \cdot \text{cm}$ as polycrystals. $\alpha_{\text{S/BOL}} \rightarrow 0.18$ to 0.22 (French Data). French and Canadian flight data available, $\Delta \alpha_{\text{S}}$ - low. $> \Delta \alpha_{\text{S}}/\text{Z}$ -93. |
| ZnTiO, | -vc TCR, semiconductor $\rho_v = 10^8 \Omega \cdot \text{cm}$ as polycrystal. $\alpha_{\text{S/BOL}} \rightarrow 0.11$ to 0.13, can be doped to improve ρ_v . Data available on simulation and synergism. (French) CNES efforts devoted to doped ZOT. |

| Table | 2-5. Results of Volumetric Electrical Resistiv HP-4329A, High Resistance Meter | Table 2-5. Results of Volumetric Electrical Resistivity Measurements HP-4329A, High Resistance Meter | nents |
|--------------------------------------|---|--|--|
| Material | p, Ocm | ρ, x t, Ω·cm² | Comments |
| ZOT (1000°C, 24 hrs) | 1.27 x 10 ¹⁰ to 3.15 x 10 ¹⁰ | 3.23 x 10 ⁸ to 8.0 x 10 ⁸ | $t = 10 \text{ mils}, \alpha_{sBOL} \rightarrow 0.10$ |
| ZOT (1000°C, 24 hrs) | 2.4 x 10 ¹⁰ to 3.9 x 10 ¹⁰ | 6.01 x 108 to 7.84 108 | $t = 10 \text{ mils}, \alpha_{sBOL} \rightarrow 0.10$ |
| Calcined ZnO, SP-500 (990°C, 16 hrs) | <0.5 z 10° to 1.5 x 10° | | Yellowish $t = 10 \text{ mils}, \alpha_{sBOL} \rightarrow 0.2$ |
| S13G (950°C, 16 hrs) | 3.7×10^{11} to 6.0×10^{10} | 5.07×10^9 to 1.03×10^9 | $\alpha_{\rm sBOL} \rightarrow 0.15$ |
| S13G (950°C, 16 hrs) | 8.4×10^{11} to 1.6×10^{11} | 1.07 x 10° to 2.15 x 10° | $\alpha_{\rm sBOL} \rightarrow 0.15$ |
| Z-93, 5 mils (100°C, 4 days) | 3.65 x 10 ¹⁴ to 9.26 x 10 ¹³ | 1.23 x 10 ¹³ to 9.2 x 10 ¹² | $\alpha_{\rm sBOL} \rightarrow 0.15$ |
| YB-71, 10 mils (100°C, 4 days) | 9.0 x 10 ¹¹ to 9.8 x 10 ¹¹ | 1.17×10^{10} to 1.34×10^{10} | $\alpha_{\rm sBOL} \rightarrow 0.15$ |
| S13GP/LO-1 8 mils | 10 ¹³ to 10 ¹⁴ | ~ 1013 | t = 8 mils $\alpha_{sBOL} \rightarrow 0.15$ |

ZOT, S13G, Z-93 are all stable to UV-Vac 1,000 ESH exposures.

stability of electrical properties depends on the stability of these oxygen deficient defect structures. (21,24,25)

Vardin and Duck's (27) efforts in experimental verification of charging provides a good insight into the stability of electrical properties of coatings based on these defect structure oxides. This work involved conductive oxide mixtures with a resistive matrix (polyamide) which were irradiated to 5 to 20 KeV (in steps of 5 KeV) electrons for different oxide loadings. The oxides used in this study were Al₂O₃, Ga₂O₃, In₂O₃, In₂SnOx, and ZnO. The charge build-up for these coatings was tested at room temperature and at -180°C. The comparison of In₂O₃ and InSnOx (In₂O₃:SnO₂) made by Vardin and Duck⁽²⁷⁾ is interesting. They argued that the function of SnO₂ in InSnOx coating is to provide conductive paths through the defect structure. We believe In₂O₃ alone can provide such conductive paths, and it may not be necessary to provide such defect structure paths. Another valuable input is that indium alone is responsible for the secondary emission with a yield of 4.8. (20) The difference in the resultant conductivities depends on oxide loading. For equal oxide loadings, InSnOx is slightly more conductive than In₂O₃ alone, e.g., for a PBR of 4:1 InSnOx-based coating exhibited an electrical resistivity of 2.7 x $10^5 \,\Omega^{\bullet}$ m versus In₂O₃-based coating having resistivity of 1.4 x $10^6 \,\Omega^{\bullet}$ m. The space environment stability of such coatings on second surface mirrors for thick films (several microns) was also studied by Dutat et al.²⁸ who reported $\Delta \alpha_s$ (InSnOx) ≈ 0.15 , whereas $\Delta \alpha_s$ (In₂O₃) ≈ 0.03 for an equivalent exposure of 3 sun years.

Vardin and Duck's⁽²⁷⁾ observations in charging experiments involving ZnO are also interesting. They observed that for electrons >20 KeV, the charging was only a few volts (< 150 V) whereas at low energies, the surfaces charged to thousands of volts. This is consistent with values shared by other researchers. These results suggest an adjustment of the solid-state chemistry of ZnO pigment by incorporating enough Zn interstitials in the ZnO matrix, so as to increase carrier concentrations to tailor in the required conductivity value. The coating made from such pigment may provide enough conductivity to mitigate charge dissipation. One needs to stabilize the Zn interstitials incorporated in the ZnO matrix, and the early work of Harada²⁹ sheds some light on this concept. The available data on ZnO pigmented potassium silicates suggests that such systems have been studied up to 270 ESH, and do show the space environment stability (Table 2-6). Let us call this candidate treatment concept Z-93CXY. The

| | Table 2- | 6. Conductive Theri Z-93CXY an | nal Control Materia d Z-93SCXY | al Systems | | | | |
|------------|------------|--|--|--------------------------|--|--|--|--|
| (A) | Z-93SXY: | (propriety formulati $\alpha_s \le 0.20$, $\epsilon_T = 0.9$ Electricity resistivity $\Delta \alpha_s$ - Per IITRI-C20 | 00 ± 0.03 $y = 10^9 \Omega/\Box$ (Tailor) | able) | | | | |
| Exp | osure, ESH | Solar Factor Reflectance, % | | | | | | |
| | | | 440 nm | 600 nm | | | | |
| | 0 | | 89.0 | 95.5 | | | | |
| | 270 | 3 85.5 94.0 | | | | | | |
| (B) | Z-93SCXY: | $\alpha_s \le 0.18$, $\epsilon_T = 0.9$ Electricity resistivity $\Delta \alpha_s$ (for 1,000 ESH | ZnO/Kasil 2130 (prop 0 ± 0.03 $\gamma = 10^9 \Omega/\Box$ (Tailor exposure) = 0.01 (Sory to Al substrates/s | able) See Figure 2-4) | | | | |

concept was further evaluated to verify feasibility during the GPS program. (18) The typical values achieved for surface and volume resistivity listed in Table 2-4 are encouraging. (See Reference 18 for details.)

Stabilization of Zn interstitial in a ZnO matrix is not a new subject. We, at IITRI, have done this routinely for decades using the method reported and for the S13G pigments. (See Zerlaut et al.⁽²²⁾) The use of such treated ZnO pigments that have been proven stable in 1,000 ESH exposures can also be used to form coatings with silicate binders. Let us call this approach Z-93SCXY. Work was undertaken recently at IITRI as part of the GPS program⁽¹⁸⁾ to verify the feasibility of this concept. The typical values observed are listed in Figure 2-4. The results obtained for surface resistivity measurements^(9,18) are encouraging.

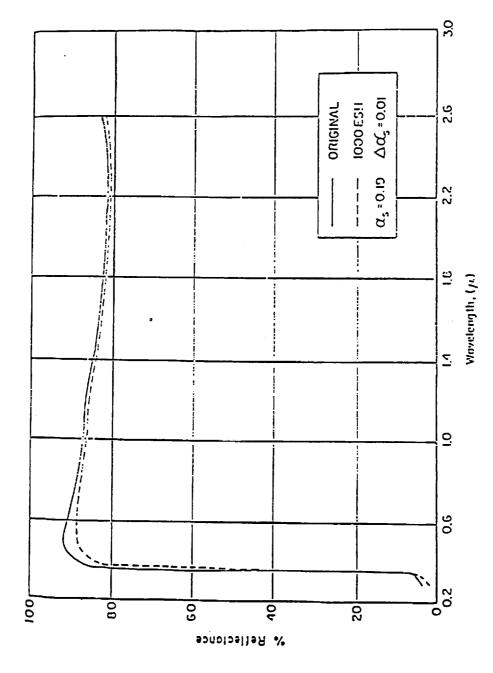


Figure 2-4. 1,000 ESH UV-Vac Stability of Z-93SC55.

The GPS hardware needs⁽¹⁸⁾ are unique. Due to the proximity of the hardware to the communication equipment, there is an added requirement that the electrical resistivity of the coating material system should not drop below 10⁷ Ω•cm to prevent interference in transmission. Due to this condition, it was decided that the candidate material system should have a BOL resistivity of the order of 10⁹ Ω•cm or slightly higher. The only data available on the space simulation testing on oxygen deficient defect structured oxides was by Dutat et al. ⁽²⁸⁾ It indicates that all oxides that depend on a defect structure for electrical conductivity exhibited a lowering of electrical resistance on the order of 2 to 3 decades due to a space environment (UV + electrons). A conscious decision was made to avoid candidate oxides that depend on oxygen deficient defect structures. This conservative approach thus reduces the probability of the coating system becoming too conductive during the lifetime of the space hardware.

Feasibility studies were also undertaken during the program⁽¹⁸⁾. Efforts were directed towards concepts based on the experience of Vardin and Duck(27), Dutat et al.(28), and the JPL work.(11,12,13,14) The qualitative discussion provided here earlier for the choice of conduction mechanisms and feedback from Garrett(20) on mitigation strategies clearly indicates that the ideal material system should provide percolation paths, as provided in Z-93CXY and Z-93SCXY with dependence on proven stabilized solid-state chemistry of pigments, along with the secondary emission provided at an optimum level. The concept is to dope the S13G pigment with a dopant that can provide maximum needed secondary emission yield at required optimum low concentration. This concept can help us to design inorganic and organic flexible thermal control coating material systems. Indium was selected as the dopant, and doped S13GPSC pigments were prepared at three candidate dopant concentrations. Three different process routes were Typical values for these doped materials are used to check feasibility of this concept. summarized in Table 2-7, indicating excellent tailorability for an inorganic coating system: Z-93SCLMXY, and for an organic coating system: S13GCLM/LO-XY. Both concepts showed excellent BOL, α_s , ϵ_T and resistivity values. The UV-vacuum stability of these concepts awaits verification.

Due to the complexity and cost of In doping of S13G pigments, a simple approach was also used for preparing "silicated indium." These trials are reported in detail in Reference 18. Such processing helps to reduce processing losses of costly indium, but demands higher

| Concept Typical Emittance* (ϵ_T) Typical Solar (ϵ_T) Typical Solar (ϵ_T) Typical Solar (ϵ_T) S13GIS/LO-41 0.90 ± 0.03 <0.18 (0.17 ± 0.03) S13GCI6/LO-41 (t = 6.1) 0.90 ± 0.03 0.16 ± 0.03 0.16 ± 0.03 S13GCI6/LO-41 (t = 7.0) 0.90 ± 0.03 0.16 ± 0.03 0.16 ± 0.03 Z-93SC1605 0.90 ± 0.03 <0.18 <0.18 | Table 2-7. | | lypical Properties of Concepts S13GC/LO-XY and Z-93SCLMXY | SCLMXY |
|--|------------------------|---|---|--|
| 0.90 ± 0.03 < 0.18 0.90 ± 0.03 0.17 ± 0.03 0.90 ± 0.03 0.16 ± 0.03 0.90 ± 0.03 0.16 ± 0.03 0.90 ± 0.03 < 0.18 | Concept | Typical Emittance* (f _T) | Typical Solar Absorptance** (α,) | Typical Surface Resistivity*** (Ω/□) |
| 0.90 ± 0.03 0.17 ± 0.03 0.90 ± 0.03 0.16 ± 0.03 0.90 ± 0.03 0.16 ± 0.03 0.90 ± 0.03 < 0.18 | S13GIS/L0-41 | 0.90 ± 0.03 | < 0.18 | <109 |
| 0.90 ± 0.03 0.16 ± 0.03 0.90 ± 0.03 0.16 ± 0.03 0.90 ± 0.03 < 0.18 | S13GC16/LO-41 (t= 4.8) | 0.90 ± 0.03 | 0.17 ± 0.03 | < 10° |
| 0.90 ± 0.03 0.16 ± 0.03 0.90 ± 0.03 < 0.18 0.90 ± 0.03 < 0.18 | S13GC16/LO-41 (t= 6.1) | 0.90 ± 0.03 | 0.16 ± 0.03 | <109 |
| 0.90 ± 0.03 <0.18 <0.90 ± 0.03 <0.18 | S13GC16/L0-41 (t= 7.0) | 0.90 ± 0.03 | 0.16 ± 0.03 | <109 |
| 0.90 ± 0.03 < 0.18 | Z-93SC1605 | 0.90 ± 0.03 | < 0.18 | < 106 |
| | Z-93SC1255 | 0.90 ± 0.03 | <0.18 | <109 |

Is the thickness of the coating in mils

Measurements were performed using Gier-Dunkel DB-100 infrared reflectometer.

With < signs, the values are Measurements were performed using Lambda 19 UV-Vis-NIR spectrophotometer. approximate.

Measurements were performed on HP-4329A High Resistance Meter, following sample preparation method (19). * *

"silicated indium" loading in the coating to achieve similar resistivity values in the coating material system. Future validation efforts are needed in this area for evaluating the impact of cost versus property tradeoffs. The feasibility study results indicate that property goals are achievable economically for the concepts selected here for screening study.

In view of the GPS hardware program goals, the concept Z-93SCXY was considered to be an established concept. It meets the required conductivity needs of the mission, and required space environment stability data is also at hand. The concept needs to be optimized to minimize cost, or tailor the electrical properties to desired values with ease.

2.2 CONDUCTIVE MATERIAL SYSTEMS: CONCEPTUAL DESIGNS

A summary of the literature review and past efforts in the area of conductive thermal control coatings were presented in the previous section. This summary clearly indicates the need for further efforts on microstructurally engineered material designs evaluated in recent feasibility studies. (18) In this section, we shall discuss the TCMS conceptual design selected for the screening study. The conceptual designs were carried out in two areas. The first one is related to the area of black conductive thermal control coatings, and the second is related to low α_i/ϵ_T reflector, white conductive thermal control coatings.

2.2.1 Black Tailorable Conductive TCMS Conceptual Design

In the area of conductive optical blacks, our recent work⁽¹⁸⁾ has resulted in the development of coating systems: MH55-IC, MH21SC/LO, D36SCB/LO, D21S/LO, D21SC/LO, and MH41SCB/LO. Typical properties of these coatings systems were given in an earlier section. They all meet the required property goals of this screening study, hence these material designs are the best candidates.

Apart from concepts from already developed in a previous study⁽¹⁸⁾, this study also included conductive thermal control material designs that depend on the concept of doping. IR&D studies at IITRI have resulted in doped B.G. and several doped "leaky" silicate binder systems. They provide unique material design opportunity to tailor electrical performance characteristics without compromise of absorption properties or penalties to the emittance. This approach can provide several material design concepts that depend on combinations of doped

pigment and doped binders. The screening study concentrated on formulations listed above from the study⁽¹⁸⁾, and only a couple concept formulations that can give feedback about doping.

2.2.2 Low α_s/ϵ_T , White Tailorable Conductive Thermal Control Coating Systems: Conceptual Design

In the area of low α_s/ϵ_T coatings, we have investigated concepts that showed potential in a prior study⁽¹⁸⁾. All proposed concepts used ZnO (SP-500) as a basic pigment material, based on the conservative rationale discussed and justified in an earlier section. The following conceptual material designs were evaluated in the screening study. The description of each material design (pigment concept) and the processes are discussed here.

- Concept Z-93CXY: This coating system used treated ZnO as the pigment and A. Kasil 2130 potassium silicate as a binder system. The goal of this pigment treatment was to stabilize the required concentration of Zn interstitials in ZnO matrix to meet the conductivity needs of the coating. Harada⁽²⁹⁾ has listed these treatments as Z-36 and Z-52 in his original work. The goal of this work was to define a processing window for the treatment so that the resultant ZnO (SP-500) has enough required interstitials remaining in ZnO without presenting any penalties in the BOL value of α_s of the resultant coating. Thus, the work involved was developing a treatment parameter matrix of calcination temperature and calcination time during the required flash calcination to control the solid-state chemistry of ZnO. Based on the work done on this concept⁽¹⁸⁾, we anticipate that for each pair of processing parameters, a different optimized pigment-to-binder ratio (PBR) is needed to tailor optical performance. The database was also developed to tailor the electrical resistivity values. This material design, due to its simplicity, offered be the most cost-effective and consistent solution.
- B. Concept Z-93CXY(IS): This concept used experimental data generated for an earlier concept along with the addition of silicated indium (In), with Kasil 2130 as a binder. The feasibility study⁽¹⁸⁾ has identified In as one of the desirable ingredients. We also found that silicated In does not provide any additional absorption in the Ultraviolet/Visible (UV/Vis) region as long as the silicated indium concentration is kept low enough and silication is carried out prudently. (18)

Efforts were made in two areas. First, to develop and optimize silicated In oxide processing. The processing route described in Reference 18 provided a good starting point. In the second area, efforts were made to determine optimum concentration of silicated indium in Z-93CXY(IS). This work involved preparing a database on resistivity versus concentration of silicated indium, and a database on α_s (BOL) versus concentration of silicated indium in the Z-93CXY(IS) coating. Silica is usually considered a strong degradable component. [See Zerlaut and Harada⁽²⁹⁾] Hence, this material concept was to receive attention only if the need arose. Thus, very little effort was devoted to this material design.

- C. Concept S13GCLM: This pigment concept deals with development of In-doped microencapsulated ZnO pigment. The study has listed two sol-gel based process routes that need optimization. Efforts were made, first, to determine the optimum dopant level (LM), by choosing three different dopant levels. Recommendations given in Reference 18 were used to economize processing. The doped pigment was processed via two routes. The first one was high temperature (1000°C) doping. Enough IR&D efforts were done on this and were summarized in the section on IR&D efforts. The second one follows the S13G route with use of doped binders.
- C-1. Z-93SCXY-LM: This concept used S13GCLM pigment with Kasil 2130 and other candidate doped binders for candidate dopant levels. The database of coating properties versus PBR was prepared for determining optimum formulation and optimum dopant levels.
- C-2. S13GCLM/LO-XY: This concept used S13GCLM pigment with stripped polydimethyl siloxane as binder for the candidate doping levels. The coating properties were measured by selecting optimum PBR formulation and the optimum dopant levels for this flexible system. This concept is named DS13N/LO-XY in the results section.
- C-3. Z-93SCXY: This concept used only the S13G and S13N pigments with Kasil 2130 and the candidate sodium silicate (SS-55) as the binder systems. The data

for this concept is available and is reported in Reference 18. The work on this concept provided a baseline database, and is presented in the results section.

2.3 CHARACTERIZATION NEEDS

2.3.1 Non-Contact Charging Tests and Needs

Brief efforts were devoted in initial IR&D during an earlier program⁽¹⁸⁾ and during screening studies to understand practices used for space charging experiments. Our efforts revealed that insufficient documentation is available. Documented test procedures were shared by only JPL⁽¹¹⁾ as an appendix to material specifications. Most of the test practices that are currently in use are based on experience with conductive materials. The usual methods to measure charge developed need to be augmented with data on charge dissipation. Such data then can help in deriving the value of conductivity in the coating due to percolation paths. The desired method would be to use existing static charge development setups to measure voltage dissipated. It is also referred to as the surface voltage decreasing method in the literature. (See Levy et al.⁽²⁶⁾) This approach is as follows:

The sample is first exposed to charging by the means of an electron beam. Sample irradiation is then interrupted and its surface is then continuously monitored by a trek voltage probe. Data are collected for surface potential versus time. The following simple analysis then applies.

The thermal control conductive coating applied on the metallic substrate is considered as a capacitor. Then the capacitance of the sample is written as:

$$C = \epsilon_0 \epsilon_r (A/t)$$

C = capacitance, A = area, t = thickness of coating, ϵ_0 and ϵ_r = usual dielectric constants for material system. The capacitor is considered discharging into its own resistance (the leakage resistance built in the thermal control system due to percolation paths)

$$R = \rho (t/A)$$

R = resistance, ρ = resistivity of TCMS. If V_s is the surface voltage, then

$$V_s = -R (dQ/dt) = -RC (dV_s/dt)$$

$$V_s = -\rho \epsilon_o \epsilon_r (dV_s/dt)$$

$$\rho = 1/\sigma = -\epsilon_o \epsilon_r (dV_s/dt)/V_s$$

Note that the conduction is not ohmic and R is dependent on voltage. The practical fallout is to present results as plot of log (resistivity) versus V_s^2 as a straight line, indicating that conduction is of the Pool-Frankel type and field assisted.

The current practices are also limited in measuring the secondary emissions. The problem of measuring the secondary emission on a dielectric material is complicated by the fact that the surface under measurement becomes charged during the measurement. This creates two problems:

- (1) Target/substrate surface potential must be accounted for to correct the electron energy at the sample level.
- Barrier effects are associated with positively built surface potentials. These barrier voltage prevents the low E secondaries to reach measurement electrodes. This must be controlled to have valid measurements. Thus, the static method that provides charge and secondary emission input for conductive samples cannot give the same reliable information on the dielectric surfaces. One needs to use a dynamic method with pulsed electron techniques. The pulse duration of such a source is critical. Pulse duration should be no longer than a threshold to be determined. To limit the barrier effects in dynamic testing, one may further have to limit peak-to-peak time in consecutive pulses to measure true material properties.

At the present time, no one in the U.S. is actively pursuing dynamic testing. The small amount of data available in literature⁽²⁶⁾ is for 10 μ sec pulses with pulse-to-pulse time of the order of 0.2 millisecond. The current practices used at IITRI are summarized in the form of an overview in the IITRI document.⁽⁹⁾ The procedure requires sample conditioning to minimize moisture contributions. The HP-4339 also collects voltage versus time data, but we have reported only steady-state resistance numbers are reported here using the method.⁽⁹⁾ These resistance values are representative of material formulation and microstructures and are considered adequate to carry out the material design exercise for this screening study.

3. RESULTS AND DISCUSSION

3.1 BASELINE MATERIALS AND RELEVANT DATA

For the evaluation of selected conductive concepts for material systems, the engineering assessment can only be fruitfully made by comparing the new suggested concepts with carefully prepared "heritage" materials that are currently being used as state-of-the-art materials. Although most of them are nonconducting, their optical, physical and mechanical performance are known with acceptable certainty. Since newer concepts are designed as the variations on the existing base materials and processes, the baseline TCMS manufacturing, sample preparation and characterization were first undertaken.

Baseline formulations were also the subject of recent requalification studies (see AMFL-TR-94-4126) to qualify new sources of raw materials. The findings from that study were to be revisited, so baseline material were prepared very carefully. Care was taken to avoid contamination during manufacturing of material systems and during their deposition as well, as during characterization. The use of newly made fresh materials was emphasized.

The following is a summary of the nonconducting baseline concepts that were selected for screening study for comparison purposes. Another motive was to learn more about reproducibility-related issues raised during the requalification efforts by comparing data that gets generated on these carefully prepared batches of baseline materials. A wide latitude was taken in selecting the baseline material systems, so that selected candidate can help in fruitful comparison during the study of their conductive counterparts.

The baseline material selection for Task A - black thermal control materials systems was straightforward. The concepts selected as a baseline were MH21-IP, MH21S/LO and a new system, D21S/LO, that used cosmic black in stripped polydimethyl siloxane polymer. The wider selection of black candidate pigments in silicate and stripped polydimethyl siloxane was the subject of investigation during recent small programs here at IITRI for GPS Block IIR. The pigments included in those studies were TiC, B₄C, CuO, MnO, and FeO. A mixture of CuO, MnO and FeO is currently being used in formulation MH2200. It was not considered as a baseline material due to its outgassing properties and also due to the fact that it uses unstripped

siloxanes. All of the above mentioned pigment materials are semiconducting in nature, hence, they can also help to formulate conductive versions. For this reason, the baseline nonconducting materials for black TCMS were limited to the three listed candidates.

The nonconducting baseline white (low $\alpha_{\rm S}/\epsilon_{\rm N}$) formulation selection was also straightforward. All of the formulations that were reformulated per AFML-TR-94-4126 were considered as baseline. The baseline candidates included are: Z-93P, S13GP/LO-1, YB-71, YB-71P, S13GP/LO-41, ZOTS/LO-41, and a ZOT pigment pellet. These candidates were selected to revisit issues from the reformulation studies such as degradation of YB-71P versus YB-71. ZOT from one freshly prepared batch was selected. A sample of ZOT in pellet form was used to evaluate behavior of pigment alone. The behavior of pigment in stripped polydimethyl siloxanes was also considered as baseline candidate per recommendation from the initial studies carried out at IITRI by Zerlaut and Gilligan. (31) Two other pigments synthesized at IITRI through IR&D studies were also added to the list of baseline candidates. They were doped zinc silicate (DZS) and Europium oxide (Eu₂O₃). Both were selected based on results from the IR&D studies that indicated that a TCMS based on these pigments is capable of providing beginning of life (BOL) solar absorptance (α_s) values less than 0.10. It is important to point out that based on past studies (AFML-TR-71-246) the Eu₂O₃ based TCMS can be acceptably stable to UV-e'-Vac type environment. The choice of Eu₂O₃ was also initiated due to the fact that Eu₂O₃ has body center cubic (BCC) structure. This fact helps in use of conductive binder systems that are doped. It presents electro-ceramists a new regime to tailor conductivity values. Thus, to evaluate conductive concepts, its use as a baseline material was an obvious choice. The pigment/binder couples considered as baseline are listed in Table 3-1.

3.1.1 Material Processing

The baseline materials were prepared using 1 lb batch of pigment and required binder for the chosen pigment-to-binder ratios (PBRs) listed in Table 3-1. Each batch formulation was sprayed onto samples of various sizes for the required thickness to obtain maximum reflectance (minimum α_s). Ten one-inch discs, eight (2"x2") plates and two (4"x4") plates were coated for each formulation and cured per the best practices developed at IITRI. All inorganic coatings need a high humidity (>50%) and ambient temperature conditions during deposition and initial cure of one day. These are then cured at ambient laboratory clean environment for 14 days.

| screening Test | PBR | | 2:1 | | 2:1 | 2:1 | | Pressed and fired at 1000°C | 4.3 | 7.1 | 7.1 | 2.1:1 | 4:1 | 4:1 | 4:1 | 5.5 | 4:1 |
|---|---------|--------------|--|-------------|---------------------------------------|---------------------------------------|---|-----------------------------|---------------------------------|-------------|-------------|---------------------------------------|---------------------------------------|---------------------------------------|---------------------------------------|----------------------|---------------------------------------|
| intion of Concepts and Material Parameters for Formulations in Screening Test | Binder | | Vacil 2130 | OCIZ HERV | MHS/LO stripped polydimethyl siloxane | MHS/LO stripped polydimethyl siloxane | | Pressed and fi | Kasil 2130 | PS-7 | Kasil 2130 | MHS/LO stripped polydimethyl siloxane | Kasil 213- | MHS/LO stripped polydimethyl siloxane |
| reintion of Concepts and Material | Digment | | | Black glass | Black glass | Cosmic black, WB-500 | atings: | ZOT # 98790 | ZnO (Calcined 625°C, 16-18 hrs) | ZOT # 98790 | ZOT # 98790 | S13GP | S13GP | SZQ | ZOT # 98790 | Eu,O, ^(A) | Eu ₂ O ₃ |
| Table 1.1 Decem | .1-0 | Concept Name | Nonconducting Black Baseline Coatings: | MH211P | MH21S/LO | D21S/LO | Name of Action White Baseline Coatings: | ZOT (Pellet) | Z-93P | VB_71 | VR-71P | S13GP/LO-1 | S13GP/LO-41 | DZSS/LO-41 | ZOT/LO-41 | 7.0 (7130 | E0S/LO-41 |

All organic flexible baseline candidates use stripped polydimethyl siloxanes as a vehicle. These coatings used the process developed for S13GP/LO-1 and documented in processing specifications IITRI-MD-C06788-SP1 and IITRI-MD-C06788-SP2, and were cured for 14 days. All baseline material processing records are available in IITRI files.

3.1.2 Materials Characterization

The samples were then characterized for optical and electrical performance. The optical properties were characterized by measuring the total hemispherical spectral reflectance and the total normal thermal emittance. The measurement method followed the best practices listed as standard operating procedures (SOPs) at IITRI. The measurement of total hemispherical spectral reflectance at IITRI was carried out on a Lambda-19 spectrophotometer, using an integrating sphere. Care was taken for installation of sphere, calibration and data collection following the guidelines given by ASTM-E-903. The spectral reflectance data was then used to calculate solar absorptance following solar spectral weighted average method recommended by ASTM, using the solar air mass zero spectral irradiance curve (ASTM-E-490). The total normal thermal emittance was measured using DB-100 reflectometer. The reported emittance was calculated by subtracting the observed DB-100 reading from one.

The resistivity values reported herewith were measured with high resistance meters provided by Hewlitt Packard. Two different models were available at IITRI. The first one, HP-4329A, represented an older model. This model of resistivity cell used an old design of spring-loaded sample clamping. The sample pressure was not known and one cannot zero out air gap contact resistance. The new high resistance meter, HP-4339A, with the new resistance cell provided the opportunity to clamp the sample at constant selected pressure by dialing dead weight pressure. It also provided the ability to zero out air gas resistances in the cell along with the ability to set limits on leakage current to protect fuses. It also collects voltage and current data as a function of time, and stores the data on computer. The steady-state resistance value for given applied voltages is displayed digitally. Only steady-state values were reported for this screening study. For the interest of this program, the baseline thermal control coatings have surface resistance $> 10^{12} \Omega/\Box$, and are inherently porous in nature. Reduction of moisture contributions to the observed resistance values and selection of proper measurement procedure was key to this program. Our IR&D efforts indicated that overnight bake-out of samples at

100°C and measurement of sample surface resistivity on samples that were carefully stored in desiccant before transferring to the resistivity cell provided surface resistance values comparable to the surface resistance values observed on Z-93/Z-93P by Hamburger et al. (8) at Cleveland State University. Their efforts were more exhaustive, where samples were left in vacuum for at least two weeks to remove moisture. The screening study needed a representative surface resistance value for comparison purposes and for designing compositions. The HP-4339 with sample with bake-out approach was used to generate the representative surface resistance values. This practice was consistent with the recommendations provided for surface resistance measurement practices used in the paper industry that produces leaky, high-resistance coated paper products. [Kiethley Instruments - (Private Communications)(32)] An important factor was to use the correct isolator material to isolate the coated sample. We attempted measurements with Kapton 6H, clear Tedlar (2 mils) and Hewlett-Packard provided polyethylene sheet. The resistance numbers vary for a given coating for each isolator, but all of them fall within half to one decade. There are no ASTMs or guidelines available in this area. Help was sought from aerospace industry representatives on these issues. It was recommended that one needs to stay away from Kapton isolators, since they can introduce errors. (33) All reported numbers were obtained by using the polyethylene isolator sheet provided with the meter for consistency.

The data generated on optical and electrical properties for the baseline materials were measured and are provided in Table 3-2. The detailed reflectance curve generated for each formulation are also provided in Figures 3-1 through 3-13. For traceability purposes, each formulation has a batch number and sample series code provided in the data tables, so that the fruitful comparison can be carried out.

| Study Batch # Batch # Batch # Batch # Batch # Study Alpha Batch Resistance I/C Black Nonconducting Baseline Coatings: Alpha Black Nonconducting Black Nonconducting Baseline Coatings: Alpha Black Nonconducting | | Table 3-2. Results of | Optical and Electrics | Results of Optical and Electrical Characterization for Baseline Concepts | Baseline Concepts | |
|---|---------------------------------------|-----------------------|-----------------------|--|-------------------|------------|
| Nonconducting Baseline Coatings: P | Screening | Batch # | Alpha | Emittance | Surface Resi | stance Ω/□ |
| Nonconducting Baseline Coatings: P | Study | | | | HP-4329A | HP-4339A |
| P | Black Nonconducting B. | aseline Coatings: | | | | |
| S/LO U-360(HZ) 0.98 0.90 LO T-023(BA) 0.98 0.91 Nonconducting Baseline Coatings: Annoconducting Baseline Coatings: 0.08 0.78 Specification of the complex of the coating signal in the coating state of the coat | MH21-IP | U-284/289(GZ) | 86:0 | 06:0 | 1.00E+14 | 1.00E+13 |
| LO T-023(BA) 0.98 0.91 Nonconducting Baseline Coatings: Algority Algority Algority igment/pellet) ZOT #98790 0.08 0.78 U-347(HX) 0.12 0.92 P U-348(HW) 0.10 0.89 P U-376(IM) 0.11 0.90 /LO-1 U-376(HD,D) 0.16 0.91 LO-41 U-276(HD,D) 0.16 0.91 LO-41 U-343 0.13 0.85 LO-41 U-403(IS) 0.14 0.90 LO-41 U-403(IS) 0.14 0.90 | MH21S/LO | U-360(HZ) | 0.98 | 06:0 | 1.00E+14 | 1.00E+13 |
| Nonconducting Baseline Coatings: igment/pellet) ZOT #98790 0.08 0.78 : U-347(HX) 0.12 0.92 0.92 P U-348(HW) 0.10 0.89 0.89 P U-376(IM) 0.11 0.90 0.91 /LO-1 U-310(HY) 0.16 0.91 0.91 LO-41 U-276(HD,D) 0.16 0.91 0.91 LO-41 U-343 0.13 0.85 0.92 LO-41 U-403(IS) 0.14 0.90 0.90 | D21S/L0 | T-023(BA) | 86.0 | 0.91 | 1.00E+10 | 2.00E+10 |
| igment/pellet) ZOT #98790 0.08 0.78 U-347(HX) 0.12 0.92 P U-348(HW) 0.10 0.89 P U-376(IM) 0.11 0.90 /LO-1 U-310(HY) 0.17 0.91 /LO-41 U-276(HD,D) 0.16 0.91 LO-41 U-208(GI,H) 0.07 0.88 LO-41 U-352(IF) 0.08 0.92 LO-41 U-403(IS) 0.14 0.90 | White Nonconducting B | aseline Coatings: | | | | |
| D-347(HX) 0.12 0.92 P U-348(HW) 0.10 0.89 P U-376(IM) 0.11 0.90 /LO-1 U-310(HY) 0.17 0.91 /LO-41 U-276(HD,D) 0.16 0.91 LO-41 U-276(HD,D) 0.16 0.91 LO-41 U-343 0.13 0.85 LO-41 U-352(IF) 0.08 0.92 LO-41 U-403(IS) 0.14 0.90 | ZOT(pigment/pellet) | ZOT #98790 | 0.08 | 0.78 | 5.00E+08 | 9.00E+07 |
| P U-348(HW) 0.10 0.89 P U-376(IM) 0.11 0.90 /LO-1 U-310(HY) 0.17 0.91 /LO-41 U-276(HD,D) 0.16 0.91 LO-41 U-208(GI,H) 0.07 0.88 LO-41 U-343 0.13 0.85 LO-41 U-352(IF) 0.08 0.92 LO-41 U-403(IS) 0.14 0.90 | Z-93P | U-347(HX) | 0.12 | 0.92 | 1.00E+14 | 11.00E+13 |
| U-376(IM) 0.11 0.90 U-310(HY) 0.17 0.91 1 U-276(HD,D) 0.16 0.91 1 U208(GI,H) 0.07 0.88 U-343 0.13 0.85 U-352(IF) 0.08 0.92 U-403(IS) 0.14 0.90 | YB-71 | U-348(HW) | 0.10 | 0.89 | 1.00E+14 | 1.00E+13 |
| 1 U-310(HY) 0.17 0.91 1 U-276(HD,D) 0.16 0.91 1 U208(GI,H) 0.07 0.88 1 U-343 0.13 0.85 1 U-352(IF) 0.08 0.92 1 U-403(IS) 0.14 0.90 | YB-71P | U-376(IM) | 0.11 | 06:0 | 1.00E+14 | 1.00E+13 |
| 1 U-276(HD,D) 0.16 0.91 0.18 0.208(GI,H) 0.07 0.88 0.13 0.85 0.13 0.85 0.352(IF) 0.08 0.92 0.14 0.90 | S13GP/LO-1 | U-310(HY) | 0.17 | 0.91 | 1.00E+14 | 1.00E+13 |
| U208(GI,H) 0.07 0.88 U-343 0.13 0.85 U-352(IF) 0.08 0.92 U-403(IS) 0.14 0.90 | S13GP/LO-41 | U-276(HD,D) | 0.16 | 0.91 | 1.00E+14 | 1.00E+13 |
| U-343 0.13 0.85 U-352(IF) 0.08 0.92 U-403(IS) 0.14 0.90 | DZSS/L0-41 | U208(GI,H) | 0.07 | 0.88 | 1.00E+14 | 1.00E+13 |
| U-352(IF) 0.08 0.92 0.403(IS) 0.14 0.90 | ZOTS/LO-41 | U-343 | 0.13 | 0.85 | 1.00E+14 | 1.00E+13 |
| U-403(IS) 0.14 0.90 | Eu ₂ O ₃ /2130 | U-352(IF) | 0.08 | 0.92 | 1.00E+14 | 1.00E+13 |
| | Eu ₂ O ₂ /LO-41 | U-403(IS) | 0.14 | 06:0 | 1.00E+14 | 1.00E+13 |
| | | | | | | |
| | | | | | | |



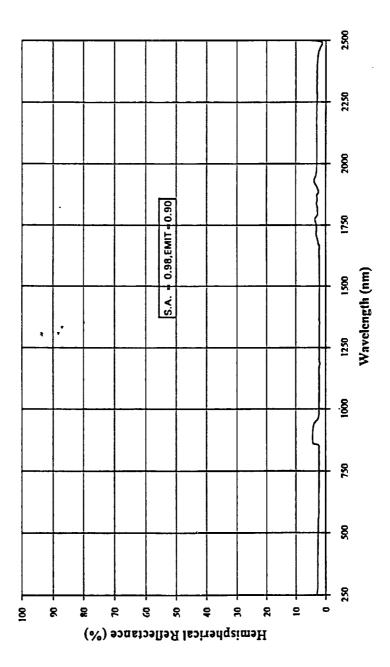


Figure 3-1. Reflectance Spectrum of MH21-IP Sample #GZM-24, Batch #U-284/289.

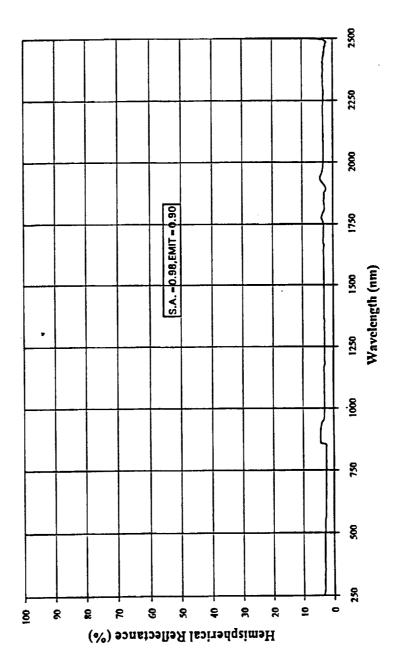


Figure 3-2. Reflectance Spectrum of MH21S/LO Sample #HZ-16, Batch #U-360.

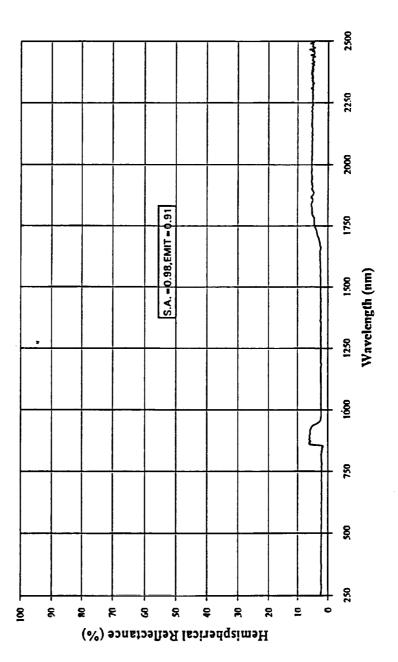


Figure 3-3. Reflectance Spectrum of D21S/LO Sample #BA-7, Batch #T-023.

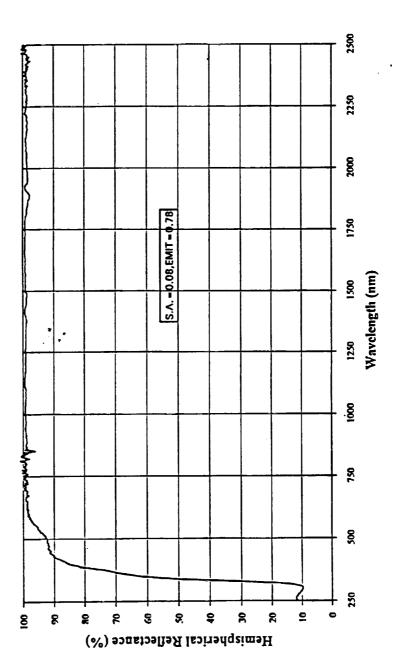


Figure 3-4. Reflectance Spectrum of ZOT/Pigment ZOT #98790.

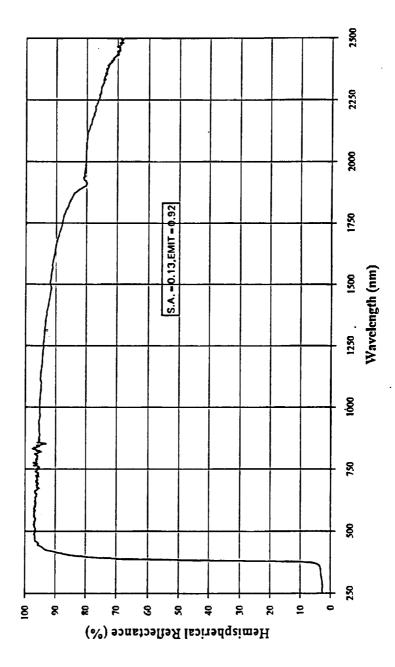


Figure 3-5. Reflectance Spectrum of Z-93P Sample #HX-14, Batch #U-347.

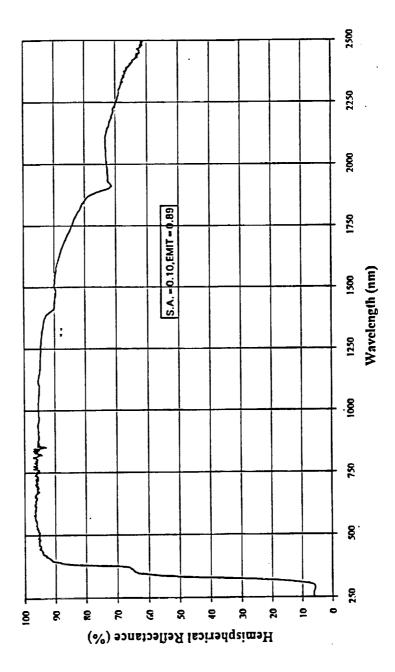


Figure 3-6. Reflectance Spectrum of YB-71 Sample #HW-7, Batch #U-348.

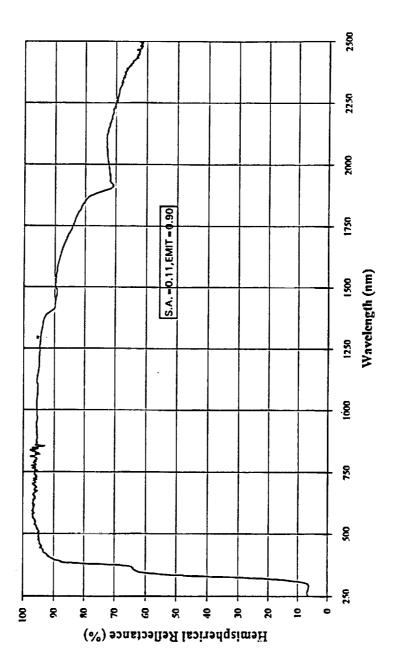


Figure 3-7. Reflectance Spectrum of YB-71P Sample #IM-25, Batch #U-376.

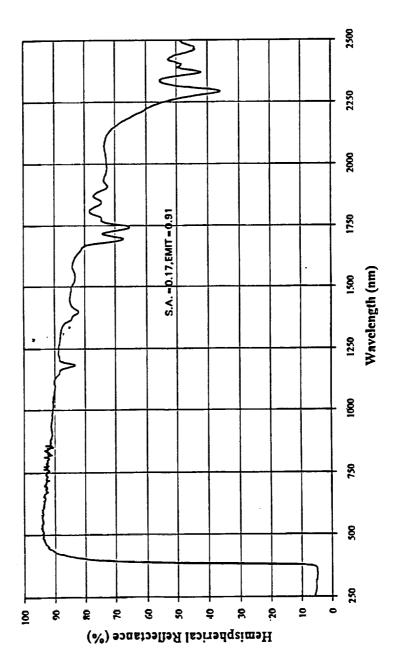


Figure 3-8. Reflectance Spectrum of S13GP/LO-1 Sample #HY-20, Batch #U-310.

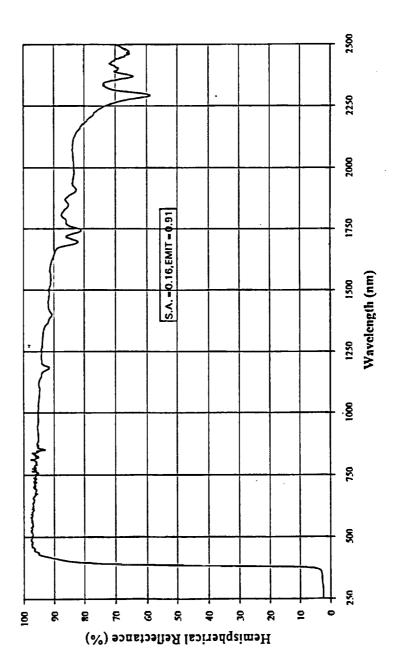


Figure 3-9. Reflectance Spectrum of S13GP/LO-41 Sample #HD-16, Batch #U-276.

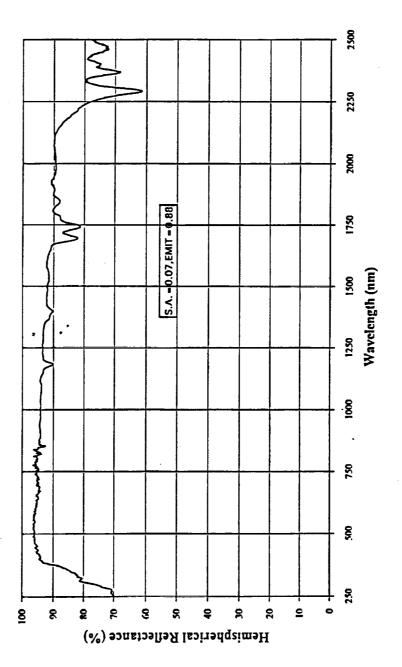
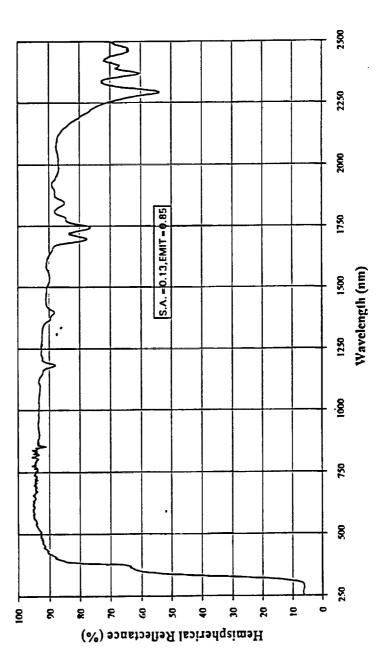


Figure 3-10. Reflectance Spectrum of DZSS/LO-41 Sample #GI-9, Batch #U-208.



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Figure 3-11. Reflectance Spectrum of ZOTS/LO-41 Sample #30, Batch #U-343.

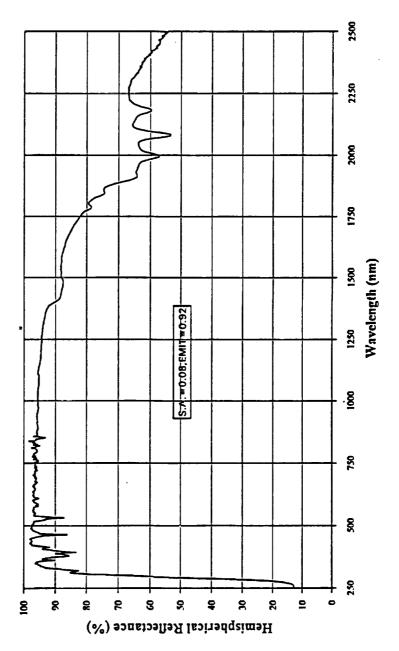


Figure 3-12. Reflectance Spectrum of Eu₂O₃/2130 Sample #IF-18, Batch #U-352.

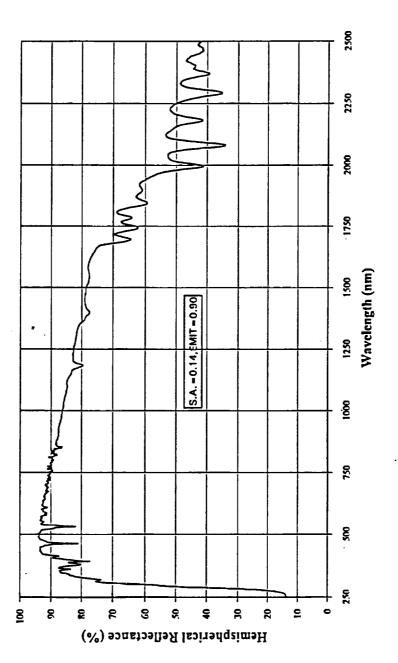


Figure 3-13. Reflectance Spectrum Eu₂O₃/LO-41 Sample #IS-17, Batch #U-403.

3.2 SELECTED CANDIDATE BINDERS AND PIGMENTS: [IR&D EFFORTS SUMMARY]

3.2.1 Binders. In this section first we describe efforts that were carried out prior to this screening study initiated at IITRI. The efforts started as early as 1991 during the reformulation efforts to replace potassium silicate binder (PS-7) used in the IITRI formulations. Most of these efforts were of short duration and were carried out as IR&D.

Sodium silicate was an obvious alternate for the potassium silicate. It was also used as a binder in very early studies funded by the Jet Propulsion Laboratory (JPL) for development of Z-93^(6,29). Sodium silicate was proven to be a suitable binder for space environment stability, but was not used because of its possible interaction with CO₂ in the atmosphere during storage to form carbonates, which can cause cracking in the coating. [See Zerlaut and Harada and ⁽²⁹⁾]. In view of these results, attempts were made to incorporate carbonate formation inhibitors to soda silicate sols to provide an alternate binder system. It was observed that the commercially available soda silicate are not pure or of electronic grade. Most of them are of industrial grade. The PQ Corporation provides soda silicate sols either too silicious for thermal control needs or with high alkali oxide content which makes them unsuitable for use in TCMS as binders. The available soda silicate sol thus needs to be considered as just precursors. The formulation of soda silicate sol was prepared using proprietary carbonate inhibitor additives (2% by weight) at IITRI via mixing and warming processes. The derived sol is named as SS-55.

Soda Silicate Sol: SS-55. The typical properties of SS-55 are as follows:

pH = 11.0 to 11.5
% solid = 35%

$$Na_2O:SiO_2 = 1:3.33$$

These key properties compare very well with the properties of PS-7 that are known to provide optimum adhesion and transmittance for use as a binder in TCMS. (See AFML-TR-94-4126) Availability of the binder system SS-55 aided this feasibility study significantly, since soda silicate provides flexibility in tailoring electrical conductivity. (6) The inherent small size of sodium ion compared to potassium ion may possibly make it easier for the binder system to participate in charge transfer, thus providing an additional mechanism for leakage in dielectric silicate.

The efforts in the binder area also were extended in doping the binder when it was realized that the presence of sodium alone can provide only limited flexibility in tailoring the surface resistance values. These efforts were initiated as IR&D during the GPS Block IIR study, (18) but were shelved because of their inherent defect structure dependent solid-state chemistry. The dopant used can only provide a defected oxide structure. The reliability of the defected structure was not known fully. This screening study provided the opportunity to investigate the two doped hybrid silicate binder systems named DHS-1 and DHS-2 prepared under IITRI's IR&D efforts. They are described here as follows:

• DHS: 50:50 mixture of Kasil 2130 and SS-55

Addition of required dopant precursor;

mixing/warming

• DHS-1: Maximum dopant level to achieve the highest surface resistance in

the desired range

• DHS-2: Minimum dopant level to achieve the highest surface resistance in

the desired range.

It was observed that the reproducibility of surface resistance values for the DHS-1 was better than for the DHS-2 formulation. Both were candidates for the feasibility study along with SS-55 to correlate the role of dopant in space environment induced degradation.

3.2.2 Pigments

3.2.2.1 <u>Doped Black Glass</u>. IR&D efforts were also made in the pigment area to enhance their conductivity. These efforts were made prior to initiation of this screening program. One of the remarkable achievements of this IR&D effort was doping black glass. Doped black glass was produced by using fumed silica or Si-backbone organic polymer as Si-precursors, Carbowax as carbon precursor, and Indium (In) precursor salts for doping. The black glass was derived following pyrolysis processing methods to yield the doped version. It was anticipated that incorporation of such a doped black glass in silicate or doped silicate binder would allow us to tailor electrical surface resistance of the coating without compromising either absorptance or emittance values.

Other IR&D efforts were in the area of white pigments. They were essentially focussed on processing aspects of pigment manufacturing. The white pigments that received attention were as follows:

- DZS: High temperature doping of Zn₂SiO₄ (Indium as dopant)
- Preparation of Eu₂O₃ from organometallic precursors of Europium
- Synthesis of Sb doped Zn₂SnO_x

The brief description of these IR&D efforts are given herewith. One to two pound batches of these pigments were processed and made available to this screening study to evaluate other defected oxide-based conductive concepts. We have included them in this study as other promising approaches.

3.2.2.2 <u>DZS: Doped Zn₂SiO₄</u>. The IR&D processing studies made to synthesize doped zinc silicate consisted of preparation of zinc precursor (ZOX) salt zinc oxalate and use of M-5 (Cab-O-Sil) fumed silica as precursor along with In precursor salts. The schedule to control grain size and morphology for tailoring maximum reflectance calls for processing the pigment as follows:

- (1) Mix precursors by dry milling 16 to 18 hrs
- (2) Precursor Decomposition/Firing: 625°C, 4 hrs
- (3) Flash calcine: 1000°C, 4 hrs

It was noted that dopant chemistry is not compatible with the high temperature firing. The retained dopant levels in pigment for tailoring targeted resistivity value demand very high usage of costly In precursor salts. The process route is an expensive route. Over all, two 1-lb batches were processed. The powders were included in the screening study evaluation to prepare coatings and data on the same are provided herewith in Section 3.3.

- 3.2.2.3 <u>Europium Oxide (Eu₂O₃)</u>. The IR&D efforts made to prepare this pigment can be summarized briefly as follows:
 - (1) Acquisition of Eu₂O₃
 - (2) Precursor Decomposition/Firing: 625°C, 4 hrs

(3) Flash calcination: 1000°C, 4 hrs

 ${\rm Eu_2O_3}$ (<400) mesh was also acquired for comparison purposes. It was found that use of <-400 mesh oxide powders provide similar optical properties as powders fired following the above schedule. The samples prepared for this study were all from europium oxide processed using the pigment synthesis steps summarized above. The powders were included in the feasibility study evaluation to prepare coatings and data on the same is provided herewith in Section 3.3.

3.2.2.4 Antimony Doped Zn₂SnO_x (Sb:Zn₂SnO_x). The IR&D processing synthesis studies carried out to prepare this pigment were based on earlier results in space environment stability of Zn₂SnO₄ reported by Zerlaut and Harada⁽³¹⁾, and to exploit economic doping approaches presented by the antimony chemistry. It is expected that the Zn₂SnO₄ invariably appears as defected oxide on the surface of pigment. SnO₂ and its defect structure has also been utilized by the semiconductor industry to develop sensing product. The antimony provides an ability to stabilize the defect structures with proper and prudent use of multiple valance nature of its ion at high pH used in precursor sol chemistry, during the synthesis.

Since this pigment synthesis was an IR&D effort, a limited attempt was made to produce Sb-doped Zn₂SnO_x. The precursors used were: (1) ZOX, zinc oxalate synthesized by IITRI following methods listed in Reference 30; (2) Sb:doped tin sol provided to IITRI by P.Q. R&D Division as an experimental product. (24) The P.Q. R&D provided data for solids content (% solids) and the thermo gravimetric analysis (TGA). These data were used to calculate proper mixing ratio to prepare precursor mixtures, which were slowly dried over 2 days at 40°C. The precursor powder mixtures were then fired following firing schedules that were known to provide needed morphology and grain size distribution for maximum reflectance. The firings were carried out as follows:

- (1) 625°, 4 hrs
- (2) 1000°C, 4 hrs flash calcination.

The powders were included in the screening study for evaluation of pigments. Various samples were prepared using this pigment and mixing Sb:Zn₂SnO_x with ZnO or DZS. The data obtained on the same are provided in Section 3.3

3.3 SELECTED CONCEPTS AND THEIR OPTICAL AND ELECTRICAL PERFORMANCE PROPERTIES

The conductive concepts considered in this study were based on two types of resistivity tailoring approaches. The first one treated TCMS as a composite. The current state-of-the-art TCMS was considered as a nonconducting component of the composite, and various candidate conductive phases were considered for addition to the nonconductive phase. An extraordinary volume of literature is available to provide guidance for tailoring composite resistance. The subject has received theoretical, numerical and quantitative modeling support, especially because of the discovery of superconductors. McLatchlan et al. (17) have reviewed the subject in depth and have provided guidance to selection of microstructural features in using conductive material as second phase to keep threshold value of volume fraction of conductive phase to a minimum and to acquire required percolation paths for achieving the target conductivity values.

Among the uses of high absorptivity material systems for space application, the electrical surface resistance needs change significantly from hardware to hardware. Tailoring high absorptance with low to moderate surface resistance have been done in the past (e.g., MH11Z at IITRI developed during 1984-1985), but such efforts have exacted penalties to absorptance and emittance values due the second conductive phase (of the order of 2 to 3%). Their RF properties also are lossy. Despite these penalties, the ease of formulating conductive high absorptivity coatings have a role to play for space hardware, because such an approach can offer an economic solution.

Some of the concepts that follow such an approach and were studied during this screening study are listed in Table 3-3. Among the conductive concepts listed in Table 3-3, the first two do not depend on the concept of tailoring conducting paths in TCMS using a second conductive phase. The first two essentially use In as a dopant in glass to provide a leaky dielectric. The two candidates considered can inform us about the role of binders, based on their space environment degradation behavior. These two cases can also provide input on the role that

| Tab | ble 3-3. Candidate Concepts for Conductive High Absorptivity TCMS | anductive High Absorptivity TCM | SI |
|----------------------------|---|---------------------------------------|-------|
| Concept Name | Pigment | Binder | PBR |
| Conducting Black Coatings: | | | |
| DBG-IP | Doped black glass (Indium doped) | Kasil 2130 | 2:1 |
| DBG/DHS-1 | Doped black glass | DHS-1 | 2:1 |
| MHSSIC | Black glass, Graphite 9035, ZnO | Kasil 2130 | 5.5 |
| MH21SC/LO | Black glass, graphite fiber, | MHS/LO stripped polydimethyl siloxane | 2.1 |
| MH41SCB/LO | Black glass/B ₄ C | MHS/LO stripped polydimethyl siloxane | 4:1 |
| D21SC/LO | Cosmic black WB-500, Graphite 9035 | MHS/LO stripped polydimethyl siloxane | 2:1 |
| MH11Z | Cosmic black WB-500 Graphite 9035, ZnO | Kasil 2130 | 1.1:1 |
| D36SCB/LO | Cosmic black WB-500 Graphite 9035, B ₄ C | MHS/LO stripped polydimethyl silicone | 3.6:1 |

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secondary emissions properties can play in tailoring surface resistances. Unfortunately, graphite and In have similar secondary emission yields, so it may not be feasible to rank the concepts easily. One can make use of RuO₂ as one more conductive phase and tailor optical properties to design new formulations and alter secondary emissions further. The importance of "secondary emissions" mechanisms and their impact on space environment stability can thus be studied. However, the cost of RuO₂ and advanced level of electro-ceramic processing required to manipulate optimized microstructures to develop conductive percolating paths, forced us to shelve this concept due to the limited scope of this screening study.

Overall eight conductive concepts listed in Table 3-4 were used to prepare a batch of conductive TCMS. Each batch was then sprayed on the Al-6061-T6 samples of various sizes (1.0" disc, 2"x2" plates, and 4"x4" plates) for optical and electrical performance characterization. The total hemispherical spectral reflectance, total normal emittance and surface electrical resistances were measured following the same methods that were used for characterizing baseline nonconducting material systems, described earlier in Section 3.1. The observed data are presented here in Table 3-4. The detailed total hemispherical spectral reflectance curves for these concepts are given in Figures 3-14 through 3-19.

3.3.1 Conductive Thermal Control Material Systems for Low $(\alpha_{\rm S}/\epsilon_{\rm N})$ Applications

The second task of these efforts was devoted to the study of conductive concepts for which conceptual material design was already carried out and reported in Section 2.3. Pigment and binder concepts discussed in the section on IR&D studies were also included in this task as other concepts. The following section describes studies carried out for each concept along with results of the material and process characterization studies.

3.3.1.1 Concept Z-93CXY. Although the rationale of this concept was discussed in detail in Section 2.3, it may be noteworthy to revisit the same along with a discussion on processing. From the studies on ZnO variators, it is known that the Zn interstitial concentration in ZnO grain control the resistivity value. Zerlaut and Harada⁽²⁹⁾ showed that removal of Zn interstitial from the grains via annealing provides space environment stability. While studying annealing parameters of temperature and time in this study⁽²⁹⁾, several schedules were attempted. The schedule picked for Z-93 provided material with the best stability to space environment.

| | Table 3-4. Sun | Table 3-4. Summary of Optical and Electrical Properties of Conductive Black Coatings | id Electrical Proper Coatings | ties | |
|----------------------------|----------------|--|----------------------------------|--------------------------|-------------|
| | | | : | Surface Resistance (n/□) | tance (Ω/□) |
| | Batch # | Alpha | Emittance | HP4329A | HP4339A |
| Conductive Black Coatings: | ngs: | | | | |
| DBG-IP | U-021(EO) | 0.97 | 0.90 | 2.00E+08 | 1.00E+08 |
| MH55-ICP | S-204(Z,AC) | 0.94 | 0.88 | 1.00E+09 | 1.00E+09 |
| DBG/DHS | U-283(GT,GU) | 96.0 | 0.00 | 8.00E+08 | 1.00E+09 |
| MH21SC/LO | U-166(FD-1) | 96.0 | 0.86 | 1.00E+06 | 7.00E+05 |
| MH41SCB/LO | U-053(FD) | 0.96 | 0.87 | 1.00E+09 | 1.00E+09 |
| D21SC/LO | T-026(BJ) | 0.95 | 0.88 | 1.00E+06 | C/M |
| D36SCB/LO | T-027 | 0.95 | 0.91 | 1.00E+06 | C/M |
| MH11Z | * | 0.89 | 0.90 | 1.00E+05 | 1.00E+04 |
| | | | | | |

* From report to JPL.

C/M = Cannot be measured (i.e., resistance is less than 1.00E+03).

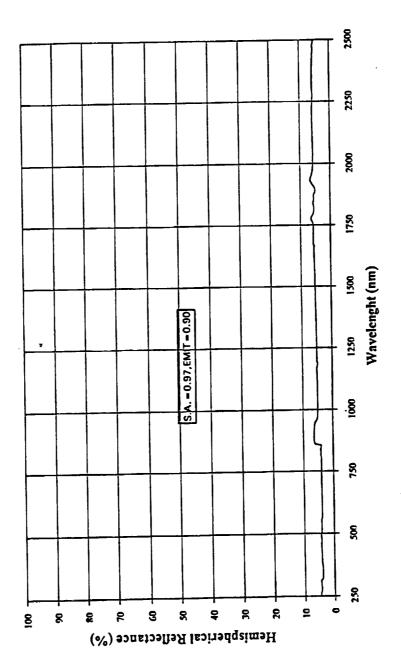


Figure 3-14. Reflectance Spectrum of DBG-IP Sample #EO-6, Batch #U-021.

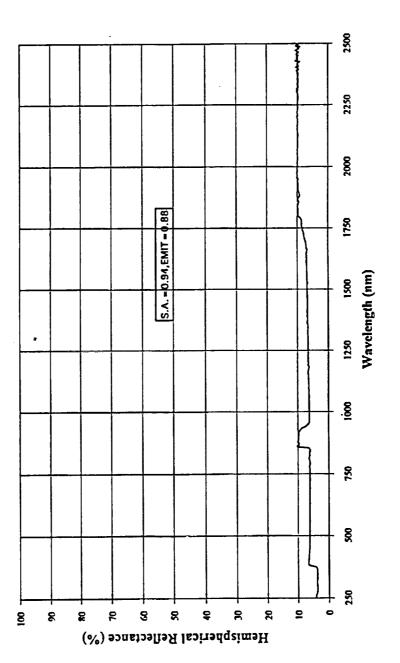


Figure 3-15. Reflectance Spectrum of MH55-IC Sample #AC-33, Batch #S-204.

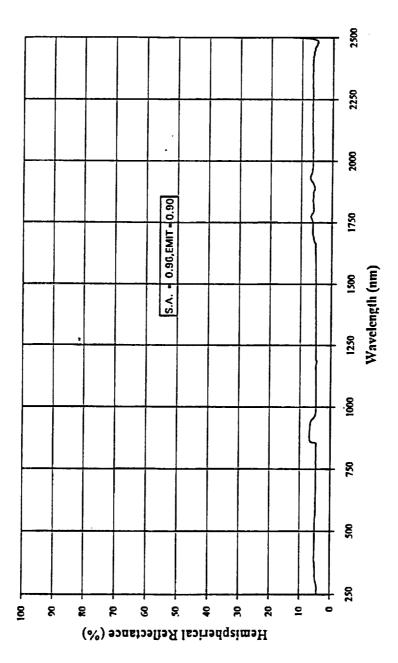


Figure 3-16. Reflectance Spectrum of DBG/DHS Sample #GU-2.

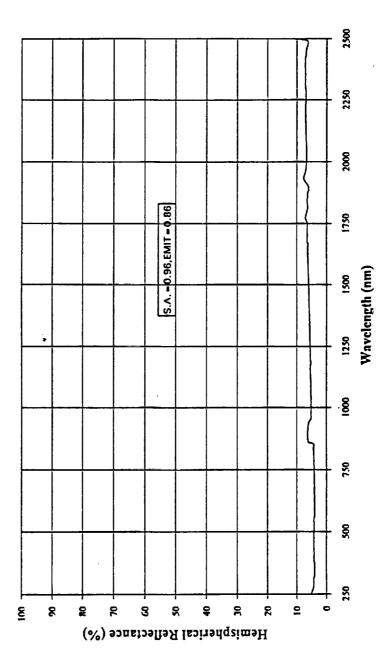


Figure 3-17. Reflectance Spectrum of MH21SC/LO Sample #FD1-15, Batch #U-166.

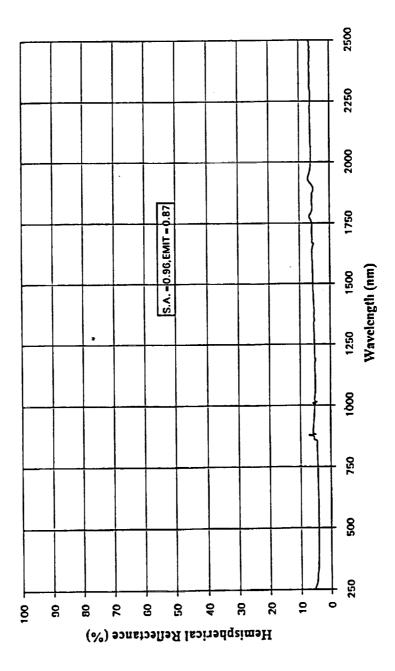


Figure 3-18. Reflectance Spectrum of MH41SCB/LO-1 Sample #FD-2, Batch #U-053.

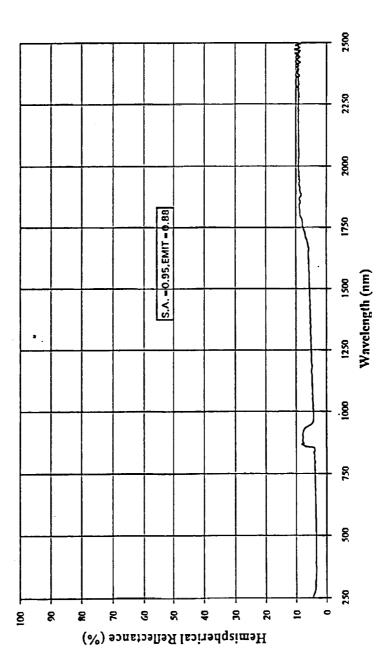


Figure 3-19. Reflectance Spectrum of D21SC/LO Sample #BJ-8, Batch #T-026.

The retention of Zn interstitials to tailor required resistivity values without compromising optical performance was the goal. ZnO (SP-500) of Zinc Corporation of America was used as source and was flash calcined at selected temperatures. Overall three temperatures (800°, 900° and 1000°C) were picked for processing trial batches and flash calcinations were carried out in quartz beakers for time periods of 15 minutes, 30 minutes, 1 hour, and 4 hours. A charge of 900 grams was used for each run. Each pigment was then mixed with Kasil 2130 as a binder for preparing samples to measure α_s , ϵ_N . Based on the data collected for a calcination temperature of 900°C, a flash calcination time of 15 minutes was selected for the screening studies. A flash calcined batch (900 grams, 900°C, 15 minutes) was prepared and used in four different binder systems. The candidate binder systems used are: Kasil 2130, SS-55 (soda silicate sol), DHS-1 and DHS-2 (both doped hybrid silicate sols developed at IITRI). The description of candidate concepts and material parameters are listed in Table 3-5.

| | Table 3-5. Conductive Concept Z93CXY | | | | | | |
|------------------|--|------------|-----|--|--|--|--|
| | Z93CXY: Flash calcined (FC), control of Zn interstitials for required conductivity in silicate binders and doped silicate binders. | | | | | | |
| Concept: Z-93CXY | Pigment | Binder | PBR | | | | |
| Pigment/Binder | | | | | | | |
| ZnO (FC)/2130 | Zno Flash Calcined (FC) | Kasil 2130 | 5.5 | | | | |
| ZnO (FC)/SS-55 | ZnO (FC) | SS-55 | 5.5 | | | | |
| ZnO (FC)/DHS-1 | ZnO (FC) | DHS-1 | 5.5 | | | | |
| ZnO (FC)/DHS-2 | ZnO (FC) | DHS-2 | 5.5 | | | | |

A batch of each material formulation was prepared for the above formulations and sprayed onto various substrates of 1.0 inch diameter discs, 2"x2" plates, 4"x4" plates for optical and electrical performance characterization. The total hemispherical spectral reflectance, total normal thermal emittance and surface resistivity were measured following the same methods that were used to characterize baseline nonconducting material described earlier in Section 3.1. The observed data are presented in Table 3-6 and the detailed reflectance curves are presented in Figures 3-20 through 3-23. All data was taken for coating thickness of 5.0 mils (nominal).

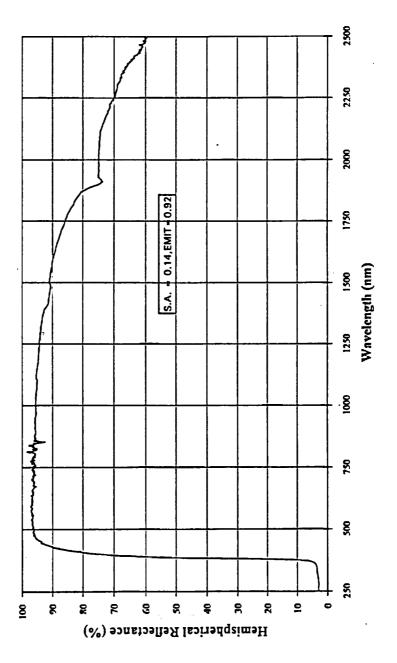


Figure 3-20. Reflectance Spectrum of ZnO(FC)/2130 Sample #HM-21, Batch #U-321.

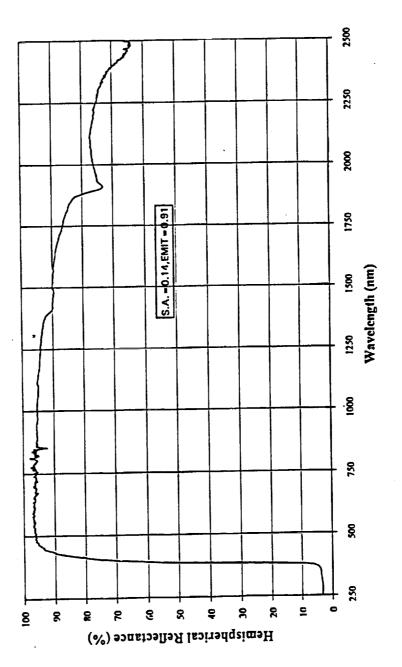


Figure 3-21. Reflectance Spectrum of ZnO(FC)/SS-55 Sample #HF-16, Batch #U-322.

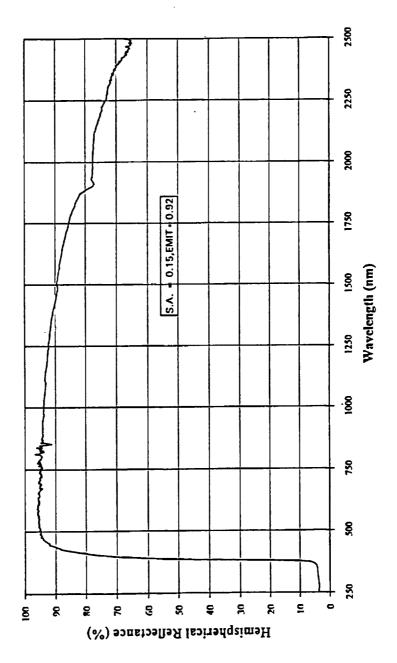


Figure 3-22. Reflectance Spectrum of ZnO(FC)/DHS-1 Sample #HG-16, Batch #U-323.

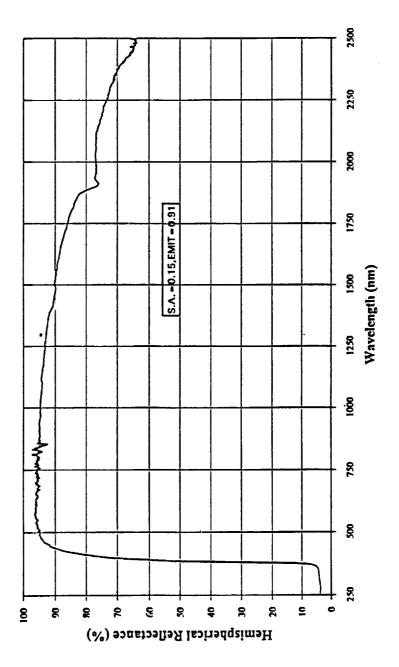


Figure 3-23. Reflectance Spectrum of ZnO(FC)/DHS-2 Sample #HL-25, Batch #U-327.

| Table 3-6. Z-93CXY: Properties of Candidate Pigment/Binder Couples | | | | | | |
|--|-----------|-------|-----------|--------------|---------------------------|--|
| Concept: Z-93CXY | Batch # | Alpha | Emittance | Surface Resi | Surface Resistivity (Ω/□) | |
| Pigment/Binder | David ,, | | | HP4329A | HP4339A | |
| ZnO(FC)/2130 | U-321(HM) | 0.14 | 0.92 | 1.00E+14 | | |
| ZnO(FC)/SS-55 | U-322(FH) | 0.14 | 0.91 | 1.00E+08 | 6.00E+08 | |
| ZnO(FC)/DHS-1 | U-323(HG) | 0.15 | 0.92 | 2.00E+09 | 2.00E+08 | |
| ZnO(FC)/DHS-2 | U-327(HL) | 0.15 | 0.91 | 1.00E+09 | 1.00E+09 | |

The resistivity values reported in Table 3-6 are the values observed immediately after 14 days of cure at ambient conditions. The coating deposition was carried out similar to Z-93P processing, and curing was carried out using high humidity environment. The samples were then conditioned to avoid moisture contribution to measurement, by purging them in dry desiccant with nitrogen. It was observed that storage of the samples using silicate binders (Kasil 2130, soda silicate SS-55) exposed to ambient conditions causes surface resistivity values to increase in the range of 10^{12} to 10^{13} Ω/\Box after 6 to 8 months. This effect was not observed for doped silicate binders even after two years. All four concepts were submitted to the space environment simulation, but only reliable resistivity concepts were submitted for ESD testing and Goddard Space Flight Center (GSFC). The results are discussed here in corresponding chapters. The results show that flash calcination alone cannot stabilize resistivity. One can speculate that dopant in binder may be helping to stabilize electrical performance of the coating via carrying out in situ doping of ZnO on grain surface. Thus, it will be interesting to review results of the space environment simulation for these candidates, which are presented in the next chapter.

3.3.1.2 Concept Z-93SCXY. The rationale for this concept still lies in control of Zn interstitials in ZnO to impart required resistivity in pigment grains. Here a material technology employs encapsulation approach to stabilize interstitials inside the grain. Our quest for such stability to space environment is not new. We have pursued such stabilization via microencapsulation routinely for S13GP pigment, with ZnO (SP-500) as the starting material.

In providing candidate material concept samples for this evaluation study, microencapsulation of ZnO (SP-500) (uncalcined) was carried out following the SOP for reactive encapsulation using Kasil 2130 as the binder system. The processing steps reported in AFML-TR-94-4126 were used, and the resultant pigment is designated S13GP. Similar processing was carried out on a bench scale setup using newly developed soda silicate sol (SS-55). This pigment was named as S13N. Both these pigments were used with two silicate binders (Kasil 2130 and SS-55) and two doped hybrid silicate binders (DHS-1 and DHS-2) to process candidate conductive TCMS. The description of material parameters is given in Table 3-7.

| Table 3-7. Conductive Concept Z93SCXY | | | | | | | |
|---------------------------------------|---|------------|-----|--|--|--|--|
| Z93SCXY: Microe | Z93SCXY: Microencapsulated pigments in silicate binders and doped silicate binders. | | | | | | |
| Concept: Z-93SCXY | Pigment | Binder | PBR | | | | |
| Pigment/Binder | | | | | | | |
| S13GP/2130 | S13GP | Kasil 2130 | 5.5 | | | | |
| S13GP/SS-55 | S13GP | SS-55 | 5.5 | | | | |
| S13GP/DHS-1 | S13GP | DHS-1 | 5.5 | | | | |
| S13GP/DHS-2 | S13GP | DHS-2 | 5.5 | | | | |
| S13N/SS-55 | S13N | SS-55 | 5.5 | | | | |
| S13N/DHS-1 | S13N | DHS-1 | 5.5 | | | | |
| S13N/DHS-2 | S13N | DHS-2 | 5.5 | | | | |

A batch of each material formulation was prepared for the above formulations and sprayed onto various substrates of 1.0 inch diameter discs, 2"x2" plates, 4"x4" plates for optical and electrical performance characterization. The total hemispherical spectral reflectance, total normal thermal emittance and surface resistivity were measured following the same methods that were used to characterize baseline nonconducting material described earlier in Section 3.1. The observed data are presented in Table 3-8 and the detailed reflectance curves are presented in Figures 3-24 through 3-30. All data were for a coating thickness of 5.0 mils (nominal).

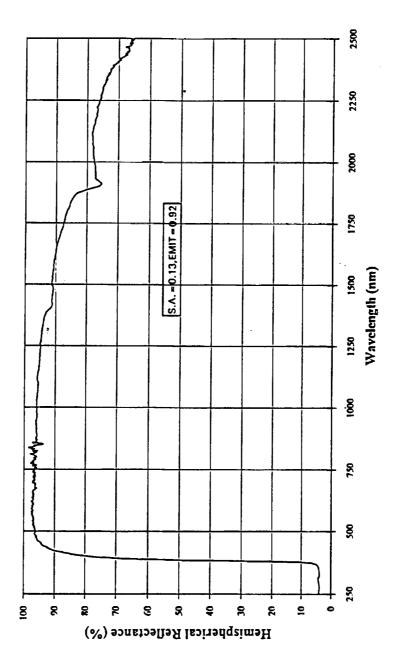


Figure 3-24. Reflectance Spectrum of S13GP/2130 Sample #FM-17.

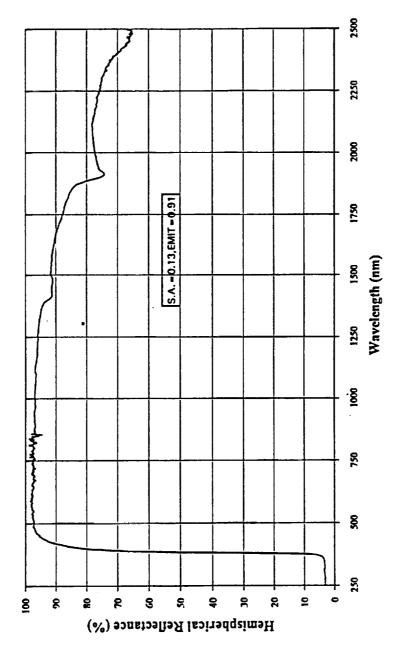


Figure 3-25. Reflectance Spectrum of S13GP/SS-55 Sample #IU-21, Batch #U-405.

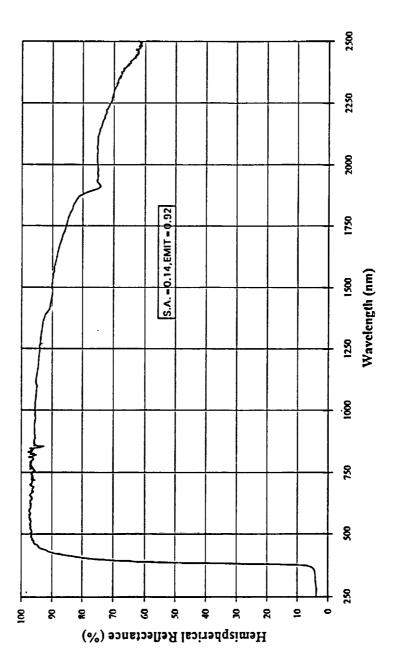


Figure 3-26. Reflectance Spectrum of S13GP/DHS-1 Sample #GS-4, Batch #U-279.

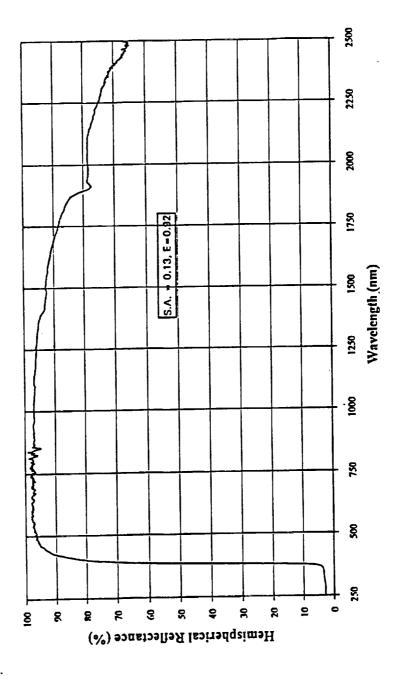


Figure 3-27. Reflectance Spectrum of S13GP/DHS-2 Sample #HH-20, Batch #U-328.

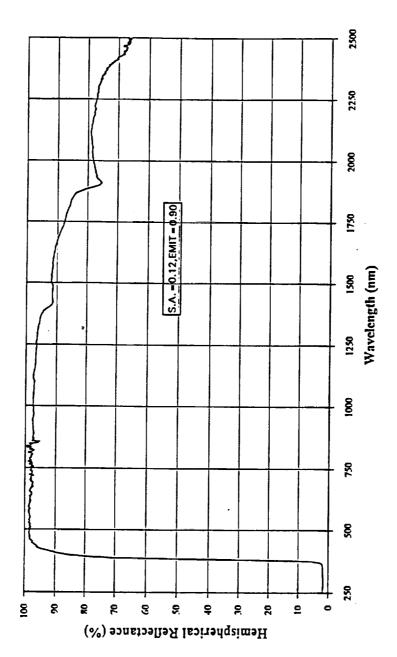


Figure 3-28. Reflectance Spectrum of S13N/SS-55 Sample #GR-7, Batch #U-260.

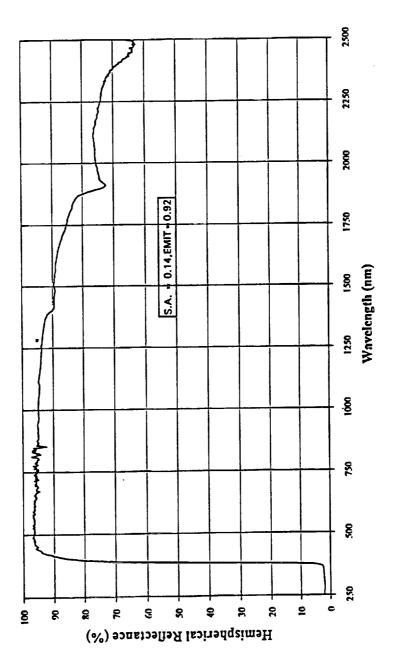


Figure 3-29. Reflectance Spectrum of S13N/DHS-1 Sample #HE-8, Batch #U-317.

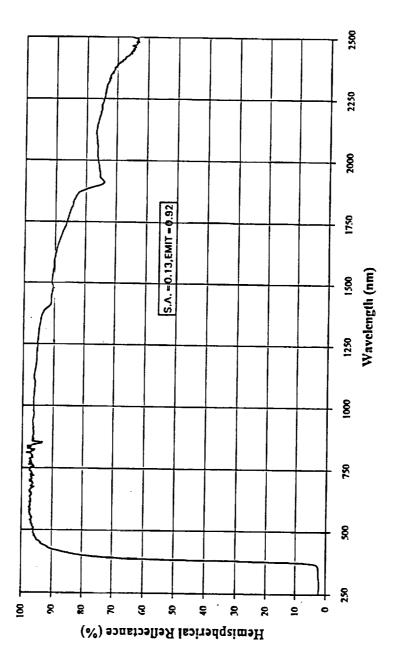


Figure 3-30. Reflectance Spectrum of S13N/DHS-2 Sample# HI-27, Batch #U-329.

| Table 3-8. Z-93SCXY: Properties of Candidate Pigment/Binder Couples | | | | | | |
|---|-----------|-------|-----------|----------------|---------------------------|--|
| Concept: Z-93SCXY | Batch # | Alpha | Emittance | Surface Resign | Surface Resistivity (Ω/□) | |
| Pigment/Binder | Daten # | При | | HP4329A | HP4339A | |
| S13GP/2130 | (FM) | 0.13 | 0.92 | 1.00E+09 | 5.00E+09 | |
| S13GP/SS-55 | U-405(IU) | 0.13 | 0.91 | 8.00E+09 | 5.00E+09 | |
| S13GP/DHS-1 | U-279(GS) | 0.14 | 0.92 | 8.00E+07 | 1.00E+09 | |
| S13GP/DHS-2 | U-328(HH) | 0.13 | 0.92 | 4.70E+08 | 1.00E+09 | |
| S13N/SS-55 | U-260(GR) | 0.12 | 0.90 | 2.00E+08 | 1.00E+09 | |
| S13N/DHS-1 | U-317(HE) | 0.14 | 0.92 | 4.00E+08 | 1.00E+09 | |
| S13N/DHS-2 | U-329(HI) | 0.13 | 0.92 | 5.60E+08 | 5.00E+09 | |

It is important to report that storage and exposure to ambient conditions do not affect the surface electrical resistance of these formulations. Thus, the microencapsulation can help in tailoring stability as well as the electrical response. We anticipate excellent stability to space environment for this concept and await results for doped binder systems that provide excellent reproducibility in the electrical surface resistance properties of the candidate TCMS pigment/binder couples.

The concept S13GP/hybrid silicate without dopants has already found applications on the GPS-Block II hardware and has been integrated on sunshades of E-Glass/Epoxy composites. The binder system for this hardware did not use dopant to avoid defect structure dependent conduction mechanisms, since we do not know enough of their behavior in extended space exposure environment. The resultant surface resistance for such systems was also 1.0E+09 as measured by HP-4339A. The $\alpha_S \approx 0.14$ and $\epsilon_N = 0.90$ are typical values for S13GP/hybrid silicate (50/50 mixture of Kasil 2130 and SS-55 without dopants) formulation.

3.3.1.3 Concept Z-93SCLMXY. The rationale for this concept is very much similar to the earlier concept of Z-93SCXY, where control of Zn interstitials was considered as a key to tailor the required resistances in pigment grains. In the earlier concept, microencapsulation of the pigment was considered as the route to stabilize the Zn interstitials in the ZnO grain. The overall process also limits the ability to tailor surface resistances, since each grain is covered

with a dielectric silicate layer. The ability to tailor resistance then is limited by our ability to subsequently leach out the dielectric silicate layer. The process of pigment surface phosphatization to make pigment compatible with the polydimethyl/siloxane also limits the ability to tailor surface resistance values of concept Z-93SCXY. The microstructural parameter that governs this ability is thickness of dielectric layer that encompasses the ZnO grain.

The concept Z-93SCLMXY was studied to relax this limit on the ability to tailor surface resistances. The goal was to provide the ability to manipulate pigments such that the dielectric silicate layer and the phosphatized layer are doped and act as a "leaky" dielectric film, which will also show the ability to protect against space environment irradiation.

To achieve this goal, we have approached "S13GP" pigment design following two paths. In the first one, we used the dopant precursor salts and doped the pigment at high temperatures using precursors of ZnO and silica (SiO₂). The related efforts have been already reported in Section 3.2.2.2 for pigment DZS developed via high temperature doping. We expect that Zn₂SiO₄ formation in such a high temperature route is unavoidable. The pigment DZS developed during IR&D also provides the ability to tailor total hemispherical solar absorptance to values less than 0.10. Thus, the pigment was incorporated in all candidate silicate binders as well as in stripped polydimethyl siloxane to prepare samples for the screening study.

The second path taken was to design a process for doping operation very much like the microencapsulation process. The uncalcined ZnO (SP-500) was reactively encapsulated into doped silicate binder. Since sodium silicate provides the flexibility in providing leaky dielectric silicate, SS-55 was doped with the dopant levels of DHS-1 and was used as a microencapsulants. The microencapsulation process reported in AFML-TR-94-4126 was followed using ZnO (SP-500) and doped hybrid silicate binder DHS-1 to produce the required candidate pigment for this screening study. The pigments produced following the second path were designated as DS13N. The resultant pigment, DS13N, was then incorporated in all candidate silicate binders as well as in the stripped polydimethyl siloxane to prepare samples for the screening study.

The description of the TCMS prepared for study can be summarized briefly in Table 3-9.

| Table 3-9. Description of Concepts and Material Parameters for Formulations in Screening Test | | | | | | |
|---|----------------|---------------------------------------|-----|--|--|--|
| Z-93SCLMXY: Doped zinc silicate (high temperature doping) in silicate binders and doped silicate binders; and doped microencapsulated pigments (low temperature doping) in silicate binders and doped silicate binders. | | | | | | |
| Concept: Z-93SCLMXY | 1 | | | | | |
| Pigment/Binder | Pigment/Binder | | | | | |
| DZS/2130 | DZS | Kasil 2130 | 5.5 | | | |
| DZS/SS-55 | DZS | SS-55 | 5.5 | | | |
| DZS/DHS-1 | DZS | DHS-2 | 5.5 | | | |
| DZS/DHS-2 | DZS | DHS-2 | 5.5 | | | |
| DS13N/SS-55 | DS13N | SS-55 | 5.5 | | | |
| DS13N/DHS-1 | DS13N | DHS-1 | 5.5 | | | |
| DS13N/DHS-2 | DS13N | DHS-1 | 5.5 | | | |
| Flexible White Conductive Coating: | | | | | | |
| DS13NSC/LO-41 | DS13N | MHS/LO stripped polydimethyl siloxane | 4.0 | | | |

Before going to the results obtained for the optical and electrical performance of these coatings, it may be worthwhile to visit the issue of involved costs of doping for each path evaluated here.

For the first path of high temperature doping of the pigment, the costs involved are dominated by the cost of dopant precursor salts and labor costs. The most desirable dopant for white (low α_s/ϵ) TCMS is In. The In precursor salts are very costly, and they are volatile at high temperatures selected for doping. Thus, high temperature doping is a costly approach for doping. The use of high temperature also causes precursor loss, which made the process very difficult to reproduce. The only advantage observed for such a processing route was its ability to provide a total hemispherical solar absorptance of less than 0.10 (as BOL α_s).

For the second path of low temperature doping following the "S13G" type microencapsulation process used in doped silicate binder of high pH, the requirement of the dopant salts per batch to produce the target resistivity was markedly less. It was approximately 1/15th of the dopant precursor salt required in the high temperature doping path. The precursor losses in the microencapsulation process were observed only during the leaching process. The dopant losses for each step of processing were not completely characterized. Analysis of "In" in collected washed fluid after leaching indicated that very minute In losses had occurred. Thus, the second process route was found to be economical. However, this approach cannot yield total hemispherical solar reflectance of less than 0.10. This is very clear from the candidate sample characterization efforts described below.

A batch of each material formulation was prepared for the candidates listed in Table 3-9 and sprayed onto various substrates of 1.0 inch diameter discs, 2"x2" plates, 4"x4" plates for optical and electrical performance characterization. The total hemispherical spectral reflectance, total normal thermal emittance and surface resistance were measured following the same methods that were used to characterize baseline nonconducting material described earlier in Section 3.1. The observed data are presented in Table 3-10 and the detailed reflectance curves are presented in Figures 3-31 through 3-38. All data was recorded for coating thickness of 5.0 mils (nominal).

It is important to report that storage and exposure to ambient conditions have not altered electrical surface resistance values over a two-year period. The conductive white flexible TCMS received a very favorable response from various industrial design engineers, who asked for samples to evaluate RF properties and inquired about the measurement methods. Although no RF properties were communicated back to us, it was apparent from the comments made by the end users that the TCMS with doped silicate binder are less lossy than doped pigment in stripped polydimethyl siloxane polymer. The feedback was received that the flexible conductive TMCS and its baseline counterpart (S13GP/LO-1) are both electroluminescent under UV-exposure. These comments are revisited in the "results and discussion" chapter.

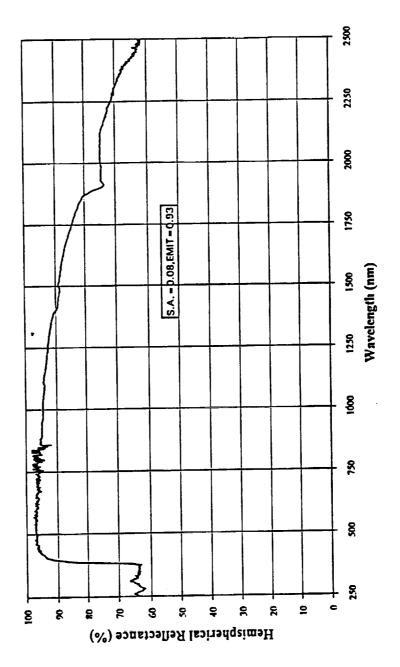


Figure 3-31. Reflectance Spectrum for DZS/2130 Sample #DK-20, Batch #T-244.

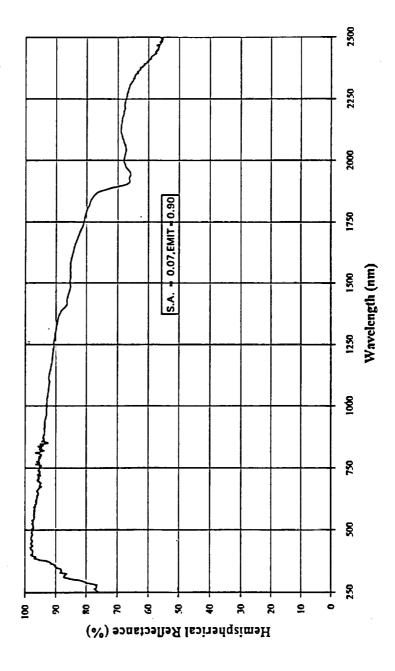


Figure 3-32. Reflectance Spectrum for DZS/SS-55 Sample #FO-4, Batch #U-206.

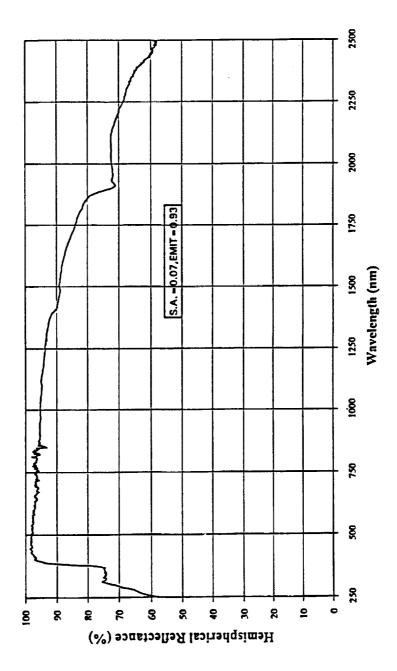


Figure 3-33. Reflectance Spectrum for DZS/DHS-1 Sample #GQ-6, Batch #U-267.

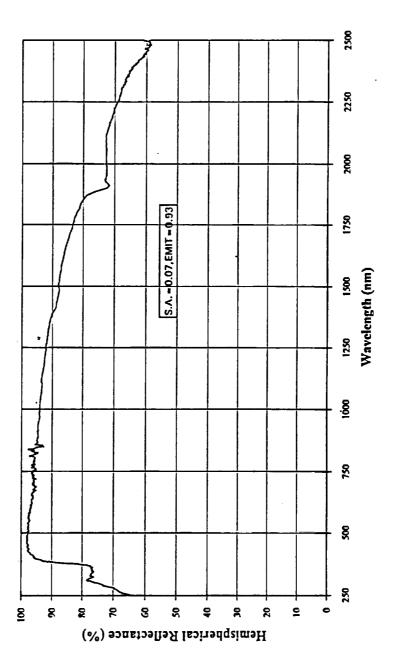


Figure 3-34. Reflectance Spectrum for DZS/DHS-2 Sample #HK-27, Batch #U-331.

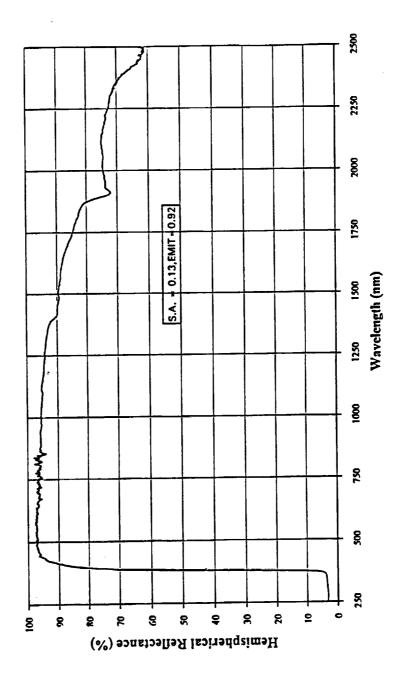


Figure 3-35. Reflectance Spectrum of DS13N/SS-55 Sample #P-1, Batch #U-315.

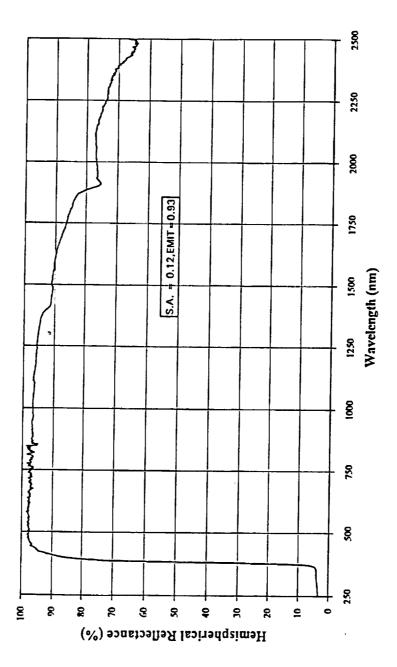


Figure 3-36. Reflectance Spectrum of DS13N/DHS-1 Sample #S, Batch #U-316.

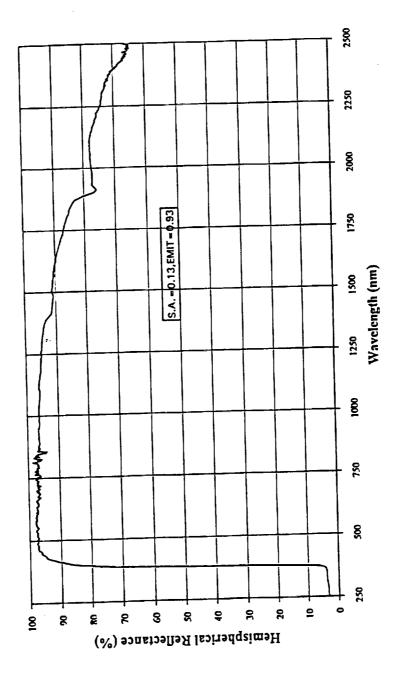


Figure 3-37. Reflectance Spectrum for DS13N/DHS-2 Sample #HJ-27, Batch #U-330.

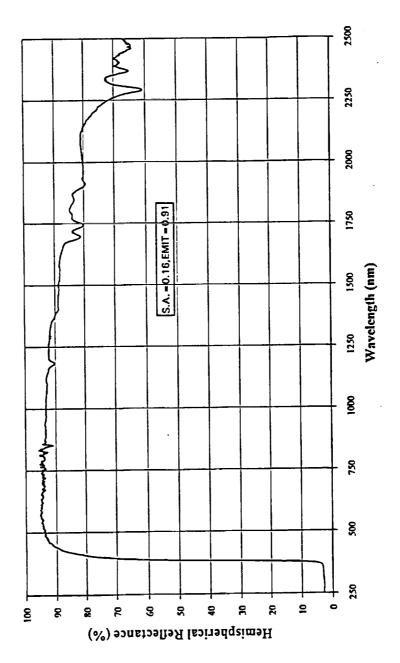


Figure 3-38. Reflectance Spectrum for DS13NSC/LO-41 Sample #HV-12, Batch #U-335.

| Table 3 | 3-10. Z-93SLMC | CXY: Propertie | s of Candidate Pig | gment/Binder Cou | ples |
|---------------------------|------------------|----------------|--------------------|------------------|---------------|
| Concept: Z-93SCLMXY | Batch # | Alpha | Emittance | Surface Resi | stivity (Ω/□) |
| Pigment/Binder | Daten " | , np. | | HP4329A | HP4339A |
| DZS/2130 | T-244(DK) | 0.08 | 0.93 | 1.00E+13 | NA |
| DZS/SS-55 | U-207(FO) | 0.07 | 0.90 | 2.00E+10 | 1.00E+09 |
| DZS/DHS-1 | U-267(GQ) | 0.07 | 0.93 | 6.00E+07 | 2.00E+08 |
| DZS/DHS-2 | U-331(HK) | 0.07 | 0.93 | 1.00E+08 | 7.00E+08 |
| DS13N/SS-55 | U-315(P) | 0.13 | 0.92 | 9.00E+08 | 1.00E+08 |
| DS13N/DHS-1 | U-316(S) | 0.12 | 0.93 | 2.00E+08 | 6.00E+08 |
| DS13N/DHS-2 | U-330(HJ) | 0.13 | 0.93 | 3.00E+08 | 1.00E+09 |
| Flexible White Con | ductive Coating: | | | | |
| DS13NSC/LO-41 (5 mils) | U-334(HV) | 0.16 | 0.91 | 1.00E+09 | 1.00E+10 |

3.3.1.4 Other White (Low α_s/ϵ_N) Conductive Concepts. In this section, we present other conductive thermal control design concepts that can help us to achieve our screening program's material performance goals for tailoring electrical resistivity without loss of optical performance. In selecting such other concepts, an attempt was made to emphasize selection of alternate conduction mechanisms or materials that provide advantages.

The first obvious choice was the use of available semiconducting pigment zinc orthotitanate (ZOT) developed at IITRI. (30,31) The ZOT pigment was also used to prepare baseline samples of TCMS YB-71 and YB-71P with potassium silicate binders PS-7 and Kasil 2130. It was decided to use ZOT from the same batches to prepare samples using the binder systems designed to achieve the target conductivity values. For this purpose, binders SS-55 and DHS-2 were selected. In earlier study of requalification efforts (AFML-TR-94-4126), some anomalous degradation of ZOT was observed with Kasil 2130 for one batch at 900 nanometers. This was discussed at length during earlier efforts along with strategies to mitigate such degradation. Among the strategies discussed, the microencapsulation and ZOT surface defect structure protonation via selection of binders promised possible success. The doped binders also present another option. The dopants can chemically stabilize defect structures present on the ZOT

surface if proper dopant chemistry is made available. Such doping has an added advantage of providing ZOT with required surface resistance. The titanate processing literature is filled with input from such studies. French efforts have been in the area of doping semiconducting Zn₂TiO₄ with lanthanum. The scope of this work requires attention to tailoring electrical surface resistance, so ZOT defect structure stabilization issue were visited here in a limited way. We have provided candidate material designs without any manipulation of pigment surface chemistry here. The ZOT from the same batch that was used to prepare baseline samples has been used to provide conductive TCMS with binders like those sol, SS-55 and DHS-2.

These efforts are also important from the comments made on YB-71 formulation by Garrett²⁰, which reflects on why it has been used in GEO orbit often. It is anticipated⁽²⁰⁾ that a coating made up of ZOT as the pigment may charge up only once, and may not charge again. Thus, electrical performance of ZOT based formulations after irradiation also need to be tested.

The second choice was the use of europium oxide. Europium oxide was used strictly because of its body center cubic structure. This aspect provides the ability to withstand diffusion of dopant from binder into pigment without forming defects that can be optically absorbing in nature, or can cause alterations in conduction band to cause deleterious effects on electrical properties. Such flexibility does not exist in semiconducting ZnO or ZOT. Thus, Eu₂O₃ can represent a different regime of conductive TCMS to address the chemical compatibility issues of pigments and doped silicate binders. The Eu₂O₃ was also of interest because of its ability to provide a reflective concept that can help to tailor total hemispherical solar absorptance less than 0.10. Because of these advantages, Eu₂O₃ was prepared for IR&D studies (Section 3.2.2.3) and was used to provide candidate conductive concepts as well as new baseline nonconductive versions.

The third choice was to evaluate a cheaper doping concept, where one avoids use of costly In precursor salts that are used in doping. Antimony-doped tin oxide sol available form P.Q. R&D Division as experimental material provided an excellent opportunity to evaluate this in the screening program. Our IR&D efforts provided antimony-doped Zn₂SnO_x [Sb:Zn₂SO_x]. Although we have not perfected or carried out efforts for synthesis of this pigment, the small amount of processed (bench scale) material was used to tailor required optical and electrical performance characteristics. From earlier work carried out by Zerlaut et al. (31) and Gilligan et

al. (6) at IITRI that antimony doping usually results in an absorbing pigment; hence, the resultant optical and electrical performance was tailored via manipulation of TCMS's composition. Calcined ZnO (pigment for Z-93P) was used as a second desirable pigment to provide optical required properties.

The last concept, using Ta_2O_5 as a pigment, was revisited during this screening study. Ta_2O_5 was studied as a candidate pigment during very initial pigment screening studies carried out by Zerlaut and Harada. The $\Delta\alpha_5$ for the Ta_2O_5 available was found to be unacceptable. No further investigations have taken place since then. We also know that Ta_2O_5 is usually added to glasses (e.g., solar panel transparent glass cover sheet) to provide required leaky dielectric response for the glass. In view of the proven use of Ta_2O_5 in transparent glasses in orbit, it was decided to purchase purest Ta_2O_5 available, calcine at $1000^{\circ}C$, and use the pigment with silicate and doped silicate binders (Kasil 2130, SS-55 and DHS-2). The prepared samples were submitted as room temperature cured and heat treated. It was interesting to note that Ta_2O_5 pigmented undoped version met the target surface resistance goal even in resistive silicate binders (i.e., Kasil 2130 and SS-55) for PBR of 5.5.

All of the above selected concepts can be summarized in tabular form to list these other candidate material designs that were considered in this screening study. (See Table 3-11) Their nonconducting versions have already been listed in Section 3.1.

A batch of each candidate material formulation was prepared for the candidates listed in Table 3-11 and sprayed onto various substrates of 1.0 inch diameter discs, 2"x2" plates, 4"x4" plates for optical and electrical performance characterization. The total hemispherical spectral reflectance, total normal thermal emittance and surface resistance were measured following the same methods that were used to characterize baseline nonconducting material described earlier in Section 3.1. The observed data are summarized here in Table 3-12 and the reflectance curves are presented in Figures 3-39 through 3-50. All data were measured for coating thickness of 5.0 mils (nominal).

| | Table 3-11. Other Conductive Material Design Concepts | e Material Design Concepts | |
|--|---|---------------------------------------|-----|
| Material Concept | Pigment | Binder | PBR |
| Pigment/Binder | | | |
| Other White Conductive Coatings: | :88: | | |
| ZOT/Phos sol | ZOT #98790 | Phos sol | 5.5 |
| ZOT/SS-55 | ZOT # 98790 | SS-55 | 5.5 |
| ZOT/DHS-2 | ZOT # 98790 | DHS-2 | 5.5 |
| Eu ₂ O ₃ /SS-55 | $\mathrm{Eu_2O_3}$ | SS-55 | 5.5 |
| Eu ₂ O ₃ /DHS-2 | Eu_2O_3 | DHS-2 | 5.5 |
| Ta ₂ O ₅ /2130 (RT Cured) | Ta_2O_5 | 2130 | 5.5 |
| Ta ₂ O ₃ /SS-55 (RT Cured) | Ta_2O_5 | SS-55 | 5.5 |
| Ta ₂ O ₅ /DHS-2 (RT Cured) | Ta_2O_5 | DHS-2 | 5.5 |
| Ta ₂ O ₅ /2130 (HT Cured) | Ta_2O_5 | 2130 | 5.5 |
| Ta ₂ O ₅ /SS-55 (HT Cured) | Ta_2O_5 | SS-55 | 5.5 |
| Ta ₂ O ₅ /DHS-2 (HT Cured) | Ta_2O_5 | DHS-2 | 5.5 |
| Composite White Conductive Coatings: | oatings: | | |
| Sb Doped | ZnO + Sb Doped Zn ₂ SnO _x | 2130 | 5.5 |
| Sb Doped | ZnO + Sb Doped Zn ₂ SnO _x | MHS/LO stripped polydimethyl siloxane | 4:1 |

| Table 3-12. Other | Conductive Materia | il Concepts: Proper | ties of Candidate P | Conductive Material Concepts: Properties of Candidate Pigment/Binder Couples | 88 |
|---|--------------------|---------------------|---------------------|--|-------------|
| Material Concept | Batch # | Alpha | Emittance | Surface Resistance (Ω/□) | tance (Ω/□) |
| Pigment/Binder | | | | HP-4329A | HP-4339A |
| Other White Conductive Coatings: | | | | | |
| ZOT/Phos. Sol. | U-377(HT) | 0.14 | 0.89 | 1.40E+10 | 5.00E+09 |
| ZOT/SS-55 | U-379(IN) | 0.12 | 0.87 | 2.10E+08 | 8.90E+08 |
| ZOT/DHS-2 | U-378(IO) | 0.13 | 0.88 | 2.40E+08 | 2.20E+08 |
| Eu ₂ O ₃ /SS-55 | U-353(ID) | 007 | 0.92 | 8.50E+09 | 1.50(E+09) |
| Eu ₂ O ₃ /DHS-2 | U-381(II) | 0.08 | 0.91 | 2.10E+07 | 1.50E+09 |
| Ta ₂ O ₃ /2150 (RT Cure) | U-358(IG) | 0.18 | 0.90 | 1.70E+09 | 4.30E+09 |
| Ta ₂ O ₃ /SS-55 (RT Cure) | U-359(IH) | 0.17 | 0.88 | 5.60E+09 | 2.20E+09 |
| Ta ₂ O ₃ /DHS-2 (RT Cure) | U-364 (IB) | 0.13 | 0.88 | 5.60E+09 | 4.00E+09 |
| Ta ₂ O ₅ /2130 (HT Cure) | U-358(IG) | 0.16 | 0.90 | 3.00E+09 | 3.00E+09 |
| Ta ₂ O ₅ /SS-55 (HT Cure) | U-359(IH) | 0.15 | 0.88 | 1.60E+10 | 5.00E+09 |
| Ta ₂ O ₄ /DHS-2 (HT Cure) | U-364(IB) | 0.18 | 0.87 | 1.50E+10 | 8.00E+09 |
| Composite CTG White Conductive Concepts: | ncepts: | | | | |
| Sb Doped | U-252(GE) | 0.13 | 0.90 | 8.50E+10 | 1.50E+09 |
| Sb Doped (Flexible) | U-377(HT) | 0.18 | 0.90 | 3.00E+09 | 3.00E+09 |

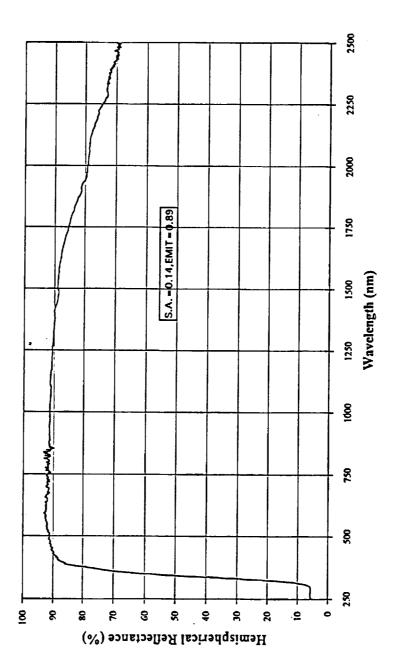


Figure 3-39. Reflectance Spectrum of ZOT/Phos. Sol. Sample #HT-14, Batch #U-377.

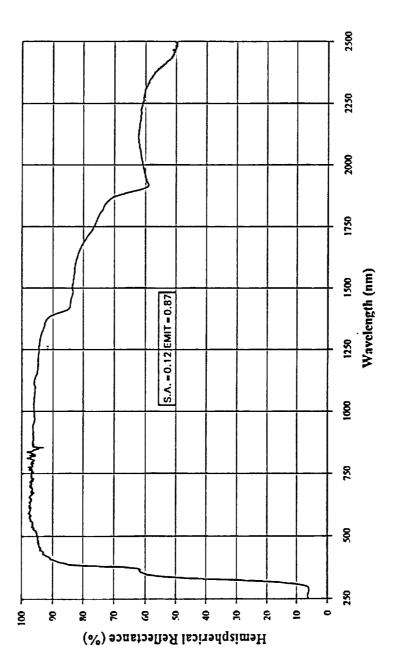


Figure 3-40. Reflectance Spectrum of ZOT/SS-55 Sample #IN-25, Batch #U-379.

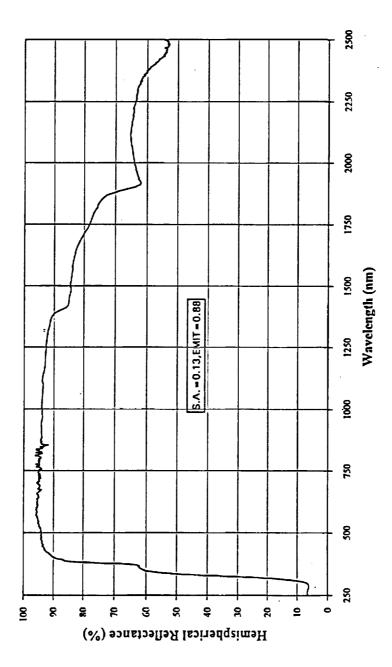


Figure 3-41. Reflectance Spectrum of ZOT/DHS-2 Sample #IO-28, Batch #U-378.

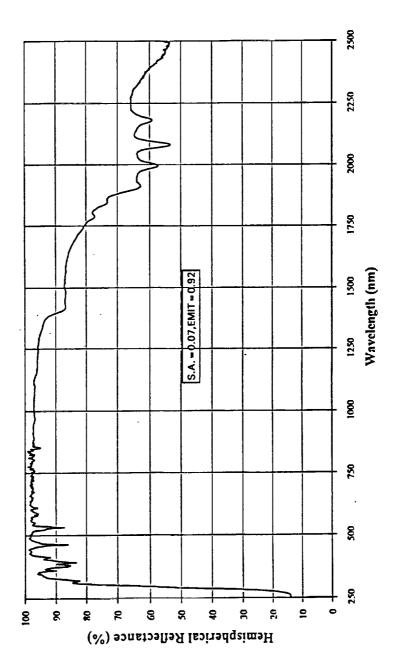


Figure 3-42. Reflectance Spectrum of Eu₂O₃/SS-55 Sample #ID-20, Batch #U-353.

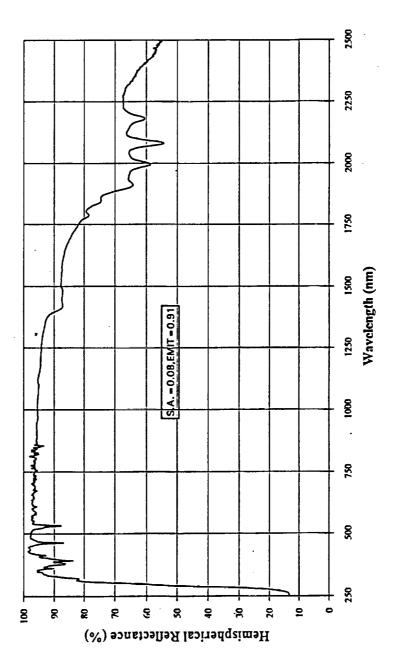


Figure 3-43. Reflectance Spectrum of E₂O₃/DHS-2 Sample #II-17, Batch #U-381.

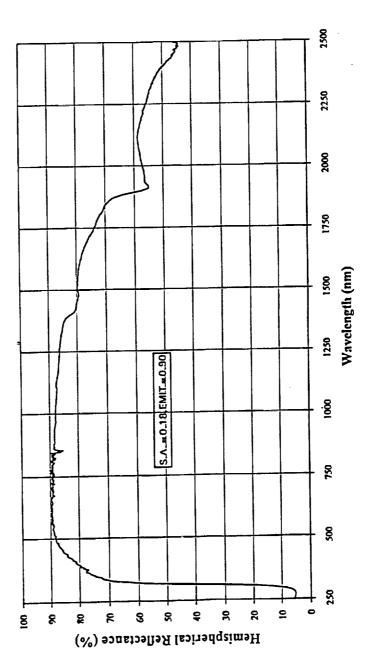


Figure 3-44. Reflectance Spectrum of Ta₂O₅/2130 (RT Cure) Sample #IG-15, Batch #U-358.

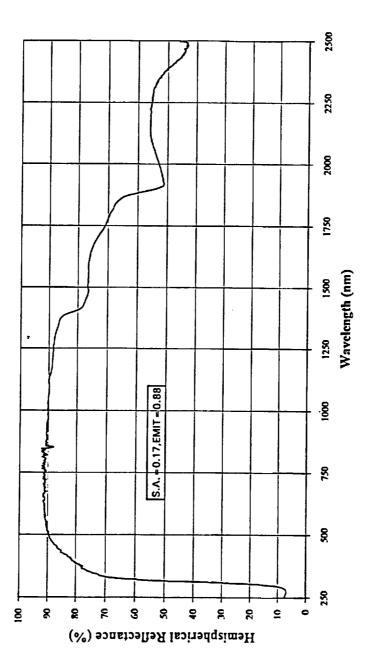


Figure 3-45. Reflectance Spectrum of Ta₂O₅/SS-55 (RT Cure) Sample #IH-21, Batch #U-359.

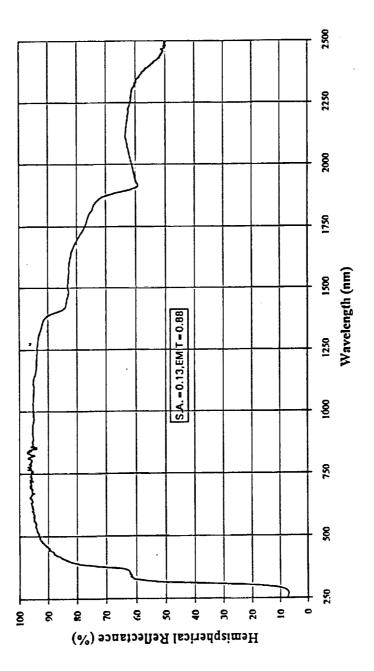


Figure 3-46. Reflectance Spectrum of Ta2O5/DHS-2 (RT Cure) Sample #IB-15, Batch #U-364.

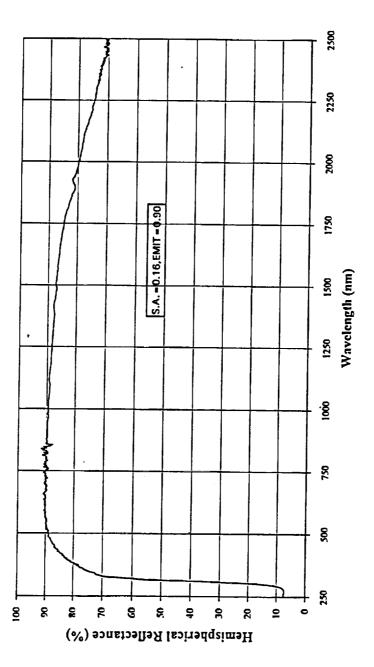


Figure 3-47. Reflectance Spectrum of Ta₂O₅/2130 (HT Cure) Sample #IG-13, Batch #U-358.

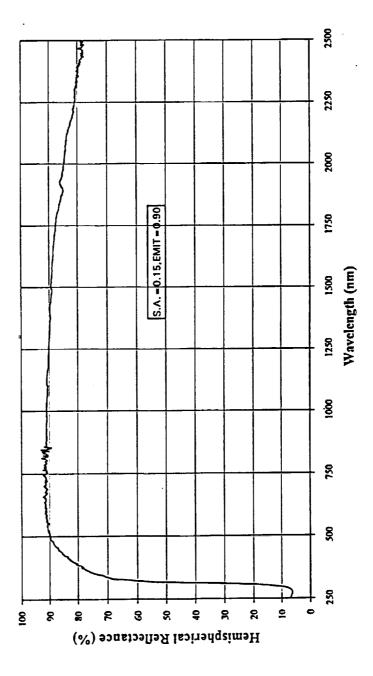


Figure 3-48. Reflectance Spectrum of Ta₂O₅/SS-55 (HT Cure) Sample #IH-18, Batch #U-359.

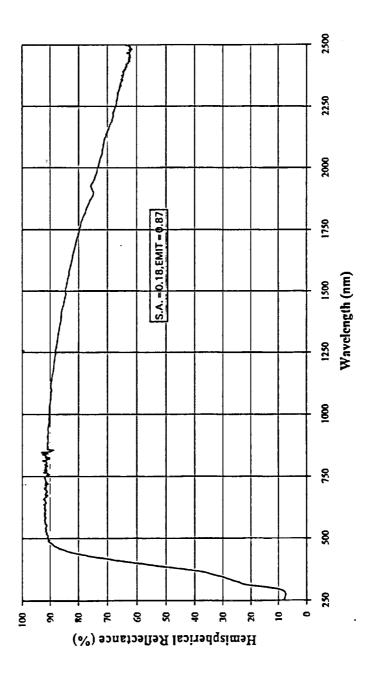


Figure 3-49. Reflectance Spectrum of Ta₂O₅/DHS-2 (HT Cure) Sample #IB-24, Batch #U-364.

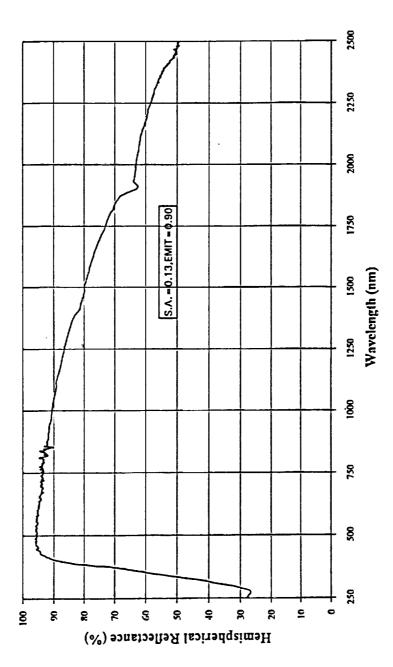


Figure 3-50. Reflectance Spectrum of Sb Doped Sample #GE-16, Batch #U-252.

3.3.1.5 Relevant Comments on Other Material Design Concepts and Data. The following important observations can be made from data presented in Section 3.3.1.4.

The last flexible material design concept based on antimony doping, was not included in screening studies because of vagarities of availability of experimental Sb.doped tin sol and variations of its electrical properties for each lot received by IITRI. The doped tin sol provided by P.Q. Corporation's R&D Division was no longer available in the same grade. A few different versions are still available. In view of this development, we were forced to end the efforts on this material concept.

From measured reflectance on ZOT pigment based concept, it was observed that the use of protonating phos sol (pH = 4.0 to 5.0) provided complete shift of reflectance spectrum to the left of 400 nanometers, a typical representative of edge for unreacted ZnO in ZOT. Thus, one can conclude that the phos sol has reacted with the unreacted ZnO present in the ZOT the pigment. The binder, phos sol, thus helps to achieve a better reflecting thermal control system. The surface electrical resistance measurements indicated the coating met the program goals, and provided a leaky dielectric. The room temperature cured samples of ZOT/phos sol was tested for adhesion following ASTM-D-3359A tape peel adhesion test. The samples passed the adhesion test with a rating of 5A. Heat treatment may provide more toughness. This binder system (phos sol) demands further attention for its use as a viable component to the TCMS.

3.3.1.6 Flexible White (Low α_s/ϵ_N) White TCMS. In this material design, we are totally dependent on the pigment to provide charge transfer processes. The pigment solid-state chemistry and pigment surface chemistry play a major role in tailoring resistances for such a material design. The binder represents a dielectric layer on the pigment. The cured polydimethyl siloxane (without catalyst) acts as a dielectric and has very high volumetric resistivity. The surface resistances are usually in the range of 10^{14} to 10^{15} Ω/\Box . When used with a catalyst such as dibutyl-tin-dilaurate used at IITRI in all silicone formulations, the surface resistance decreases to 10^{12} to 10^{13} Ω/\Box . Petrach Chemicals has introduced organo zirconates and organo titanates as alternate catalyst agents. Petrach Chemicals claims that these catalysts can bring surface resistance down to 10^{10} to 10^{11} Ω/\Box , but may have adverse impact on pot life and cure kinetics of the resin system used in thermal control coatings along with some penalties in α_s . In view of the limited scope of this screening study, emphasis was placed on the use of

doped pigment alone, with the highest possible pigment-to-volume ratio to formulate the flexible TCMS. No attempts were made to manipulate flexible binder dielectric system so that it can participate in the charge transfer process as a leaky dielectric. Such a manipulation calls for an entire research program devoted towards such a goal.

The screening study efforts were very limited in producing the needed resistance tailoring results. Our dependence on pigment to tailor surface resistance required the material design to depend on very high dopant levels to meet the surface resistance goals. This was more apparent when interested industrial parties evaluated RF losses of candidate flexible and silicate-based TCMS, and gave qualitative feedback. The DS13N/LO-41 was a border line acceptable compared to Sb-doped versions. The RF losses in Sb:Zn₂SnO₄ were very high. (33) Thus, only DS13N/LO-41 was included in the screening study.

The comments made to us⁽³³⁾ citing European Space Agency (ESA) studies on S13GP/LO-1 for inquiring reasons for electroluminescence were very important from material design issues. There has been no effort to evaluate the behavior of this material under charged conditions. Such behavior needs to be fully characterized along with the electrical performance. It may be worthwhile to visit this phenomenon of electroluminescence in a qualitative way.

3.3.1.7 Electroluminescence (EL) in Thermal Control Systems. The phenomenon of EL means conversion of electrical energy directly into light. The word direct is important, because it means no intermediate process (or mechanism). Many textbooks can give detailed definitions, candidate mechanisms, and lists of materials that show such characteristics. All pigments used in the candidate TCMS are either ZnO based or titanate (ZOT) based, which can be found on such lists. Thus, all are capable of showing EL performance under favorable charging conditions.

The essential element of EL grain in the thermal control system can be stated briefly as follows: The electric energy imparted to grain is used to excite the grain into a more energetic state. This stored energy tends to relax by a variety of processes. One of which is radiative recombination, i.e., that free electrons and holes combine and fraction of liberated energy is emitted as photon. The two essential mechanisms are then electrical excitation and radiative

recombination. They both need to be studied. Radiative recombination competes with the other process which are bulk and are nonradiative-relaxation type to lose excitation energy.

In the case of candidate electrically conductive TCMS with doped silicate or soda silicate binders, the relaxation process is built into the binder system. In the case of flexible candidate thermal control systems, the polydimethyl siloxane cannot participate in a relaxation process until one alters the polymer chemically to make it as a leaky dielectric. The impact of such a manipulation on space environment stability of siloxane polymer is not known. Hence, the approach taken here was <u>not</u> to alter polymer chemistry. This leads to providing limits on relaxation processes available as a mechanism of charge dissipation in flexible conductive thermal control coatings.

The EL effects also need to be studied for the state-of-the-art TCMS and for the conductive versions where we provide doped relaxation paths. The time constants of recombination and relaxation processes and nature of incoming space irradiation (UV, e', P⁺) determines resultant charge development. From this discussion, one can conclude that it is essential to characterize baseline and candidate material designs for their response to pulses. No such study was planned during this effort. The goal for this screening study was to provide candidate engineering material system, and to check its response to select promising candidates for future scaleup and validation and use on space hardware.

Lastly, interested researcher should refer to the treaties on EL to assess role of dielectric behavior, interface effects, contact formation and junctions and their impact on radiative recombination. We expect that EL does play a role for the inorganic version of the conductive concepts investigated in this study as well as a significant role for flexible versions. Future studies need to address these issues, if they are important for the hardware. If EL serves as an effective mechanism, one can consider incorporating EL effect enhancing efficient whisker-type innovative material forms (e.g., ZnS whiskers) in existing flexible TCMS for innovative material designs and formulations.

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4. RESULTS OF SPACE ENVIRONMENT SIMULATIONS STUDIES

The space environment simulation did not happen as planned for this screening study. Samples for simulations were submitted as opportunities were made available during this research program. At times, we were ready to submit samples per requests, and at times, we were not completely ready to take advantage of such an opportunity. The detailed screening study on all candidate samples was carried out later at the Aerospace Corporation. In this chapter, results are presented from space environment simulation studies carried out on the candidates material concepts and baseline materials presented earlier in Chapter 3.

4.1 SPACE SIMULATION STUDIES AT AEROSPACE CORPORATION

The space environment simulation at Aerospace Corporation was conducted as a combined radiation test, typically representing low earth orbit (LEO) mission like the DMSP orbit consisting of 2000 ESH of UV + 8.26 x 10¹³ e'/cm² (100 KeV) + 5.18 x 10¹³ e'/cm² (35 KeV). The detailed report of exposures, and results of pre- and post-total hemispherical solar absorptance measurements were generated under a sub-contract for this material design and screening study. The report is provided in Appendix A for completeness.

The results are summarized here on baseline materials and candidate conductive material design concepts, and presented here in tabular form (Tables 4-1 to 4-5) for completeness and ease of comparison.

| Table 4-1. Re | sults of Vacuum | -UV, Electr | on Irradi: | ation of Bl | ack TCMS at | Aerospace (| Corporation |
|------------------------|------------------------------|---------------|------------|-------------|------------------------|-----------------------------|----------------------|
| TCMS | Pigment | Binder | Batch # | Sample # | Pretest α _s | Post Test α _s | $\Delta lpha_{ m S}$ |
| Baseline Nonco | nducting TCMS | Candidates | | | | , | |
| MH21IP ^a | Black Glass | Kasil 2130 | U-284 | GZ-21 | 0.981 | 0.981 | 0 |
| MH21S/LO ² | Black Glass | LO-1 | U-360 | HZ-18 | 0.982 | 0.982 | 0 |
| D21S/LO ^a | Cosmic Black WB-500 | LO-1 | T-023 | BA-8 | 0.98 | 0.98 | 0 |
| Conductive TC | MS Candidates | | | | | | |
| DBG-IP* | In Doped Black Glass | Kasil 2130 | U021 | EO-7 | 0.974 | 0.975 | 0.001 |
| DBG-IPb | In Doped Black Glass | Kasil 2130 | U021 | EO-8 | 0.973 | 0.974 | 0.001 |
| MH55IC ^a | Black Glass, Graphite | Kasil 2130 | S-024 | Z-9 | 0.94 | 0.939 | -0.001 |
| MH55ICP ^b | Black Glass, Graphite | Kasil 2130 | S-024 | Z-10 | 0.942 | 0.941 | -0.001 |
| DBG/DHS ² | Doped Black Glass | DHS-1 | U-283 | GU-13 | 0.958 | 0.957 | -0.001 |
| DBG/DHSb | Doped Black Glass | DHS-1 | U-283 | GU-14 | 0.956 | 0.956 | 0 |
| MH21SC/LO ² | Black Glass, Graphite | LO-1 | U-166 | FD-20 | 0.963 | 0.964 | 0.001 |
| MH21SC/LOb | Black Glass, Graphite | LO-1 | U-166 | FD-21 | 0.965 | 0.965 | 0 |
| MH41SC/LO ² | Black Glass/B4C | LO-1 | U-053 | FD-5 | 0.951 | 0.95 | -0.001 |
| MH41SC/LOb | Black Glass/B4C | LO-1 | U-053 | FD-6 | 0.95 | 0.953 | 0.003 |
| D21SC/LO | Cosmic Black, Graphite | LO-1 | T-026 | BJ-1 | 0.955 | 0.956 | 0.001 |
| D21SC/LO ^b | Cosmic Black, Graphite | LO-1 | T-026 | ВЈ-2 | 0.956 | 0.957 | 0.001 |

⁽a) UV intensity (0.94 to 0.98 Suns): 2000 ESH + 5.18x10¹³ e'/cm² (35 KeV) + 8.26x10¹³ e'/cm² (100 KeV). (b) UV intensity (0.60 to 0.90 Suns): 1500 to 1800 ESH + 5.18x10¹³ e'/cm² (35 KeV) + 8.26x10¹³ e'/cm² (100 KeV).

Table 4-2. Results of Vacuum-UV, Electron Irradiation of White (Low α_s/ϵ_N) TCMS at Aerospace Corporation for Baseline Material Systems and Z-93CZY **Batch** Sample Post Binder $\Delta \alpha_{\rm S}$ **TCMS** Pigment Pretest α_s Test α_s # # **Baseline Nonconducting TCMS** 9890 ZOT#1 0.025 ZOT² ZOT 0.079 0.104 None #98790 Z-93P^a Zinc Oxide U-347 HX-10 0.133 0.14 0.007 Kasil 2130 • YB-71 ZOT HW-3 0.107 Kasil U-348 0.14 0.033 #98790 2130 YB-71P* ZOT Kasil U-376 IM-27 0.112 0.123 0.011 #98790 2130 S13GP/LO-1ª S13GP LO-1 U-310 **U-27** 0.167 0.177 0.01 S13GP/LO-41^a S13GP LO-41 U-276 HC-20 0.149 0.007 0.156 DZSS/LO-41^a **DZSS** LO-41 U-208 GH-8 0.077 0.243 0.166 ZOTS/LO-41^a ZOT LO-41 U-343 28 0.133 0.196 0.063 #98790 Eu203/2130^a Eu203 Kasil U-352 IF-18 0.092 0.12 0.028 2130 Eu203/LO-41* Eu203 LO-41 U-403 IS-19 0.136 0.178 0.042 Conductive Concept Z-93CXY-Based TCMS ZnO(FC)/2130^a ZnO(FC) Kasil U-321 HM-14 0.148 0.154 0.006 2130 ZnO(FC)/2130b ZnO(FC) Kasil U-321 HM-15 0.148 0.149 0.001 2130 ZnO(FC)/SS-55² ZnO(FC) SS-55 U-322 HF-26 0.147 0.144 -0.003 ZnO(FC)/SS-55b ZnO(FC) SS-55 -0.002 U-322 HF-27 0.146 0.144 ZnO(FC)/DHS-1* ZnO(FC) DHS-1 U-323 HG-23 0.157 0.164 0.007 ZnO(FC)/DHS-1^b ZnO(FC) DHS-1 U-323 **HG-22** 0.149 0.156 0.007

U-327

U-327

HL-14

HL-25

0.139

0.146

0.141

0.146

DHS-2

DHS-2

ZnO(FC)

ZnO(FC)

ZnO(FC)/DHS-2^a

ZnO(FC)/DHS-2b

0.002

0

⁽a) UV intensity (0.94 to 0.98 Suns): $2000 \text{ ESH} + 5.18 \times 10^{13} \text{ e}'/\text{cm}^2$ (35 KeV) + $8.26 \times 10^{13} \text{ e}'/\text{cm}^2$ (100 KeV). (b) UV intensity (0.60 to 0.90 Suns): $1500 \text{ to } 1800 \text{ ESH} + 5.18 \times 10^{13} \text{ e}'/\text{cm}^2$ (35 KeV) + $8.26 \times 10^{13} \text{ e}'/\text{cm}^2$ (100 KeV).

| Table 4-3. Resul | Corporation | on for Basel | ine Mater | ial Systems | and Z-93SC2 | ZY | |
|--------------------------|-------------|---------------|------------|-------------|------------------------|-----------------------------|----------------------|
| TCMS | Pigment | Binder | Batch # | Sample # | Pretest α _s | Post Test α _s | $\Delta lpha_{ m S}$ |
| Conductive Conce | pt Z-93SCXY | -Based TCN | 1 S | | | | |
| S13GP/2130 ^a | S13GP | Kasil 2130 | NA | FM-18 | 0.131 | 0.148 | 0.017 |
| S13GP/2130b | S13GP | Kasil 2130 | NA | FM-21 | 0.129 | 0.137 | 0.008 |
| S13GP/SS-55* | S13GP | SS-55 | U-405 | IU-22 | 0.125 | 0.14 | 0.015 |
| S13GP/SS-55b | S13GP | SS-55 | U-405 | IU-25 | 0.127 | 0.134 | 0.007 |
| S13GP/DHS-1* | S13GP | DHS-1 | U-279 | GS-16 | 0.147 | 0.153 | 0.006 |
| S13GP/DHS-1b | S13GP | DHS-1 | U-279 | GS-17 | 0.142 | 0.148 | 0.006 |
| S13GP/DHS-2 ^a | S13GP | DHS-2 | U-328 | HH-20 | 0.13 | 0.155 | 0.025 |
| S13GP/DHS-2b | S13GP | DHS-2 | U-328 | HH-27 | 0.134 | 0.146 | 0.012 |
| S13N/SS-55* | S13N | SS-55 | U-260 | GR-3 | 0.115 | 0.141 | 0.026 |
| S13N/SS-55b | S13N | SS-55 | U-260 | GR-7 | 0.114 | 0.119 | 0.005 |
| S13N/DHS-1* | S13N | DHS-1 | U-317 | HE-19 | 0.15 | 0.173 | 0.023 |
| S13N/DHS-1b | S13N | DHS-1 | U-317 | HE-23 | 0.143 | 0.16 | 0.017 |
| S13N/DHS-2* | S13N | DHS-2 | U-329 | HI-14 | 0.134 | 0.161 | 0.027 |
| S13N/DHS-2b | \$13N | DHS-2 | U-329 | HI-27 | 0.132 | 0.16 | 0.028 |

⁽a) UV intensity (0.94 to 0.98 Suns): $2000 \text{ ESH} + 5.18 \times 10^{13} \text{ e}'/\text{cm}^2$ (35 KeV) $+ 8.26 \times 10^{13} \text{ e}'/\text{cm}^2$ (100 KeV). (b) UV intensity (0.60 to 0.90 Suns): $1500 \text{ to } 1800 \text{ ESH} + 5.18 \times 10^{13} \text{ e}'/\text{cm}^2$ (35 KeV) $+ 8.26 \times 10^{13} \text{ e}'/\text{cm}^2$ (100 KeV).

| Table 4-4. Re | | | | | ite (Low α _s /ε _N nd Z-93SCLM | | Aerospace |
|--------------------------|--------------|-------------|------------|-------------|--|-----------------------------|-----------------|
| TCMS | Pigment | Binder | Batch # | Sample # | Pretest α ₂ | Post Test α _s | Δα _s |
| Conductive Cond | cept Z-93SCL | MXY-Based T | CMS | | | | |
| DZS/2130 ² | DZS | Kasil 2130 | T-244 | DK-11 | 0.077 | 0.132 | 0.055 |
| DZS/2130b | DZS | Kasil 2130 | T-244 | DK-10 | 0.081 | 0.122 | 0.041 |
| DZS/SS-55 ^a | DZS | SS-55 | U-207 | FO-20 | 0.107 | 0.204 | 0.097 |
| DZS/SS-55b | DZS | SS-55 | U-207 | FO-23 | 0.1 | 0.185 | 0.085 |
| DZS/DHS-1* | DZS | DHS-1 | U-278 | GQ-4 | 0.075 | 0.202 | 0.127 |
| DZS/DHS-1b | DZS | DHS-1 | U-278 | GQ-8 | 0.078 | 0.163 | 0.085 |
| DZS/DHS-2ª | DZS | DHS-2 | U-331 | HK-15 | 0.086 | 0.189 | 0.103 |
| DZS/DHS-2b | DZS | DHS-2 | U-331 | HK-27 | 0.089 | 0.18 | 0.091 |
| DS13N/SS-55 ² | DS13N | SS-55 | U-315 | P-26 | 0.138 | 0.173 | 0.035 |
| DS13N/SS-55b | DS13N | SS-55 | U-315 | P-27 | 0.137 | 0.153 | 0.016 |
| DS13N/DHS-1* | DS13N | DHS-1 | U-316 | S-7 | 0.123 | 0.133 | 0.01 |
| DS13N/DHS-1b | DS13N | DHS-1 | U-316 | S-1 | 0.124 | 0.161 | 0.037 |
| DS13N/DHS-2* | DS13N | DHS-2 | U-330 | HJ-20 | 0.122 | 0.128 | 0.006 |
| DS13N/DHS-2b | DS13N | DHS-2 | U-330 | НЈ-20 | 0.129 | 0.149 | 0.02 |

⁽a) UV intensity (0.94 to 0.98 Suns): $2000 \text{ ESH} + 5.18 \times 10^{13} \text{ e}'/\text{cm}^2 (35 \text{ KeV}) + 8.26 \times 10^{13} \text{ e}'/\text{cm}^2 (100 \text{ KeV})$. (b) UV intensity (0.60 to 0.90 Suns): $1500 \text{ to } 1800 \text{ ESH} + 5.18 \times 10^{13} \text{ e}'/\text{cm}^2 (35 \text{ KeV}) + 8.26 \times 10^{13} \text{ e}'/\text{cm}^2$

⁽¹⁰⁰ KeV).

Table 4-5. Results of Vacuum-UV, Electron Irradiation of White (Low α_{S}/ϵ_{N}) TCMS at Aerospace Corporation for Baseline Material Systems for Other Conductive Concepts Post Sample Batch Test $\Delta \alpha_{\rm s}$ Pretest as Binder **Pigment TCMS** # # $\alpha_{\rm S}$ Other Conductive Concepts Based TCMS 0.149 0.018 U-377 HT-21 0.131 Phos. ZOT #98790 ZOT/Phos. Sol.^a Sol 0.023 0.151 HT-22 0.128 Phos. U-377 ZOT #98790 ZOT/Phos. Sol.^b Sol 0.039 0.156 IN-13 0.117 **SS-55** U-379 **ZOT #98790** ZOT/SS-55^a 0.156 0.044 0.112 IN-15 SS-55 U-379 ZOT #98790 ZOT/SS-55b 0.033 0.125 0.158 U-378 IO-21 DHS-2 ZOT/DHS-2^a **ZOT #98790** 0.059 **IO-28** 0.122 0.181 DHS-2 U-378 ZOT #98790 ZOT/DHS-2b ID-11 0.087 0.138 0.051 **SS-55** U-353 Eu203 Eu203/SS-55^a 0.026 ID-20 0.1 0.126 SS-55 U-353 Eu203 Eu203/SS-55b 0.158 0.065 0.093 U-381 II-12 DHS-2 Eu203 Eu203/DHS-2* 0.032 II-15 0.076 0.108 DHS-2 U-381 Eu203 Eu203/DHS-2b 0.304 0.141 IG-11 0.163 Kasil 2130/RT U-358 Ta205 Ta205/2130^a 0.106 0.286 0.18 Kasil 2130/RT **IG-25** U-358 Ta205 Ta205/2130b 0.067 IG-12 0.165 0.232 U-358 Ta205 Kasil Ta205/2130^a 2130/HT 0.05 0.231 **IG-22** 0.181 U-358 Ta205 Kasil Ta205/2130b 2130/HT 0.085 0.137 0.222 IH-19 U-359 Ta205/SS-55* Ta205 55/RT 0.226 0.083 0.143 U-359 IH-20 Ta205 Ta205/SS-55b 55/RT 0.199 0.05 IH-17 0.149 U-359 Ta205/SS-55^a Ta205 55/HT 0.216 0.049 0.167 U-359 IH-27 SS-Ta205/SS-55^b Ta205 55/HT 0.159 0.018 0.141 U-252 GE-9 ZnO+Sb:Zn2SnO4 Kasil Sb Doped^a 2130 0.144 0.176 0.032 GE-8 ZnO+Sb:Zn2SnO4 U-252 Kasil Sb Doped^b 2130 Flexible Conductive Concept Based TCMS 0.278 0.117 HQ-10 0.161 LO-41 U-334 DS13N DS13N/LO-41' 0.076 0.239 HO-12 0.163 LO-41 U-334 DS13N DS13N/LO-41b

⁽a) UV intensity (0.94 to 0.98 Suns): $2000 \text{ ESH} + 5.18 \times 10^{13} \text{ e}'/\text{cm}^2$ (35 KeV) $+ 8.26 \times 10^{13} \text{ e}'/\text{cm}^2$ (100 KeV). (b) UV intensity (0.60 to 0.90 Suns): $1500 \text{ to } 1800 \text{ ESH} + 5.18 \times 10^{13} \text{ e}'/\text{cm}^2$ (35 KeV) $+ 8.26 \times 10^{13} \text{ e}'/\text{cm}^2$ (100 KeV).

4.2 SPACE ENVIRONMENT AT GODDARD SPACE FLIGHT CENTER (GSFC)

The space environment simulation at GSFC was conducted for UV-vacuum in situ reflectance degradation studies for 1000 ESH exposure. It refers to the original research plan, the screening study at the Aerospace Corporation which was to provide candidate concepts for this simulation exercise, but the program and scheduling did happen as originally planned. Dr. Lon Kauder offered additional test spots very early in the program to insert Z-93SC55 (S13GP/DHS) samples into the UV-vacuum test that was being carried out at GSFC. The Z-93SCXY:Z-93SC55 (S13GP/DHS) samples from GPS Block IIR hardware witness samples were provided along with samples based on the material design concept Z93SCLMXY (Z93SC1655:[DZS/SS-55]). GSFC (thermal branch) also processed Z93SC55 samples (T-251). The following samples were submitted to GSFC and were included in UV-vacuum exposure tests for up to 4300 ESH. In situ reflectances were measured at every 500 ESH. The report on this testing can be found in Appendix B, titled "Ultraviolet Degradation of IITRI's Six White Paints." The results from this study can be summarized in a tabular form (Tables 4-6 and 4-7) as follows.

| | Table 4-6. Car | ndidate Baseline a | nd Conductive Concept | s for 4000 ESH Study | |
|-------------|-----------------------------|--------------------|---------------------------------------|----------------------------|-----|
| Item No. | Material Concept | Pigment | Binder | Batch # Sample # | PBR |
| 1 | Z-93SCXY: (Z-93SC55) | S13GP | DHS 2130 + SS-55 (50:50) | EI-2 (T-342) | 5.5 |
| 2 | Z-93SC55 | S13GP | DHS 2130 + SS-55 (50:50) | T-251 processed by GSFC | 5.5 |
| 3 | Z-93SCLMXY: (Z-93SC1655) | DCS | SS- 55 | EK-20 (T-344) | 5.5 |
| 4 | Z-93P | calcined ZnO | 2130 | N/A | 4.3 |
| 5 | S13GP/LO-1 | S13GP | MHS/LO stripped polydimethyl siloxane | N/A | 2.1 |

| Table 4-7 .] | Results of In S | Situ Reflect | ance Degradation Baseline TCMS | 1 Study [4300] | ESH (UV-Vacu | ium)] |
|-----------------------|--|--------------|-----------------------------------|-----------------|--------------|----------|
| Sample Description | Optical Properties | 0 ESH | 1010 ESH | 2041 ESH | 3042 ESH | 4331 ESH |
| Nonconductive Whi | te TCMS | | | | <u> </u> | |
| S13GP/LO-1 | $\alpha_{\scriptscriptstyle S}$ | .181 | .186 | .184 | .195 | .192 |
| | € _N | .922 | No Measurement | .919 | .917 | .918 |
| Z93P (Kasil 2130) | $\alpha_{\scriptscriptstyle{	extsf{S}}}$ | .172 | .175 | .178 | .181 | .189 |
| | € _N | .930 | .931 | .929 | .928 | .929 |

| Table 4-8. Re | esults of In Sit | u Reflectance for Con | e Degradation ductive TCMS | Study [4300] | ESH (UV-Vacu | ıum)] |
|---------------------|-----------------------|--------------------------|-------------------------------|---------------|--------------|-------------------|
| Sample Description | Optical Properties | 0 ESH | 1010 ESH | 2041 ESH | 3042 ESH | 4331 ESH |
| Conductive White TO | CMS | | | | | |
| Z93SC55 (T251) | $\alpha_{\rm s}$ | .140 | .151 | .157 | .175 | .177 |
| | ϵ_{N} | .935 | .929 | .930 | .931 | .929 |
| Z93SC55 (EI-2) | α_{s} | .151 | .164 | .177 | .200 | .207 |
| | ϵ_{N} | .934 | .928 | .926 | .928 | .929 |
| Z93SC1655 (EK-20) | α_{s} | .128 | .220 | .261 | .304 | sample removed |
| | € _N | .935 | .926 | .927 | .925 | sample removed |

The SEE Project Office has also contracted GSFC to assist this screening study to study reflectance degradation for 1000 ESH exposure of UV-vacuum irradiation. Dr. Lon Kauder and Ms. Wanda Peters at GSFC received companion samples of the selected material concepts that were also submitted to Aerospace Corporation. The material concepts submitted to GSFC are listed here in Table 4-9.

Unfortunately, selection of conductive concepts for this study do not benefit from the input from the screening test at Aerospace Corporation as planned. The scheduling conflicts precluded us from doing that. Thus, the choices were made strictly based on conservative

| Table 4-9. Electrically | | 00 ESH UV-Vacuum Exposure ion Study | Induced Reflectance |
|--|-------------|---|---------------------------------------|
| Concept Name | Sample # | Pigment | Binder |
| Black Conductive TCMS | | | |
| DBG-IP | GTM-29 | Doped black glass (Indium doped) | Kasil 2130 |
| MH55-ICP | Z-6 | Black glass, Graphite 9035, ZnO | Kasil 2130 |
| MH21SC/LO | GX-16 | Black glass, graphite fiber, Graphite 9035 | MHS/LO stripped polydimethyl siloxane |
| MS41SCB/LO | U-356 IC-20 | Black glass/B ₄ C | MHS/LO stripped polydimethyl siloxane |
| D36SCB/LO | JV-17 | Cosmic black WB-500, B ₄ C | MHS/LO stripped polydimethyl siloxane |
| White Conductive TCMS | 3 | | |
| ZnO(FC)/DHS-2 | HL-26 | ZnO(FC) | DHS-2 |
| S13GP/DHS-2 | НН-19 | S13GP | DHS-2 |
| S13N/DHS-2 | HI-21 | S13LN | DHS-2 |
| DS13N/DHS-2 | НЈ-25 | DS13N | DHS-2 |
| DS13N/SS-55 | P-31 | DS13N | SS-55 |
| DS13N/LO-51 | JX-16 | DS13N | MHS/LO stripped polydimethyl siloxane |
| ZOT/SS-55 | IW-19 | ZOT #98790 | SS-55 |
| Eu ₂ O ₃ */SS-55 | HD-26 | Eu ₂ O ₃ | SS-55 |
| Sb Doped | GE-12 | ZnO + Sb doped Zn₂SnO₄ | Kasil 2130 |

engineering judgement where electrical resistance behavior was more reproducible. We could have also picked candidates like ZOT/Phos. Sol and D21SC/LO from reproducibility results, but the space limitations in the chamber did not allow more samples. In conclusion, one should not assume that these concepts were considered as promising candidates.

The report for the degradation behavior results can be found in Appendix C titled, "Ultraviolet Degradation on Study of IITRI Electrically Conductive Thermal Control

Formulations". The summary of their results is given here in Table 4-10 for quick comparison purposes.

| | of Candidate | TCMS at GSFC | ure Induced Degrada | |
|---|---------------------------------|--------------|---------------------|-----------------|
| TCMS/Sample # Description | Optical Properties | 0 ESH | 1068 ESH | Δα _s |
| DBG-IP/[GTM-29] | α_{s} | .957 | .951 | -0.006 |
| • | $\epsilon_{ m N}$ | .911 | .914 | |
| MH55-IC/[Z-6] | $\alpha_{\rm s}$ | .928 | .938 | 0.010 |
| | $\epsilon_{ m N}$ | .886 | .885 | |
| MH21SC/LO/[GX-16] | $\alpha_{\rm s}$ | .951 | .955 | 0.004 |
| | $\epsilon_{ m N}$ | .862 | .864 | |
| MS41SCB/LO/[IC-20] | $\alpha_{\scriptscriptstyle S}$ | .939 | .947 | 0.008 |
| | € _N | .866 | .867 | |
| D36SCB/LO/[JV-17] | $\alpha_{\scriptscriptstyle S}$ | .962 | .967 | 0.005 |
| | € _N | .908 | .909 | |
| ZnO(FC)/DHS-2/[HL-26] | $lpha_{	ext{s}}$ | .158 | .168 | 0.010 |
| | € _N | .925 | .925 | |
| S13GP/DHS-2/[HH-19] | $\alpha_{\scriptscriptstyle S}$ | .148_ | .165 | 0.017 |
| | € _N | .934 | .932 | |
| S13N/DHS-2/[HI-21] | $\alpha_{\rm s}$ | .152 | .162 | 0.010 |
| | $\epsilon_{ m N}$ | .935 | .932 | |
| DS13N/DHS-2/[HJ-25] | $\alpha_{\rm s}$ | .148 | .154 | 0.006 |
| | $\epsilon_{ m N}$ | .937 | .935 | |
| DS13N/SS-55/[P-31] | α_{ς} | .152 | .172 | 0.020 |
| | $\epsilon_{ m N}$ | .930 | .930 | |
| DS13N/LO-51/[JX-16] | $\alpha_{\rm s}$ | .210 | .434 | 0.224 |
| · · • • | $\epsilon_{	extsf{N}}$ | .926 | .926 | |
| ZOT/SS-55/[IW-19] | $\alpha_{\rm s}$ | .129 | .180 | 0.051 |
| | $\epsilon_{ m N}$ | .881 | .881 | |
| Eu ₂ O ₃ /SS-55/[HD-26] | $\alpha_{\scriptscriptstyle S}$ | .096 | .129 | 0.033 |
| | $\epsilon_{ m N}$ | .923 | .921 | |
| Sb Doped Concept/[GE-12] | $\alpha_{\rm s}$ | .156 | .243 | 0.087 |
| no noben comerbatan 1-1 | $\epsilon_{ m N}$ | .930 | .931 | |

4.3. SPACE ENVIRONMENT SIMULATION AT AIR FORCE MATERIALS LABORATORY, (AFML) SCEPTRE RESULTS

During the very beginning of the program, interest was also shown by Mr. P. Carlin of AFML to provide assistance for some in situ reflectance degradation studies of the selected concept in the SCEPTRE facility. The program had not progressed enough to make the best possible judgements and to select concepts and samples to provide to AFML. Mr. P. Carlin (AFML) and Mr. C. Cerbus (UDRI) requested two candidates of doped conductive TCMS concept for the space environment simulations and offered to provide six spots for passive samples, where EUV intensities were high. The material concepts submitted to the Air Force Materials Laboratory for their test 96QV01 are described in Table 4-11 for ready reference, along with the tabulated results in Table 4-12. The exposure consists of irradiation in vacuum (5.0E-08 torr) with UV exposure at 2 suns and 1.6 suns along with electrons of 1.0 KeV at a flux of 6.0E+09 e'/cm²/sec and electrons of 10.0 KeV at a flux of 3.0E+09 e'/cm²/sec for total fluence of 1.2E+16e'/cm². The sample temperature was maintained from 135° to 146°F with loss of chiller cooling causing temperatures to rise up to 185°F for 12 hr. period. The total exposure happened over 314 hours. The report provided by Mr. C. Cerbus can be found in Appendix D.

| Material Concept | Pigment | Binder | PBR |
|----------------------------------|--|---------------------------------------|-----|
| Conductive Concepts for | "In Situ" Studies | | |
| DS13N/LO-51 | DS13N | MHS/LO stripped polydimethyl siloxane | 5.0 |
| DS13N/SS-55 | DS13N | SS-55 | 5.5 |
| Black Conductive TCMS | for "Passive" Tests | | |
| DGB/IP | Doped Black Glass DBG | Kasil 2130 | 2.0 |
| D36SCB/LO | Cosmic Black WB-500 graphite 9035 B ₄ C | MHS/LO stripped polydimethyl Siloxane | 3.6 |
| White Conductive TCMS | for "Passive" Test | | |
| Z-93CXY:Z9355 | ZnO(FC) | DHS-2 | 5.5 |
| Z-93SCXY:Z-93SC55 | S13N | DHS-2 | 5.5 |
| Z-93SCLMXY:Z-931655 | DS13N | DHS-2 | 5.5 |
| Other Conductive White | TCMS for "Passive" Test | | |
| Eu ₂ O ₃ * | Eu ₂ O ₃ | SS-55 | 5.5 |

| Tal | ole 4-12. Sum | mary of Resul | its from | n SCEI | TRE Screening | Test 96QV01 | |
|---|---------------|----------------|-------------------|------------------------------|------------------------|----------------------|-----------------|
| TCMS | Batch | Specimen ID | _ | IV osure | Pretest α _s | Post Test α_s | Δα _s |
| Measured In Situ | Conductive V | White Coatings | 5 | | | | |
| DS13N/LO-51 | V-032 | JZ-17 | 1.6] | EUVS | 0.193 | 0.423 | 0.230 |
| DS13N/LO-51 | U-315 | P-25 | 2.1 1 | EUVS | 0.157 | 0.267 | 0.110 |
| Passively Exposed | Conductive | Black TCMS | | | | | |
| DGB-IP | U-021 | EO-17 | 4.0 1 | EUVS | 0.970 | 0.978 | 0.008 |
| D36SCB/LO | V-036 | JV-15 | 3.9 | EUVS | 0.966 | 0.956 | -0.010 |
| Conductive White | e TCMS | | | | | | |
| Z-93CXY | | | | | | | _ |
| ZnO(FC/DHS-2 | U-327 | HL-23 | 1.1 | EUVS | 0.143 | 0.225 | 0.082 |
| Z-93SCXY | | | | | | | |
| S13N/DHS-2 | U-329 | HI-17 | 1.0 | EUVS | 0.130 | 0.196 | 0.066 |
| Z-93SCLMXY | | | | | | | |
| DS13N/DHS-2 | U-330 | НЈ-16 | 3.3 | EUVS | 0.117 | 0.237 | 0.120 |
| Other Conductive | e White Coati | ngs | | | | | |
| Eu203/SS-55 | U-353 | ID-19 | 3.6 | EUVS | 0.076 | 0.161 | 0.085 |
| Test Environment | Parameters: | | | <u>Tempe</u> | rature Range: | | |
| Vacuum: 5.0E-08 tour Total Exposure Time: 314 hrs. | | | In Sitt Passiv | | -146°F F | | |
| 1.0 KeV electron flux 6.0E+09 e'/cm2/sec 10.0 KeV electron flux 3.0E+09 e'/cm2/sec | | | | chiller for 12 hr ?-185°F | s and temperatu | res reached | |
| Total Fluence: | 1.2E+ | 16 e-/cm2 | | | | | |

4.4 IRRADIATION INDUCED CHARGE DEVELOPMENT [ESD] TESTING

Dr. Lon Kauder at GSFC was also contracted by the SEE Project Office to perform ESD testing on the selected candidates from the screening study conducted at IITRI/AMCL. The samples used for the ESD testing are described below.

IITRI submitted 4"x4" coated samples on Al6061-T6 substrates. The Al-6061-T6 substrates were chosen to assure good electrical contact to substrate. The sample were exposed

to vacuum for several hours to ensure minimum moisture contributions. The electron gun(s) were used to carry out required exposure of the samples to electrons of various energy for various fluxes (currents). The energies used were from 5 KeV to 20 KeV and currents used were from 1.0 nanoamps/cm² to 10 nanoamps/cm². The charge induced potential (volts) developed on the sample was measured by tracking with a voltage trek probe over the sample. These measurements were carried out at various temperatures between the range of 77 to 325K. The results received from Dr. L. Kauder can be found in Appendix E. The conductive material concepts tested are summarized here in Table 4-13.

| Table 4-13. Conductive Material Concepts for ESD Testing at GSFC | | | | | | |
|--|--|---------------------------------------|-----|--|--|--|
| Conductive Material TCMS | Pigment | Binder | PBR | | | |
| Black Conductive TCMS | | | | | | |
| DBG-IP | Doped Black, Glass | Kasil 2130 | 2.0 | | | |
| MH55ICP | Black Glass, Graphite 9035 | Kasil 2130 | 5.5 | | | |
| MH41SCB/LO | Black Glass, Graphite 9035 B ₄ C | MHS/LO stripped polydimethyl siloxane | 4.0 | | | |
| D36SCB/LO | WB-500 Cosmic Black graphite 9035 B ₄ C | MHS/LO stripped polydimethyl siloxane | 3.6 | | | |
| MH21SC/LO | Black Glass, Graphite fibers | MHS/LO stripped polydimethyl siloxane | 2.5 | | | |
| White Conductive TCMS | | | | | | |
| Z-93CXY:Z-93C55 | ZnO (FC) | DHS-2 | 5.5 | | | |
| Z-93SCXY:Z-93SC55 | S13GP | DHS-2 | 5.5 | | | |
| Z-93SCLMXY:Z-93SC1255 | DS13N | DHS-2 | 5.5 | | | |
| Flexible Conductive TCMS | | | | | | |
| DS13N/LO-51 | DS13N | MHS/LO stripped polydimethyl siloxane | 5.0 | | | |
| Other Conductive TCMS | | | | | | |
| DZS/SS-55 | DZS | SS-55 | 5.5 | | | |
| ZOT/SS-55 | ZOT | SS-55 | 5.5 | | | |
| Eu ₂ O ₃ /SS-55 | Eu ₂ O ₃ | SS-55 | 5.5 | | | |

The flexible conductive white sample charged to very high voltages (at 77K) in this study. The results of DS13N/LO-51 are hence, discouraging. The material design does not charge at room temperature. Further investigations are necessary for carrying out root cause analysis, and concurrently carrying out volumetric and surface resistance measurements on the conductive candidate material concepts. The results from GSFC for the ESD test on DS13N/LO-51 are given in Appendix E.

From review of the data from Appendix E, one can conclude that all conductive black coating concepts do not charge under irradiation of electrons for the studied energies and current at the selected temperatures which are representative of typical mission conditions. Table 4-14 summarizes the data. All conductive coatings concepts charged to maximum voltage at 77K and were conductive at room temperature showing charging below 20 volts.

The candidates concepts for the ESD charging testing studies were selected and sent for testing at GSFC without having knowledge of results for $\Delta\alpha_s$ from screening studies done at Aerospace Corporation. Because of this, we took a very conservative approach in selecting candidates for ESD testing. The concepts that are resistive, but have less dopant concentration, were expected to be more stable. Therefore, ESD testing was carried out only on those that were slightly resistive. In light of the screening study results and the charging study, we are very hopeful about the next phase of work. We recommend that more effort be devoted to ESD testing. The limited ESD testing in this program has shown that the conservative material designs also performed satisfactorily. We need to carry out testing on more conductive concepts that have shown remarkable stability to the space environment. Thus, much more remains to be done in this area for the identified promising candidates.

Table 4-14. Results of ESD Test at GSFC

DBG-IPDate Tested: 11/26/96
Charging Potential Developed on Sample (Volts)

| | Electron Flux | | | | | | |
|--------------------|---------------------|--------------------|----------------------|---------------------|---------------------|--|--|
| Temperature (K) | 5KeV @ 1.0nA/cm² | 5KeV @ 10nA/cm² | 10KeV @ 1.0nA/cm² | 10KeV @ 10nA/cm² | 20KeV @ 10nA/cm² | | |
| 130 | -17 | -39 | -66 | -71 | -87 | | |
| 179 | -20 | -32 | -63 | -70 | -80 | | |
| 227 | -25 | -30 | -51 | -64 | -74 | | |
| 273 | -10 | -16 | -16 | -28 | -17 | | |
| 320 | <u>-1</u> | -5.9 | -3.5 | -8.1 | -3 | | |

MH55IC Date Tested: 11/24/96 Charging Potential Developed on Sample (Volts)

| | Electron Flux | | | | | |
|--------------------|---------------------|--------------------|----------------------|---------------------|---------------------|--|
| Temperature (K) | 5KeV @ 1.0nA/cm² | 5KeV @ 10nA/cm² | 10KeV @ 1.0nA/cm² | 10KeV @ 10nA/cm² | 20KeV @ 10nA/cm² | |
| 129 | -33 | -35 | -71 | -68 | -68 | |
| 178 | -30 | -30 | -60 | -56 | -61 | |
| 226 | -22 | -31 | -60 | -60 | -60 | |
| 273 | -18 | -22 | -22 | -44 | | |
| 321 | -3 | -7.3 | -3.9 | -9.5 | | |

MH41SCB/LO1 Date Tested: 11/30/96 Charging Potential Developed on Sample (Volts)

| | Electron Flux | | | | | | |
|--------------------|---------------------|--------------------|----------------------|---------------------|---------------------|--|--|
| Temperature (K) | 5KeV @ 1.0nA/cm² | 5KeV @ 10nA/cm² | 10KeV @ 1.0nA/cm² | 10KeV @ 10nA/cm² | 20KeV @ 10nA/cm² | | |
| 128 | -10 | -12.6 | -55 | -60 | -62 | | |
| 173 | -13 | -13 | -62 | -60 | -61 | | |
| 224 | -16 | -16 | -61 | -63 | -63 | | |
| 268 | -17 | -17 | -65 | -62 | | | |
| 323 | -22 | -22 | -65 | -73 | -60 | | |

Table 4-14. Results of ESD Test at GSFC (Continued)

D36SCBB/LO
Date Tested: 11/20/96
Charging Potential Developed on Sample (Volts)

| | Electron Flux | | | | | | |
|--------------------|---------------------|--------------------|----------------------|---------------------|--|--|--|
| Temperature (K) | 5KeV @ 1.0nA/cm² | 5KeV @ 10nA/cm² | 10KeV @ 1.0nA/cm² | 10KeV @ 10nA/cm² | | | |
| 133 | -6.3 | -8.5 | -8.4 | -9.9 | | | |
| 178 | -6.1 | -6.2 | -6.5 | -7 | | | |
| 226 | -5.8 | -6.1 | -5.6 | -6.3 | | | |
| 275 | -4.1 | -4.9 | -3.9 | -4.8 | | | |
| 321 | -2.4 | -3.4 | -3 | -3.7_ | | | |

MH21SC/LO Date Tested: 10/28/96 Charging Potential Developed on Sample (Volts)

| | Electron Flux |
|--------------------|---------------------|
| Temperature (K) | 5KeV @ 1.0nA/cm² |
| 275 | -190 |
| 322 | -175 |

ZnO(CF)/DHS-2 Date Tested: 11/13/96 Non-contact Charging Potential Developed on Sample (Volts)

| | Electron Flux | | | | | | |
|--------------------|---------------------|--------------------|----------------------|---------------------|--|--|--|
| Temperature (K) | 5KeV @ 1.0nA/cm² | 5KeV @ 10nA/cm² | 10KeV @ 1.0nA/cm² | 10KeV @ 10nA/cm² | | | |
| 130 | -7 | -11 | -7 | -17 | | | |
| 179 | -8 | -15 | -9 | -14 | | | |
| 228 | -13 | -22 | -17 | -25 | | | |
| 275 | -9 | | -35 | -56 | | | |
| 322 | -3 | | -3 | -20 | | | |

Table 4-14. Results of ESD Test at GSFC (Continued)

S13GP/DHS-2
Date Tested: 10/23/96
Non-contact Charging Potential Developed on Sample (Volts)

| | Electron Flux | | | | |
|--------------------|---------------------|--------------------|----------------------|--|--|
| Temperature (K) | 5KeV @ 1.0nA/cm² | 5KeV @ 10nA/cm² | 10KeV @ 1.0nA/cm² | | |
| 128 | | | | | |
| 178 | -160 | | | | |
| 225 | -123 | | | | |
| 275 | -60 | -190 | -130 | | |
| 321 | -2.5 | -13 | -70 | | |

S13GP/DHS-2 Date Tested: 10/31/96 Non-contact Charging Potential Developed on Sample (Volts)

| _ | Electron Flux | | | | |
|--------------------|-----------------|-----------------|-------------------|------------------|-------------------|
| Temperature (K) | 5KeV @ 0.44n | 5KeV @ 4.4nA | 10KeV @ 0.44nA | 10KeV @ 4.4nA | 20KeV @ 0.44nA |
| 275 | -80 | -130 | | | |
| 323 | -45 | -56 | | | |

S13GP/DHS-2 Date Tested: 11/4/96 Non-contact Charging Potential Developed on Sample (Volts)

| | | <u></u> | | | |
|--------------------|---------------------|--------------------|----------------------|---------------------|---------------------|
| Temperature (K) | 5KeV @ 1.0nA/cm² | 5KeV @ 10nA/cm² | 10KeV @ 1.0nA/cm² | 10KeV @ 10nA/cm² | 20KeV @ 10nA/cm² |
| 129 | -60 | -180 | -120 | | -120 |
| 179 | -45 | -170 | -110 | | |
| 228 | -70 | -160 | -110 | | |
| 275 | -30 | -100 | -65 | -160 | -60 |
| 322 | -6 | -27 | -20 | -70 | -30 |

Table 4-14. Results of ESD Test at GSFC (Continued)

ZOT/SS-55 Date Tested: 1/7/97 Non-contact Charging Potential Developed on Sample (Volts)

| Temperature (K) | 5KeV @ 0.44nA | 10KeV @ 0.44nA | 10KeV @ 4.4nA | 20KeV @ 0.44nA |
|--------------------|------------------|-------------------|------------------|-------------------|
| 129 | -174 | | | |
| 179 | -160 | | | |
| 227 | -100 | -170 | | |
| 271 | -1.7 | -2 | | -100 |
| 319 | -0.5 | -0.6 | | 0 |

Eu₂0₃/SS-55 Date Tested: 11/19/96 Charging Potential Developed on Sample (Volts)

| | | | Electron Flux | | |
|--------------------|----------------------------------|---------------------|-----------------------|----------------------|---------------------|
| Temperature (K) | 5KeV @ 0.44nA/cm ² | 5KeV @ 4.4nA/cm² | 10KeV @ 0.44nA/cm² | 10KeV @ 4.4nA/cm² | 20KeV @ 10nA/cm² |
| 273 | -180 | | | | |
| 322 | -25 | | | | |

ZnO + Sb doped Zn₂SnOx in Kasil 2130 Date tested 12/2/96 Charging Potential Developed on Sample (Volts)

| | Electron Flux | | | | | | |
|--------------------|---------------------------|---------------------|-----------------------|----------------------|-----------------------------------|--|--|
| Temperature (K) | Flux 5KeV @ 0.44nA/cm² | 5KeV @ 4.4nA/cm² | 10KeV @ 0.44nA/cm² | 10KeV @ 4.4nA/cm² | 20KeV @ 0.44nA/cm ² | | |
| 129 | -77 | -134 | -143 | | -140 | | |
| 179 | -73 | -110 | -120 | | -163 | | |
| 227 | -66 | -105 | -100 | | -137 | | |
| 275 | -30 | -120 | -60 | -160 | -70 | | |
| 323 | -14 | -19 | -34 | -36 | -48 | | |

Table 4-14. Results of ESD Test at GSFC (Concluded)

DS13N/LO-41
Date tested 12/4/96
Charging Potential Developed on Sample (Volts)

| | | | Electron Flux | | |
|--------------------|---------------------------------------|---------------------|-----------------------------------|----------------------|-----------------------|
| Temperature (K) | Flux 5KeV @ 0.44nA/cm ² | 5KeV @ 4.4nA/cm² | 10KeV @ 0.44nA/cm ² | 10KeV @ 4.4nA/cm² | 20KeV @ 0.44nA/cm² |
| 274 | | | | | |
| 320 | -20 | -76 | -41 | -145 | -40 |

In the ESD tests carried out, the charging observed for MH21SC/LO and DS13N/LO-41 was not expected. These concepts need further investigation. We expect better ESD performance from these formulations based on their surface electrical resistances. ESD testing carried out for this program was very limited and could not provide information on some more conductive concepts that have shown excellent space environment stability. Thus, it may be premature to carry out data analysis and provide conclusions just based on these tests. Tests are expensive and time consuming and need to be carried out on the full spectrum of material design concepts, so that these results can be used for future material selection • Our approach to tailor steady-state electrical surface resistance at room temperature may also need to be modified for material design. The approach of resistance tailoring may need to consider RF properties in the material design issue. Thus, the material design attempt made for the screening study was limited and has not been completely tested for ESD performance.

4.5 DISCUSSION OF RESULTS

In this section, we present observations, comparison of data, and qualitative comments as applied to the thermal control material system design. Each material design concept was addressed attention in this discussion. For testing and related issues detailed comments were made by each collaborating researcher in their report. One can refer to their reports in the appendices of this final reports. (Appendices A through E).

4.5.1 High Absorptivity Conductive TCMS

The comparison of baseline and conductive concepts reveals that all of the conductive concepts behaved satisfactorily showing excellent stability to the space environment, and satisfactorily mitigating charging for the tested conditions. Table 4-15 provides summary of the typical performance of the promising conductive candidates that are inorganic and flexible with low outgassing siloxane polymer as a vehicle. The thermal designers can use them on spare hardware. The following qualitative comments can be made for the evaluated concepts:

- Candidate conductive thermal control materials systems behaved very similar to the baseline materials and are recommended as space environment stable in low earth orbit (LEO) type exposures studied in this screening effort.
- Doping of black glass provides the ability to tailor surface resistance values without compromising α_s or ϵ_N of the TCMS. This concept should be studied further with emphasis on dopant cost issues and processing issues. This provides material designers additional flexibility to tailor surface resistances along with BRDF.
- If component design and location of hardware on the spacecraft demands that the conductive TCMS should need to depend on tailoring percolating leaky paths only. Then, the use of semi-conducting grade B₄C is a viable option as a conductive phase. It provides the ability to provide percolating path along with additional ability to control RF losses.
- AO resistance studies are recommended on all of the electrically conductive TCMS candidates, with emphasis on cost reducing processing steps. The role of processing steps in AO resistance of TCMS needs to receive attention.

| Table 4-15. Typical Performance of Promising Conductive Absorber Optical Coatings in Low Earth Orbit (LEO) | | | | | |
|--|-------------------------------|--------------------------|----------------------|-----------------------|--|
| Material | Pre-exposure $\alpha_{\rm S}$ | Post-exposure α_s | $\Delta lpha_{ m S}$ | Tested By | |
| Inorganic-Condu | ıctive | | | | |
| MH55-IC | 0.940 | 0.938 | -0.001 | Aerospace Corporation | |
| MH55-IC | 0.942 | 0.941 | -0.001 | Aerospace Corporation | |
| MH55-IC | 0.928 | 0.938 | 0.010 | GSFC | |
| DBG-IP | 0.974 | 0.975 | 0.001 | Aerospace Corporation | |
| DBG-IP | 0.973 | 0.974 | 0.001 | Aerospace Corporation | |
| DBG-IP | 0.957 | 0.951 | -0.006 | GSFC | |
| DBG-IP | 0.970 | 0.978 | 0.008 | AFML/SCEPTRE | |
| DBG-DHS | 0.958 | 0.957 | -0.001 | Aerospace Corporation | |
| DBG-DHS | 0.956 | 0.956 | 0 | Aerospace Corporation | |
| Organic-Flexible | -Conductive | | | | |
| D21S/LO | 0.980 | 0.980 | 0 | Aerospace Corporation | |
| D21SC/LO | 0.956 | 0.957 | 0.001 | Aerospace Corporation | |
| D21SC/LO | 0.955 | 0.956 | 0.001 | Aerospace Corporation | |
| MH21SC/LO | 0.963 | 0.964 | 0.001 | Aerospace Corporation | |
| MH21SC/LO | 0.965 | 0.965 | 0 | Aerospace Corporation | |
| MH41SCB/LO | 0.951 | 0.950 | -0.001 | Aerospace Corporation | |
| D36SCB/LO | 0.962 | 0.967 | 0.005 | GSFC | |
| D36SCBLO | 0.966 | 0.956 | -0.010 | AFML/SCEPTRE | |
| MH41SCB/LO | 0.939 | 0.947 | 0.008 | GSFC | |
| MH21SC/LO | 0.951 | 0.955 | 0.004 | GSFC | |

All tests at GSFC involved exposure to 1000 ESH of UV and Vacuum.

All tests at AFML/SCEPTRE involved exposure at 2.0 Suns of EUV and 6.78 x 10^{15} e'/cm² (1 KeV) + 3.39 x 10^{15} e'/cm² (10 KeV) and Vacuum.

All tests carried out at Aerospace Corporation involved exposure to 2000 ESH of UV + $8.26 \times 10^{13} \text{ e}'/\text{cm}^2$ [35 KeV] (typical DMSP orbit) and Vacuum.

• The ESD performance for MH21SC/LO does not correlate with the low steadystate surface resistance value of the sample. The material design (MH21SC/LO) should receive more attention during the next phase to verify this. The comparable surface resistance formulations of other concepts show no charging, hence the sample may be suspect. No surface or sample analysis was carried out to identify root cause of this charging behavior.

4.5.2 Low (α_s/ϵ_N) TCMS

In the area of white thermal control material systems, this screening study has provided significant input in two areas. First, it has provided new insight on baseline nonconducting materials by testing them one more time after recent reformulation study (AFML-TR-94-4126), and secondly, it has evaluated proposed conductive material design concepts to provide guidance to electro-ceramists on designing required surface resistance values in the TCMS.

First, we discuss the nonconducting baseline material systems and their degradation in space environment. The following observations are important.

- (1) S13GP/LO-1 and S13GP/LO-41 showed excellent stability in studies carried out at Aerospace Corp and GSFC approaching very close to performance of Z-93P.
- (2) Zn₂TiO₄ (ZOT) containing thermal control material systems showed excellent space environment stability. YB-71P turns out to be slightly more stable than YB-71. This is an encouraging result.
- showed an interesting phenomena. They show a reverse correlation of degradation with solar exposure. That is, the samples exposed in intense area of beam showed lesser degradation than their counterparts placed in weaker area of beam in the Aerospace Corporation study. The differences in Δα_s are significant. This phenomena needs to receive further attention. If this is true, it has more importance from a material selection point of view for orbits that have significant electron fluxes (e.g., GEO, polar). The other implication of this finding may relate to ground simulation testing. If electron exposure causes degradation and UV acts as a healing or restoring means for reflectance, then during accelerated test one needs to keep ratio of one radiation flux to other as a constant for

simulating similar synergism for a given orbit. If exposures are to be limited to 2 suns to avoid contributions from thermal effects in degradation of reflectance, then from the similarity principles in acceleration and to achieve correct synergism for a given orbit, there needs to be a limit on e' or particulate radiation acceleration. The same opinion has been expressed by Zerlaut and Gilligan⁽²²⁾, Zerlaut et al.⁽³¹⁾ in the past while discussing the combined radiation exposures study for ZOT type materials using UV and protons. Their study concluded that radiation flux ratios and rates are important.

- (4) Among the baseline nonconducting TCMS concepts, two pigments were studied that can provide beginning of life solar absorptance less than 0.10. The Eu₂O₃ was promising. Although Δα_S for Eu₂O₃ were greater than that of Z-93P, one can observe that α_S after [2000 ESH UV + 5.18 x 10³ e'/cm² (35 KeV) + 8.26 x 10¹³ e'/cm² (100 KeV] exposure was 0.12 in the Aerospace Corporation study, and 0.16 in AFML's in situ degradation study. The α_S for Z-93P for the same exposure was also 0.16. Thus, future study should include baseline Eu₂O₃/2130 nonconducting material along with its conducting versions to explore if Eu₂O₃ can provide this performance consistently. The Eu₂O₃ may also provide additional flexibility of processing for thermal spraying or plasma spraying of the pigment. Eu₂O₃ is not semiconducting in nature and, hence, should be more conducive to thermal or plasma spraying via microencapsulation.
- (5) DZS high temperature route doped (Zn₂SiO₄) based nonconducting baseline material system processed in Kasil 2130 showed extraordinary degradation, thus ruling out its use in any future study.

Among electrically conducting thermal control material systems designs, there were several encouraging observations, as well as a few that are difficult to rationalize. The root causes can either be in material concept or in the selection of the parameters for processing. Table 4-16 is presented here to provide insight on space environment stability. The following qualitative remarks can help us to appropriately assess the material design issues.

| Table 4-16. Typical Performance of Promising Conductive White Thermal Control Coatings in Low Earth Orbit (LEO) | | | | | |
|---|---|--|--|--|--|
| TCMS | Pre-exposure α_s | Post-exposure $\alpha_{\rm S}$ | $\Delta lpha_{ m S}$ | Tested By | |
| Concept: Fla | sh Calcination of Zno h doped hybrid, silicat | O (SP-500) to retain content by the binder. | ontrolled Zn intersti | itials, and stabilization | |
| ZnO(FC)/DHS-2 | 0.139 | 0.141 | 0.002 | Aerospace Corporation | |
| ZnO(FC)/DHS-2 | 0.146 | 0.146 | 0.000 | Aerospace Corporation | |
| ZnO(FC)/DHS-2 | 0.168 | 0.168 | 0.010 | GSFC* | |
| ZnO(FC)/DHS-2 | 0.143 | 0.225 | 0.082 | AFML/SCEPTRE(1) | |
| Concept: 7-0 | 93SC55 - Stabilization orid silicate and doped | of Zn interstitials via r hybrid silicate binders | nicroencapsulation | and incorporation in | |
| S13GP/DHS-1 | 0.147 | 0.153 | 0.006 | Aerospace Corporation | |
| S13GP/DHS-1 | 0.142 | 0.148 | 0.006 | Aerospace Corporation | |
| S13GP/SS-55 | 0.125 | 0.140 | 0.015 | Aerospace Corporation | |
| S13GP/SS-55 | 0.127 | 0.137 | 0.010 | Aerospace Corporation | |
| S13GP/SS-55 | 0.140 | 0.177 | 0.037 | GSFC** | |
| S13GP/SS-55 | 0.151 | 0.206 | 0.055 | GSFC** | |
| S13N/DHS-2 | 0.13 | 0.196 | 0.066 | AFML/SCEPTRE(1) | |
| Concept: Z-hy | 93SCLMXY - Stabiliz brid silicate binders ander. | ation of Zn interstitials d incorporation in hybr | via microencapsula id silicate binder o | ation using doped r doped hybrid silicate | |
| DS13N/SS-55 | 0.136 | 0.173 | 0.037 | Aerospace Corporation | |
| DS13N/SS-55 | 0.137 | 0.153 | 0.016 | Aerospace Corporation | |
| DS13N/DHS-2 | 0.124 | 0.161 | 0.037 | Aerospace Corporation | |
| DS13N/DHS-2 | 0.122 | 0.128 | 0.006 | Aerospace Corporation | |
| DS13N/DHS-2 | 0.148 | 0.154 | 0.006 | GSFC* | |
| DS13N/DHS-2 | 0.12 | 0.24 | 0.120 | AFML/SCEPTRE(2) | |
| DS13N/SS-55 | 0.16 | 0.27 | 0.11 | AFML/SCEPTRE(3) | |

| Table 4-16 | 5. Typical Performand in Low E | ce of Promising Condu | | ermal Coatings |
|--|-----------------------------------|--------------------------|----------------------|--------------------------|
| TCMS | Pre-exposure α_s | Post-exposure α_s | $\Delta lpha_{ m S}$ | Tested By |
| Concept: Fl | exible Conductive Wh | ite TCC with DS13N | in stripped silicon | ie |
| DS13N/LO-41 | 0.161 | 0.278 | 0.117 | Aerospace Corporation |
| DS13N/LO-41 | 0.163 | 0.239 | 0.076 | Aerospace Corporation |
| DS13N/LO-41 | 0.210 | 0.430 | 0.22 | GSFC* |
| DS13N/LO-51 | 0.193 | 0.423 | 0.230 | AFML/SCEPTRE(3) |
| Conductive Concer | ots with Other Candid | ate Pigments | | |
| ZOT/Phos. Sol. | 0.131 | 0.149 | 0.018 | Aerospace Corporation |
| ZOT/Phos. Sol. | 0.128 | 0.151 | 0.023 | Aerospace Corporation |
| ZOT/SS-55 | 0.117 | 0.156 | 0.039 | Aerospace Corporation |
| ZOT/SS-55 | 0.112 | 0.156 | 0.044 | Aerospace Corporation |
| ZOT/SS-55 | 0.129 | 0.180 | 0.051 | Aerospace Corporation |
| Eu ₂ O ₃ */SS-55 | 0.087 | 0.138 | 0.051 | Aerospace Corporation |
| Eu ₂ O ₃ */SS-55 | 0.100 | 0.126 | 0.028 | Aerospace Corporation |
| Eu ₂ O ₃ */DHS-2 | 0.093 | 0.158 | 0.065 | Aerospace Corporation |
| Eu ₂ O ₃ */DHS-2 | 0.076 | 0.108 | 0.032 | Aerospace Corporation |
| Eu ₂ O ₃ */SS-55 | 0.096 | 0.129 | 0.033 | GSFC* |
| | | | | _ |

^{*}GSFC test involved exposure to 1000 ESH of UV-Vacuum.
**This GSFC test involves exposure to 4300 ESH of UV-Vacuum.

0.076

Eu₂O₃*/SS-55

All tests carried out at Aerospace Corporation involved exposure to 2000 ESH of UV + 8.26×10^{13} e'/cm² (100 KeV) + 5.18×10^{13} e'/cm² (35 KeV) (typical DMSP orbit) and Vacuum.

0.161

0.085

- (1) Test at AFML/SCEPTRE involved exposure at 1 Sun of EUV-Vacuum (for Z-93CXY and Z-93SCXY) + 6.78 x 10¹⁵ e'/cm² (1 KeV) + 3.39 x 10¹⁵ e'/cm² (10 KeV).
- (2) Test at AFML/SCEPTRE involved exposure at 3.3 to 3.6 Suns of EUV-Vacuum (for DS13N/DHS-2 and Eu₂O₃*/SS-55) + 6.78 x 10^{15} e'/cm² (1 KeV) + 3.39 x 10^{15} e'/cm² (10 KeV).
- (3) The in situ test at AFML/SCEPTRE involved exposure at 1.6 to 2.1 Suns of EUV-Vacuum + $6.78 \times 10^{15} \text{ e}'/\text{cm}^2$ (1 KeV) + $3.39 \times 10^{15} \text{ e}'/\text{cm}^2$ (10 KeV).

AFML/SCEPTRE(2)

The first concept suggested for the screening study was of flash calcination of ZnO (SP-500) to retain Zn interstitials in grain so as to tailor grain resistivity, and hence, the resultant surface resistance. The material resistivity tailoring exercise showed that the concept met the program goals with all candidate binder systems (i.e., SS-55, DHS-1, DHS-2). The conductive concept Z-93C55: ZnO(FC)/DHS-2, and Z-93(FC)/DHS-1 showed best reproducibility in surface resistance values along with compatibility with ambient storage conditions. Studies at Aerospace Corporation, GSFC, and AFML/SCEPTRE show that both concept have acceptable stability to the space environment, along with acceptable ESD performance. Concept ZnO(FC)/DHS-1 and ZnO(FC)/DHS-2 both should receive attention for further scale up, validation and reliability studies.

The second concept suggested was to microencapsulate ZnO(SP-500) following well know S13G processing steps and to incorporate them in candidate silicate binder systems. GSFC studied this for up to 4300 ESH exposure of UV-Vacuum. The study at Aerospace Corporation (Appendix A), and the study at AFML/SCEPTRE (Appendix D) show that the concept has the required stability to the space environment. The ESD behavior studied shows promising characteristics along with the required RF properties. The concepts Z93SC55:S13GP/DHS-1 and S13GP/SS-55 are suggested for further studies of scaleup, validation and reliability, since they show excellent stability in resistance values for ambient dry storage conditions.

The third concept was stabilization of Zn particles via microencapsulation using doped silicate binder and incorporation of pigment in doped hybrid silicate binders. The screening studies have showed excellent reproducibility for the electrical resistivity values and surface resistance values. The ambient storage had no effect on the observed surface resistance values. The degradation observed for this system at Aerospace Corporation, GSFC, and AFML provide guidance in selecting candidates for the future study of scaleup, validation, and reliability. The pigment/binder couples DS13N/SS-55, DS13N/DHS-1, and DS13N/DHS-2 are excellent candidate for these further studies. The high temperature doping should not receive any further consideration.

The flexible conductive concept showed an acceptable space environment stability for its use, but it can also be improved. The stability may need to be revisited again, because the formulation showed conflicting results. It charged beyond anticipated design values at cold

temperatures but showed acceptable RF loss behavior at room temperature. These results warrant further investigations. The processing of this coating can be a very tricky art. Primer chemistry poses situations that are not process friendly and interfere with the reproducible surface resistance values. The deposition process specifications development should go hand in hand with future "validation and reliability" analysis studies. The result demands that the program should devote the efforts to design a cured resin system that can act as a "leaky dielectric", and preferably with leakage paths. Such a program should be devoted to chemistry manipulations to achieve this goal of designing a "leaky flexible dielectric". It will fill a material technology gap and have considerable impact on space hardware design.

Among other concepts studied, no future efforts should consider tantalum oxide (Ta₂O₅). The pigment showed unacceptable degradation. The Eu₂O₃ based conductive concepts have acceptable electrical properties, but their space environment induced degradation behavior showed mixed results. One needs to keep this material in future studies to access Eu₂O₃ material technology maturity, only because of the flexibility that this pigment brings to electro-ceramist for tailoring surface resistances, as well as for processing options like plasma spray or thermal spray.

The Zn₂Ti0₄ (ZOT) pigmented conductive concept showed great promise with acceptable behavior. The ZOT pigmented materials have been traditionally used in GEO successfully. Their use on long life missions has already been proven (e.g., missions like DSP). Our attempt to measure resistivity of exposed YB-71 or YB-71P revealed that the after exposure surface resistance of ZOT pigmented thermal control material system are ideal for the leakage current requirement [earlier communication with by Dr. Purvis⁽¹⁹⁾] for typical space hardware. In view of this background, the pigment/binder couples: ZOT/Phos sol, ZOT/SS-55, and ZOT/DHS-2 results are promising for tailoring surface resistance and their stability. The surface resistance manipulation of defected ZOT may also provide additional freedom in material design to electroceramists. Based on the results of the screening study, one must revisit solid-state chemistry, surface chemistry, and synthesis of titanate to tailor and stabilize defect structures that are hardened. Thus, chemical stabilization of ZOT may provide the sought "very long life" for reliable conductive TCMS.

Lastly, antimony doping efforts may provide cheaper and stable system that can mitigate charging phenomena, but higher $\Delta\alpha_s$ are to be expected for such materials system than the ZnO based or the ZOT based conductive TCMS. No further effort for such a material design is recommended. Future efforts are worthwhile only if the thermal designers are ready to compromise $\Delta\alpha_s$ for electrical performance at less cost. The Sb-doping precursor chemistry does pose market vagarities and inconsistent lot-to-lot variations. Future SEE program funding will be needed to address "source" related issues for such doping chemistry, if there is interest in Sb doping of TCMS pigments by thermal designers.

5. SUMMARY AND CONCLUDING REMARKS

The objective of this program was to address the material needs for two classes of materials: (1) black thermal control coating, where property goals are $\alpha_S \ge 0.90$, $\epsilon_N \approx 0.90$, with space environment stability for long term use; and (2) low (α_S/ϵ_N) white thermal control coatings, where property goals are $\alpha_S \sim$ minimum as possible (≈ 0.18 , i.e., equivalent to Z-93); $\epsilon_N \approx 0.90$ with space environment stability for long term use. The resistivity goal was 10^4 - 10^9 Ω/\Box tailorable. Our screening study on suggested various material designs concept and the findings on them have resulted in the following important conclusions.

5.1 HIGH ABSORPTIVITY CONDUCTIVE TCMS

All electrically conductive TCMS designs proposed for high absorptivity applications met the program goals. New In-doped Black Glass material technology was developed along with three doped silicate binder systems. They provide a much needed flexibility in tailoring electrical performance characteristics without penalties in optical performance. These material technologies are recommended for further scaleup and validation, so that the following conductive material concepts can be offered as products in a cost-effective economic manner. The conductive concepts recommended for scaleup are: DBG-IP, DBG/DHS, MH55ICP, MH21SC/LO, MH41SCB/LO, D21S/LO, D21SC/LO, and D36SCB/LO. They all meet the above program goal, and showed acceptable ESD performance. (See Table 4-15)

5.2 LOW (α_s/ϵ_n) WHITE CONDUCTIVE TCMS

The white, low (α_s/ϵ_N) TCMS designs studied in this screening program fall into two categories. One is a baseline nonconducting TCMS design, and the other, the electrically conducting TCMS design. The following important conclusions can be drawn for each individual category found in Table 4-16.

5.2.1 Nonconducting Baseline TCMS

- (1) The reformatted versions of baseline white TCMS were evaluated again. The S13GP/LO-1 and S13GP/LO-41 showed space environment stability approaching that of Z-93P.
- YB-71P showed more or less equal stability to that of YB-71. The Kasil 2130 is now recommended as a replacement in this formulation.
- (3) Zn₂TiO₄ (ZOT) pigmented coating showed reversal in degradation with UV intensity; this should be investigated further, and may have consequences in material selection for thermal control for orbits like GEO.
- (4) Eu₂O₃ showed promise as a simple BCC pigment. This can provide flexibility in material design for using doped binders. This material should receive more attention in future studies, to verify some conflicting data obtained in this screening study.
- (5) Surface resistance values of YB-71 and YB-71P TCMS after irradiation are: 5.0E+07 to 8.0E+09 Ω/\Box .

5.2.2 Electrically Conductive TCMS (α_s/ϵ_N)

The following material technologies were developed as IR&D at IITRI to meet the goals of the screening program:

- (1) Material technologies for new binder systems SS-55 and doped binders (DHS-1 and DHS-2) are now available. They have played a significant role in providing flexibility in tailoring surface resistances in the candidate TCMS along with no interference in optical performance of TCMS.
- (2) Material and process technology for doped microencapsulation has been developed to provide new pigment DS13N. This provides a pigment that is essentially a ZnO grain coated with doped silica dielectric layer, that provides hardening for the space environment irradiation, along with freedom to tailor dopant levels and dielectric behavior of coating over the grain.

The above mentioned two material technologies need attention to mature to the required reproducibility and reliability. They are excellent material candidates for scaleup and validation efforts.

Along with these material technologies, the following are important findings on individual conductive material design concepts.

- (1) The material concept Z-93CXY calls for the flash calcination of ZnO (SP-500) and use of doped binders. This provides an excellent approach to tailor desired surface resistance values in white TCMS. This material design shows acceptable stability to the space environment along with the required ESD properties. This may be the most amenable to scaleup material design concept, and is recommended for future validation and reliability studies. The pigment/binder couples to be studied are: ZnO(FC)/DHS-1 and ZnO(FC)/DHS-2. Future studies should also include characterization of the RF properties for these couples.
- (2) The material concept Z-93SCXY calls for microencapsulated ZnO (SP-500) as a pigment and use of doped binders to tailor resistance values for white TCMS. Although this material design limits tailorability of electrical performance, the concept provides much needed RF properties along with ESD performance. The scaleup of this material concept is recommended. The concept needs to be considered in the future validation and reliability studies. The pigment/binder couples to be studied are: S13GP/DHS, S13GP/DHS-1, and S13GP/DHS-2.
- (3) The material concept Z-93SCLMXY call for microencapsulation of ZnO (SP-500) with doped silicate binder as pigment and use of doped binders to tailor desired resistance values for white conductive TCMS. Tailorability of resistance values is excellent. The concept provides acceptable ESD performance. RF properties need attention during the next phase of study for this concept. The concept needs to be studied in the future scaleup, validation and reliability studies. The pigment/binder couples to be studied are DS13N/SS-55, DS13N/DHS-1, and DS13N/DHS-2.
- (4) ZOT-based conductive TCMS should receive attention based on their performance in the screening studies. The solid-state chemistry of defect structures in ZOT needs attention in processing to improve on our ability to control them, so that tailorability of electrical properties become appealing. Such an understanding of ZOT can provide best, long life electrically conductive TCMS. The pigment/binder couples that should receive attention in scaleup, validation and reliability studies are: ZOT/Phos Sol, ZOT/DHS-2 and chemical stabilization of

ZOT surface defect structures.

- Eu₂O₃-based conductive material system provides the ability to use high dopant levels in a binder system. The results of the screening system indicate that Eu₂O₃ is the only non-semiconducting pigment that provides such a ability. Its performance is conflicting, and it remains to be seen if it can mature into an engineering quality material system. The use of Eu₂O₃ provides a different material design regime to electro-ceramists for tailoring properties, especially increase in dopant levels without worrying about its impact on band gap and a resultant impact on material properties. Hence, it is recommended for future scaleup, validation and reliability studies. The pigment/binder couples Eu₂O₃/DHS-1, Eu₂O₃/DHS-2, Eu₂O₃/SS-55 along with Eu₂O₃/2130 as baseline material design should be added to such studies.
- behavior. The TCMS still shows charging behavior at low temperatures. The future developments are now identified for implementation in the next phase of the program. The results demand that the program should devote efforts to design a cured resin system that can act as a "leaky dielectric", preferably with leakage paths. Such a program should be devoted to chemistry manipulations to achieve the goal of designing a "leaky flexible dielectric". It will fill a material technology gap for conductive flexibles and have a considerable impact on space hardware design.

6. RECOMMENDATIONS FOR FUTURE WORK

The work accomplished during this research program was to study thermal control design concepts to tailor resistivity. We have only finished the screening study on the selected conductivity tailoring concepts for two classes of TCMS (black and white). We have identified material design concepts and approaches that promise success or acceptable performance. This work can be transferred to the aerospace community only through a dedicated effort that can help to generate "reliability" and "long life" into the identified concepts along with a systematic approach and commitment to integrating these types of coatings on space hardware and space structures.

For the material design studies carried out here, overall 18 different material designs (pigment/binder couples) have been identified as very promising material concepts for scaleup, validation and reliability studies. [See section 4.5 and Chapter 5 for details] The pigment/binder couples for future effort are discussed in an earlier section along with promising materials and processes. Some of them need attention as a scaleup study to find economic processing routes that can provide reliability. The following materials related technology needs attention for such a scaleup and reliability study.

- Doped hybrid silicate binder technology/large batches
- Black Glass and doped Black Glass economic routes
- ZOT Reproducibility issues and reliability, including but not limited to chemical stabilization of defect structures that promise "long life" along with electrical performance
- Eu₂O₃ Bench scale manufacturing and reliability
- Microencapsulation via doped hybrid silicates
- Flash calcination facility development for "large batches"
- Laboratory synthesis capability for producing alkali silicates and/or polydimethyl siloxane polymers.

The testing community also needs to address the needs of test method development. The screening study identifies the following two needs for future development work.

- Laboratory test methods for electrical measurements that co-relates with ESD type test carried out at GSFC. Such methods need to mature for <u>material certification needs</u> (i.e., very cost effective and economical), and not as a research tool, or validation tool.
- Space environment simulation testing requirements of electrically conductive material systems. The role of conduction mechanisms in space environment induced degradation and synergism, and limits posed by them on effective feasible acceleration rates in the laboratory setup.

The above mentioned items are examples of some of the very broad areas that need to be addressed for the new electrically conductive TCMS to mature fully. At IITRI, we are also involved in integrating appealing material concepts on space hardware for the aerospace industry. It may be worthwhile to present some of the ideas on the use of these conductive thermal control material designs on space hardware, for "reliability" and "long life". In Figure 6-1, we have presented a synthesis of our ideas as a block diagram. Although the figure is fairly self-explanatory, the following discussion may provide insight into the system approach necessary to carry out systematic concurrent engineering exercise. The discussion and block diagram may also apply to all types of thermal control material technologies. Such a concurrent approach will help industry to build long term reliable space hardware and space structures.

The figure shows a dotted oblong-shaped zone called "Effects Zone" of the material system. We have listed here three important effects blocks for conductive material systems. The existence of effects depends on the material technology in question. For example, EL effects may not play a role for all material designs. Thus, one needs to be prudent in interpreting and using this flow diagram. The effect zone is fed by several legs which have several elements on them. Each element has either material technology name or testing technology name or system engineering issue listed in them that is relevant. Each leg is also identified with flow of information related arrows. A two-way flow usually provides an opportunity for concurrent engineering exercise which is needed for cost-effective, reliable integration, and design.

From this flow diagram one can easily see that we have just started to generate information that is required for such a reliable integration in this screening study. The information available for traditional nonconducting TCMS is also fragmentary. Efforts would be needed to generate information for each of these blocks, and industry/government may need

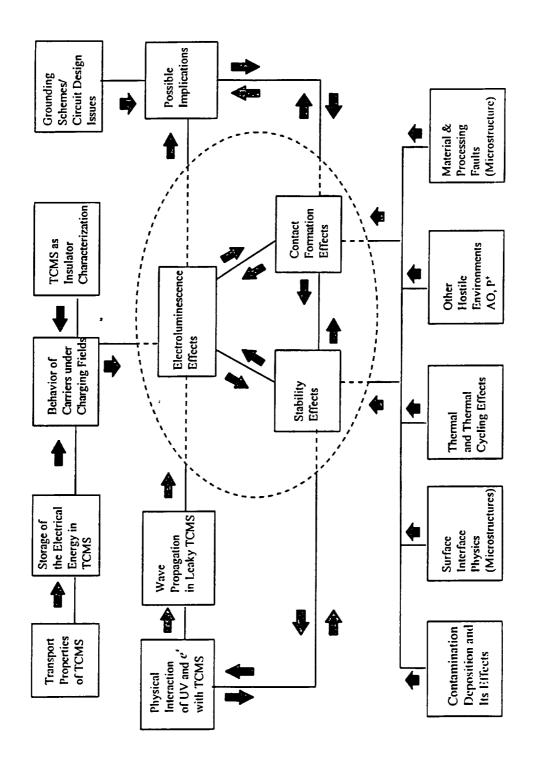


Figure 6-1. Synthesis of Ideas for the Use of Electrically Conductive TCMS on Space Hardware for Reliability and Long Life.

to cooperate and emphasize sharing of analyzed information for "reliability". SEE Project Office can play a significant role towards these efforts.

In view of the suggested philosophy in the block diagram (Figure 6-1), the following are some of the areas that need to receive attention in future studies of the material design concepts identified here:

- AO resistance testing of conductive TCMS, and impact of AO on dopant chemistry
- UV-VUV-protons-vacuum testing
- Material processing/microstructural features and their role in tailoring electrical properties
- Role of surfaces and contact formations for each applicable material design (formulation) concept
- RF properties of applicable TCMS concepts
- Synergism in UV and e' as applied to semiconducting pigments and acceleration issues in laboratory simulation of space environment
- Thermal cycling effects/life with use of doped hybrid silicate binders
- Behavior of TCMS under charging fields/EL effects
- Electroluminescence of TCMS/if applicable
- Complete characterization of electrical behavior of TCMS [including response to pulses and secondary emission properties as needed in NASCAP charging codes]
- Role of contamination on functional aspects of the electrically conductive TCMS
- Lastly, survivability analysis of conductive TCMS for defense mission needs.

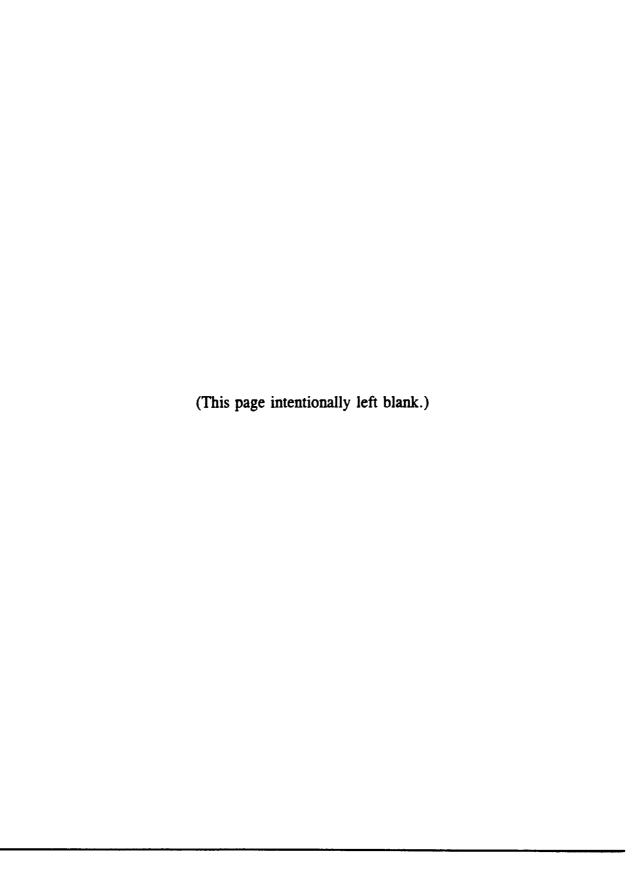
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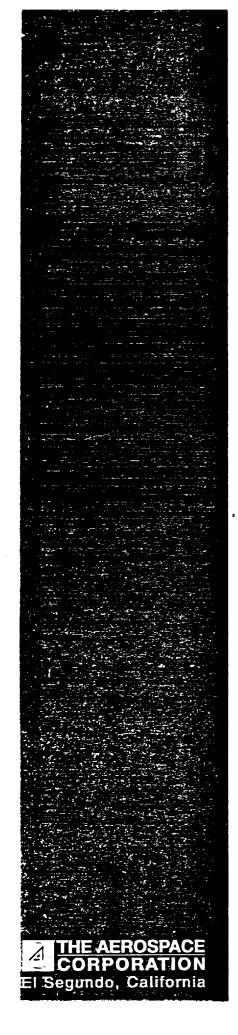


APPENDIX A

SPACE ENVIRONMENTAL EXPOSURE OF IITRI TAILORABLE ELECTRICALLY CONDUCTIVE THERMAL CONTROL MATERIAL SYSTEMS

FINAL REPORT

THE AEROSPACE CORPORATION



Space Environmental Exposure of IITRI Tailorable Electrically Conductive Thermal Control Material Systems

Final Report

15 September 1996

Prepared by

M. MESHISHNEK
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Technology Operations

Prepared for

IIT RESEARCH INSTITUTE Chicago, IL 60616

Contract No. C06804-S001-00

Engineering and Technology Group

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TECHNOLOGY OPERATIONS

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SPACE ENVIRONMENTAL EXPOSURE OF IITRI TAILORABLE ELECTRICALLY CONDUCTIVE THERMAL CONTROL MATERIAL SYSTEMS

FINAL REPORT

Prepared by

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15 September 1996

Engineering and Technology Group THE AEROSPACE CORPORATION El Segundo, CA 90245-4691

Prepared for

IIT RESEARCH INSTITUTE Chicago, IL 60616

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SPACE ENVIRONMENTAL EXPOSURE OF ITTRI TAILORABLE ELECTRICALLY CONDUCTIVE THERMAL CONTROL MATERIAL SYSTEMS

Prepared

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Abstract

This report documents the results and data obtained from an extended space simulation test of 85 electrically conductive thermal control paints. This work was performed under a contract with the Illinois Institute of Technology Research Institute (IITRI) and was part of the effort funded by the National Aeronautics and Space Administration/Marshall Space Flight Center (NASA/MSFC) for IITRI to develop new advanced coatings for spacecraft. Part of the IITRI development process involved a simulated exposure of these coatings to ultraviolet (UV) radiation in vacuum. The test was designed to simulate an exposure of a material to low Earth orbit (LEO) conditions. Additionally, since many satellite systems are in polar orbits, a representative dose of electrons was also included in the test. The parameter used to measure the coating performance was the solar absorptance. This parameter was calculated from reflectance spectra measured on each sample pre- and posttest. The results of this test can be used to rank the performance of these materials and their stability to space exposure. While there were clear differences in the response of the various materials to this UV/electron exposure, most, if not all, materials performed well, showing little degradation in reflectance. A control specimen of Chemglaze A-267 polyurethane white paint degraded to a value expected from flight and previous ground test UV data. Contamination monitors included in the test matrix also indicated no appreciable changes due to contamination-radiation effects.

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1. Introduction

In 1994, the Illinois Institute of Technology Research Institute (IITRI) was selected by the National Aeronautics and Space Administration (NASA), under the Space Environment Effects (SEE) program, for a research award to develop new electrically conductive thermal control materials. Part of the award included funds for space environmental exposure testing of the newly developed materials. The Aerospace Corporation, because of its unique large-area exposure capability, was selected by IITRI to perform this testing under a subcontract. This unique capability allows the exposure of up to roughly 90 1 in. specimens in one test. This work was performed in July 1996.

2. Background

The initial purpose of the space environmental test was to evaluate the performance of new advanced electrically conductive thermal control paints, both white and black, under simulated ultraviolet (UV) radiation, typical of a low Earth orbit (LEO) exposure. The performance of the paints was measured against baseline industry standard nonconductive thermal control coatings developed by IITRI. These coatings have been used on several NASA, Department of Defense (DOD), and commercial spacecraft over the last 30 years and continue to be used today. Aerospace suggested that, in addition to the UV exposure, a small dose of electrons, typical of that encountered in a LEO polar orbit commonly used by weather satellites, be included to obtain more information regarding the space radiation stability of the new coatings. The IITRI contract called for a minimum of 500 h exposure of the coatings to an approximate 2 sun level of UV radiation.

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3. Description

The candidate developmental coatings tested in the project consisted of 10 baseline nonconductive white thermal control paints and 35 new white thermal control paints exhibiting varying electrical conductivity. One sample of each of the baseline materials was tested, along with two samples of each of the new coatings. Three baseline nonconductive black coatings were submitted for test, along with six new black thermal control coatings exhibiting varying levels of electrical conductivity. Again, one sample of each of the baseline black materials was tested, along with two samples of each of the new black coatings. In addition, two samples of a white porcelain enamel were included in the test at IITRI's request. A sample of Chemglaze A-276 polyurethane white thermal control paint, an optical solar reflector (OSR) from Optical Coating Laboratories, Inc. (OCLI), and a fused silica substrate were also included in the exposure as controls and contamination monitors. The total sample complement was 88 specimens.

The conductive coatings were formulated from either doped pigments and nonconductive binders, from nonconductive pigments using doped binders, or from both doped pigments and doped binders. The coatings were both ceramic, silicate-based, and organic, silicone-based materials. Many of the coatings were derivatives of the existing industry standard IITRI white and black paints currently used on spacecraft.

4. Exposure Conditions

Since the agency funding the development of these new coatings was NASA, and the prime use of the coatings will be in LEO environments, UV radiation was identified as the main component of the space environment that should be simulated. Use of these coatings, in either elliptical or geosynchronous orbits, would require the addition of electrons and protons to the simulation. Both of these orbits have large fluxes of low and high energy particles that cause degradation of the coatings. However, many DOD and National Oceanic and Atmospheric Administration (NOAA) weather satellites, as well as other proposed NASA missions, use polar low earth orbits and hence encounter significant fluxes of electrons. Thus, modeling of the electron radiation environment, and the deposition of energy through the coating, was performed to assess the extent of this type of orbital exposure.

The orbit chosen for simulation was the one used by the Defense Meteorological Satellite Program (DMSP) and is roughly 460 nmi at an inclination of 90°. The model used for the electron environment was AE8 MAX. This model is tabulated in the Appendix. Energy deposition in the coatings was modeled using the code, Integrated Tiger Series (ITS), v. 3.0, available from Oak Ridge National Laboratory. This code was developed at Sandia National Laboratories. The code was run for the two baseline materials, Z-93P and S13GP/LO-1. Dose-depth profiles for a 10 year exposure in a DMSP orbit are given in the Appendix. These curves indicate that the surface dose in these two materials is approximately 25 Mrad for the 10 year exposure.

As the AE8 MAX model indicates, the energies of the electrons vary from less than 40 keV to over 6.5 MeV. The problem that must be solved is how to best simulate this on-orbit exposure in the laboratory using a variable monoenergetic electron beam. Use of the ITS code allows selection of electron energies and fluences that approximate the orbital dose-depth profile. Trial and error selection of energies and fluences are used to fit this profile. The results indicate that energies of 35 and 100 keV produce an adequate simulation. For a 10 year exposure, the corresponding fluences are 5.0×10^{13} /cm² and 8.0×10^{13} /cm², respectively. These results are shown on accompanying graphs included in the Appendix. The two-energy simulation is quite good for the first two mils of coating and is slightly better for the Z-93P material. Since most of the coatings are silicate based, this was the best choice.

The UV simulation source chosen was the xenon high pressure short arc, since it simulates the near-solar UV (200-400 nm) better than any other source. Vacuum UV radiation (<200 nm) was not included in the simulation although it is a significant part of the solar UV spectrum.

5. Test Method

Candidate test specimens were received from IITRI in labeled brown envelopes with the sample enclosed in heat-sealed inner plastic bags. Two samples of each material were received, coated on 15/16 in. aluminum disks, with a third specimen of each material on a 2×2 in. square aluminum substrate. Each sample was marked with a code on the back. For this test, only the disk samples were used. The UV-VIS-NIR reflectance of each sample to be tested was measured on a Perkin-Elmer Lambda-9 spectrometer equipped with a 6 in. Labsphere, Inc., Spectralon coated integrating sphere. The diffuse reflectance measurements were made using a 480 nm/min scan rate, 2 nm slit width, 1 s response, with the scan range being 250-2500 nm.

The reflectance spectra were background corrected and referenced to a NIST 2019d white tile diffuse reflectance standard. This procedure references the sample spectrum to a NIST perfect diffuser rather than to the Spectralon integrating sphere coating. Correction factors are calculated and applied to the sample spectrum from the measured NIST standard. The accuracy of the reflectance measurements is $\pm 2\%$. The reflectance curves are included in the Appendix to this report.

The calculation of the solar absorptance was made from the reflectance spectrum (assuming that the samples are opaque) over the range of measurement using the trapezoidal rule to approximate the integral in Eq. (1) below:

$$\alpha_{\mathbf{S}}(\lambda_{1},\lambda_{2}) = \begin{cases} \lambda_{2} \\ \int S_{\lambda} (1-\rho_{\lambda})d\lambda \\ \lambda_{1} \\ \int S_{\lambda} d\lambda \\ \lambda_{1} \end{cases}$$
(1)

where

 $\alpha_s(\lambda_1,\lambda_2)$ = solar absorptance calculated from λ_1 to λ_2 (250–2500 nm)

 $S\lambda$ = solar air mass zero spectral irradiance curve (ASTM E490)

Pλ = measured wavelength-dependent hemispherical reflectance of sample

Inspection of the solar air mass zero spectral irradiance curve in ASTM E490 reveals that 3.9% of the sun's energy falls outside the wavelength region measured by the Lambda-9 spectrometer (250–2500 nm). Depending on the absorptive properties of the samples outside of the measurement range, an error of up to 0.039 (absolute) can be incurred in the solar absorptance calculations. To estimate this error, the calculations using Eq. (1) were also performed, assuming total sample absorption outside the 250–2500 nm measurement range. This worst-case calculation is important for these paints, since the silicate and silicone binders in them absorb strongly in the infrared (IR) region beyond 2500 nm. Similarly, these, like most materials, are also highly absorptive below 250 nm. These calculations indicate that, for the

white materials, the true solar absorptance is approximately +0.03 higher than the value in Table A-2 in the Appendix. There is little difference for the black coatings since they are highly absorptive.

A listing of the samples that were tested, along with their pigment and binder compositions, sample numbers, batch numbers, pretest solar absorptance (250–2500 nm), and corresponding data file is included in Table A-2 in the Appendix to this report.

The exposure test was conducted in the "Deathstar" Facility. A drawing of this facility is included in the Appendix. The chamber is cylindrical, with a 24 in. diameter and a 5 ft length. The chamber is equipped with a water or LN₂ cooled sample table and quartz-halogen IR bake-out heaters. The chamber is pumped with both a turbomolecular pump and a cryopump to achieve a baseline pressure in the mid to high 10⁻⁹ torr range. The facility is equipped with a data system that monitors temperature, pressure, and electron beam current. The radiation sources used in this test are the xenon solar simulator and the 120 keV electron flood gun. Temperature is measured via five K-type thermocouples, pressure is measured via a nude ion gauge, and the electron beam current is monitored using up to six Faraday cups. The data system integrates the dose rate to obtain the total fluence in real time and estimates the remaining time needed to reach a predetermined dose.

Prior to sample layout, the solar simulator was refurbished, and a new arc lamp was installed. The output UV beam is filtered to remove excess IR radiation with a water filter, cooled via a continuous purge of deionized water. The intensity of the solar irradiance at the table was measured via a calibrated spectroradiometer. Measurements were taken at 5 nm intervals from 200 to 400 nm. These readings were then integrated and normalized to the ASTM E409 solar air mass zero curve to obtain the UV solar intensity in suns, to the nearest 0.1 sun level. The intensity of the source was adjusted to give 3.0 suns at the sample table in the hottest area of the beam. A UV intensity distribution map was generated using a solar cell as an irradiance detector. Short-circuit current was measured for the cell at several locations on the sample table. These readings were then normalized to the highest value to give solar intensity values from 0 to 1. The total solar irradiance, or equivalent UV sun hours (EUVSH) on a sample, is therefore exposure time × 3.0 × intensity factor. The sample layout and the UV intensity map are included in the Appendix. A table of the test specimens with the EUVSH for each sample (Table A-2) is also included. The solar cell was included in the exposure test to monitor UV irradiance during the test and to allow adjustment of the UV source to maintain a 3.0 sun intensity as the lamp degraded.

The electron beam was adjusted for beam size and homogeneity prior to sample introduction. After the samples were laid out, the Faraday cups were added for monitoring of the beam homogeneity, the dose, and the dose rate. During the exposure, several variables are used to maintain the desired flux level: filament power, beam focus, beam divergence, and trap setting. All of these variables are dependent on the electron energy, which is held constant during an exposure. When the energy is changed, all the other variables must also be changed with it. This exposure involved two electrons energies: 100 and 35 keV.

After the reflectance of the samples was measured, they were laid out on the sample table in a hexagonal close-packing arrangement. The samples were arranged so that one each of the candidate new coatings was in the prime area of the UV beam; the corresponding sister sample of each coating was then placed in an area of lesser irradiance. The baseline samples were also placed in a prime area of the beam. Preference was given to the white coatings in the layout, which were placed in a better area of the beam. A videotape record was then started and was updated daily during the test. Roughly 1-3 min of videotape were recorded under varying lighting and exposure conditions.

After all necessary adjustments were made, the chamber was evacuated, and the data system was initiated. The samples were pumped on overnight. A plot of the chamber pressure during the test is included in the Appendix. The chamber was vented with GN_2 after roughly 16 h to add the Chemglaze A-276 sample. Pumping was then continued for another day, at which time chamber pressure was in the range of 5×10^{-8} torr, and the solar simulator was started. The chamber was then vented again briefly to adjust the position of the solar cell. Pumping and solar exposure were then continued. After about 4 days' exposure, the 100 keV electrons were started. These electrons were applied only during the normal working hours. After the fluence of $8.26 \times 10^{13}/\text{cm}^2$ was reached, the gun settings were changed to allow exposure to 35 keV electrons. This exposure was performed similarly to the first exposure, until the fluence of $5.18 \times 10^{13}/\text{cm}^2$ was reached. The test was then continued, with only the UV source running. After roughly 500 h exposure, the arc lamp was replaced, and the exposure was continued. Plots of the sample table temperature and the Faraday cup electron dose and dose rate are shown in the Appendix. The total exposure time for the solar simulator was 722 h. The test time is longer due to pump-down cycles and venting cycles for sample measurements.

At the end of the UV test exposure, the arc lamp was shut down, and the chamber was vented with GN₂. The samples were measured on the Lambda-9 in somewhat random order. The samples that had been in good light were measured first. The white coatings were also measured first. After measurement, the samples were returned to the chamber and held under nitrogen purge. The time that they were each exposed to air was roughly 10 min. The measurements took roughly 2 1/2 days. At the completion of a day's measurements, the chamber was pumped back down to store all samples overnight under vacuum.

After all measurements were taken, the samples were placed in labeled polyethylene bags, and each was stapled to the original brown paper bag for identification. Samples that indicated unusual patterns or colors were photographed.

6. Results

The values for the pretest and posttest solar absorptance are tabulated, together with the sample number and its pigment/binder composition. The solar intensity factor and the corresponding EUVSH values are included in Table A-2 in the Appendix. A curve showing the pretest and posttest reflectance spectrum is given for each sample tested. These 88 curves are presented in the Appendix. In all cases, the posttest reflectance curve is the dashed line.

As can be seen from the data in the table, most samples displayed very little degradation. However, some trends are apparent. The samples containing the DZS pigment generally exhibited higher degradation, as much as a factor of 10, relative to the ZnO pigmented materials. Similar results were seen with the materials that were pigmented with Ta₂O₅. Samples that were pigmented with ZOT or Eu₂O₃ gave mixed values for their degradation. By far, the least degradation was observed in the Z-93P derived materials. Another very positive result is the very small degradation observed with the baseline materials S13GP/LO-1 and S13GP/LO-41. These materials exhibited very small degradation equal to that seen with Z-93P.

One disturbing and peculiar result is the reversal of the degradation in YB-71 and YB-71P. In this exposure, YB-71 indicated three times the degradation of YB-71P, opposite that expected from previous test data. These two samples were exposed side by side in the test; therefore, differences in exposure levels can be ruled out. The other very interesting result observed with only the ZOT-containing materials is the reverse correlation of degradation with solar exposure (EUVSH). That is, the samples exposed in a more intense area of the beam, and therefore received more EUVSH than their counterparts, exhibited less degradation. This result was observed with every ZOT pair tested and with no other pairs. The differences in degradation levels are significant.

The black samples essentially displayed no significant changes due to the exposure. There was also essentially no change for the OSR and the SiO₂ sample. The Chemglaze A-276 sample, which was only one-half exposed at the edge of the beam, gave a degradation, in alpha, of about 0.1 and turned brown, as expected. The white material showing the least degradation is ZnO(FC)/SS-55, and the white material displaying the most degradation is Ta₂O₅/2130 (RT).

M. J. Meshishnek, C. S. Hemminger, and S. R. Gyetvay, Space Environmental Exposure of IITRI White Thermal Control Paints, TR-95(5904)-2, The Aerospace Corporation, El Segundo, CA (27 April 1995).

7. Discussion

Generally, one can conclude from this test that all of the candidate conductive thermal control coatings should perform well optically in a LEO environment. The qualifiers are that the effects of vacuum UV light have not been simulated and hence exposure to this radiation on orbit should increase the degradation. Also, the exposure time is only about 2100 EUVSH in this test, and if the degradation rates of these coatings are slow, they may not be near plateau of the solar absorptance. However, Z-93/Z-93-P has been shown to be stable on orbit. The inference is that these materials performed similarly to Z-93/Z-93P and therefore should also do well on orbit. In some LEO missions, the significant atomic oxygen flux could offset, via bleaching or erosion, UV damage in these materials, providing better overall performance. Previous work on similar coatings has indicated that the performance of materials in an elliptical or geosynchronous orbit may be very poor for those not containing ZnO based pigments. This is due to the significantly higher charged particle fluxes in these orbits.

After only a few days exposure, and before the electrons were started, some samples (those containing Ta₂O₅ and DZS) began to show changes in color relative to the others on the video. A few samples also developed gray splotches and/or spots. The samples were photographed posttest. The markings did not disappear upon venting or even after air exposure during measurement. The appearance of blemishes or spots on some of the samples would indicate some problem in cleanliness, purity, or processing. These phenomena have been observed in previous IITRI tests, but the cause has never been discovered.

Previous extensive testing of Chemglaze A-276 in this laboratory indicates that a UV exposure of roughly 2000 EUVSH should give a value of 0.38 for solar absorptance. This value compares well with the value of 0.36 that was obtained in this test. Since the sample was only half lit at the edge of the UV beam, this correspondence in values is amazing. The lack of any significant degradation for the mirror or fused silica sample indicates that contamination was essentially nonexistent during this exposure. It would also strongly indicate that cross contamination from outgassing of the IITRI silicone-based materials is exceedingly small. However, surface analysis techniques would be required to validate this assertion, which was beyond the scope of this test.

There may be a possible explanation for the reversal of ZOT pigmented materials' degradation with EUVSH. If the degradation in the ZOT materials is mostly due to electrons and not UV, and if the effects of UV counteract or heal some of the damage due to the electrons, then extended UV exposure will restore the reflectance of the ZOT materials. If this is true, it has important implications for use of these materials in orbits with significant electron fluxes.

Errors involved in the repeatability of the reflectance measurements are estimated at ± 0.002 . It should be remembered, though, that the reflectance spectrum is measured from 250 to 2500 nm and hence does not include the entire solar spectrum. Thus, the values calculated for solar absorptance will be lower than actual or calorimetric numbers if the sample absorbs above 2500 nm or below 250 nm. This error is roughly + 0.03 in solar absorptance. Since most of the silicate and silicone paints absorb above 2500 nm, this + 0.03 should be added to all pretest and posttest numbers. The delta alpha numbers, however, should not change.

The error in the electron dose is less than 5%. The errors in the UV exposure factors are more difficult to quantify. Since the solar cell exposure area is about 3 in² and the test sample area is about 1 in², there

can be a significant intensity gradient across the area measured by the cell. Thus, the values for the intensity factors are probably only good to ± 0.1 . This is especially true for the samples placed in the dimmer areas of the beam. Intensity factors for samples that were in good light are probably accurate to ± 0.02 .

Currently, we do not have a method for determining the amount of recovery of reflectance due to air exposure. Experience has indicated that this recovery is small for the exposure times involved in these tests. In a test designed to screen or rank various candidate materials for relative space environmental stability, this recovery loses much of its importance. Since many materials exhibited significant degradation, wholesale bleaching of radiation damage can be discounted. However, the very small changes that were observed in the zinc oxide pigmented materials are perhaps suspiciously low. To explain this phenomenon, one has to invoke preferential bleaching of only zinc oxide based materials, which is highly unlikely. It should also be remembered that LDEF results indicated a degradation of only about 0.02 for Z-93. Similar degradation has also been seen in some ground tests. The very low degradation seen in the S13GP derived materials is somewhat more surprising since large changes in alpha were observed on LDEF. However, the materials that were flown on LDEF used a different silicone binder material, which may account for the difference in degradation. Previous work at Aerospace and other laboratories has shown that ground test data on S13G type materials often underpredicts the flight data. Table 1 presents some of the relevant flight and ground data on these paints:

Table 1. Selected Flight and Ground Test Data on Z-93/S13G Materials

| Paint | Data Type | Time | Delta a |
|--------------|-------------------|---------------|---------|
| Z-9 3 | LDEF | 69 months | 0.017 |
| Z-9 3 | Pegasus II | 2000 h | 0.002 |
| Z-93 | Pegasus III | 1000 h | 0.002 |
| Z-93 | OSO III | 1580 h | 0.005 |
| Z-93 | Ground, UV | 7067 h | 0.02 |
| Z-93P | Ground, UV | 3100 h | 0.039 |
| Z-93 | Ground, UV | 5000 h | 0.028 |
| Z-93P | UV/e- (This work) | 2100 h | 0.007 |
| S13G/LO | LDEF | 69 months | 0.32 |
| \$13G/LO-1 | Ground, UV/e- | 70 0 h | 0.025 |
| S13GP/LO-1 | Ground, UV | 3100 h | 0.036 |
| S13GP/LO-1 | UV/e- (This work) | 2100 h | 0.01 |

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^{*} See footnote on page 13.

8. Conclusions

While nearly all of the new material systems tested have shown promise, it would seem that the materials utilizing Ta₂O₅ or DZS should probably be screened out in favor of the other materials. Mixed results seem to have been obtained with the Eu₂O₃ materials; hence, it would seem that they would warrant further investigation before they are discarded.

The conductivity measurements of these samples, pre- and postexposure to UV, are more important than the exposure testing. Careful unambiguous measurements need to be performed to assess the true conductivities of these candidate materials and their relative stability to the space environment. A detailed study of these properties and materials should provide data to focus on a few of the most promising materials for stable conductive thermal control paints.

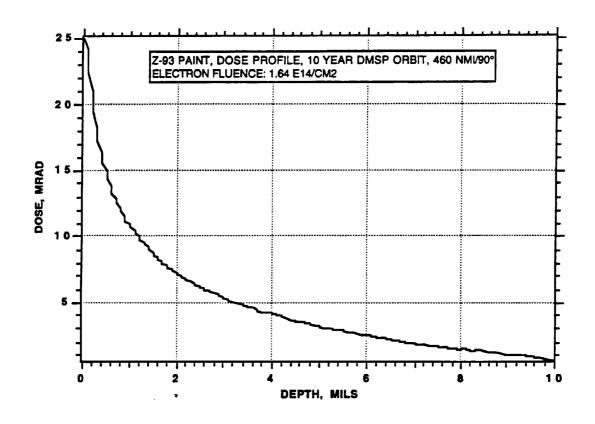
Future work should focus on the stability of these materials to environments more severe than LEO, such as geosynchronous Earth orbit (GEO) or medium Earth orbit (MEO), where military spacecraft operate. Conductive coatings should prove invaluable for such military systems. These candidate developmental materials may go a long way to solving charging problems for these systems, provided that these materials retain their properties in these higher radiation environments.

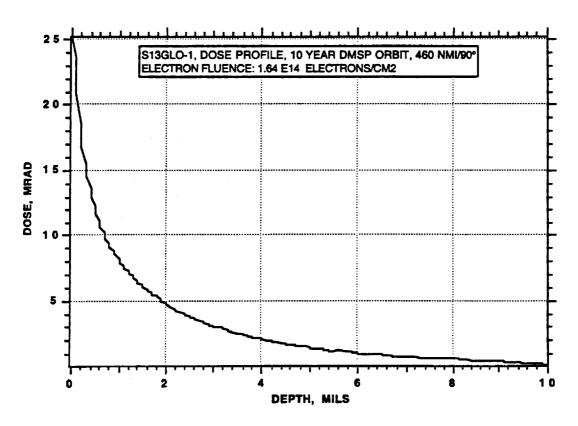
Appendix

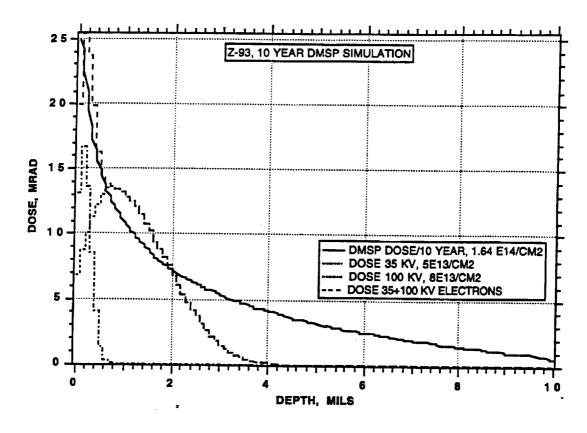
Test Data, Environmental Model, and Analytical Predictions for the UV/Electron Exposure

Table A-1. Electron Environment Model AE8 MAX, DMSP Orbit, 460 nmi/90°

| E1-E2,MEV | Flux/day>E1 | Integral Flux | Flux/day <e1< th=""><th>%<e1< th=""></e1<></th></e1<> | % <e1< th=""></e1<> |
|-----------|-------------|---------------|---|---------------------|
| 0 | 9.04E+10 | 2.50E+10 | 0.00E+00 | .000000 |
| 0.04 | 6.54E+10 | 1.29E+10 | 2.50E+10 | .276602 |
| 0.07 | 5.25E+10 | 1.01E+10 | 3.79E+10 | .419328 |
| 0.1 | 4.24E+10 | 2.39E+10 | 4.80E+10 | .531075 |
| 0.2 | 1.85E+10 | 1.02E+10 | 7.19E+10 | .795507 |
| 0.3 | 8.29E+10 | 4.34E+09 | 8.21E+10 | .908360 |
| 0.4 | 3.95E+09 | 1.85E+09 | 8.64E+10 | .956378 |
| 0.5 | 2.09E+09 | 6.25E+08 | 8.83E+10 | .976847 |
| 0.6 | 1.47E+09 | 4.10E+08 | 8.89E+10 | .983762 |
| 0.7 | 1.06E+09 | 2.47E+08 | 8.93E+10 | .988298 |
| 0.8 | 8.10E+08 | 1.58E+08 | 8.96E+10 | .991031 |
| 0.9 | 6.52E+08 | 1.25E+08 | 8.97E+10 | .992779 |
| 1 | 5,27E+08 | 1.86E+08 | 8.99E+10 | .994162 |
| 1.25 | 3.41E+08 | 1.19E+08 | 9.00E+10 | .996220 |
| 1.5 | 2.23E+08 | 7.69E+07 | 9.02E+10 | .997537 |
| 1.75 | 1.46E+08 | 5.01E+07 | 9.02E+10 | .998388 |
| 2 | 9.56E+07 | 3.17E+07 | 9.03E+10 | .998942 |
| 2.25 | 6.39E+07 | 2.10E+07 | 9.03E+10 | .999293 |
| 2.5 | 4.29E+07 | 1.45E+07 | 9.03E+10 | .999525 |
| 2.7 | 2.84E+07 | 1.27E+07 | 9.04E+10 | .999685 |
| 3 | 1.57E+07 | 9.94E+06 | 9.04E+10 | .999826 |
| 3.5 | 5.79E+06 | 3.86E+06 | 9.04E+10 | .999936 |
| 4 | 1.93E+06 | 1.36E+06 | 9.04E+10 | .999979 |
| 4.5 | 5.71E+05 | 4.13E+05 | 9.04E+10 | .999994 |
| 5 | 1.58E+05 | 1.23E+05 | 9.04E+10 | .999998 |
| 5.5 | 3.51E+04 | 2.99E+04 | 9.04E+10 | 1.000000 |
| 6 | 5.13E+03 | 5.13E+03 | 9.04E+10 | 1.000000 |
| 6.5 | 0.00E+00 | 0.00E+00 | 9.04E+10 | 1.000000 |







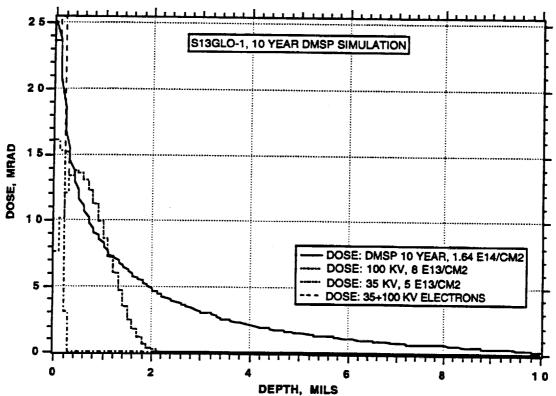
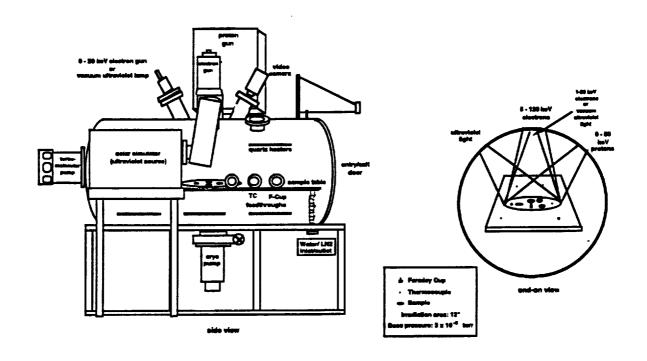


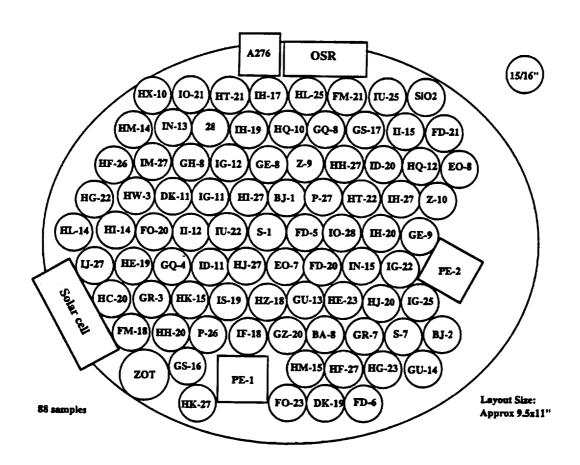
Table A-2. IITRI Test Samples

| Coating | Pigment | Binder | Batch # Samp | 1= | e# File# Pretes | Pretest | File # | Posttest | 1 | 2 | EIIVEH |
|---------------|-------------|------------|--------------|---|-------------------|---------|---------|----------|--------|--------|--------|
| 101 | OCCOUNT VOE | | | | | Alpha | | Alpha | Alpha | Factor | |
| 107 | 08/98#107 | None | 98/90 | 18107 | DS40801 | 0.079 | DS41801 | 0.104 | 0.025 | 96.0 | 2079 |
| Z:93P | Zinc Oxide | Kasil 2130 | U-347 | HX-10 | DS40802 | 0.133 | DS41840 | 0.14 | 0.007 | 96.0 | 2079 |
| YB-71 | ZOT #98790 | PS-7 | U-348 | HW-3 | DS40803 | 0.107 | DS41826 | 0.14 | 0.033 | 0.98 | 2123 |
| YB-71P | ZOT #98790 | Kasil 2130 | U-376 | IM-27 | DS40804 | 0.112 | DS41831 | 0.123 | 0.011 | 0.98 | 2123 |
| S13GP/LO-1 | S13GP | LO-1 | U-310 | 13-51 | DS40805 | 0.167 | DS41814 | 0.177 | 0.01 | 0.99 | 2144 |
| S13GP/LO-41 | S13GP | LO-41 | U-276 | HC-20 | DS40806 | 0.149 | DS41810 | 0.156 | 0.007 | 0.97 | 2101 |
| 0255/LO-41 | SS70 | 10-41 | U-208 | GH.8 | DS40807 | 0.077 | DS41832 | 0.243 | 0.166 | 0.97 | 2101 |
| 2015/LO-41 | 06/86# 107 | LO-41 | U-343 | 28 | DS40808 | 0.133 | DS41837 | 0.196 | 0.063 | 0.95 | 2058 |
| Eu203/2130 | Eu203 | Kasil 2130 | U-352 | IF-18 | DS40809 | 0.092 | DS41808 | 0.12 | 0.028 | 0.93 | 2014 |
| Eu203/LO-41 | Eu203 | LO-41 | U-403 | IS-19 | DS40810 | 0.136 | DS41813 | 0.178 | 0.042 | 96.0 | 2079 |
| ZnO(FC)/2130 | ZnO(FC) | Kasil 2130 | U-321 | HM-14 | DS40811 | 0.148 | DS41835 | 0.154 | 900.0 | 0.97 | 2101 |
| ZnO(FC)/2130 | ZnO(FC) | Kasil 2130 | U-321 | HM-15 | DS40812 | 0.148 | DS41845 | 0.149 | 0.001 | 9.0 | 1733 |
| ZnO(FC)/SS-55 | ZnO(FC) | SS-55 | U-322 | HF-26 | DS40813 | 0.147 | DS41830 | 0.144 | -0.003 | 0.98 | 2123 |
| ZnO(FC)/SS-55 | ZnO(FC) | SS-55 | U-322 | HF-27 | DS40814 | 0.146 | DS41846 | 0.144 | -0.002 | 20 | 1516 |
| ZnO(FC)/DHS-1 | ZnO(FC) | DHS-1 | U-323 | HG-23 | DS40815 | 0.157 | DS41847 | 0.164 | 0.007 | 0.64 | 1386 |
| ZnO(FC)/DHS-1 | ZnO(FC) | DHS-1 | U-323 | HG-22 | DS40816 | 0.149 | DS41825 | 0.156 | 0.007 | 0.99 | 2144 |
| ZnO(FC)/DHS-2 | ZnO(FC) | DHS-2 | U-327 | HL-14 | DS40817 | 0.139 | DS41819 | 0.141 | 0.002 | 0.99 | 2144 |
| ZnO(FC)/DHS-2 | ZnO(FC) | DHS-2 | U-327 | HL-25 | DS40818 | 0.146 | DS41850 | 0.146 | 0 | 98.0 | 1863 |
| S13GP/2130 | S13GP | Kasil 2130 | Ϋ́ | FM-18 | DS40819 | 0.131 | DS41805 | 0.148 | 0.017 | 0.97 | 2101 |
| S13GP/2130 | S13GP | Kasll 2130 | Ϋ́ | FM-21 | DS40820 | 0.129 | DS41851 | 0.137 | 900.0 | 0.78 | 1689 |
| S13GP/SS-55 | S13GP | SS-55 | U-405 | IU-22 | DS40821 | 0.125 | DS41823 | 0.14 | 0.015 | - | 2166 |
| S13GP/SS-55 | S13GP | SS-55 | U-405 | 10-25 | DS40822 | 0.127 | DS41852 | 0.134 | 0.007 | 0.78 | 1689 |
| S13GP/DHS-1 | S13GP | DHS-1 | U-279 | GS-16 | DS40823 | 0.147 | DS41803 | 0.153 | 900.0 | 0.94 | 2036 |
| S13GP/DHS-1 | S13GP | DHS-1 | U-279 | GS-17 | DS40824 | 0.142 | DS41854 | 0.148 | 900.0 | 8.0 | 1733 |
| S13GP/DHS-2 | S13GP | DHS-2 | U-328 | HH-20 | DS40825 | 0.13 | DS41806 | 0.155 | 0.025 | 96.0 | 2079 |
| S13GP/DHS-2 | S13GP | DHS-2 | U-328 | HH-27 | DS40826 | 0.134 | DS41856 | 0.146 | 0.012 | 0.91 | 1971 |
| S13N/SS-55 | S13N | SS-55 | U-260 | GR-3 | DS40827 | 0.115 | DS41811 | 0.141 | 0.026 | 0.97 | 2101 |
| S13N/SS-55 | S13N | SS-55 | U-260 | GR-7 | DS40828 | 0.114 | DS41848 | 0.119 | 0.005 | 0.82 | 1776 |
| S13N/DHS-1 | S13N | DHS-1 | U-317 | HE-19 | DS40929 | 0.15 | DS41815 | 0.173 | 0.023 | 96.0 | 2123 |
| S13N/DHS-1 | S13N | DHS-1 | U-317 | HE-23 | DS40830 | 0.143 | DS41867 | 0.16 | 0.017 | 0.85 | 1841 |
| S13N/DHS-2 | S13N | DHS-2 | U-329 | HI-14 | DS40831 | 0.134 | DS41820 | 0.161 | 0.027 | 0.99 | 2144 |
| S13N/DHS-2 | S13N | DHS-2 | U-329 | HI-27 | DS40832 | 0.132 | DS41829 | 0.16 | 0.028 | - | 2166 |
| DZS/2130 | SZO | Kasil 2130 | T-244 | DK-11 | DS40833 | 0.077 | DS41727 | 0.132 | 0.055 | 96.0 | 2079 |
| DZS/2130 | SZO | Kasil 2130 | T-244 | DK-19 | DS40834 | 0.081 | DS41844 | 0.122 | 0.041 | 0.78 | 1689 |
| 075/55 | S70 | SS-55 | 0.207 | 20-53 5-53 5-53 5-53 5-53 5-53 5-53 5-53 | DS40835 | 0.107 | DS41821 | 0.204 | 0.097 | 0.95 | 2058 |
| 0.28/83-55 | \$70 | SS-55 | 0.207 | FO-23 | DS40836 | 0.1 | DS41804 | 0.185 | 0.085 | 0.83 | 1798 |
| DZS/DHS-1 | 920 | DHS-1 | 0-278 | 60-4 | DS40837 | 0.075 | DS41816 | 0.202 | 0.127 | 0.95 | 2058 |
| DESOURS-1 | 023 | DHG 9 | 0-270 | 200 | DS40838 | 0.078 | DS41853 | 0.163 | 0.085 | 0.8 | 1733 |
| 025/0HS-2 | 078 | DHS-2 | 1531 | HK-97 | DSAOBAO | 0.080 | 0541812 | 0.189 | 0.103 | 96.0 | 2079 |
| DS13N/SS-55 | DS13N | SS-55 | U-315 | P-26 | DS40841 | 0.003 | DS41807 | 0.10 | 10.09 | 96.0 | 2036 |
| DS13N/SS-55 | DS13N | SS-55 | U-315 | P-27 | DS40842 | 0 137 | DS41850 | 0.153 | 0.033 | 8 6 | 5000 |
| DS13N/DHS-1 | DS13N | DHS-1 | U-316 | S-7 | DS40843 | 0.123 | DS41849 | 0.133 | 100 | 7.00 | 1516 |
| DS13N/DHS-1 | DS13N | DHS-1 | U-316 | S-1 | DS40844 | 0.124 | DS41824 | 0.161 | 0.037 | 800 | 2144 |
| DS13N/DSH-2 | DS13N | DHS-2 | U-330 | HJ-20 | DS40845 | 0.122 | DS41868 | 0.128 | 9000 | 0.8 | 1733 |
| DS13N/DSH-2 | DS13N | DHS-2 | U-330 | HJ-27 | DS40846 | 0.129 | DS41818 | 0.149 | 0.02 | 0.98 | 2123 |
| ZOT/SS-55 | ZOT #98790 | 58-55 | U-379 | N-13 | DS40847 | 0.117 | DS41836 | 0.156 | 0.039 | 0.95 | 2058 |
| ZOT/SS-55 | ZOT #98790 | SS-55 | U-379 | N-15 | DS40848 | 0.112 | DS41865 | 0.156 | 0.044 | 9.0 | .1733 |
| ZOT/DHS-2 | ZOT #98790 | DHS-2 | U-378 | 10-21 | DS40849 | 0.125 | DS41841 | 0.158 | 0.033 | 0.95 | 2058 |
| | | | | | | | | | | | |

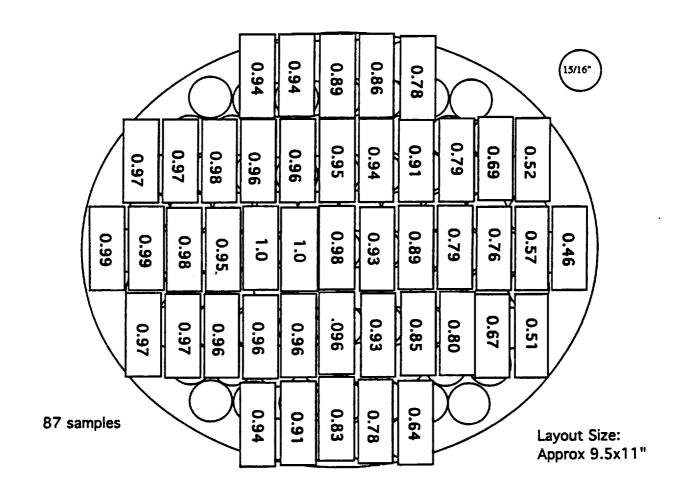
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|---------------------------|-------------------|------------|----------------|----------------|-------------|-------------|------------|-------------|---------------|---------------|---------------|---------------|-------------|-------------|--------------|--------------|---------------|--------------|----------------|----------------|------------------|-----------------|-------------|-------------|-------------|------------------|-------------------------|-------------------------|--------------------------|--------------------------|-------------------|-------------------|--------------------------|--------------------------|-----------------|-----------------|---------------------------|---------------|-----------------------|----------------|--------------|---------------|
| | EUVSH | 1300 | 2036 | 1733 | 2144 | 1733 | 2166 | 25.100 | 1310 | 4300 | 2000 | 6078 | 004 | 1993 | 200 | 1350 | 1070 | 646 | 1495 | 1451 | 2058 | 1993 | 1083 | 2014 | 2079 | 1841 | 2058 | 1191 | 2036 | 1300 | | 1461 | 2014 | 1191 | 1.00 | 4007 | 2036 | 1191 | | 2036 | 1841 | 1083 |
| | ۸n ع | | 9 6 | 3 6 | 800 | 80 | - | | 300 | 200 | 9 6 | 0.30 | 3 6 | 26.0 | 0,0 | 60.00 | 200 | | 200 |) (0.6 | 0.95 | 0.92 | 0.5 | 0.93 | 96.0 | 0.85 | 0.95 | 0.55 | 0.94 | 9.0 | 5 | 0.57 | 0.93 | 0.55 | 5 | 25.5 | 0.94 | 0.55 | 3 | 0.94 | 0.85 | 0.5 |
| | Delta | 0.050 | 0.038 | 0.023 | 0.051 | 0.026 | 0.065 | 0 032 | 0 141 | 0 108 | 0.00 | 905 | 2800 | 200 | 30.0 | 200 | 117 | 0000 | 0.0/0 | 0.018 | 0.032 | 0.025 | 0.024 | 0 | 0 | 0 | 0.001 | 0.001 | -0.001 | -0.001 | 4 | 3 | 0.001 | 0 | 100 | 200 | 0.001 | 0.001 | | 0.103 | 0.00 | 0 |
| | Posttest Ainha | 181 | 0.149 | 0.151 | 0.138 | 0.126 | 0.158 | 0.108 | 0.304 | 0.286 | 0 232 | 0 231 | 0 222 | 0 226 | 0 100 | 0.216 | 0.278 | 0000 | 0.538 | 2 | 0.176 | 0.284 | 0.289 | 0.981 | 0.982 | 96.0 | 0.975 | 0.974 | 0.939 | 0.941 | 0.057 | 9560 | 0.964 | 0.965 | 285 | 0.053 | 0.956 | 0.957 | | 0.364 | 0.084 | 0.069 |
| | File # | DS41862 | DS41842 | DS41860 | DS41817 | DS41857 | DS41822 | DS41855 | DS41828 | DS41869 | DS41833 | DS41866 | DS41838 | DS41863 | DS41843 | DS41861 | DS41839 | DS41858 | DS41864 | 20041004 | DS41834 | US4 1809 | 0541870 | DS41886 | DS41883 | DS41887 | DS41882 | DS41879 | DS41876 | DS41880 | DS41885 | DS41874 | DS41884 | DS41878 | DS41881 | DS41873 | DS41877 | DS41875 | | DS41871 | DS41872 | DS41888 |
| mples | Pretest Alpha | 0.122 | 0.131 | 0.128 | 0.087 | 0.1 | 0.093 | 0.076 | 0.163 | 0.18 | 0.165 | 0.181 | 0.137 | 0.143 | 0.149 | 0.167 | 0.161 | 0 163 | 0 141 | | 0.144 | 8030 | 0.200 | 0.981 | 0.982 | 0.98 | 0.974 | 0.973 | 0.94 | 0.942 | 0.958 | 0.956 | 0.963 | 0.965 | 0.951 | 0.95 | 0.955 | 0.956 | | 0.261 | 0.065 | 0.069 |
| IITRI Test Samples | # elle | DS40850 | DS40851 | DS40852 | DS40853 | DS40854 | DS40855 | DS40856 | DS40857 | DS40858 | DS40859 | DS40860 | DS40861 | DS40862 | DS40863 | DS40864 | DS40865 | DS40866 | DS40867 | Devoge | DS40860 | 0000000 | 0340070 | US408/1 | DS40872 | DS40873 | DS40874 | DS40875 | DS40876 | DS40877 | DS40878 | DS40879 | DS40880 | DS40881 | DS40882 | DS40883 | DS40884 | DS40885 | 000777 | NM11306 | 034/711 | DS42006 |
| | Sample # | 10-28 | HT-21 | HT-22 | ID-11 | ID-20 | 11-12 | 11-15 | G-11 | IG-25 | 16-12 | IG-22 | H-19 | IH-20 | H-17 | IH-27 | HQ-10 | HO-12 | GE-9 | G.E.B | | | 27.31 | 12-75 | HZ-18 | B-49 | EO-7 | EO-8 | 6-2 | 2-10 | GU-13 | GU-14 | FD-20 | FD-21 | FD-5 | FD-6 | - | BJ-2 | 1,141 | | E 50 | NO2 |
| Table A-2. | Batch # | U-378 | U-377 | U-377 | U-353 | U-353 | U-381 | U-381 | U-358 | U-358 | U-358 | U-358 | U-359 | U-359 | U-359 | U-359 | U-334 | U-334 | U-252 | 11.252 | NA | | 11.284 | 1000 | 0-360 | T-023 | U021 | U021 | S-024 | S-024 | U-283 | U-283 | U-166 | U-166 | U-053 | U-053 | T-026 | T-026 | 4 | ž ž | ¥ 2 | ¥. |
| | Binder | DHS-2 | Phos. Sol. | Phos. Sol. | SS-55 | SS-55 | DHS-2 | DHS-2 | Kasil 2130/RT | Kasil 2130/RT | Kasil 2130/HT | Kasil 2130/HT | SS-55/RT | SS-55/RT | SS-55/HT | SS-55/HT | LO-41 | 10-41 | Kasil 2130 | Kasil 2130 | | | Kaeil 2130 | Masil £ 130 | 3 | | Kasil 2130 | Kasil 2130 | Kasil 2130 | Kasii 2130 | DHS-1 | DHS-1 | 10-1 | 1-0-1 | -63 | 1.0.1 | LO-1 | 1:0:1 | Dolumenthan | - Chydreurarre | ¥ 2 | Y. |
| | Pigment | ZOT #98790 | ZOT #98790 | ZOT #98790 | Euzoa | Eu203 | Eu203 | Eu203 | Ta205 | Ta205 | Ta205 | Ta205 | Ta205 | Ta205 | Ta205 | Ta205 | DS13NSC | DS13NSC | ZnO+Sb+Zn2SnO4 | ZnO+Sb+Zn2SnO4 | | | Black Glass | Disch Class | DIRCK GIASS | Cosmic Black WB- | In Doped Black Glass | In Doped Black Glass | Black Glass, Graphite | Black Glass, Graphite | Doped Black Glass | Doped Black Glass | Black Glass, Graphite | Black Glass, Graphite | Black Glass/B4C | Black Glass/B4C | Cosmic Black, Graphite | Cosmic Black, | Graphite Tina/cina | Coated Sing/Al | Freed SiO | LUSEU OIOE |
| | Coating | ZOT/DHS-2 | ZOT/Phos. Sol. | ZOT/Phos. Sol. | EUZO3/55-55 | EUZO3/55-55 | EUZUSUHS-2 | EUZU3/DHS-2 | Ta2O5/2130 | Ta2O5/2130 | Ta205/2130 | Ta2O5/2130 | Ta205/SS-55 | Ta205/SS-55 | Ta205/SS-55 | Ta2Q5/SS-55 | DS13NSC/LO-41 | DS13NSCAO-41 | Sb Doped | Sb Doped | Porcelain Enamel | Porcelain Ename | MH211P | MH21SAO | Dated | DZISALO | DBG-IP | DBG-IP | MH55IC | MHSSIC | | | MH21SC/LO | MH21SC/LO | MH41SC/LO | MH41SC/LO | D21SC/LO | D21SCALO | A.978 | 8501100 | Fired Silica | ו משפח כאונים |



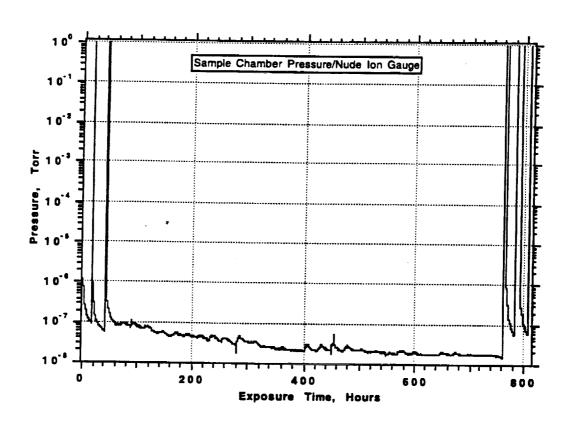
D E A T H S T A R
Space Environment Effects Chamber

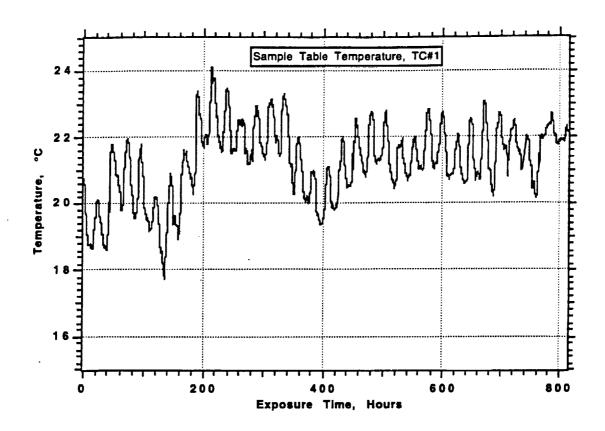


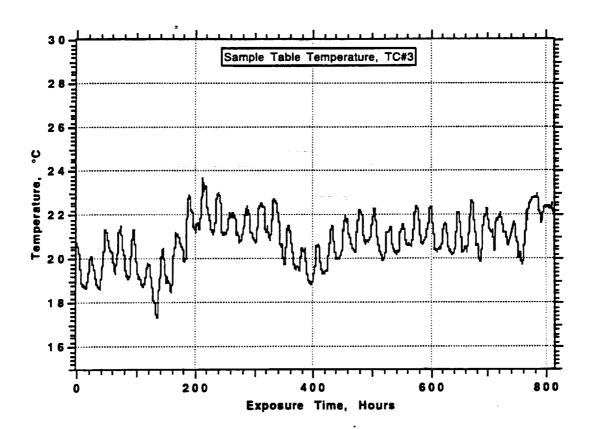
Sample Layout for IITRI UV Test.

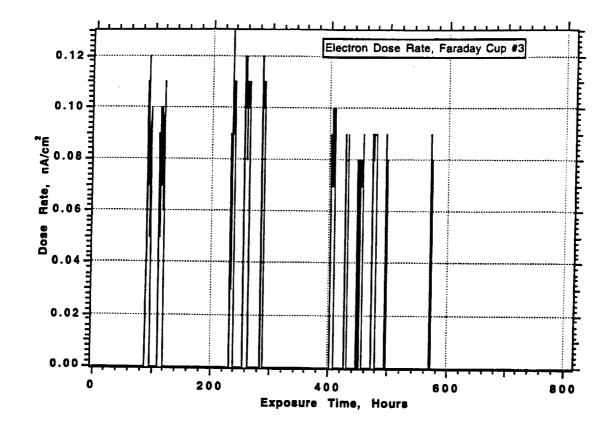


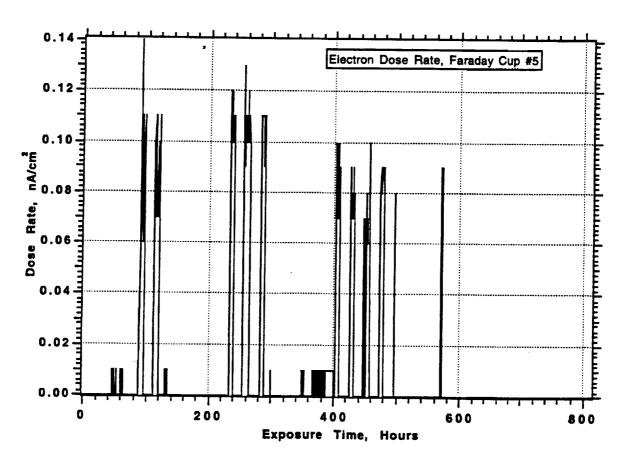
UV Intensity Map for IITRI Test

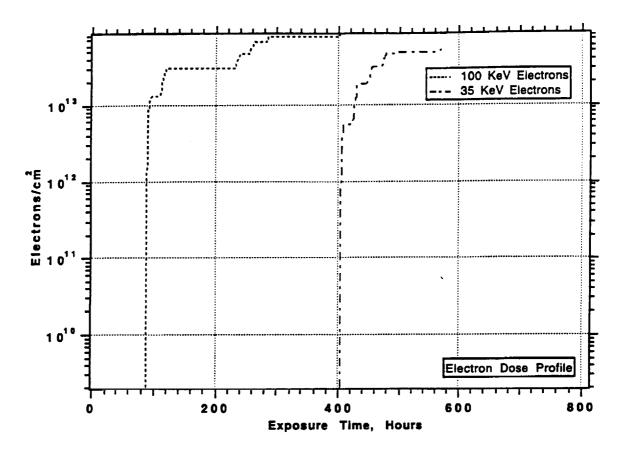


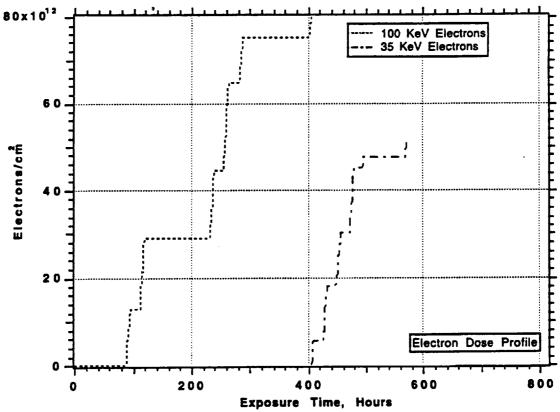




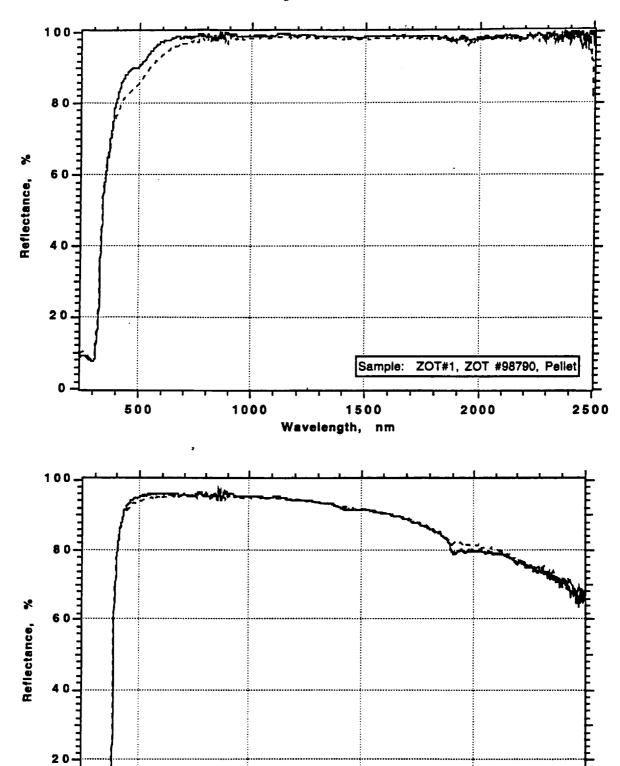








Test Sample Reflectance Plots



Wavelength, nm

1500

1000

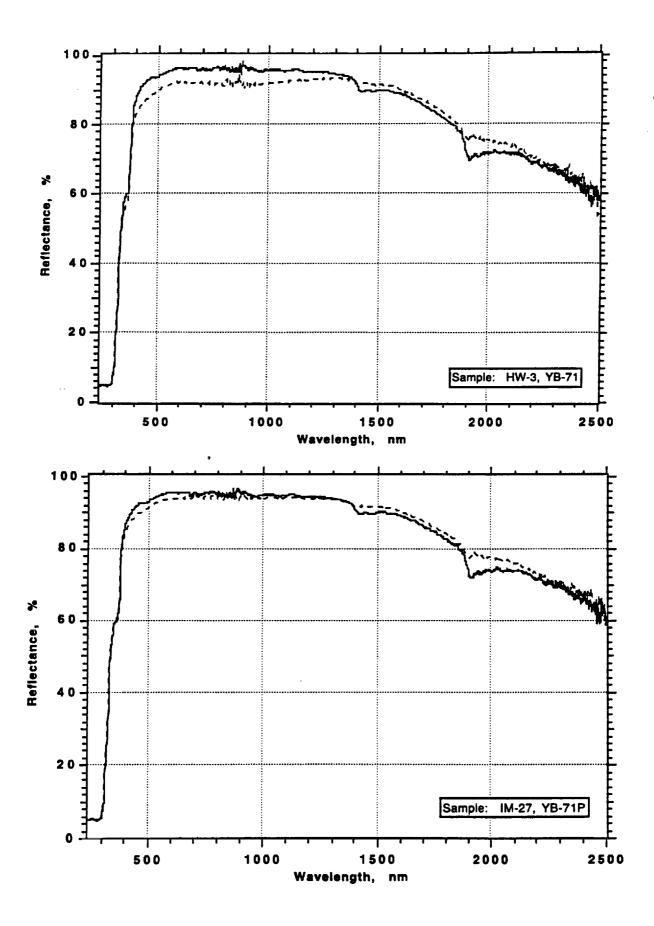
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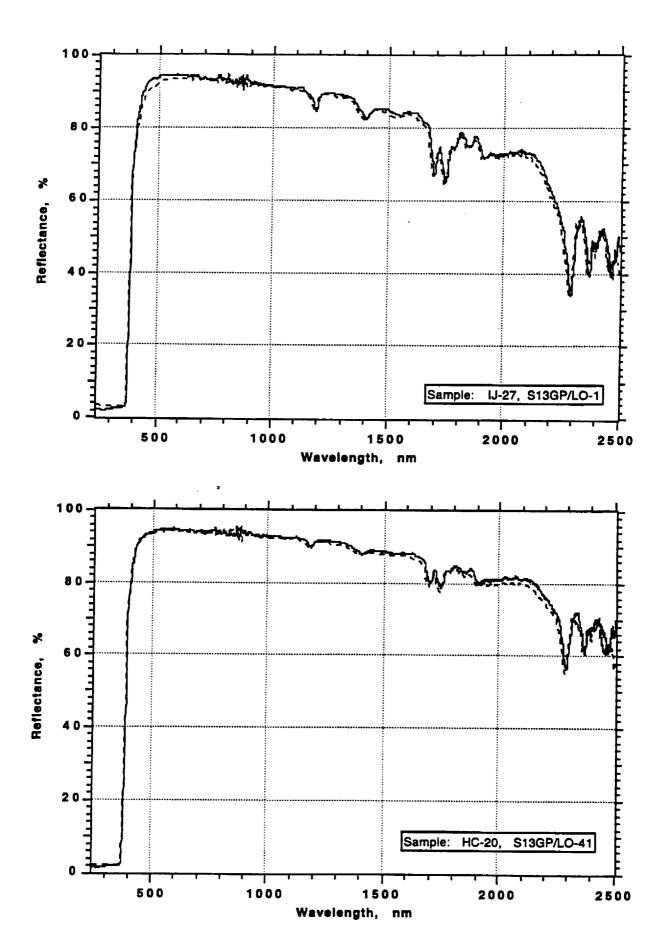
500

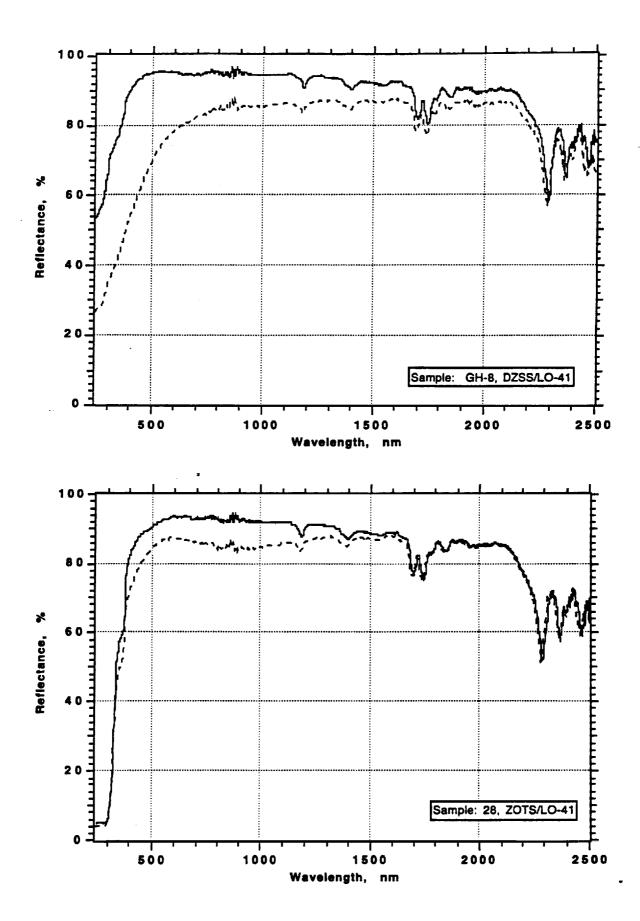
Sample: HX-10, Z-93P

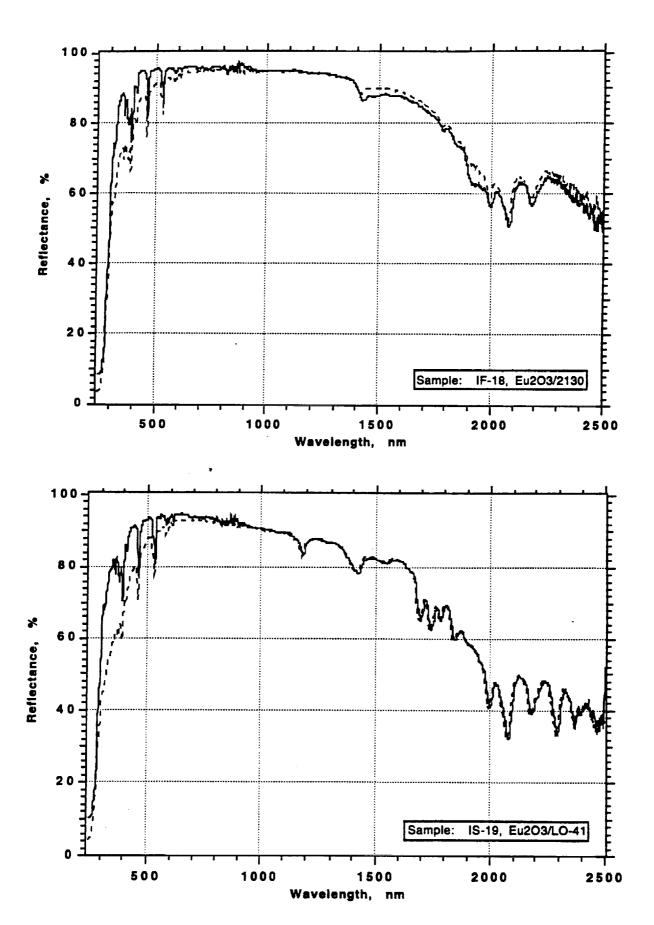
2500

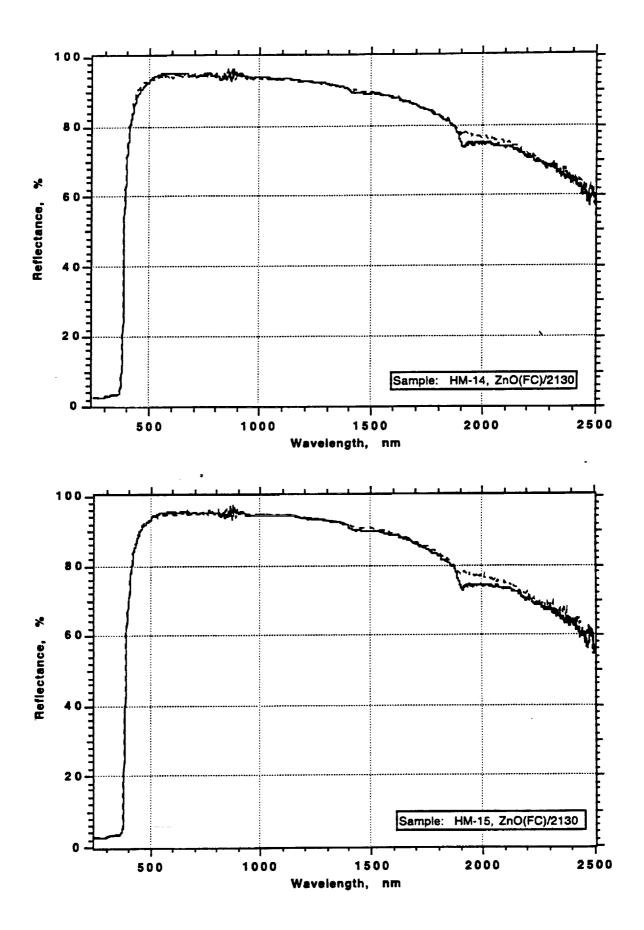
2000

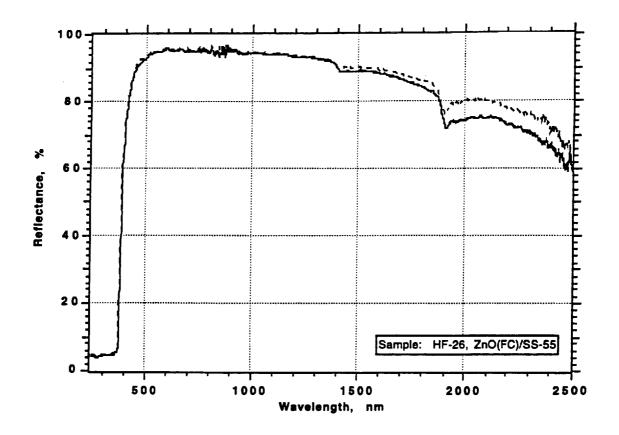


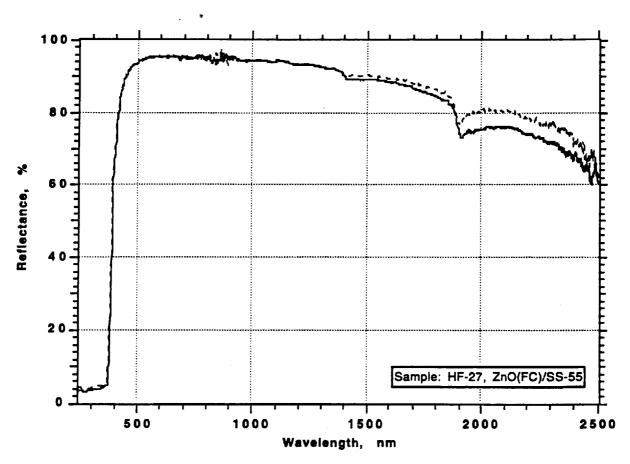


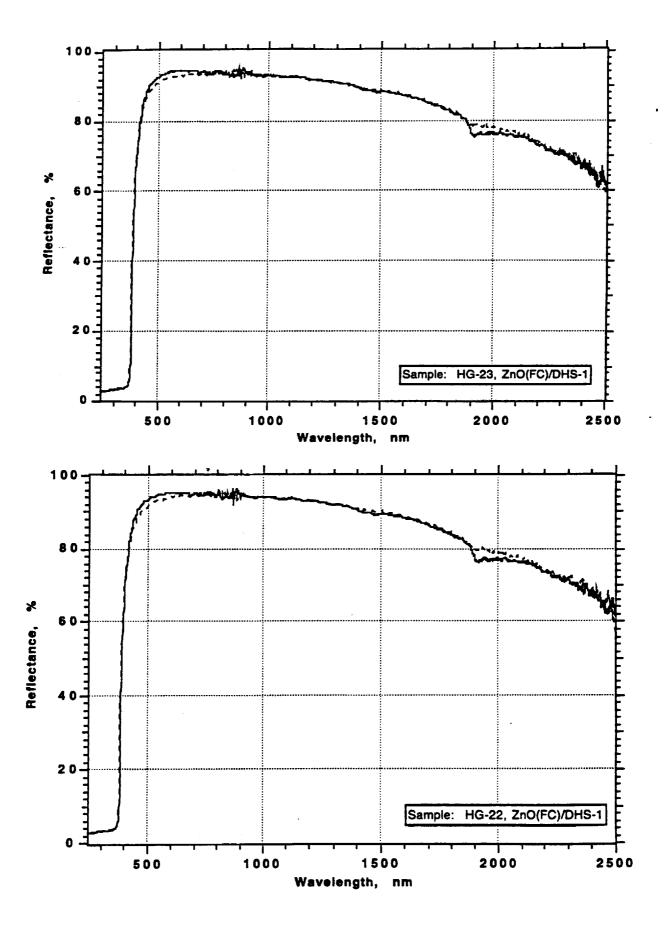


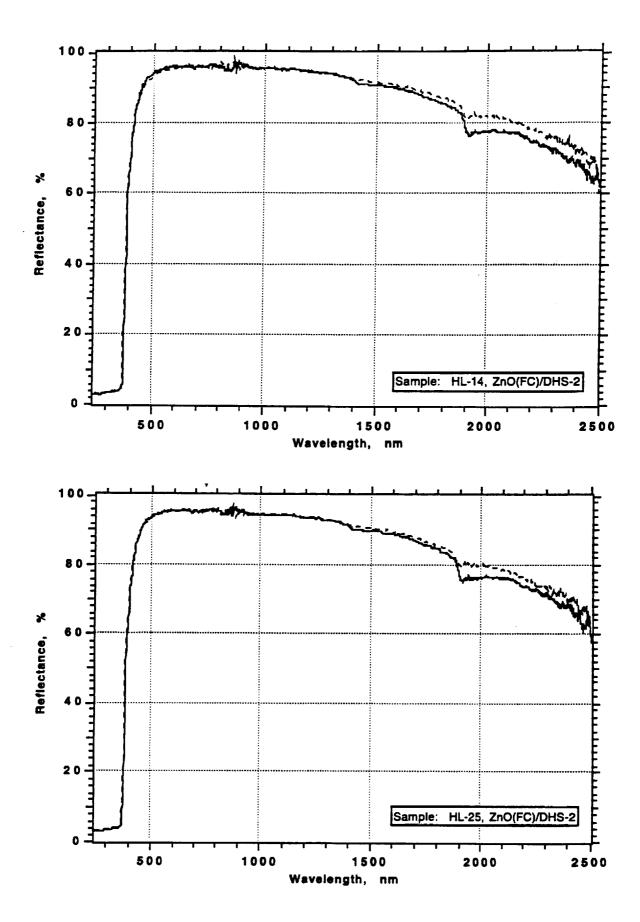


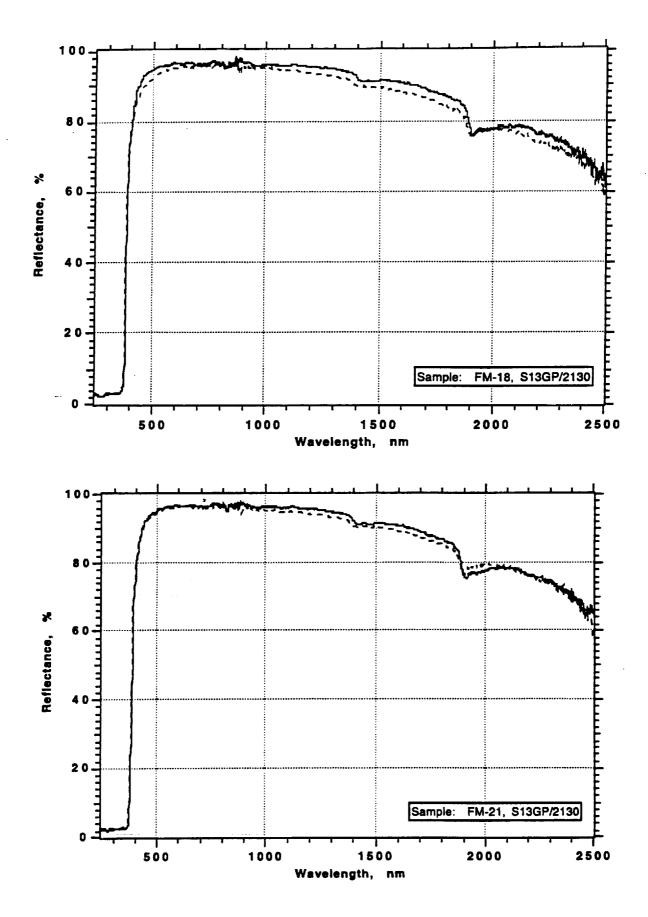


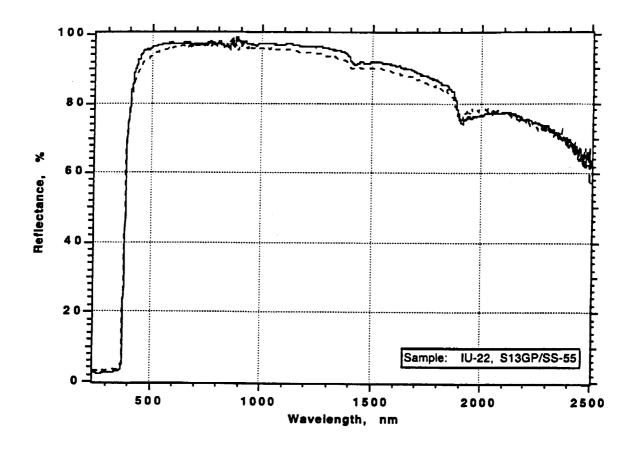


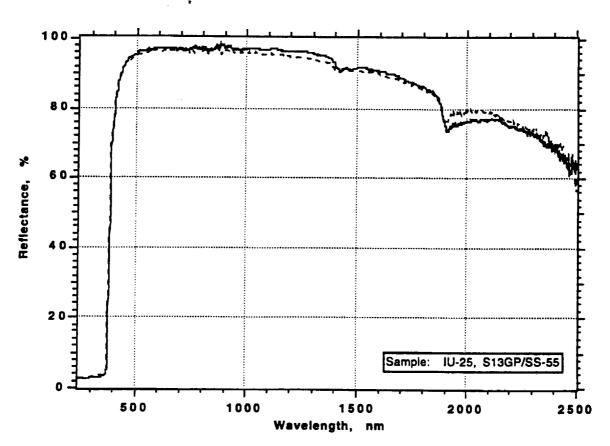


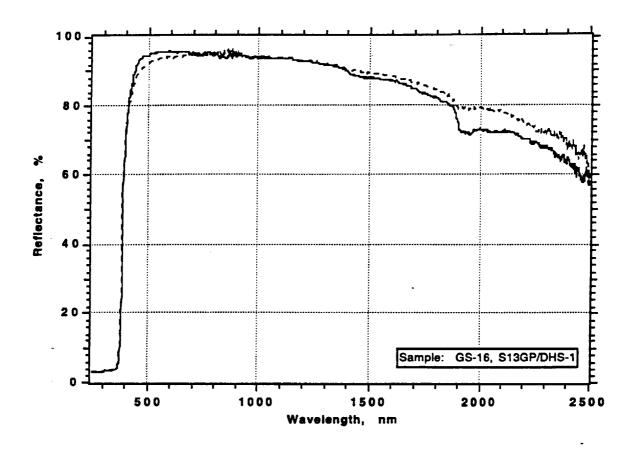


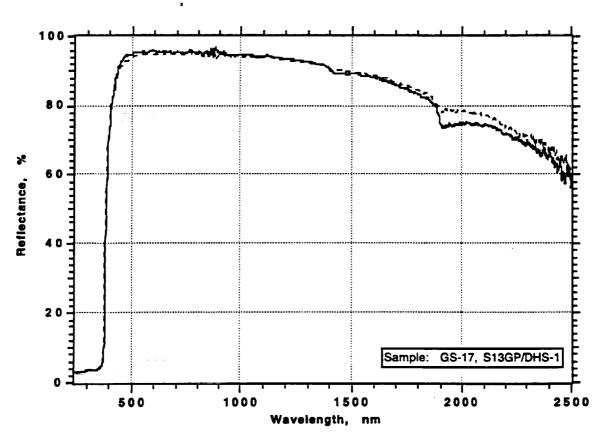


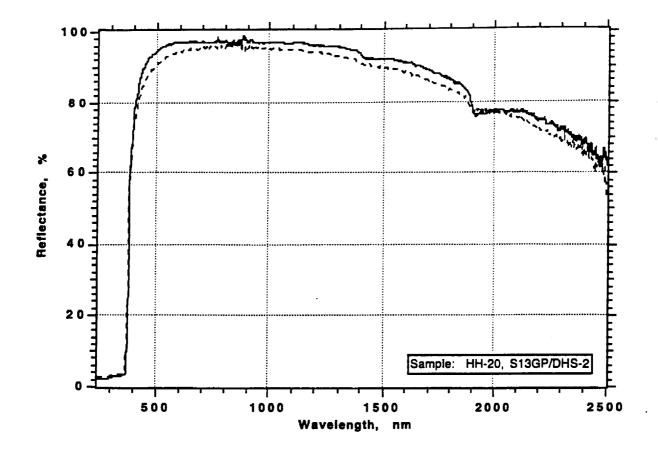


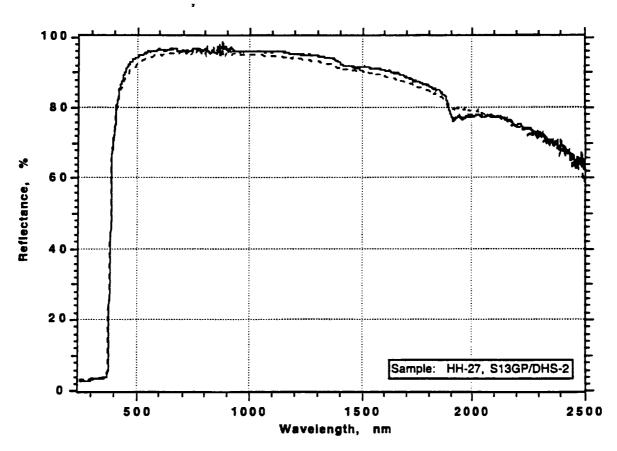


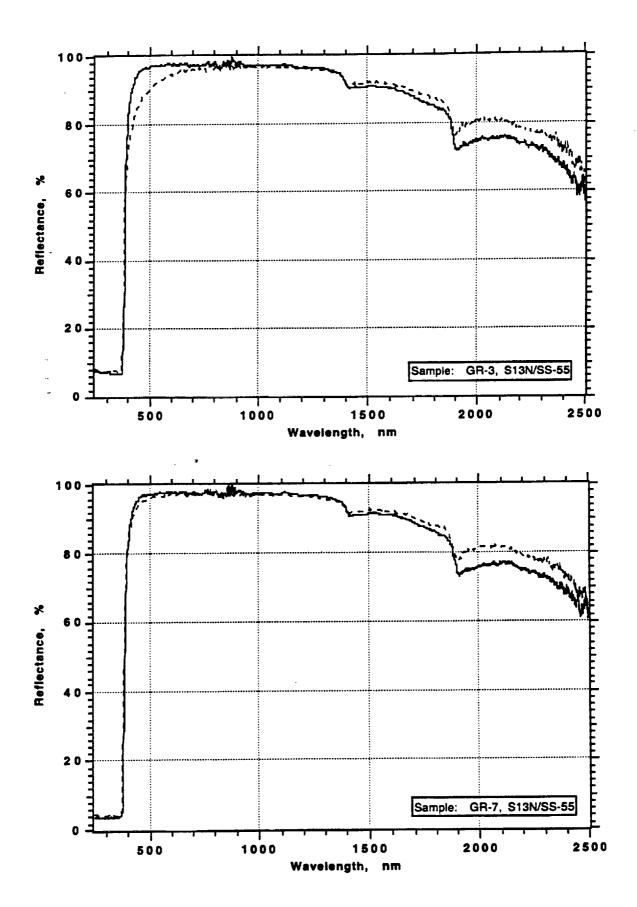


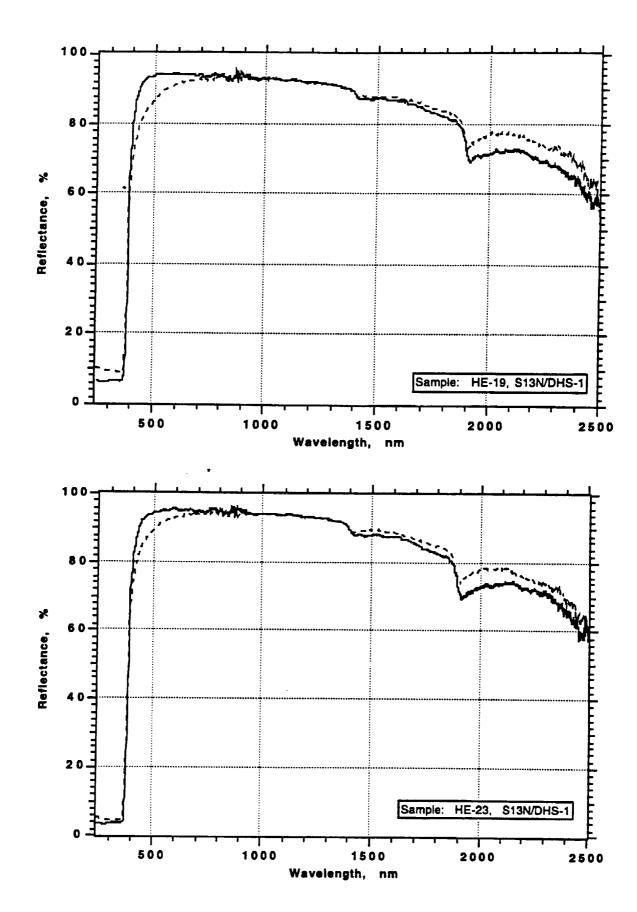


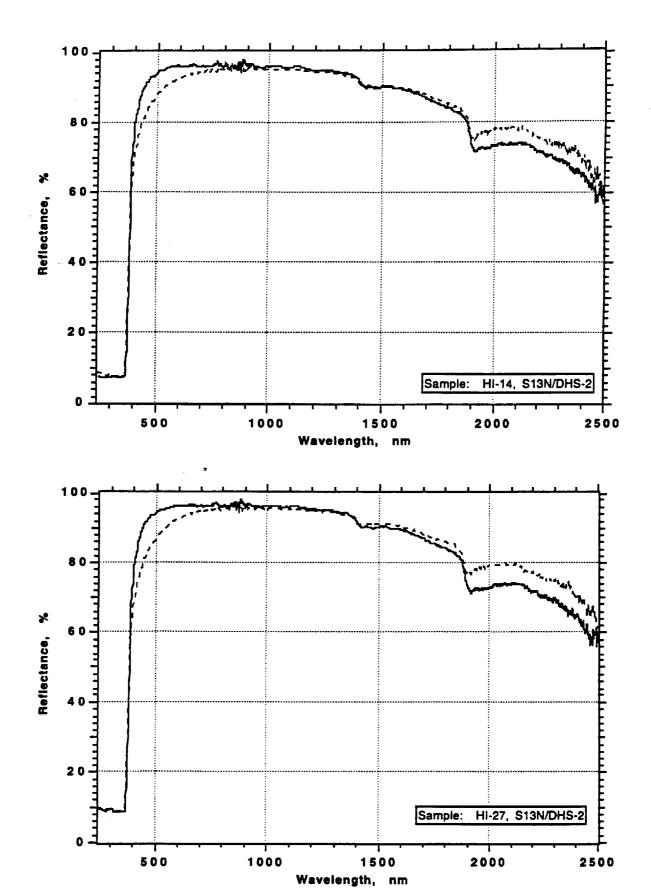


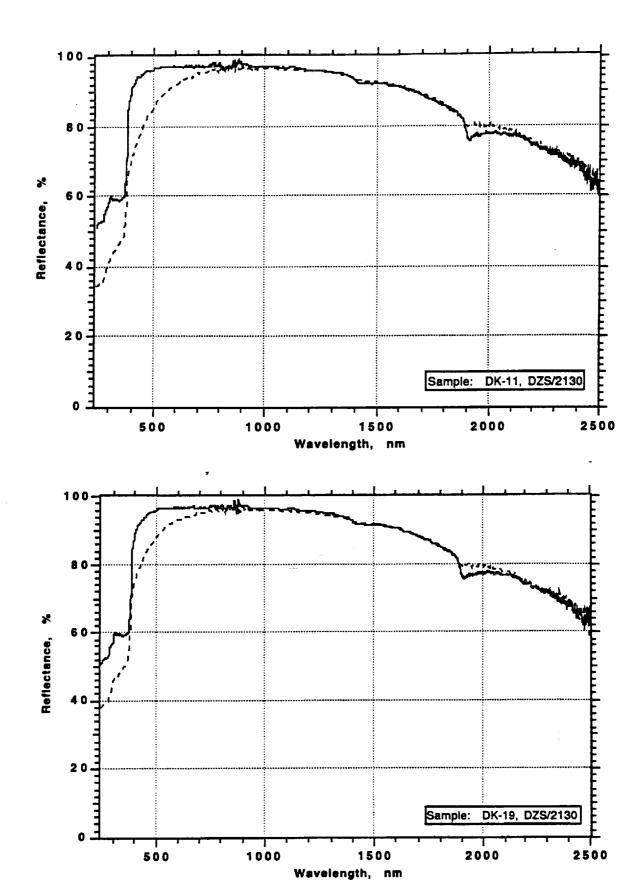


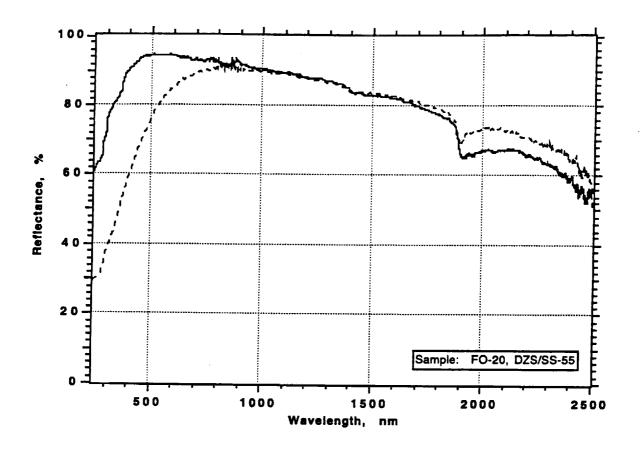


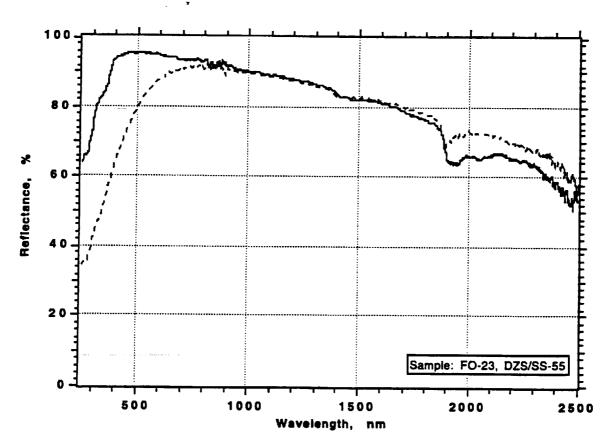


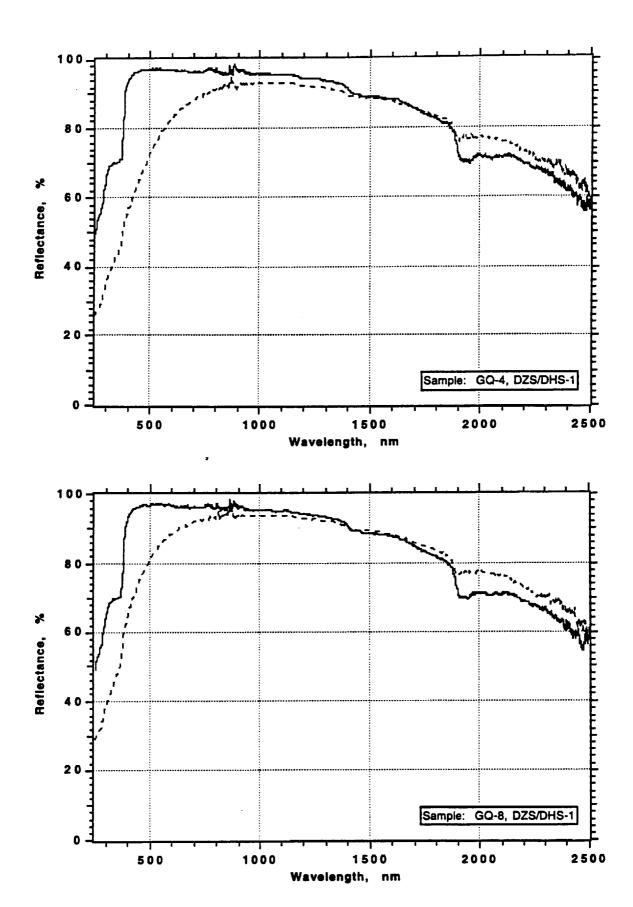


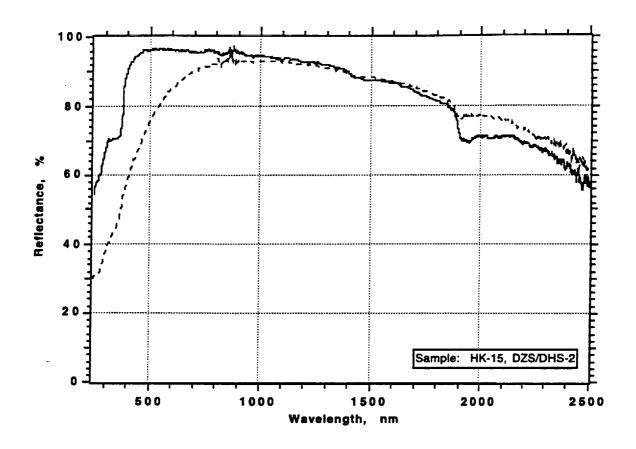


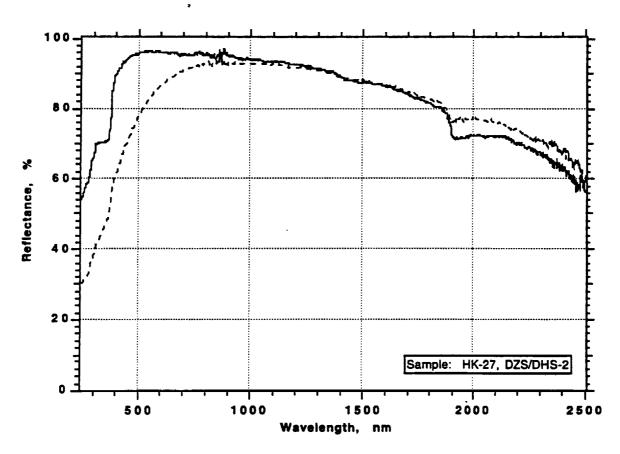


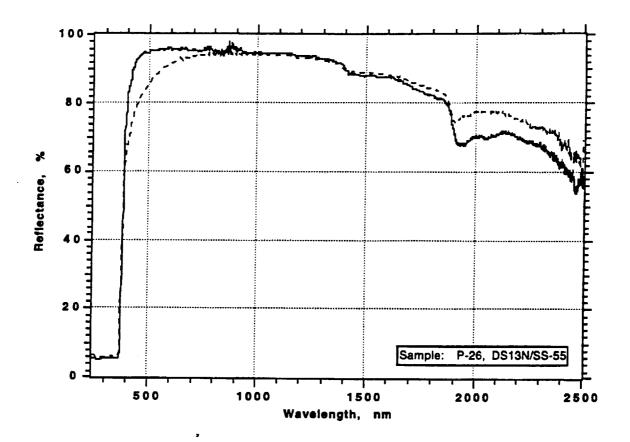


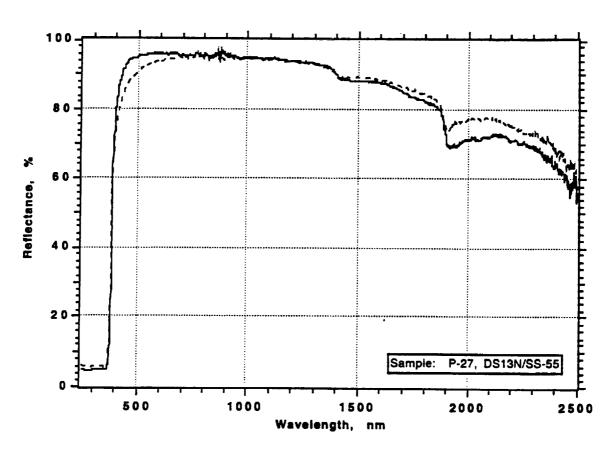


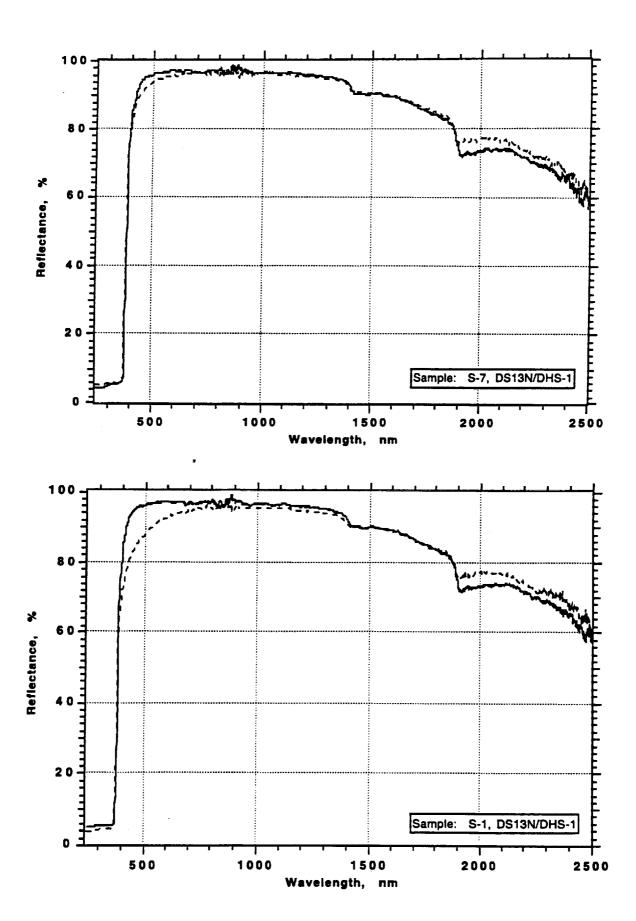


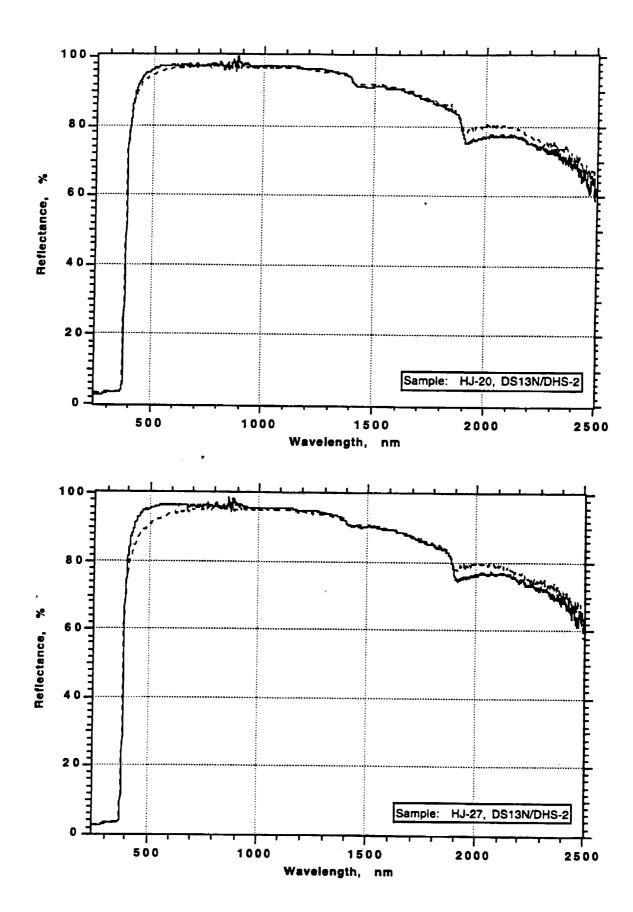


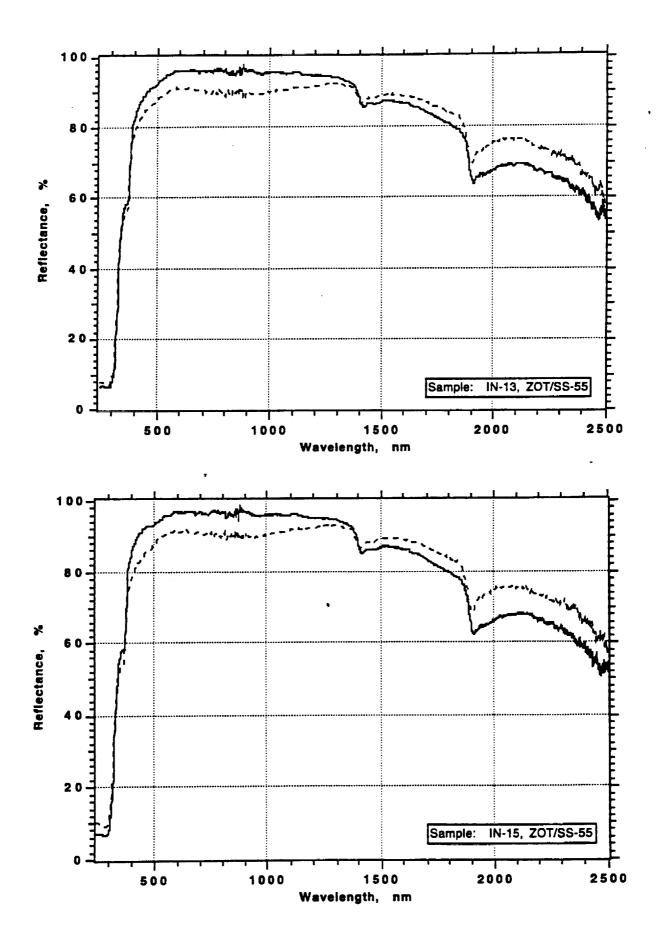


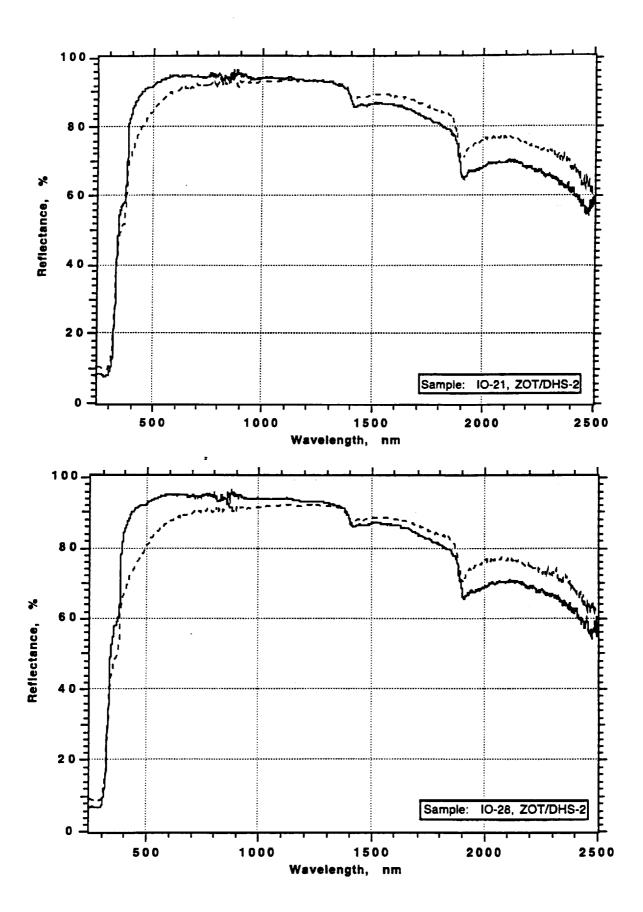


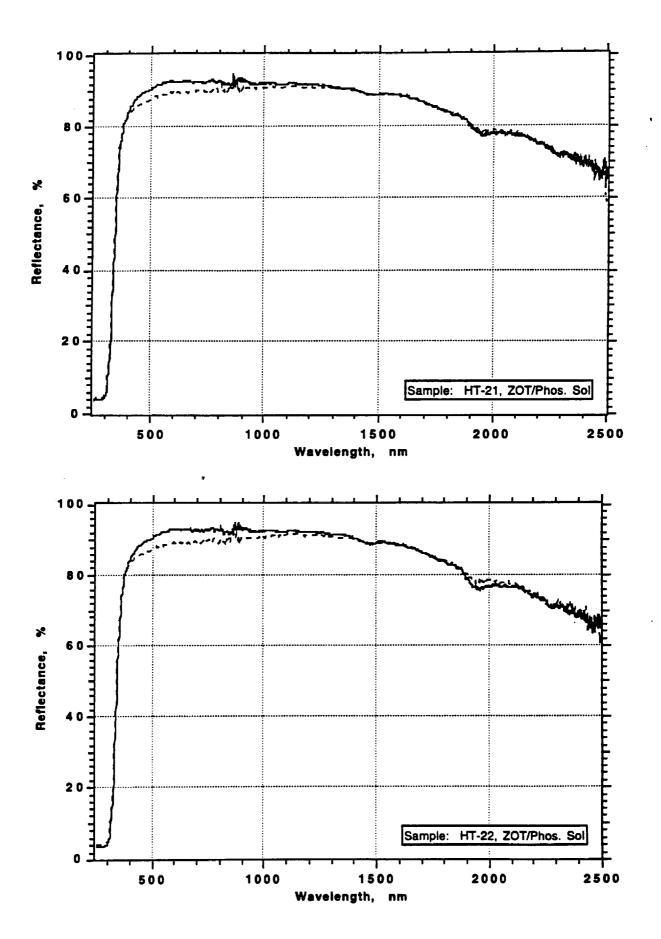


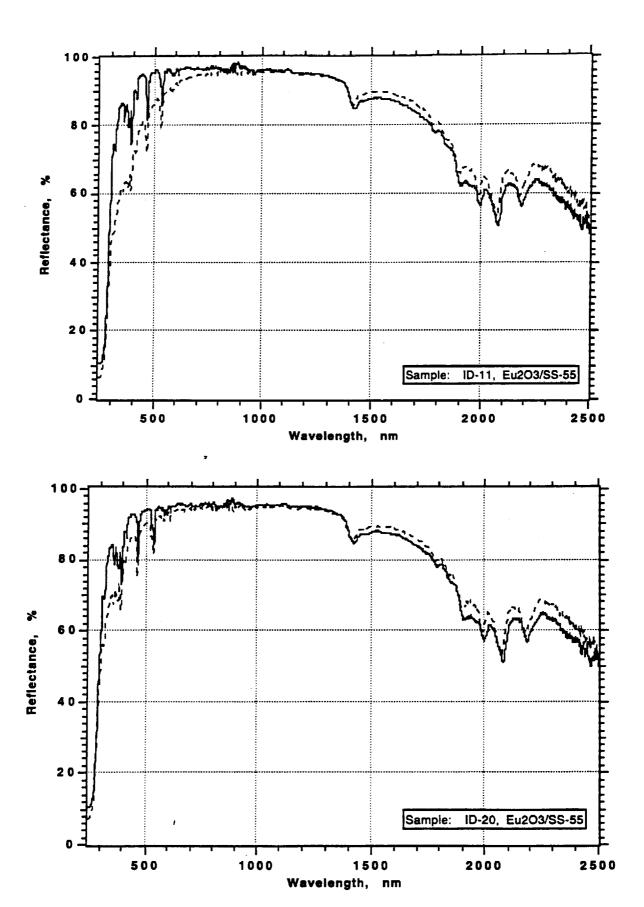


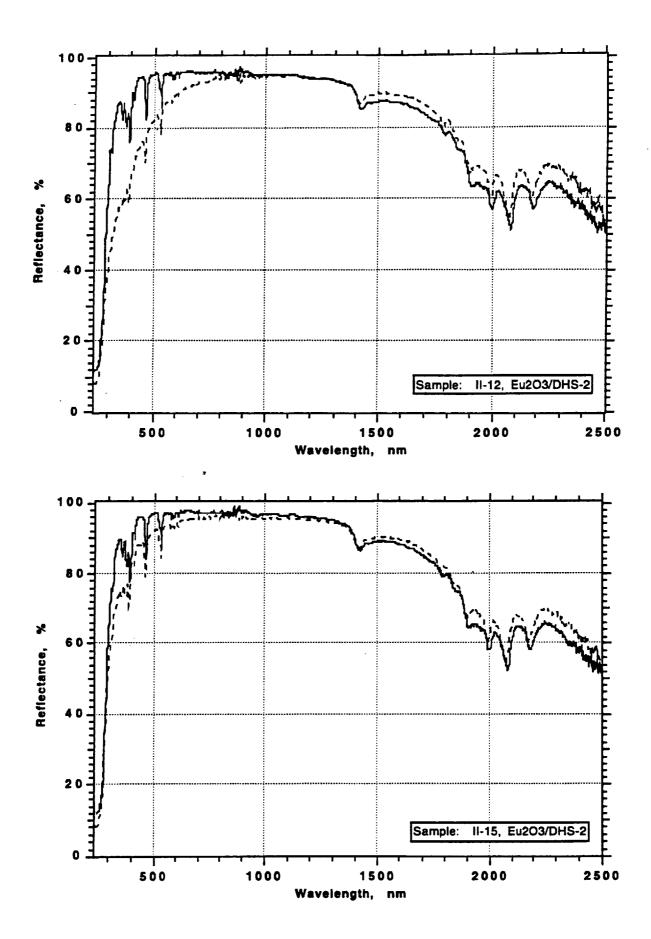


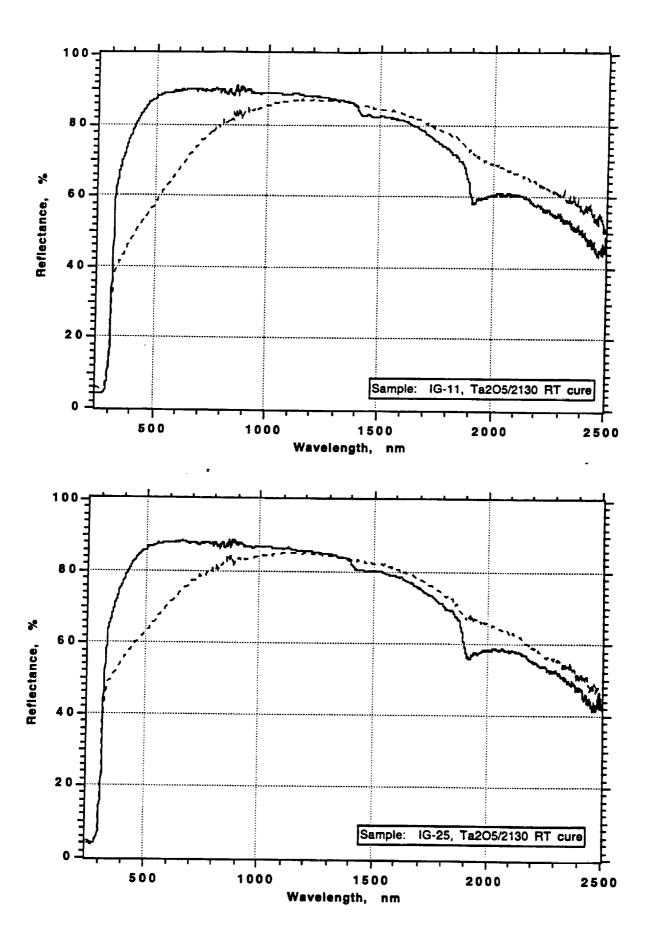


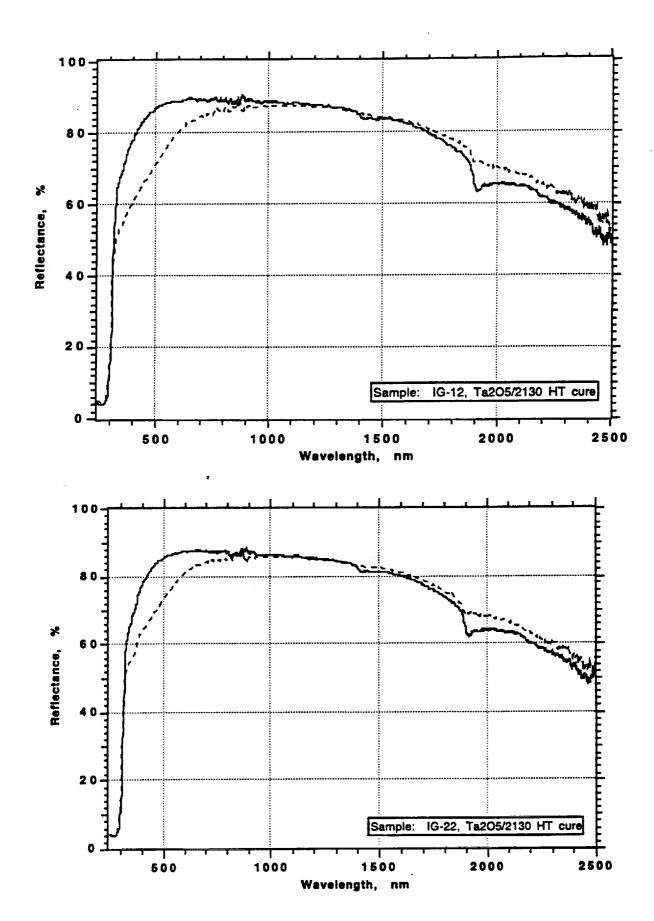


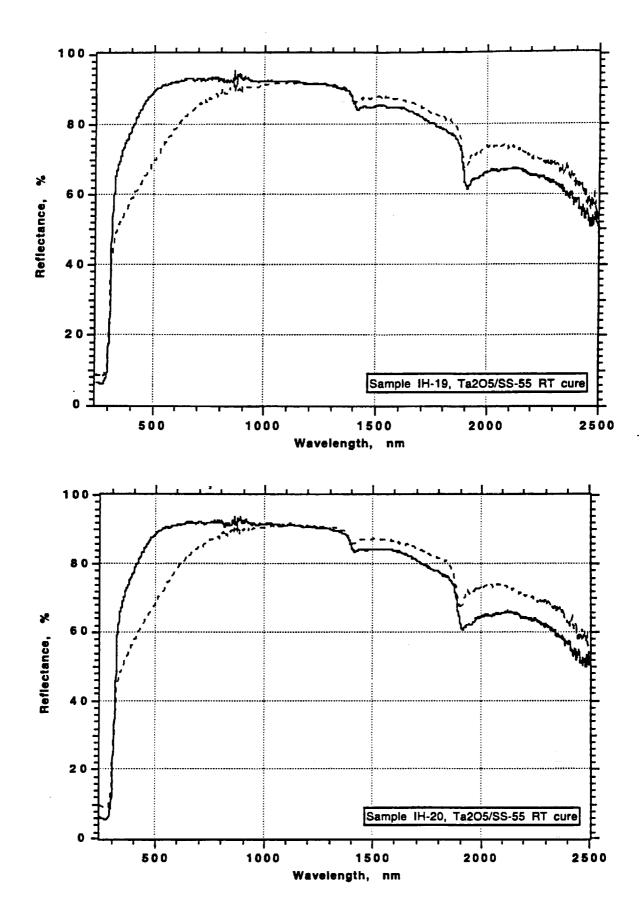


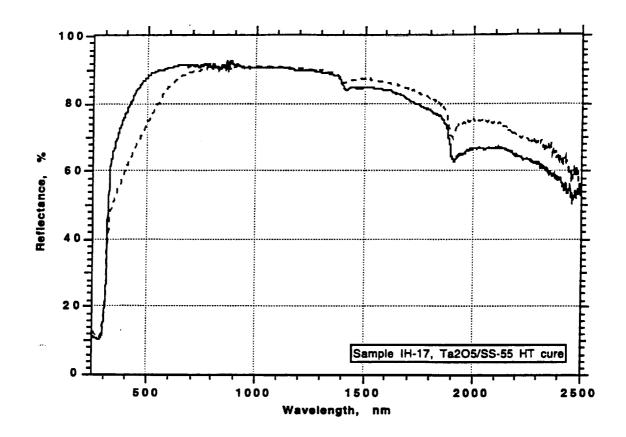


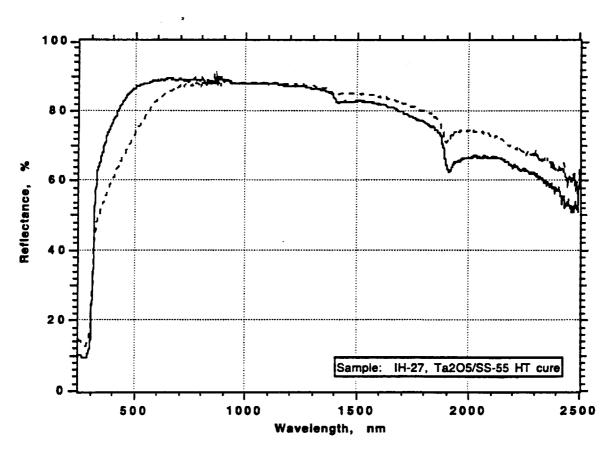


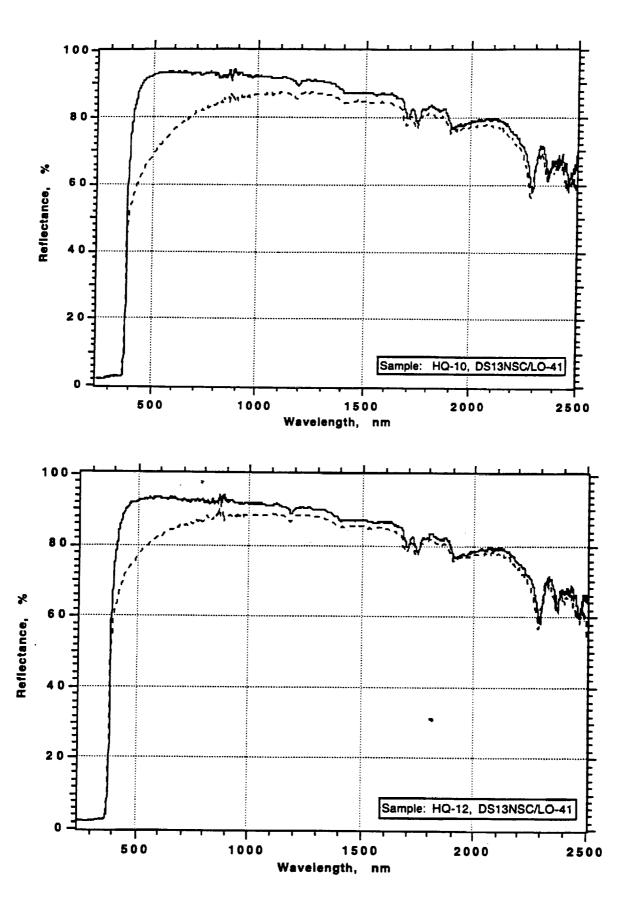


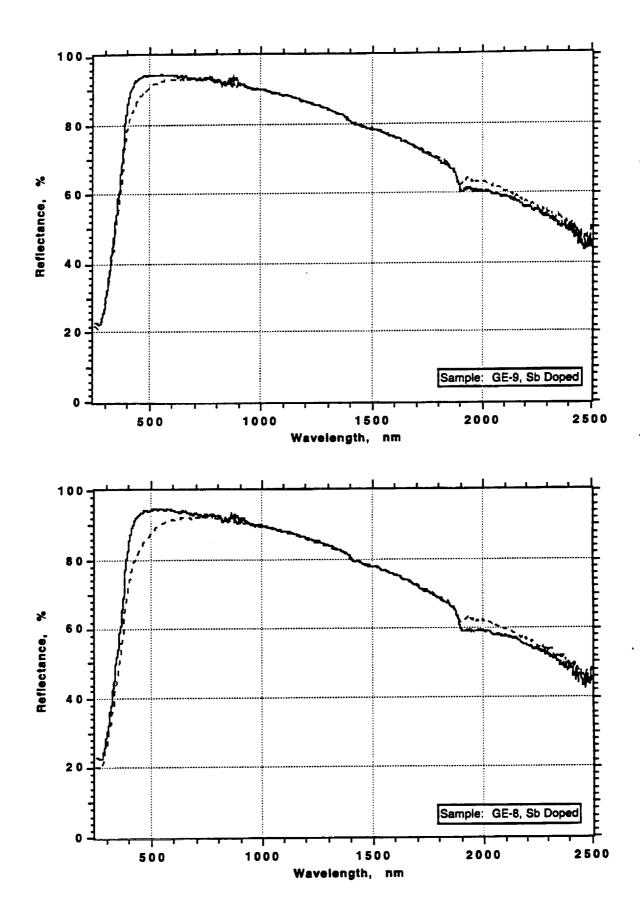


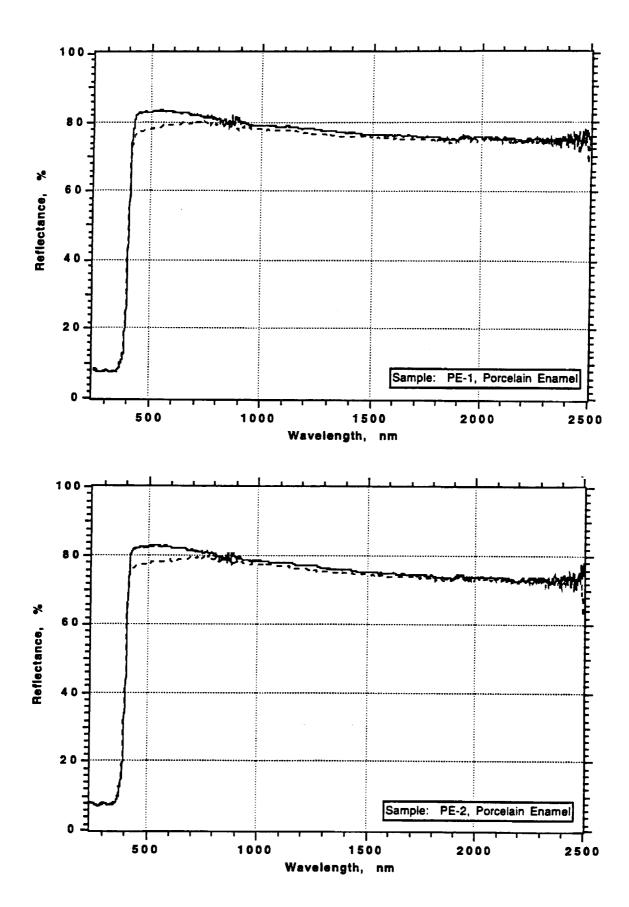


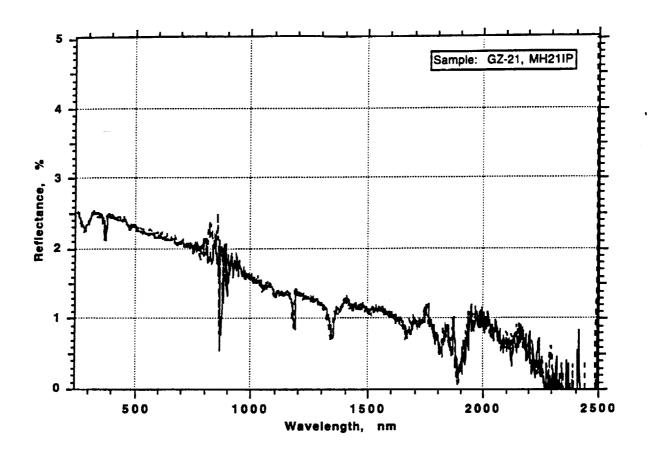


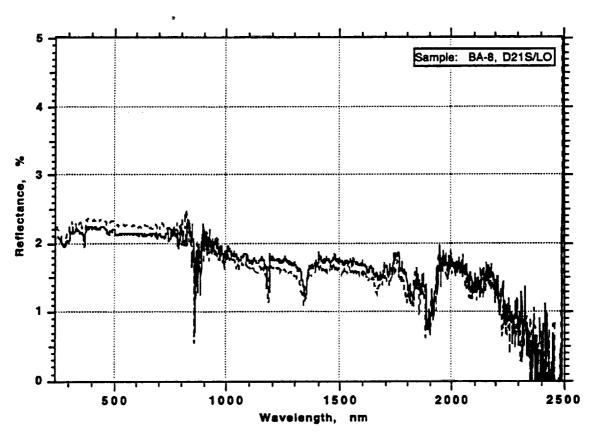


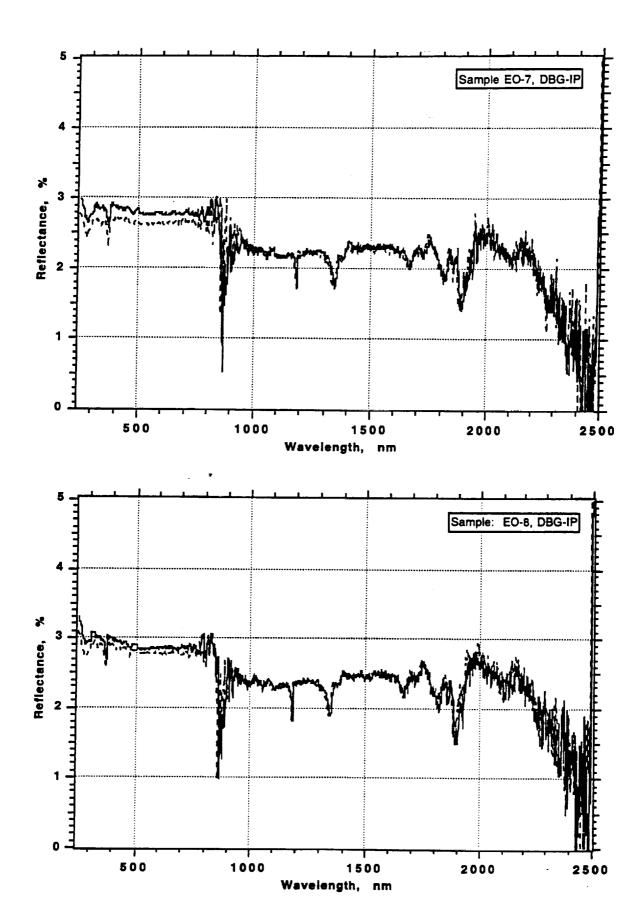


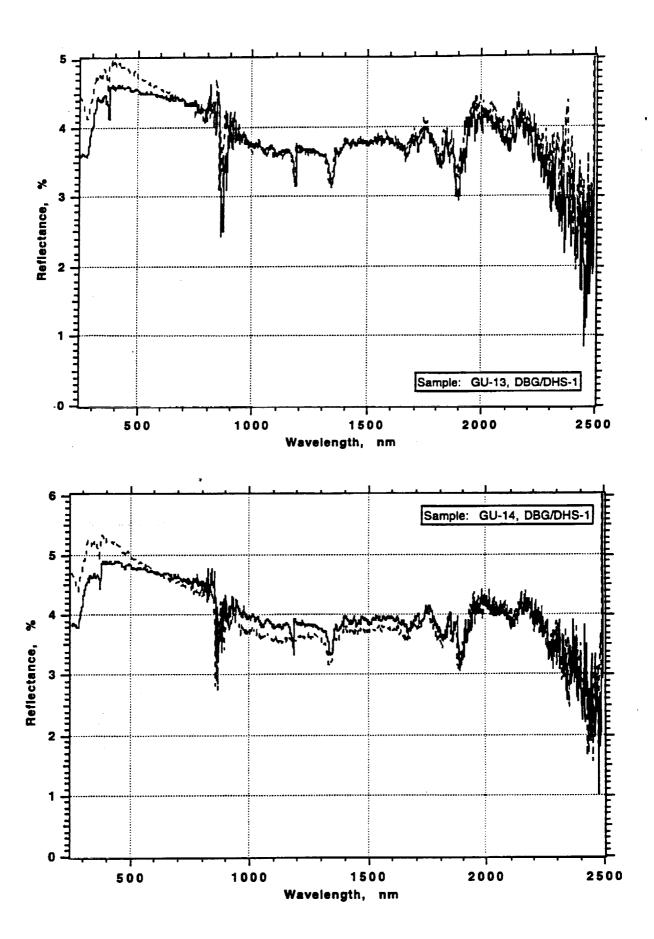


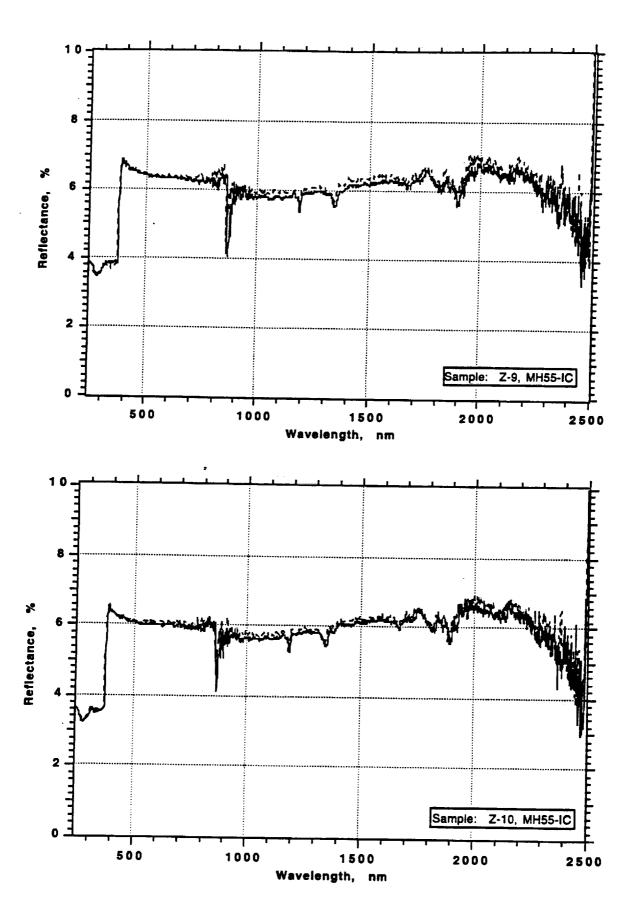


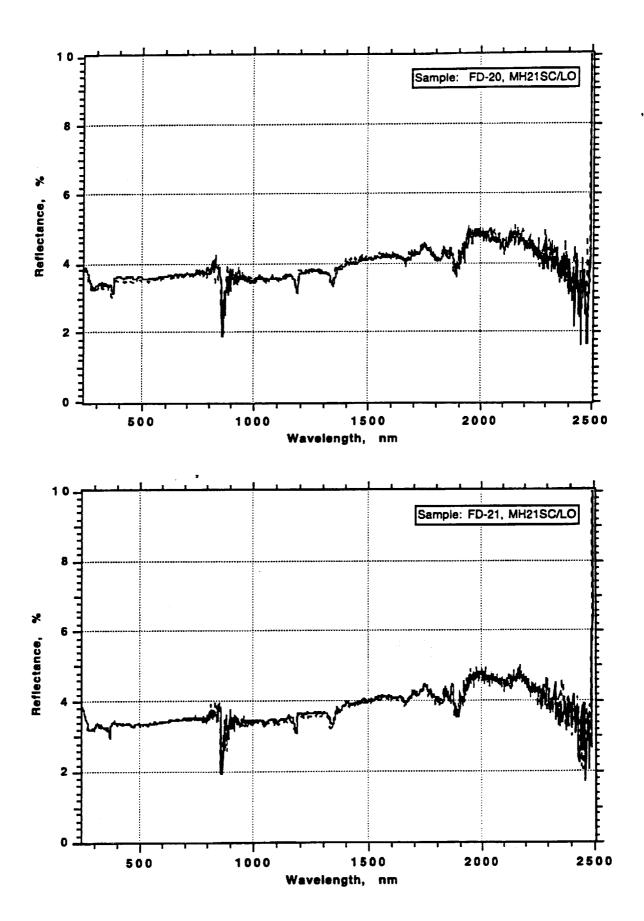


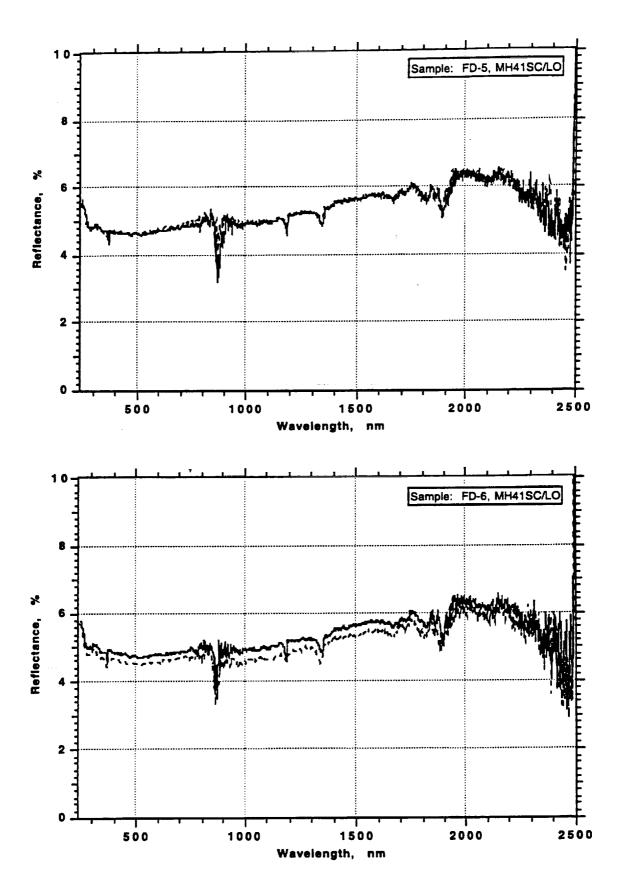


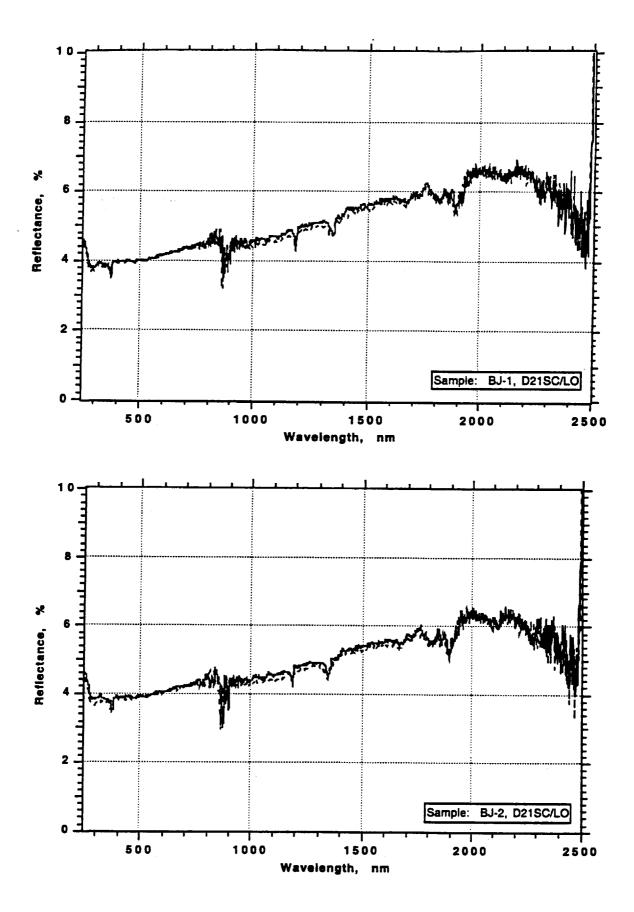


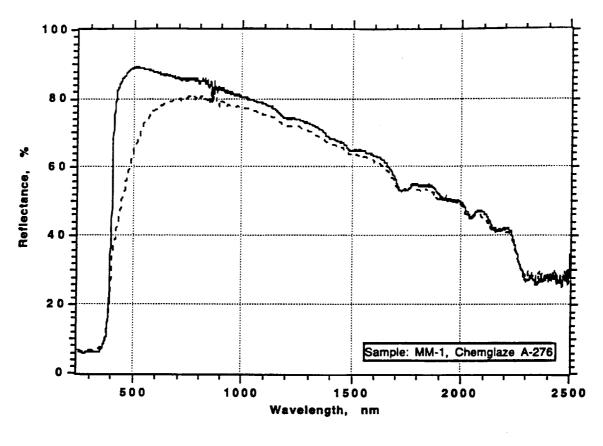


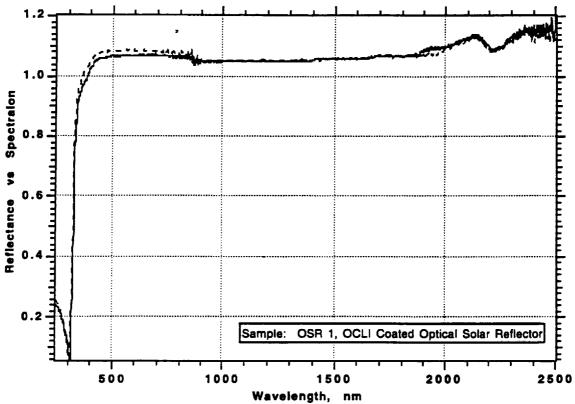


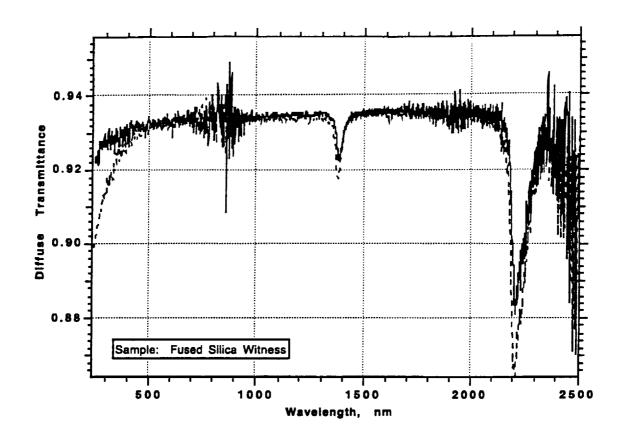












APPENDIX B

ULTRAVIOLET DEGRADATION STUDY OF IITRI'S SIX WHITE PAINTS

SWALES AEROSPACE

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Ultraviolet Degradation Study of IITRI's Six White Paints

5 Aug 1996

Prepared by:

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ULTRAVIOLET DEGRADATION STUDY OF SIX WHITE PAINTS

INTRODUCTION

In July 1995, six (6) formulations of thermal control white paints were selected for ultraviolet (UV) degradation testing by the NASA, Goddard Space Flight Center's (GSFC) Thermal Engineering Branch, Code 724. The purpose of the test was to obtain information on the UV stability of the paints. UV stability is critical to the space application of thermal control white paints. The results of the test will assist in the qualification of the white paints for space flight use. This report contains the test conditions and results of the UV degradation study.

BACKGROUND

The space program uses white paints for controlling temperatures of spacecraft systems and calibrating space optical systems. Some of the commonly used white paints, for space flight missions, are Z93, S13GLO-1, and YB71 (ZOT). These three paints are manufactured by Illinois Institute of Technology Research Institute (IITRI). In 1989, IITRI began researching a replacement for the binder, potassium silicate (PS-7), that was currently used in the formulation of the three white paints. The company, Osram Sylvannia, that supplied IITRI with the binder had discontinued the manufacture of PS-7 for economic reasons. Therefore, a replacement for the binder was necessary because of DOD and NASA current use of the three white paints.

In 1991, the Space and Missile Systems Center's Chief Engineer's Office funded IITRI to reformulate the white paints. The reformulate paints were designated as: Z93P, S13GP/LO-1, and YB71P. At the completion of the first phase of the reformulation process, IITRI with support from Wright Laboratory, Aerospace Corporation, NASA and industry performed test to requalify the reformulated paints. Information obtained from the testing of the reformulated paints were presented at the Aerospace Corporation/Space and Missile Systems Center Industry Briefing for IITRI White Thermal Control Coating Requalification in October 1994. As a result of the requalification process, IITRI decided to replace PS-7 with a potassium silicate binder called Kasil 2130. The requalification process also revealed that further reformulation and testing of the paints was needed. Replacement of Kasil 2130 in Z93P and S13GP/LO-1 gave similar test results to test results of the original formulations. Whereas, testing showed that substitution of Kasil 2130 in YB71P was not an acceptable replacement. During this period, IITRI was also developing a series of conductive paints called Z93SC55 and Z93SC1655.

The IITRI white paints requalification process revealed the need for further simulated space environment testing. Therefore NASA - GSFC's Thermal Engineering Branch, Code 724, requested ultraviolet testing of all newly formulated and reformulated white paints.

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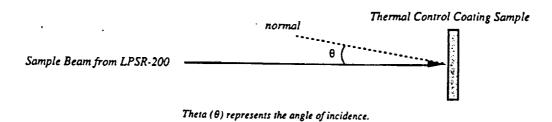


Diagram 1: Reflectance Measurement's Angle of Incidence

The Geir-Dünkle DB-100 InfraRed Reflectometer was used to measure the normal emittance of the samples. The Geir-Dünkle DB-100 measures the normal emittance of the surface from 5 to 40 microns. This measurement is made at room temperature.

The Perkin-Elmer Lambda 9 NearIR/Vis/UV Spectrophotometer was used to measure all "in-situ" measurements. The Lambda 9 measures the reflectance of the sample's surface from 250 to 2500 nm. The light source of the Lambda 9 passes through the sapphire chamber window in order to measure the reflectance of the sample while in the Multisedes chamber. Therefore, these reflectance measurements made by the Lambda 9, while the samples are located in the Multisedes sample chamber, are influenced by the reflectance of the sapphire window.

The Spectrolab X25 solar simulator was used to simulate solar UV light. The X25 solar simulator has a two sun capability. The level of sun intensity radiated by the X25 is used to determines the number of equivalent sun hours (ESH) during a specified period of time. The IL1700 Radiometer is used to measure the sun intensity level of the X25 solar simulator. From the IL1700 reading, the ESH is calculated as a function of time.

AZ Technology's LPSR-200 was used to perform the post-vacuum reflectance measurements and solar absorptance calculations. The solar absorptance is calculated in accordance with ASTM E903-82. The LPSR-200 measures the reflectance from 250 to 2500 nm of the sample's surface at a 15° angle of incidence. The instrument measurement accuracy is $^+/_-$.02 for both α and ϵ_n values.

TEST PROCEDURE

Before the sample is placed in the Multisedes chamber, the sample's reflectance is measured using the AZ Technology's LPSR-200 to obtained the pre-vacuum solar absorptance value. The reflectance data is graphed and the solar absorptance (α) is calculated and recorded. The normal emittance (α) of the sample is measured and recorded.

At the completion of the pre-vacuum measurements, the samples are then placed into the Multisedes sample chamber. The samples are mounted against a metal cold plate, which has a water reservoir located directly behind the metal plate. The water reservoir is used to maintain the Multisedes' samples at room temperature. When the test samples are all mounted, the chamber is sealed. The following diagram denotes the positions of the samples location in the Multisedes.

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TEST OBSERVATIONS

| Date | Comments |
|---|---|
| 1 Aug 1995 | Six white paint control and test samples' reflectances, solar absorptances, and normal emittances were measured. |
| 7 Aug 1995 | Pre-testing photographs were taken of the control and test samples. |
| 10 Aug 1995 | Test samples were placed in the Multisedes chamber. |
| 16 Aug 1995 | Began "in air" reflectance measurements of six test samples. |
| 17 Aug 1995 | Completed "in air" measurements. |
| *************************************** | Began evacuating the Multisedes system. |
| 18 Aug 1995 | Began 0 ESH "in-situ" reflectance measurements. |
| 21 Aug 1995 | Completed 0 ESH "in-situ" measurements. |
| *************************************** | Began UV exposure. |
| 2 Sep 1995 | Made 505 ESH "in-situ" reflectance measurements. |
| 14 Sep 1995 | Made 1010 ESH "in-situ" reflectance measurements. |
| 15 Sep 1995 | Back-filled chamber with gaseous nitrogen and removed samples from Multisedes. Made 1010 ESH reflectance, solar absorptance, and normal emittance measurements. |
| | Visual Examination of Samples: |
| | YB71P*2 showed signs of discoloration (tanish beige) |
| *************************************** | - Z93SC1655 showed discoloration (beige) |
| 15 Sep 1995 | Continued UV exposure of samples. |
| 18 Sep 1995 | UV exposure was halted for recalibration of the lamp and optics. |
| 21 Sep 1995 | S13GP/LO-1 sample was removed for reflectance measurement. During correlation of the data measured on September 15, 1995, it was discovered that this sample had not been measured. The test had reached 1137 ESH of UV exposure. |
| · | The S13GP/LO-1 sample was replaced in the chamber and the system was evacuated. |

TEST RESULTS

The solar absorptance (α) and normal emittance (ϵ_n) values of the samples were measured at 0 ESH, 1010 ESH, 2041 ESH, 3042 ESH and 4331 ESH. These measurement were made after removal of the test samples from the Multisedes chamber.

Table 2 contains the measured α and ϵ_n values of the non-conductive white paint samples. Table 3 contains the measured α and ϵ_n values of the conductive white paint samples. Table 4 contains the $\Delta\alpha$ of the six white paints. Reflectance curves of each paint test sample are located in the appendix. A comparison curve of the change in solar absorptance over time is also located in the appendix.

Table 2 - Non-Conductive White Paints

| Sample Description | Optical Properties | 0 ESH | 1010 ESH | 2041 ESH | 3042 ESH | 4331 ESH |
|-------------------------|-----------------------|-------|-------------------|----------|----------|-------------------|
| S13GP/LO-1 (Kasil 2130) | α | .181 | .186* | .184 | .195 | .192 |
| | ε _n | .922 | No Measurement | .919 | .917 | .918 |
| Z93P (Kasil 2130) | α | .172 | .175 | .178 | .181 | .189 |
| | , ε _n | .930 | .931 | .929 | .928 | .929 |
| YB71P*2 (Kasil 2130) | α | .141 | .259 | .393 | .524 | sample removed |
| t 512CD // O t | ϵ_{n} | .933 | .932 | .927 | .928 | sample removed |

^{*} S13GP/LO-1 was inadvertently not measured at 1010 ESH. The sample was removed at 1137 ESH for measurement.

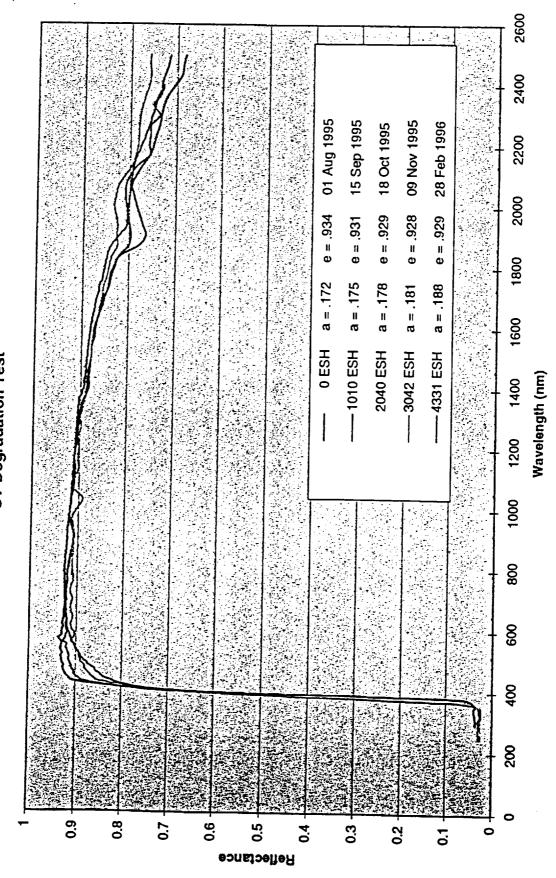
Table 3 - Conductive White Paints

| Sample Description | Optical Properties | 0 ESH | 1010 ESH | 2041 ESH | 3042 ESH | 4331 ESH |
|--------------------|-----------------------|-------|----------|----------|----------|-------------------|
| Z93SC55 (T251) | α | .140 | .151 | .157 | .175 | .177 |
| | $\varepsilon_{\rm n}$ | .935 | .929 | .930 | .931 | .929 |
| Z93SC55 (EI-2) | α | .151 | .164 | .177 | .200 | .207 |
| | $\epsilon_{\rm n}$ | .934 | .928 | .926 | .928 | .929 |
| Z93SC1655 (EK-20) | α | .128 | .220 | .261 | .304 | sample removed |
| | ε_{n} | .935 | .926 | .927 | .925 | sample removed |

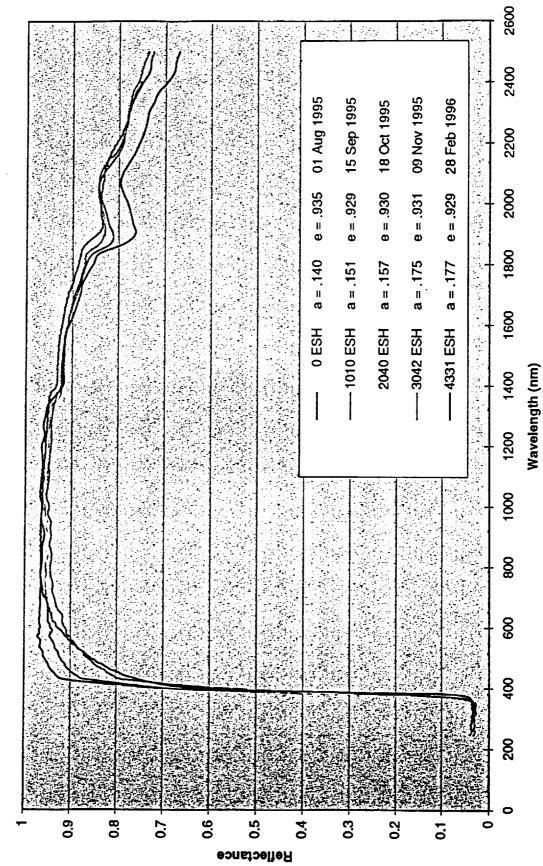
APPENDIX

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Z93P White Paint (Primer Z93P; Binder Kasil 2130) UV Degradation Test



Z93SC55 (T-251) Conductive White Paint UV Degradation Test



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2600 18 Oct 1995 01 Aug 1995 15 Sep 1995 09 Nov 1995 2400 2200 e = .927e = .934e = .926e = .9252000 a = .220a = .129a = .261a = .3041800 0 ESH 1010 ESH 2040 ESH 3042 ESH 1600 Wavelength (nm) 1400 1200 1000 8 009 8 200 0.9 0.7 0.5 0.3 0.4 0.2 0.1 Reflectance

Z93SC1655 #EK-20 Conductive Paint UV Degradation Test

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APPENDIX C

ULTRAVIOLET DEGRADATION STUDY OF IITRI ELECTRICALLY CONDUCTIVE THERMAL CONTROL FORMULATIONS

SWALES AEROSPACE

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Of IITRI Electrically Conductive Thermal Control Formulations

15 October 1996

Prepared by:

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ULTRAVIOLET DEGRADATION STUDY OF IITRI ELECTRICALLY CONDUCTIVE THERMAL CONTROL FORMULATIONS

INTRODUCTION

In February 1996, fourteen (14) conductive conceptual formulations of thermal control coatings were submitted for ultraviolet (UV) degradation testing by the NASA, Goddard Space Flight Center's (GSFC) Thermal Engineering Branch, Code 724. The fourteen paint samples were developed and supplied by IIT Research Institute (IITRI). The purpose of the test was to obtain information on the UV stability of the paints. UV stability is critical to the space application of thermal control white paints. The results of this test will assist in determining future qualification test. This report contains the test conditions and results of the UV degradation study.

TEST INFORMATION

Test Requestor:

NASA - Marshall Space Flight Center (MSFC)

Project:

None

Type of Test:

Ultraviolet Degradation Study 1000 Equivalent Sun Hours (ESH)

Test Duration: Results Required:

Reflectance, Solar Absorptance, Normal Emittance

Special Instructions:

Measure reflectance, solar absorptance, and normal emittance "in-situ" reflectance measurement, removal and

thermal-optical properties measurements at 1000 ESH.

SAMPLE DESCRIPTION

Table 1: White Coatings.

| Concept Name | Batch # | Pigment | Binder |
|--|---------|---|---------------------------------------|
| S13GP/DHS-2 | HH-19 | S13GP | DHS-2(a) |
| Sb Doped | GE-12 | ZnO + Sb doped Zn ₂ SnO ₄ (b) | Kasil 2130 |
| S13N/DHS-2 | HI-21 | S13N ^(c) | DHS-2 |
| DS13N/LO-51 | JX-16 | DS13N | MHS/LO stripped polydimethyl siloxane |
| DS13N/DHS-2 | HJ-25 | DS13N | DHS-2 |
| DS13N/SS-55 | P-31 | DS13N(d) | SS-55(e) |
| ZnO(FC)/DHS-2 | HL-26 | ZnO (FC) | DHS-2 |
| Eu ₂ O ₃ */SS-55 | HD-26 | Eu ₂ O ₃ (f) | SS-55 |
| ZOT/SS-55 | IW-19 | ZOT # 98790 | SS-55 |

Table 2: Black Coatings

| Concept Name | Batch # | Pigment | Binder | | |
|--------------|-------------|--|---------------------------------------|--|--|
| D36SCB/LO | JV-17 | Cosmic black-WB-500, B ₄ C | MHS/LO stripped polydimethyl siloxane | | |
| MS41SCB/LO | U-356 IC-20 | Black glass/B ₄ C | MHS/LO stripped polydimethyl siloxane | | |
| MH21SC/LO | GX-16 | Black glass, graphite fiber, Graphite 9035 | MHS/LO stripped polydimethyl siloxane | | |
| MH55-IC | Z-6 | Black glass, Graphite 9035, ZnO | Kasil 2130 | | |
| DBG-IP | GTM-29 | Doped black glass (Indium doped) | Kasil 2130 | | |

Sample Size:

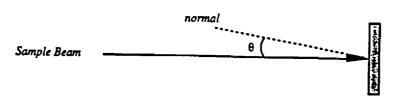
One inch diameter circular disc

Control samples are maintained for comparison purposes at the completion of the UV degradation test. The control samples are stored in a dark environment to prevent UV exposure from fluorescent lights.

TEST INSTRUMENTATION

The GSFC Thermal Engineering Branch's Multisedes Vacuum System was used for the UV degradation studies. The system consists of two sorption pumps, a sputter-ion pump, and a sample chamber. The sample chamber consists of 16 testing positions. The system has the capability of testing a maximum of 14 samples at one time. The remaining two testing positions contain a reference mirror and a blank.

The Beckman DK-2 Spectrophotometer was used to perform the pre-vacuum reflectance measurements. The DK-2 measures the reflectance from 290 to 2400 nm of the sample's surface at a 20° angle of incidence. A computer program is used to calculate the solar absorptance in accordance with ASTM E903-82.



Theta (θ) represents the angle of incidence.

Diagram 1: Reflectance Measurement's Angle of Incidence

The Geir-Dünkle DB-100 InfraRed Reflectometer was used to measure the normal emittance of the samples. The Geir-Dünkle DB-100 measures the normal emittance of the surface from 5 to 40 microns. This measurement is made at room temperature.

The Perkin-Elmer Lambda 9 NearIR/Vis/UV Spectrophotometer was used to measure all "in-situ" measurements. The Lambda 9 measures the reflectance of the sample's surface from 250 to 2500 nm. The light source of the Lambda 9 passes through the sapphire chamber window in order to measure the reflectance of the sample while in the Multisedes chamber. Therefore, these reflectance measurements, while the samples are located in the Multisedes sample chamber, are influenced by the reflectance of the sapphire window.

The Spectrolab X25 solar simulator was used to simulate solar UV light. The X25 solar simulator has a two sun capability. The level of sun intensity radiated by the X25 is used to determines the number of equivalent sun hours (ESH) during a specified period of time. The IL1700 Radiometer is used to measure the sun intensity level of the X25 solar simulator. From the IL1700 reading, the ESH is calculated as a function of time.

AZ Technology's LPSR-200 was used to perform the post-vacuum reflectance measurements and solar absorptance calculations. The solar absorptance is calculated in accordance with ASTM E903-82. The LPSR-200 measures the reflectance from 250 to 2500 nm of the sample's surface at a 15° angle of incidence. The instrument measurement accuracy is $^{+}/_{-}$.02 for both α and ϵ_{n} values.

TEST PROCEDURE

Before the sample is placed in the Multisedes chamber, the sample's reflectance is measured using the AZ Technology's LPSR-200 to obtained the pre-vacuum solar absorptance value. The reflectance data is graphed and the solar absorptance (α) is calculated and recorded. The normal emittance (ϵ_n) of the sample is measured and recorded.

At the completion of the pre-vacuum measurements, the samples are then placed into the Multisedes sample chamber. The samples are mounted against a metal cold plate, which has a water reservoir located directly behind the metal plate. The water reservoir is used to maintain the Multisedes' samples at room temperature. When the test samples are all mounted, the chamber is sealed. The following diagram denotes the positions of the samples location in the Multisedes.

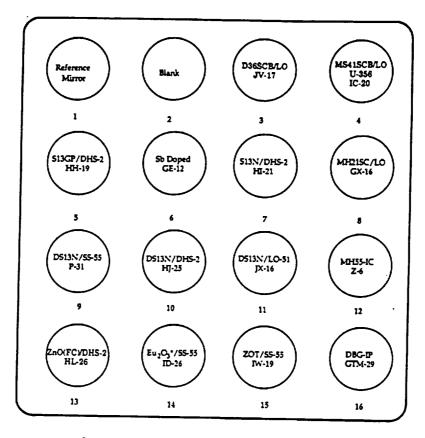


Diagram 2: Thermal Coating Samples UV Test Positions

Before evacuation begins, an "in air" reflectance measurement is taken of each sample using the Perkin-Elmer Lambda 9 NearIR/Vis/UV Spectrophotometer. At the complete of the "in air" measurement, evacuation of the system is started.

When the system pressure reaches 2×10^{-5} torr, the zero equivalent sun hours (0 ESH) reflectance measurement is made of each sample. The 0 ESH is the first reflectance measurement made of the sample while under vacuum. After the 0 ESH measurements are completed, UV exposure begins.

The X-25 solar simulator is turned on to begin UV exposure. At specified ESH intervals, "in-situ" reflectance measurements are taken of each sample. At 1000 ESH, the reflectance of the samples are measured "in-situ" and then removed from the Multisedes for the post-vacuum reflectance, solar absorptance, and normal emittance measurements. The chamber is back-filled with gaseous nitrogen prior to removal of the samples.

TEST OBSERVATIONS

| Date | Comments |
|---|---|
| 22 Mar 1996 | Nine (9) white paint control and test samples' reflectances, solar absorptances, and normal emittances were measured. |
| *************************************** | Five (5) black paint control and test samples' reflectances, solar absorptances, and normal emittances were measured. |
| 2 Apr 1996 | Test samples were placed in the Multisedes chamber. |
| *************************************** | Began "in air" measurements of paint samples. |
| 3 Apr 1996 | Completed "in air" measurements. |
| | Began evacuating the Multisedes system. |
| 4 Apr 1996 | Made 0 ESH "in-situ" measurements. |
| *************************************** | Began UV exposure. |
| 10 Apr 1996 | Made 256 ESH "in-situ" reflectance measurements. |
| | DS13N/LO-51 was beige in color. |
| 3 May 1996 | Made 674 ESH "in-situ" reflectance measurements. |
| .• | DS13N/LO-51 was brown in color. |
| *************************************** | Experienced problem with the PE Lambda 9 during "in-situ" reflectance measurements. |
| 9 May 1996 | Corrected PE Lambda 9 problem. Completed 674 ESH "in-situ" reflectance measurements. |
| 20 May 1996 | Made 1068 ESH "in-situ" reflectance measurements. |
| • | Back-filled chamber with gaseous nitrogen and removed samples from Multisedes. Made 1068 ESH reflectance, solar absorptance, and normal emittance measurements. |

TEST RESULTS

The solar absorptance (α) and normal emittance (ϵ_n) values of the samples were measured at 0 ESH, 256 ESH, 673 ESH, and 1068 ESH. These measurement were made after removal of the test samples from the Multisede chamber.

Table 2 contains the measured α and ϵ_n values of the nine white paint samples. Table 3 contains the measured α and ϵ_n values of the five black paint samples. Table 4 contains the $\Delta\alpha$ of all 14 paints. Reflectance curves of the test samples are located in the appendix.

Table 2 - White Paints

| Sample Description | Optical Properties | 0 ESH | 1068 ESH |
|--|-----------------------|-------|----------|
| S13GP/DHS-2 [HH-19] | α | .148 | .165 |
| | ϵ_{n} | .934 | .932 |
| Sb Doped [GE-12] | α | .156 | .243 |
| · | ε _n | .930 | .931 |
| S13N/DHS-2 [HI-21] | α | .152 | .162 |
| | $\epsilon_{ m n}$ | .935 | .932 |
| DS13N/LO-51 [JX-16] | α | .210 | .434 |
| | $\epsilon_{\rm n}$ | .926 | .926 |
| DS13N/DHS-2 [HJ-25] | α | .148 | 154 |
| | $\epsilon_{ m n}$ | .937 | .935 |
| DS13N/SS-55 [P-31] | α | .152 | .172 |
| | επ | .930 | .930 |
| ZnO(FC)/DHS-2 [HL-26] | α | .158 | .168 |
| | $\epsilon_{\rm n}$ | .925 | .925 |
| Eu ₂ O ₃ */SS-55 [HD-26] | α | .096 | .129 |
| | εη | .923 | .921 |
| ZOT/SS-55 [IW-19] | α | .129 | .180 |
| | εη | .881 | .881 |

Table 3 - Black Paints

| Sample Description | Optical Properties | 0 ESH | 1068 ESH | |
|--------------------------|-----------------------|--------------|----------|--|
| D36SCB/LO [JV-17] | α | .962 | .967 | |
| | ε _n | .908 | .909 | |
| MS41SCB/LO U-356 [IC-20] | α | .939 | .947 | |
| | ε _n | .866 | .867 | |
| MH21SC/LO [GX-16] | α | .951 | .955 | |
| | εη | .862 | .864 | |
| MH55-IC [Z-6] | α | .928 | .938 | |
| | $\epsilon_{\rm n}$ | .886 | .885 | |
| DBG-IP [GTM-29] | α | .95 <i>7</i> | .951 | |
| | ε _n | .911 | .914 | |

Table 4 - Change in Solar Absorptance After UV Exposure

| Sample Description | Δα after 1068 ESH |
|--|-------------------------|
| S13GP/DHS-2 [HH-19] | .017 |
| Sb Doped [GE-12] | .087 |
| S13N/DHS-2 [HI-21] | .010 |
| DS13N/LO-51 [JX-16] | .224 |
| DS13N/DHS-2 [HJ-25] | .006 |
| DS13N/SS-55 [P-31] | .020 |
| ZnO(FC)/DHS-2 [HL-26] | .010 |
| Eu ₂ O _{3*} /SS-55 [HD-26] | .033 |
| ZOT/SS-55 [IW-19] | .051 |
| D36SCB/LO [JV-17] | .005 |
| MS41SCB/LO U-356 [IC-20] | .008 |
| MH21SC/LO [GX-16] | .004 |
| MH55-IC [Z-6] | .010 |
| DBG-IP [GTM-29] | 006 |

DATA INTERPRETATION

After 1068 ESH of UV exposure, DS13N/LO-51 [JX-16] and Sb Doped [GE-12] showed significant degradation. Based on repeatability of the measurement (± .01), coatings with changes of .02 and greater may display long term instability when life time requirements of greater than ten (10) years are considered.

Wanda C. Peters, Senior Thermal Coatings Engineer

Distribution List:

724/Dennis Hewitt 724/Lonny Kauder

724/Ted Śwanson 724/Mark Kobel SAI/Jack Triolo SAI/Grace Miller

IITRI/Michael Desphande

APPENDIX

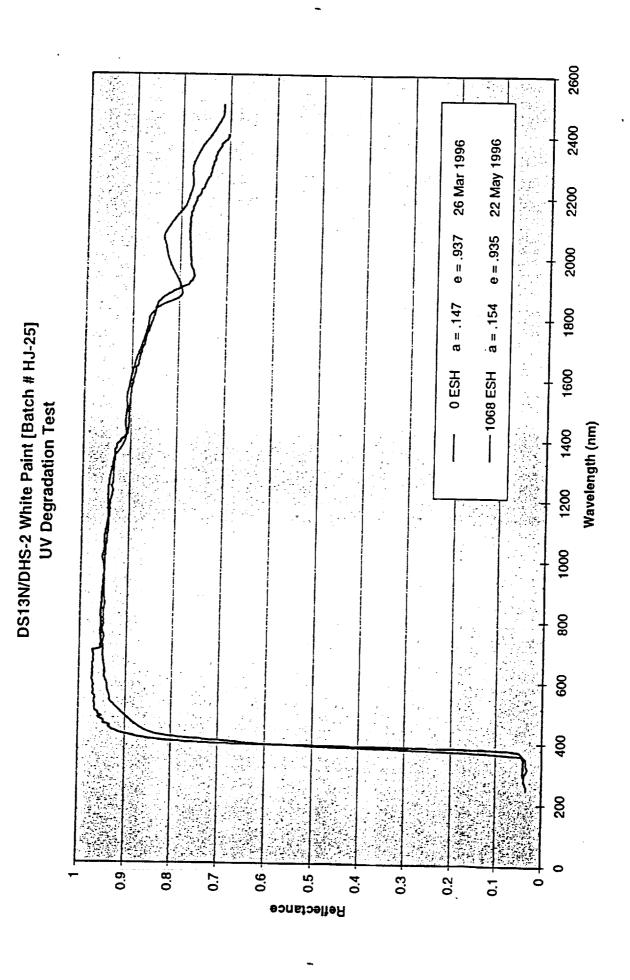
2600 22 May 1996 26 Mar 1996 2400 2200 e = .934e = .9322000 a = .148a = .1651800 S13GP/DHS-2 White Paint [Batch # HH-19] 0 ESH ---- 1068 ESH 1600 **UV Degradation Test** Wavelength (nm) 1400 1200 1000 800 900 400 200 0.8 9.0 0.7 0.5 0.2 Reflectance

22 May 1996 1 Apr 1996 2400 2200 e = .930e = .9312000 a = .156a = .2431800 0 ESH ---- 1068 ESH 1600 **UV Degradation Test** Wavelength (nm) 1400 1200 1000 8 8 400 200 0.9 0.8 9.0 0.5 0.7 0.4 0.3 0.2 0.1 eonstoelleR

Sb Doped White Paint [Batch # GE-12]

2600 2400 22 May 1996 26 Mar 1996 2200 e = .931e = .9352000 a = .162a = .1521800 S13N/DHS-2 White Paint [Batch # HI-21] - 1068 ESH 0 ESH 1600 **UV Degradation Test** Wavelength (nm) 1400 1200 1000 800 009 400 200 0.8 0.7 0.5 0.4 0.3 0.2 0.1 eonstoelteR

2600 2400 27 Mar 1996 22 May 1996 2200 e = .926e = .9262000 a = .434 1800 a = .210DS13N/LO-51 White Paint [Batch # JX-16] 1600 0 ESH - 1068 ESH **UV Degradation Test** Wavelength (nm) 1400 1200 1000 80 000 400 200 0.8 0.7 0.5 0.3 0.2 0.1 Reflectance



2600 2400 22 May 1996 26 Mar 1996 2200 e = .930e =.930 2000 a = .172a = .1511800 0 ESH 1600 - 1068 ESH **UV Degradation Test** Wavelength (nm) 1400 1200 1000 8 8 400 200 6.0 0.8 0.7 9.0 0.5 0.3 0.4 0.2 0.1 Reflectance

DS13N/SS-55 White Paint [Batch # P-31]

2600 2400 22 May 1996 1 Apr 1996 2200 e = .925e = .9252000 a = .168 a = .1581800 0 ESH - 1068 ESH 1600 **UV Degradation Test** Wavelength (nm) 1400 1200 1000 800 9 400 200 0.9 0.8 9.0 0.5 0.7 0.3 0.4 0.2 0.1 eonstoelleR

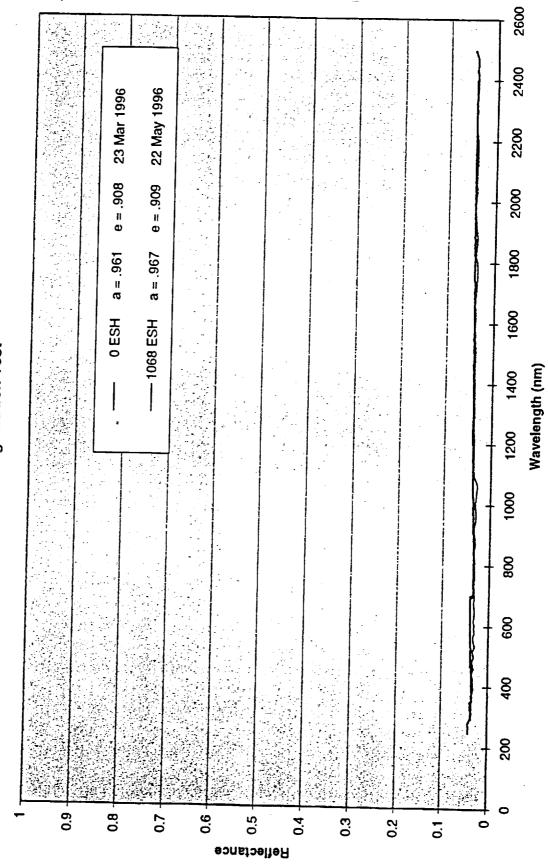
ZnO(FC)/DHS-2 White Paint [Batch # HL-26]

2400 22 May 1996 1 Apr 1996 2200 e = .923e = .9212000 a = .129 a = .0951800 Eu203*/SS-55 Whtie Paint [Batch # HD-26] 1600 0 ESH - 1068 ESH **UV Degradation Test** Wavelength (nm) 1400 1200 1000 800 009 8 200 6.0 0.8 9.0 0.7 0.5 0.4 0.3 0.2 0 0.1 Reflectance

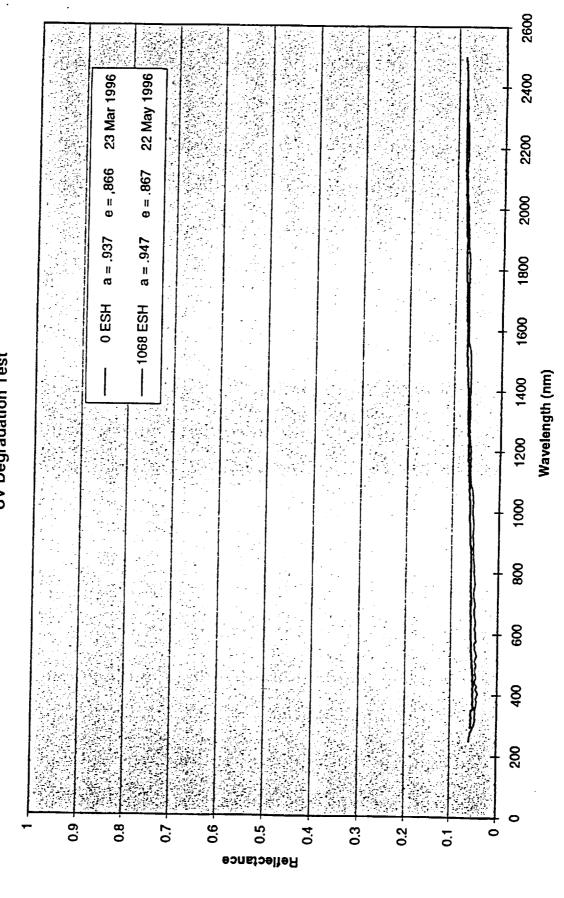
2600 2400 22 May 1996 1 Apr 1996 2200 e = .881e = .881 2000 1800 a = .180a = .1291600 0 ESH ---- 1068 ESH **UV Degradation Test** Wavelength (nm) 1400 1200 1000 800 8 400 200 0.8 0.5 0.7 0.2 Reflectance

ZOT/SS-55 White Paint [Batch # IW-19]
UV Degradation Test

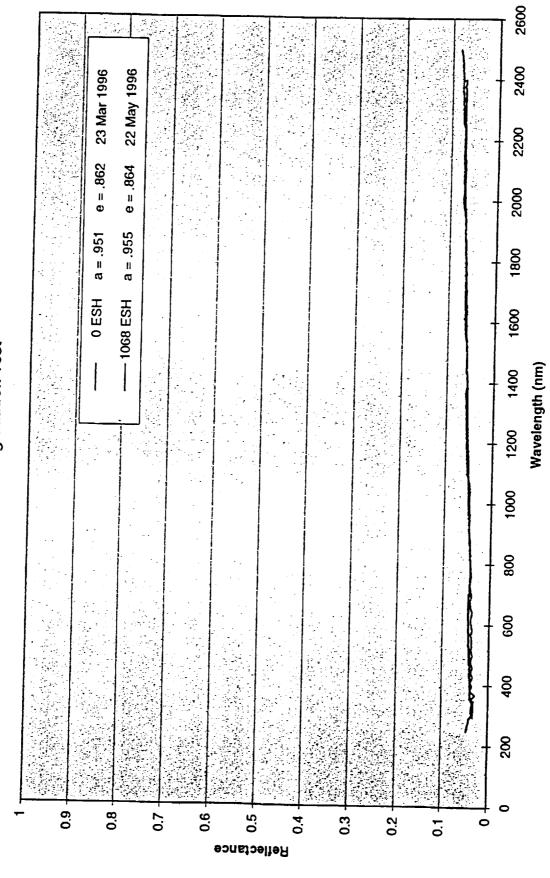
D36SCB/LO Black Paint [Batch # JV-17] UV Degradation Test



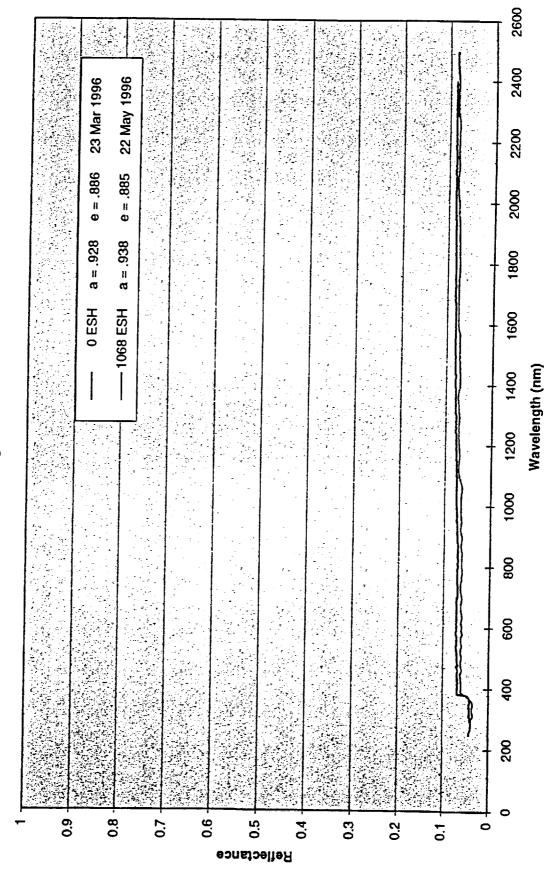
MS41SCB/LO Black Paint [Batch # U-356 IC-20] UV Degradation Test



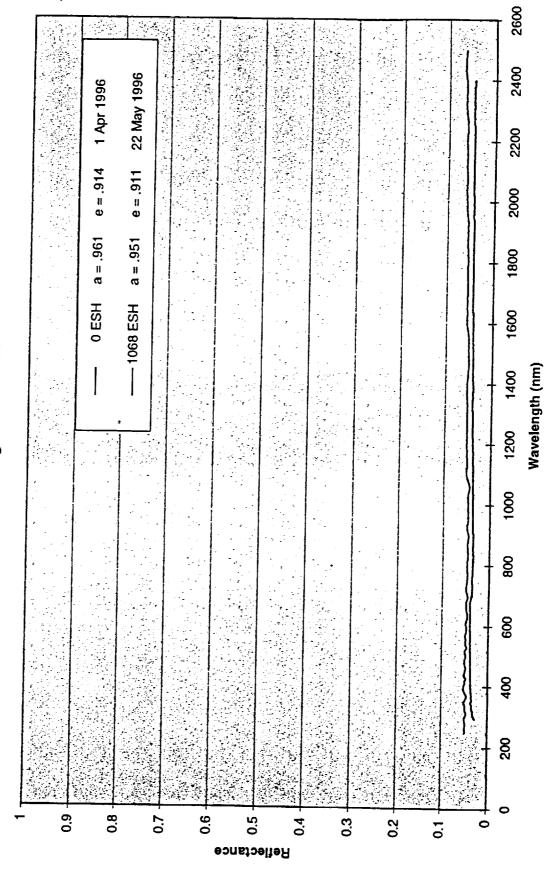
MH21SC/LO Black Paint [Batch # GX-16] UV Degradation Test



MH55-IC Black Paint [Batch #Z-6] UV Degradation Test



DBG-IP Black Paint [Batch # GTM-29] UV Degradation Test



APPENDIX D

USAF WL/ML SCEPTRE SCREENING TEST 96QV01

USAF WL/ML SCEPTRE Screening Test 96QV01 April 9 - May 1, 1996

Test Environmental Parameters

Temperature Range

In Situ* · 135-146 deg. F

Passive = 185 deg. F

*lossed chiller for 12 hrs. & temperatures reached 182-185 deg. F

1.0 KeV electron flux - 6.0E+09 e-/cm2/sec

total exposure time: 314 hrs.

vacuum: 5.0E-08 torr

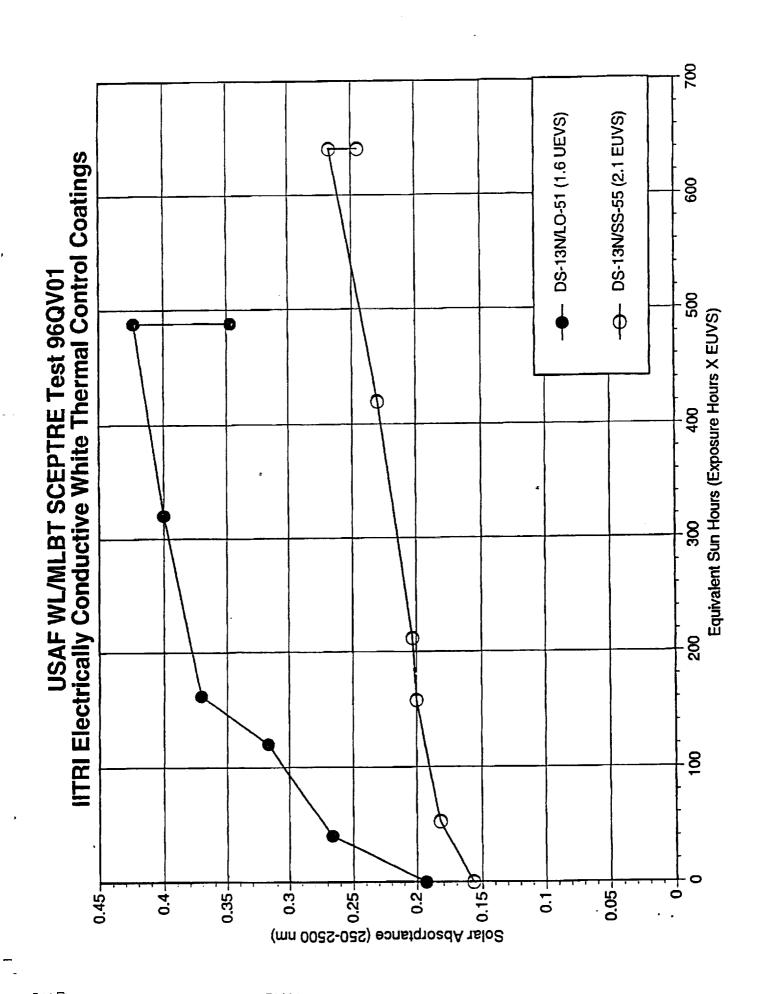
10.0 KeV electron flux - 3.0E+09 e-/cm2/sec

total fluence- 1.2E+16 e-/cm2

Measured In Situ

Conductive White Coatings

| | | Specimen | UV | α | α | change in |
|-------------|-------|----------|----------|---------|----------|-----------|
| Coating | Batch | ID | Exposure | pretest | posttest | α |
| DS13N/LO-51 | V-032 | JX-17 | 1.6 EUVS | 0.193 | 0.423 | 0.230 |
| DS13N/SS-55 | U-315 | P-25 | 2.1 EUVS | 0.157 | 0.267 | 0.110 |



USAF WL/ML SCEPTRE Screening Test 96QV01

April 9 - May 1, 1996

Test Environmental Parameters

Temperature Range

vacuum: 5.0E-08 torr

total exposure time: 314 hrs.

In Situ* · 135-146 deg. F

Passive = 185 deg. F

*lossed chiller for 12 hrs. & temperatures reached 182-185 deg. F

1.0 KeV electron flux - 6.0E+09 e-/cm2/sec

10.0 KeV electron flux - 3.0E+09 e-/cm2/sec total

total fluence- 1.2E+16 e-/cm2

Measured In Situ

Conductive White Coatings

| | | Specimen | UV | α | α | change in |
|-------------|-------|----------|----------|---------|----------|-----------|
| Coating | Batch | ID | Exposure | pretest | posttest | α |
| DS13N/LO-51 | V-032 | JX-17 | 1.6 EUVS | 0.193 | 0.423 | 0.230 |
| DS13N/SS-55 | U-315 | P-25 | 2.1 EUVS | 0.157 | 0.267 | 0.110 |

Passively Exposed

Conductive Black Coatings

| | | Specimen | UV | α | α | change in |
|-----------|-------|----------|----------|---------|----------|-----------|
| Coating | Batch | ID | Exposure | pretest | posttest | α |
| DBG-IP | U-021 | EO-17 | 4.0 EUVS | 0.970 | 0.978 | 0.008 |
| D36SCB/LO | V-036 | JV-15 | 3.9 EUVS | 0.966 | 0.956 | -0.010 |

Conductive White Coatings

| | | Specimen | W | α | α | change in |
|---------------|-------|----------|----------|---------|----------|-----------|
| Coating | Batch | ID | Exposure | pretest | posttest | α |
| Z-93CXY | | | | | | Ì |
| ZnO(FC)/DHS-2 | U-327 | HL-23 | 1.1 EUVS | 0.143 | 0.225 | 0.082 |
| Z-93SCXY | | | | | | |
| S13N/DHS-2 | U-329 | HI-17 | 1.0 EUVS | 0.130 | 0.196 | 0.066 |
| Z-93SCLMXY | | | | | | |
| DS13N/DHS-2 | U-330 | HJ-16 | 3.3 EUVS | 0.117 | 0.237 | 0.120 |

Other Conductive White Coating

| | | Specimen | UV | α | α | change in |
|--------------|-------|----------|----------|---------|----------|-----------|
| Coating | Batch | _ ID | Exposure | pretest | posttest | α |
| Eu2O3*/SS-55 | U-353 | ID-19 | 3.6 EUVS | 0.076 | 0.161 | 0.085 |

APPENDIX E

RESULTS OF ELECTROSTATIC DISCHARGE TESTING AT GODDARD SPACE FLIGHT CENTER

DS13N/LO

Date Tested: 12/4/96

| Flux | Non-contact Charging Potential Developed on Sample (Volts) | | | | |
|-----------------|--|-------------|----------------|---------------|----------------|
| | 5KeV @ 0.44nA | 5KeV @4.4nA | 10KeV @ 0.44nA | 10KeV @ 4.4nA | 20KeV @ 0.44nA |
| Temperature (K) | _ | | | | |
| 274 | | | | | |
| 320 | -20 | -76 | -41 | -145 | -40 |

ZnO(CF)DHS-2

Date Tested: 11/13/96

| Flux | Non-contact Charging Potential Developed on Sample (Volts) | | | | | | |
|-----------------|--|-----------------|--------------------|-------------------|--|--|--|
| | 5KeV @ 1.0nA/cm^2 | 5KeV @10nA/cm^2 | 10KeV @ 1.0nA/cm^2 | 10KeV @ 10nA/cm^2 | | | |
| Temperature (K) | | | | | | | |
| 130 | -7 | -11 | -7 | -17 | | | |
| 179 | -8 | -15 | -9 | -14 | | | |
| 228 | -13 | -22 | -17 | -25 | | | |
| 275 | -9 | | -35 | -56 | | | |
| 322 | -3 | | -3 | -20 | | | |

S13GP/DHS-2

Date tested: 10/23/96

| Flux | Non-contact Charging Potential Developed on Sample (Volts) | | | | |
|-----------------|--|-----------------|--------------------|--|--|
| | 5KeV @ 1.0nA/cm^2 | 5KeV @10nA/cm^2 | 10KeV @ 1.0nA/cm^2 | | |
| Temperature (K) | | | | | |
| 128 | | | | | |
| 178 | -160 | | | | |
| 225 | -123 | | | | |
| 275 | -60 | -190 | -130 | | |
| 321 | -2.5 | -13 | -70 | | |

S13N/DHS-2

Date tested: 10/31/96

| Flux | Non-contact Charging | Potential Developed on Sample (Volts | ;) |
|-----------------|----------------------|--------------------------------------|----|
| | 5KeV @ 0.44nA | 5KeV @4.4nA | |
| Temperature (K) | | | |
| 275 | -80 | -130 | |
| 323 | -45 | -56 | |

ZnO(CF)DHS-2

Date Tested: 11/13/96

| Flux | Non-contact Charging Potential Developed on Sample (Volts) | | | | | |
|-----------------|--|-----|-----|-----|--|--|
| | 5KeV @ 1.0nA/cm^2 5KeV @10nA/cm^2 10KeV @ 1.0nA/cm^2 10KeV @ 10nA/cm^2 | | | | | |
| Temperature (K) | | | | | | |
| 130 | -7 | -11 | -7 | -17 | | |
| 179 | -8 | -15 | -9 | -14 | | |
| 228 | -13 | -22 | -17 | -25 | | |
| 275 | -9 | | -35 | -56 | | |
| 322 | -3 | | -3 | -20 | | |

DS13N/DHS2

Date tested: 11/4/96

| Flux | Non-contact Charging Potential Developed on Sample (Volts) | | | | |
|-----------------|--|-----------------|--------------------|-------------------|--------------------|
| | 5KeV @ 1.0nA/cm^2 | 5KeV @10nA/cm^2 | 10KeV @ 1.0nA/cm^2 | 10KeV @ 10nA/cm^2 | 20KeV @ 1.0nA/cm^2 |
| Temperature (K) | | | | | |
| 129 | -60 | -180 | -120 | | -120 |
| 179 | -45 | -170 | -110 | | |
| 228 | -70 | -160 | -110 | | |
| 275 | -30 | -100 | -65 | -160 | -60 |
| 322 | -6 | -27 | -20 | -70 | -30 |

ZOT/SS-55

Date tested: 1/7/97

| Flux | Non-contact Charging Potential Developed on Sample (Volts) | | | |
|-----------------|--|----------------|---------------|----------------|
| | 5KeV @ 0.44nA | 10KeV @ 0.44nA | 10KeV @ 4.4nA | 20KeV @ 0.44nA |
| Temperature (K) | | | | |
| 129 | -174 | | | |
| 179 | -160 | | | |
| 227 | -100 | -170 | | |
| 271 | -1.7 | -2 | | -100 |
| 319 | -0.5 | -0.6 | | 0 |

Eu2O3/SS-55

Date Tested: 11/19/96

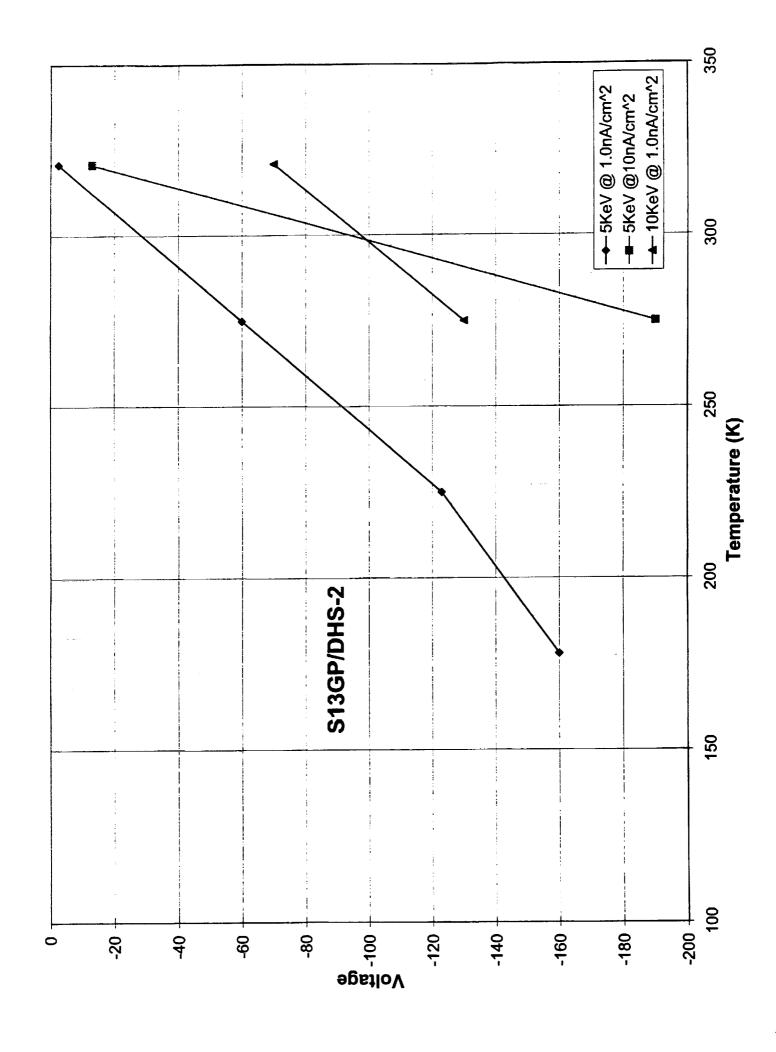
Flux Non-contact Charging Potential Developed on Sample (Volts) 5KeV @ 0.44nA

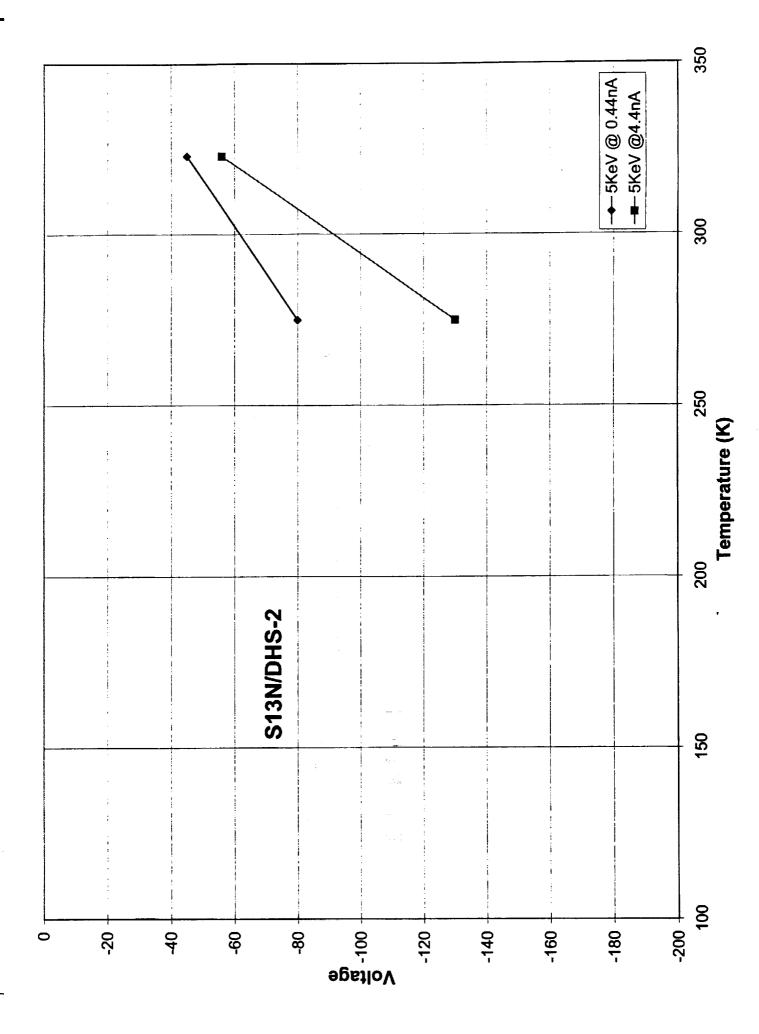
Temperature (K)

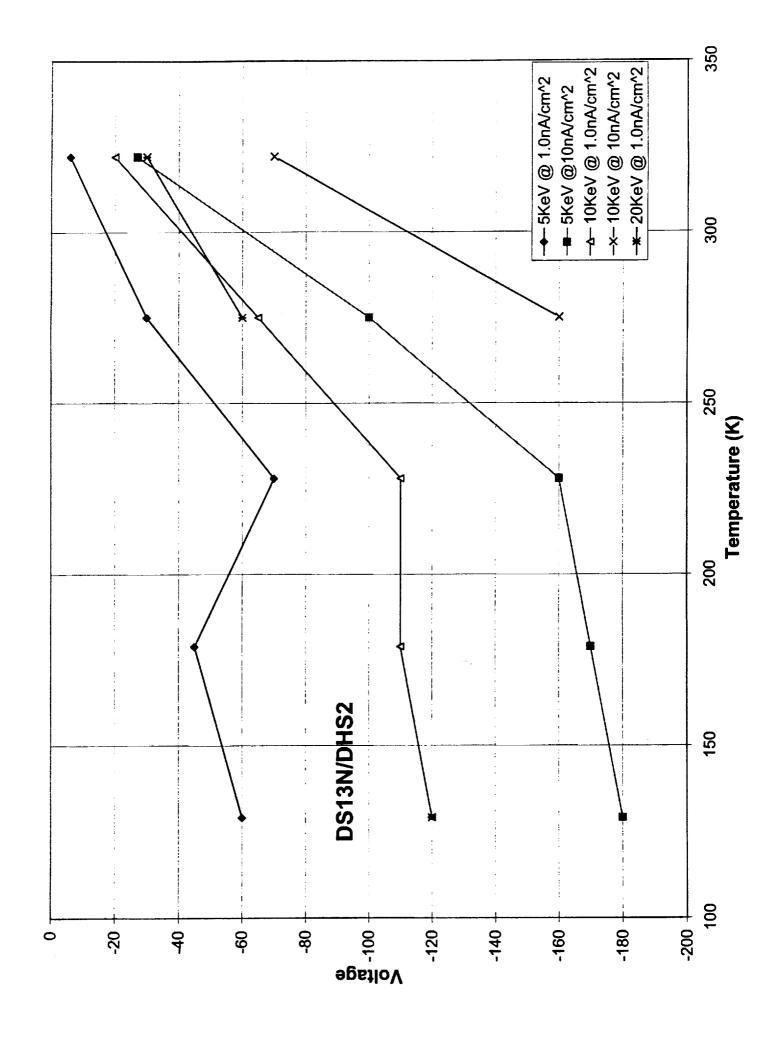
273 -180 322 -25

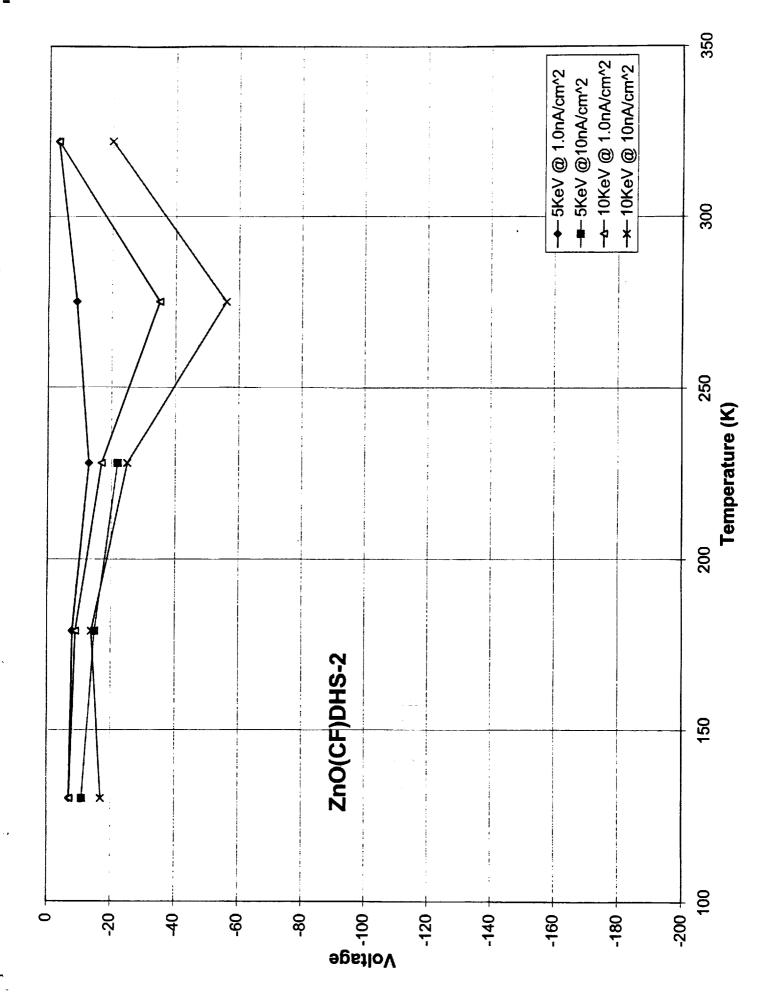
ZnO + Sb : Zn2SnOx Date tested: 12/2/96

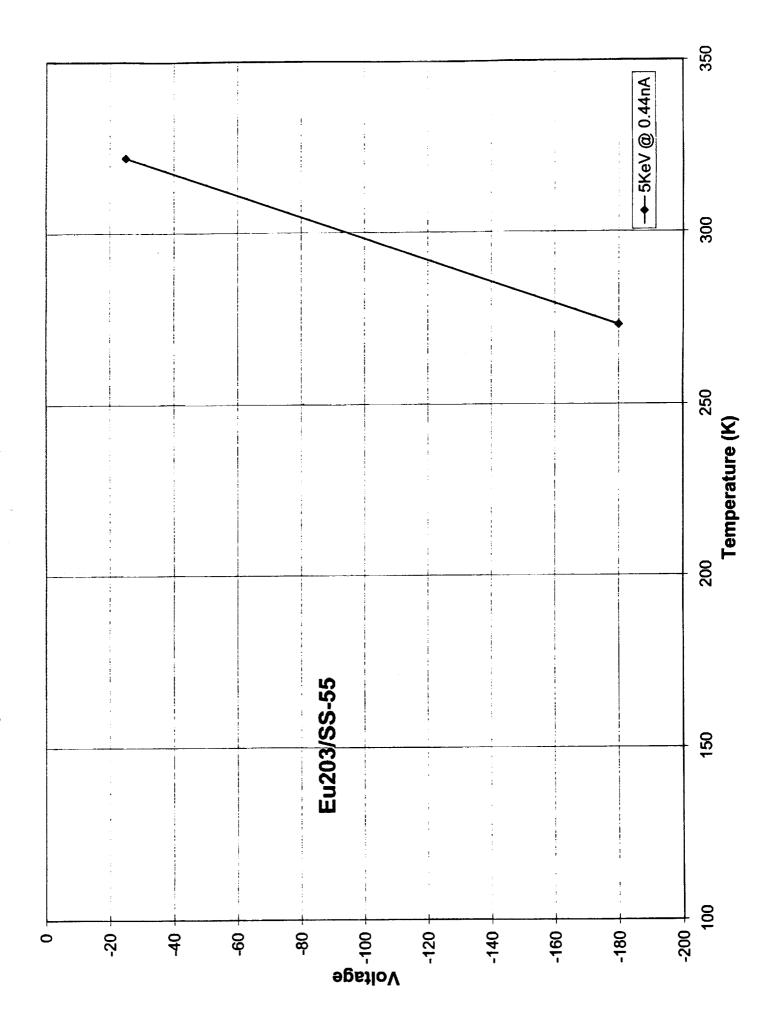
| Flux | Non-contact Charging | Potential Developed | | | |
|----------------|----------------------|---------------------|----------------|---------------|----------------|
| | 5KeV @ 0.44nA | 5KeV @4.4nA | 10KeV @ 0.44nA | 10KeV @ 4.4nA | 20KeV @ 0.44nA |
| Temperature (K | 3) | | #11 mm | | |
| 129 | -77 | -134 | -143 | | -140 |
| 179 | -73 | -110 | -120 | | -163 |
| 227 | -6 6 | -105 | -100 | | -137 |
| 275 | -30 | -120 | -60 | -160 | -70 |
| 323 | -14 | -19 | -34 | -36 | -48 |

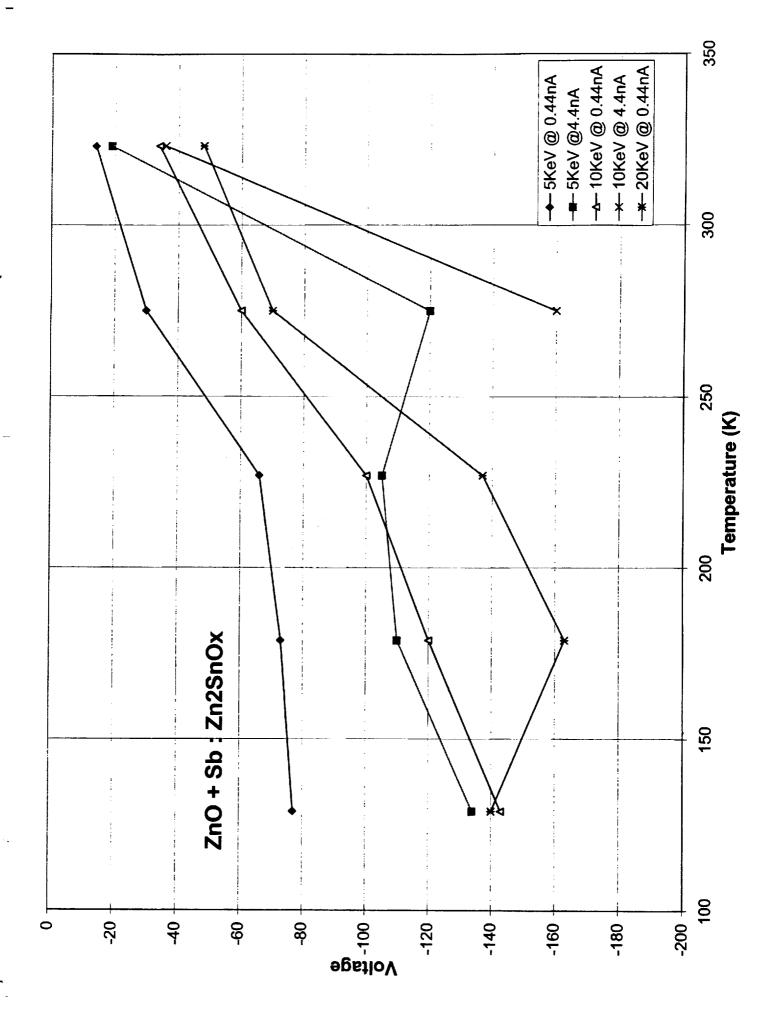


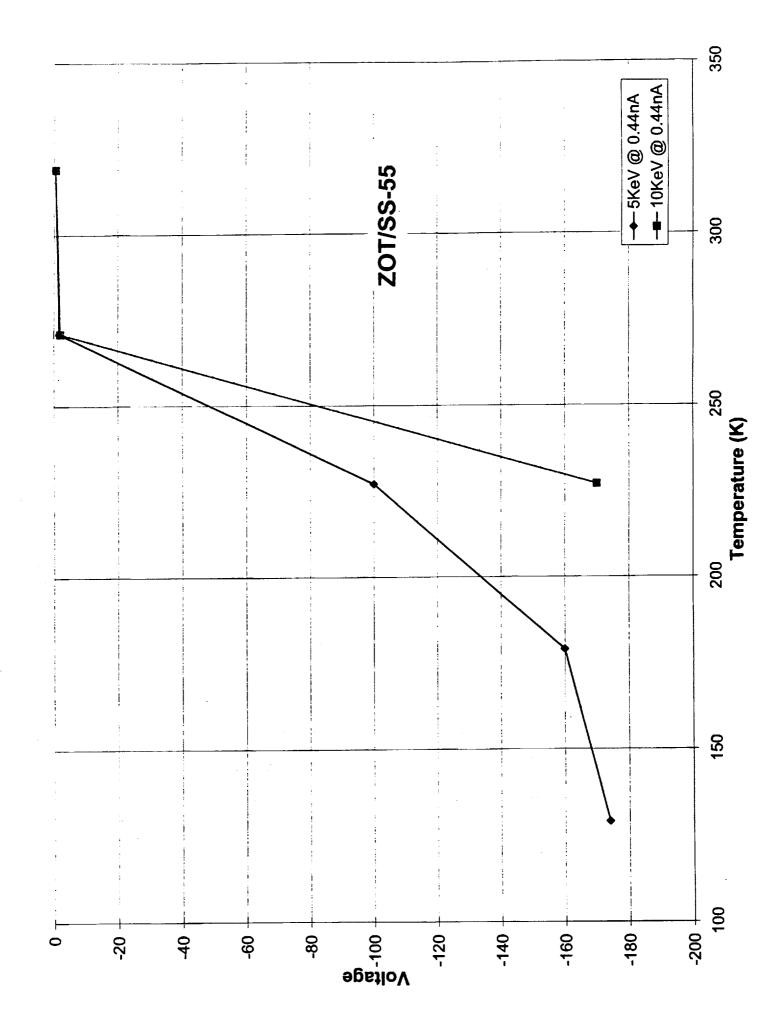












MH21SC/LO-2

Date Tested: 10/28/96

Flux Non-contact Charging Potential Developed on Sample (Volts) 5KeV @ 0.44nA

Temperature (K)

275 -190 322 -175

D36SCB/LO

Date tested: 11/20/96

| Flux | Non-contact Charging Potential Developed on Sample (Volts) | | | | | |
|-----------------|--|-----------------|--------------------|---|--|--|
| | 5KeV @ 1.0nA/cm^2 | 5KeV @10nA/cm^2 | 10KeV @ 1.0nA/cm^2 | 10KeV @ 10nA/cm^2 | | |
| Temperature (K) | | _ | | • | | |
| 133 | -6.3 | -8.5 | -8.4 | -9.9 | | |
| 178 | -6.1 | -6.2 | -6.5 | -7 | | |
| 226 | -5.8 | -6.1 | -5.6 | -6.3 | | |
| 275 | -4 .1 | -4 .9 | -3.9 | -4.8 | | |
| 321 | -2.4 | -3.4 | -3 | -3.7 | | |

MH55IC

Date Tested: 11/24/96

| Flux | Non-contact Charging | Potential Developed | on Sample (Volts) | | |
|-----------------|----------------------|---------------------|--------------------|-------------------|--------------------|
| | 5KeV @ 1.0nA/cm^2 | 5KeV @10nA/cm^2 | 10KeV @ 1.0nA/cm^2 | 10KeV @ 10nA/cm^2 | 20KeV @ 1.0nA/cm^2 |
| Temperature (K) | | | _ | • | • |
| 129 | -33 | -35 | -71 | -68 | -68 |
| 178 | -30 | -30 | -60 | -56 | -61 |
| 226 | -22 | -31 | -60 | -60 | -60 |
| 273 | -18 | -22 | -22 | -44 | •• |
| 321 | -3 | -7.3 | -3.9 | -9 .5 | |

DBG-IP

Date Tested: 11/26/96

| Flux | Non-contact Charging Potential Developed on Sample (Volts) | | | | | | |
|-----------------|--|-----------------|--------------------|---|--------------------|--|--|
| | 5KeV @ 1.0nA/cm^2 | 5KeV @10nA/cm^2 | 10KeV @ 1.0nA/cm^2 | 10KeV @ 10nA/cm^2 | 20KeV @ 1.0nA/cm^2 | | |
| Temperature (K) | | _ | | • | 2 | | |
| 130 | -17 | -39 | -6 6 | -71 | -87 | | |
| 179 | -20 | -32 | -63 | -70 | -80 | | |
| 227 | -25 | -30 | -51 | -64 | -74 | | |
| 273 | -10 | -16 | -16 | -28 | -17 | | |
| 320 | -1 | -5.9 | -3.5 | -8.1 | -3 | | |

MH41SCB/LO1

Date Tested: 11/30/96

| Flux | Non-contact Charging Potential Developed on Sample (Volts) | | | | | |
|-----------------|--|-----------------|--------------------|-------------------|--------------------|--|
| | 5KeV @ 1.0nA/cm^2 | 5KeV @10nA/cm^2 | 10KeV @ 1.0nA/cm^2 | 10KeV @ 10nA/cm^2 | 20KeV @ 1.0nA/cm^2 | |
| Temperature (K) | | | • | | | |
| 128 | -10 | -12.6 | -55 | -60 | -62 | |
| 173 | -13 | -13 | -62 | -60 | -61 | |
| 224 | -16 | -16 | -61 | -63 | -63 | |
| 268 | -17 | -17 | -65 | -62 | •• | |
| 323 | -22 | -22 | -65 | -73 | -60 | |

